

DAFTAR PUSTAKA

- [1] D. Guan, X. He, R. Zhang, and X. Qu, “Microstructure and tensile properties of in situ polymer-derived particles reinforced steel matrix composites produced by powder metallurgy method,” *Mater. Sci. Eng. A*, vol. 705, pp. 231–238, 2017, doi: 10.1016/j.msea.2017.07.084.
- [2] J. R. Pickens, “Aluminium powder metallurgy technology for high-strength applications,” *J. Mater. Sci.*, vol. 16, no. 6, pp. 1437–1457, 1981, doi: 10.1007/BF02396863.
- [3] V. V. Vani and S. K. Chak, “The effect of process parameters in Aluminum Metal Matrix Composites with Powder Metallurgy,” vol. 7, 2018.
- [4] J. R. Davis, *ASM Handbook Aluminum and Aluminum Alloys*. 1993.
- [5] W. D. Callister and D. G. Rethwisch, *Fundamentals of materials science and engineering : an integrated approach*. 2016. [Online]. Available: <https://www.wiley.com/en-us/Fundamentals+of+Materials+Science+and+Engineering%3A+An+Integrated+Approach%2C+5th+Edition-p-9781119175506R120>
- [6] P. R. Matli, U. Fareeha, R. A. Shakoor, and A. M. A. Mohamed, “A comparative study of structural and mechanical properties of Al-Cu composites prepared by vacuum and microwave sintering techniques,” *J. Mater. Res. Technol.*, vol. 7, no. 2, pp. 165–172, 2018, doi: 10.1016/j.jmrt.2017.10.003.
- [7] A. Gökçe, F. Findik, and A. O. Kurt, “Microstructural examination and properties of premixed Al-Cu-Mg powder metallurgy alloy,” *Mater. Charact.*, vol. 62, no. 7, pp. 730–735, 2011, doi: 10.1016/j.matchar.2011.04.021.
- [8] A. I. Y. Tok, F. Y. C. Boey, and X. L. Zhao, “Novel synthesis of Al₂O₃ nano-particles by flame spray pyrolysis,” *J. Mater. Process. Technol.*, vol. 178, no. 1–3, pp. 270–273, 2006, doi: 10.1016/j.jmatprotec.2006.04.007.
- K. Abbass and B. F. Sultan, “Effect of Al₂O₃ Nano Particles on Corrosion Behaviour of Aluminum Alloy (Al-4.5wt% Cu-1.5wt% Mg) Fabricated by Powder Metallurgy,” vol. 11, no. 1, pp. 25–31, 2019.



- [10] G. Rodríguez-Cabriales *et al.*, “Synthesis and characterization of Al-Cu-Mg system reinforced with tungsten carbide through powder metallurgy,” *Mater. Today Commun.*, vol. 22, no. July, p. 100758, 2020, doi: 10.1016/j.mtcomm.2019.100758.
- [11] B. Zlaticanin, B. Radonjic, and M. Filipovic, “Characterization of Structure and Properties of As-cast AlCuMg Alloys,” *Mater. Trans.*, vol. 45, no. 2, pp. 440–446, 2004, doi: 10.2320/matertrans.45.440.
- [12] B. V. Ramnath *et al.*, “Aluminium metal matrix composites - A review,” *Rev. Adv. Mater. Sci.*, vol. 38, no. 1, pp. 55–60, 2014.
- [13] M. Rahimian, N. Ehsani, N. Parvin, and H. reza Baharvandi, “The effect of particle size, sintering temperature and sintering time on the properties of Al-Al₂O₃ composites, made by powder metallurgy,” *J. Mater. Process. Technol.*, vol. 209, no. 14, pp. 5387–5393, 2009, doi: 10.1016/j.jmatprotec.2009.04.007.
- [14] M. Rahimian, N. Parvin, and N. Ehsani, “Investigation of particle size and amount of alumina on microstructure and mechanical properties of Al matrix composite made by powder metallurgy,” *Mater. Sci. Eng. A*, vol. 527, no. 4–5, pp. 1031–1038, 2010, doi: 10.1016/j.msea.2009.09.034.
- [15] D. M. Nuruzzaman, S. N. S. Jamaludin, F. F. B. Kamaruzaman, S. Basri, and N. A. M. B. Zulkifli, “Fabrication and mechanical properties of aluminium-aluminium oxide metal matrix composites,” *Int. J. Mech. Mechatronics Eng.*, vol. 15, no. 6, pp. 68–75, 2015.
- [16] C. D. Boland, R. L. Hexemer, I. W. Donaldson, and D. P. Bishop, “Industrial processing of a novel Al-Cu-Mg powder metallurgy alloy,” *Mater. Sci. Eng. A*, vol. 559, pp. 902–908, 2013, doi: 10.1016/j.msea.2012.09.049.
- [17] G. Pitchayyapillai, P. Seenikannan, K. Raja, and K. Chandrasekaran, “Al6061 Hybrid Metal Matrix Composite Reinforced with Alumina and Molybdenum Disulphide,” vol. 2016, 2016.



Zabihi, M. Reza, and A. Shafyei, “Materials Science & Engineering A Application of powder metallurgy and hot rolling processes for manufacturing aluminum / alumina composite strips,” *Mater. Sci. Eng. A*,

- vol. 560, pp. 567–574, 2013, doi: 10.1016/j.msea.2012.09.103.
- [19] T. Suwanda, “Dan Waktu Sintering Terhadap Kekerasan Dan Berat Jenis Aluminium Pada Proses,” *J. Ilm. Semesta Tek.*, vol. 9, pp. 187–198, 2006.
- [20] W. D. Callister, *Materials Science and Engineering: An Introduction*, Wiley Asia. 2007.
- [21] S. Hadi, “Teknologi Bahan Lanjut,” no. Komposit, p. 103, 2017.
- [22] Budiarto, “Sintesis Paduan Alumunium (6061) dengan Metalurgi Serbuk Bahan Fin Rocket,” 2019.
- [23] R. M. German, *Powder Metallurgy Science*, Second Edi. 1994.
- [24] S. Kalpakjian, *Manufacturing Engineering and Technology*. 1989.
- [25] J. T. S. Iman, P. Agus, H. Ricki, “Studi Karakteristik Sifat Mekanik Alumunium Matrix Composite (AMC) Paduan AL,5% Cu, 12%Mg, 15%SiC Hasil Proses Stir Casting dengan Variasi Temperatur Pendinginan,” vol. 12, no. 2, pp. 151–164, 2018.
- [26] M. K. Surappa, “Aluminium matrix composites : Challenges and opportunities,” vol. 28, no. April, pp. 319–334, 2003.
- [27] X. Du, R. Liu, X. Xiong, and H. Liu, “Effects of sintering time on the microstructure and properties of an Al-Cu-Mg alloy,” *J. Mater. Res. Technol.*, vol. 9, no. 5, pp. 9657–9666, 2020, doi: 10.1016/j.jmrt.2020.06.083.
- [28] R. . Smallman and R. . Bishop, *Metalurgi Fisik Modern dan Rekayasa Material*. Erlangga, 2000.
- [29] E. Medvedovski, “Alumina-mullite ceramics for structural applications,” *Ceram. Int.*, vol. 32, no. 4, pp. 369–375, 2006, doi: 10.1016/j.ceramint.2005.04.001.
- [30] J. Campbell, *Castings*, Second Edi. Elsevier, 2003.
- [31] M. Kuts, *Handbook of Materials Selection*. Jhon Wiley & Sons, 2002.
- [32] A. International, “ASTM G99— 95a: Standard test method for wear testing with a pin-on-disk apparatus,” *ASTM Int.*, vol. 1, no. Reapproved, pp. 1–5,



17.

Leng, *Materials Characterization: Introduction to Microscopic and Spectroscopic Methods*. Hongkong: John Wiley & Sons (Asia) Pte Ltd,

- 2008.
- [34] A. Setiabudi, R. Hardian, and A. Muzakir, *Karakterisasi Material: Prinsip dan Aplikasinya dalam Penelitian Kimia*, vol. 1. 2012.
 - [35] Widyastuti, H. Ardhyananta, H. Purwaningsih, and R. M. Quluq, *Karakterisasi Material Bagian 1 : Komposisi, Topografi dan Sifat Mekanik*. ITS Press, 2019.
 - [36] P. Bindu and S. Thomas, “Estimation of lattice strain in ZnO nanoparticles: X-ray peak profile analysis,” *J. Theor. Appl. Phys.*, vol. 8, no. 4, pp. 123–134, 2014, doi: 10.1007/s40094-014-0141-9.
 - [37] L. F. Francis, *Materials Processing: A Unified Approach to Processing of Metals, Ceramics and Polymers*. Elsevier, 2016.
 - [38] C. Suryanarayana, “Mechanical alloying and milling,” *Prog. Mater. Sci.*, vol. 46, no. 1–2, pp. 1–184, 2001, doi: 10.1016/S0079-6425(99)00010-9.
 - [39] Z. Razavi Hesabi, H. R. Hafizpour, and A. Simchi, “An investigation on the compressibility of aluminum/nano-alumina composite powder prepared by blending and mechanical milling,” *Mater. Sci. Eng. A*, vol. 454–455, pp. 89–98, 2007, doi: 10.1016/j.msea.2006.11.129.
 - [40] J. Raharjo and S. Rahayu, “Pengaruh Penambahan MgO dan SiO₂ Serta Suhu Sintering Terhadap Sifat Fisis dan Mekanis Komposit Keramik α – Alumina,” *Pros. Semin. Nas. Tek. Kim. “Kejuangan” Pengemb. Teknol. Kim. untuk Pengolah. Sumber Daya Alam Indones.*, pp. 1–6, 2015.
 - [41] R. N. Lumley, T. B. Sercombe, and G. B. Schaffer, “Surface oxide and the role of magnesium during the sintering of aluminum,” *Metall. Mater. Trans. A Phys. Metall. Mater. Sci.*, vol. 30, no. 2, pp. 457–463, 1999, doi: 10.1007/s11661-999-0335-y.
 - [42] G. Tosun and M. Kurt, “The porosity, microstructure, and hardness of Al-Mg composites reinforced with micro particle SiC/Al₂O₃ produced using powder metallurgy,” *Compos. Part B Eng.*, vol. 174, no. February 2019, p. 106965, 2019, doi: 10.1016/j.compositesb.2019.106965.
- M. Sankhla *et al.*, “Effect of mixing method and particle size on hardness and compressive strength of aluminium based metal matrix composite prepared through powder metallurgy route,” *J. Mater. Res.*



- Technol.*, vol. 18, pp. 282–292, 2022, doi: 10.1016/j.jmrt.2022.02.094.
- [44] R. Simanjuntak, “Pembuatan dan Karakterisasi Keramik Konstruksi dengan Memanfaatkan Limbah Padat Pulp dengan Bahan Baku Bentonit,” Universitas Sumatra Utara, 2010.
- [45] J. Anggono, “Penyusutan dan Densifikasi Keramik Alumina: Perbandingan antara Hasil Proses Slip Casting dengan Reaction Bonding,” Universitas Kristen Petra, 2008. [Online]. Available: https://www.researchgate.net/profile/Juliana-Anggono/publication/43649397_Penyusutan_dan_Densifikasi_Keramik_Alumina_Perbandingan_Antara_Hasil_Proses_Slip_Casting_dengan_Reaction_Bonding/links/565b2b3208aeafc2aac60d15/Penyusutan-dan-Densifikasi-Keramik-Al
- [46] G. Manohar, A. Dey, K. M. Pandey, and S. R. Maity, “Fabrication of metal matrix composites by powder metallurgy: A review,” *AIP Conf. Proc.*, vol. 1952, 2018, doi: 10.1063/1.5032003.
- [47] A. Slipenyuk, V. Kuprin, Y. Milman, V. Goncharuk, and J. Eckert, “Properties of P/M processed particle reinforced metal matrix composites specified by reinforcement concentration and matrix-to-reinforcement particle size ratio,” *Acta Mater.*, vol. 54, no. 1, pp. 157–166, 2006, doi: 10.1016/j.actamat.2005.08.036.
- [48] R. N. Lumley and G. B. Schaffer, “The effect of additive particle size on the mechanical properties of sintered aluminium-copper alloys,” *Scr. Mater.*, vol. 39, no. 8, pp. 1089–1094, 1998, doi: 10.1016/S1359-6462(98)00278-4.
- [49] A. M. Al-Qutub, A. Khalil, N. Saheb, and A. S. Hakeem, “Wear and friction behavior of Al6061 alloy reinforced with carbon nanotubes,” *Wear*, vol. 297, no. 1–2, pp. 752–761, 2013, doi: 10.1016/j.wear.2012.10.006.
- [50] F. H. Latief and E. S. M. Sherif, “Effects of sintering temperature and graphite addition on the mechanical properties of aluminum,” *J. Ind. Eng. Em.*, vol. 18, no. 6, pp. 2129–2134, 2012, doi: 10.1016/j.jiec.2012.06.007.
- M. Y. K. & A. K. Khanra, “Hot Deformation Studies of Al-Cu-Mg



Powder Metallurgy Alloy Composite," *Lect. Notes Mech. Eng.*, pp. 75–81, 2019, doi: 10.1007/978-981-13-1780-4_9.



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LAMPIRAN 1

Perhitungan Beban Kompaksi

Diketahui :

- Diameter lubang cetakan (d) $= 20 \text{ mm} = 20 \times 10^{-3} \text{ m}$
- Tekanan Kompaksi (P) $= 200 \text{ MPa} = 200 \times 10^6 \text{ N/m}$

Ditanyakan :

- Besar beban kompaksi (dalam Ton)

Penyelesaian ;

$$\begin{aligned}
 F &= PA \\
 &= 200 \times 10^6 [\text{N/m}^2] \times 0.25 \times \pi \times (20 \times 10^{-3})^2 [\text{m}^2] \\
 &= 200 \times 10^6 [\text{N/m}^2] \times 0.25 \times \pi \times 400 \times 10^{-6} [\text{N}] \\
 &= 62.800 \text{ N} \\
 &= 6.280 \text{ Kg} \\
 &= \mathbf{6,28 \text{ Ton}}
 \end{aligned}$$



LAMPIRAN 2

Perhitungan Densitas Teoritis

Diketahui

- Massa Jenis / Densitas Alumunium ($\rho_{Alumunium}$) = $2,700 \text{ g/cm}^3$
- Massa Jenis / Densitas Tembaga ($\rho_{Tembaga}$) = $8,960 \text{ g/cm}^3$
- Massa Jenis / Densitas Magnesium ($\rho_{Magnesium}$) = $1,738 \text{ g/cm}^3$
- Massa Jenis / Densitas Alumina ($\rho_{Alumina}$) = $3,950 \text{ g/cm}^3$

Ditanyakan

- Massa Jenis / Densitas Teoritis Logam Paduan 1 ($\rho_{Al-4.5Cu-1.5Mg-1Al^2O_3}$)
- Massa Jenis / Densitas Teoritis Logam Paduan 2 ($\rho_{Al-4.5Cu-1Mg-1Al^2O_3}$)

Penyelesaian

$$\rho_c = \rho_m \times v_{fm} + \rho_{p1} \times v_{p1} + \rho_{p2} \times v_{p2} + \rho_{p3} \times v_{p3}$$

$$\rho_{Al-4.5Cu-1.5Mg-1Al^2O_3} = 2,700 \times 93\% + 8,960 \times 4,5\% + 1,738 \times 1.5\% + 3,950 \times 1\%$$

$$= 2,511 + 0,403 + 0,026 + 0,039$$

$$= 2,980 \text{ g/cm}^3$$

$$\rho_{Al-4.5Cu-1Mg-1Al^2O_3} = 2,700 \times 93.5\% + 8,960 \times 4,5\% + 1,738 \times 1\% + 3,950 \times 1\%$$

$$= 2,524 + 0,403 + 0,017 + 0,039$$

$$= 2,983 \text{ g/cm}^3$$

Keterangan:

ρ_c = Densitas Paduan

v_{fm} = Volume Fraksi Bahan Utama

ρ_m = Densitas Bahan Utama

v_{p1} = Volume Fraksi Bahan Paduan 1

ρ_{p1} = Densitas Bahan Paduan 1

v_{p2} = Volume Fraksi Bahan Paduan 2

ρ_{p2} = Densitas Bahan Paduan 2

v_{p3} = Volume Fraksi Bahan Paduan 3

ρ_{p3} = Densitas Bahan Paduan 3



LAMPIRAN 3

Perhitungan Volume dan Berat Komposit Matriks Alumunium

Diketahui :

- | | | | |
|-----------------------------|----------|-----------------------------|---------|
| - Jari-jari Spesimen | = 10 mm | - Fraksi Volume Magnesium 1 | = 1.5 % |
| - Tinggi Spesimen | = 6 mm | - Fraksi Volume Magnesium 2 | = 1 % |
| - Fraksi Volume Alumunium 1 | = 93 % | - Fraksi Volume Tembaga | = 4.5 % |
| - Fraksi Volume Alumunium 2 | = 93.5 % | - Fraksi Volume Alumina | = 1 % |

Ditanya :

- Volume Komposit Matriks Alumunium
- Volume Matrik Alumunium
- Volume Filler (Cu, Mg, Al₂O₃)
- Berat Matrik Alumunium
- Berat Filler (Cu, Mg, Al₂O₃)

Penyelesaian

a. Volume Komposit Matrik Alumunium

Cetakan berbentuk tabung dengan ukuran :

Diameter : 20 mm

Tinggi : 6 mm

$$\begin{aligned}
 \text{Volume Komposit Matrik Alumunium} &= \pi r^2 t \\
 &= 3,14 \times 10^2 \times 6 \\
 &= 1.884 \text{ mm}^3 \\
 &= 1,884 \text{ cm}^3
 \end{aligned}$$

b. Volume Alumunium (Al) 93% dan 93.5%

$$\text{I. } V_{\text{Al}} = 93 \% \times \text{volume komposit matrik alumunium}$$

$$= 93 \% \times 1,884 \text{ cm}^3$$

$$= 1,752 \text{ cm}^3$$

$$\text{II. } V_{\text{Al}} = 93.5 \% \times \text{volume komposit matrik alumunium}$$

$$= 93.5 \% \times 1,884 \text{ cm}^3$$

$$= 1,762 \text{ cm}^3$$

$$\text{III. } V_{\text{Al}} = 100 \% \times \text{volume komposit matrik alumunium}$$

$$= 100 \% \times 1,884 \text{ cm}^3$$

$$= 1,884 \text{ cm}^3$$

c. Volume Tembaga (Cu) 4.5 %

$$V_{\text{Cu}} = 4.5 \% \times \text{volume komposit}$$

$$= 4.5 \% \times 1,884 \text{ cm}^3$$

$$= 0,085 \text{ cm}^3$$



d. Volume Magnesium (Mg) 1 % dan 1.5 %

$$\text{I. } V_{\text{Mg}} = 1.5 \% \times \text{volume komposit}$$

$$= 1.5 \% \times 1,884 \text{ cm}^3$$

$$= 0,028 \text{ cm}^3$$

$$\text{II. } V_{\text{Mg}} = 1 \% \times \text{volume komposit}$$

$$= 1 \% \times 1,884 \text{ cm}^3$$

$$= 0,019 \text{ cm}^3$$

e. Volume Alumina (Al_2O_3) 1 %

$$V_{\text{Al}_2\text{O}_3} = 1 \% \times \text{volume komposit}$$

$$= 1 \% \times 1,884 \text{ cm}^3$$

$$= 0,019 \text{ cm}^3$$

Berat Material yang dibutuhkan

Berat material dapat ditentukan menggunakan persamaan berat jenis :

$$\rho = \frac{m}{v}$$

Maka berat material : $m = \rho \times v$

Dimana : ρ = massa jenis

v = Volume jenis

m = massa

a. Berat Alumunium (Al) 93 %

$$m_{\text{Al}} = 2,7 \text{ g/cm}^3 \times \text{volume Alumunium I}$$

$$= 2,7 \text{ g/cm}^3 \times 1,752 \text{ cm}^3$$

$$= 4,731 \text{ gram}$$

c. Berat Alumunium (Al) 100 %

$$m_{\text{Al}} = 2,7 \text{ g/cm}^3 \times \text{volume Alumunium III}$$

$$= 2,7 \text{ g/cm}^3 \times 1,884 \text{ cm}^3$$

$$= 5,087 \text{ gram}$$

d. Berat Tembaga (Cu) 4.5 %

$$m_{\text{Cu}} = 8,96 \text{ g/cm}^3 \times \text{volume Tembaga}$$

$$= 8,96 \text{ g/cm}^3 \times 0,085 \text{ cm}^3$$

$$= 0,762 \text{ gram}$$

e. Berat Magnesium (Cu) 1.5 %

$$m_{\text{Mg}} = 1,738 \text{ g/cm}^3 \times \text{volume Magnesium I}$$

$$= 1,738 \text{ g/cm}^3 \times 0,028 \text{ cm}^3$$

$$= 0,049 \text{ gram}$$

b. Berat Alumunium (Al) 93.5 %

$$m_{\text{Al}} = 2,7 \text{ g/cm}^3 \times \text{volume Alumunium II}$$

$$= 2,7 \text{ g/cm}^3 \times 1,762 \text{ cm}^3$$

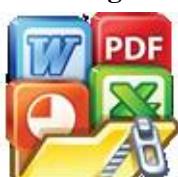
$$= 4,757 \text{ gram}$$

g. Berat Alumina (Al_2O_3) 1 %

$$3,95 \text{ g/cm}^3 \times \text{volume Alumina}$$

$$3,95 \text{ g/cm}^3 \times 0,019 \text{ cm}^3$$

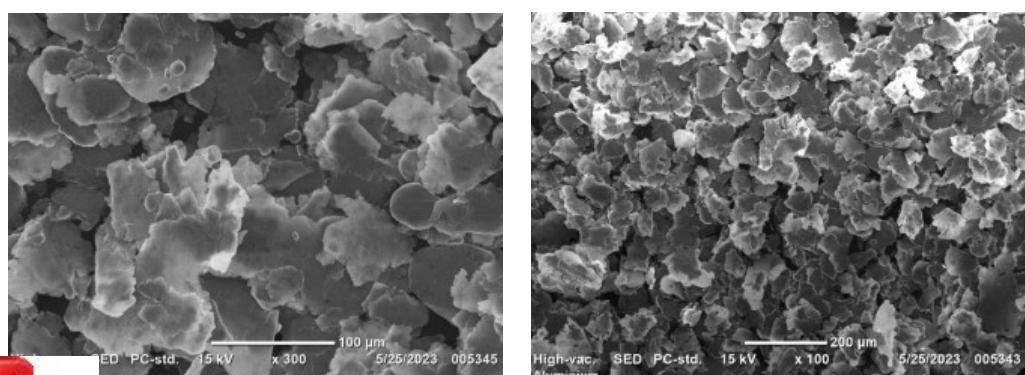
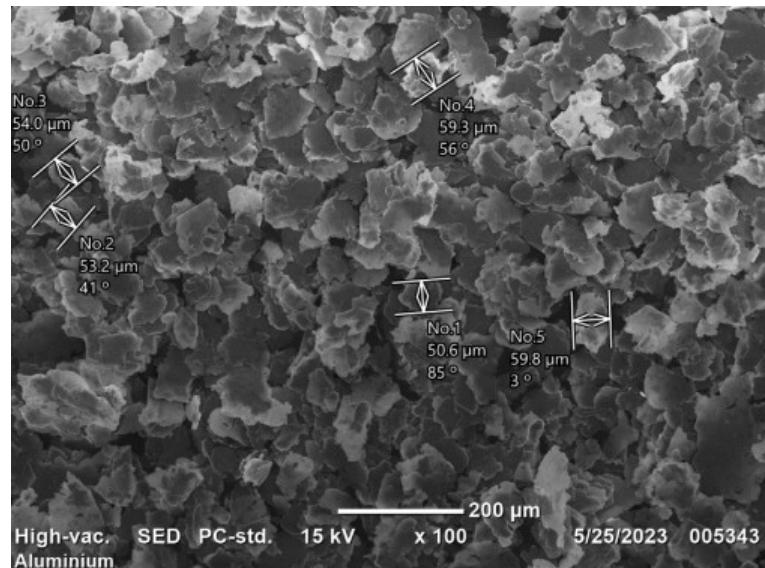
$$0,075 \text{ gram}$$



LAMPIRAN 4
ESTIMASI UKURAN SERBUK BAHAN UTAMA PENELITIAN

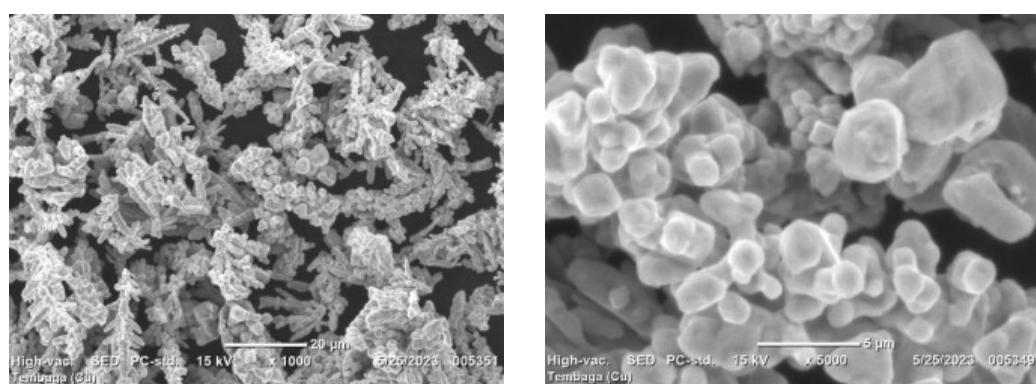
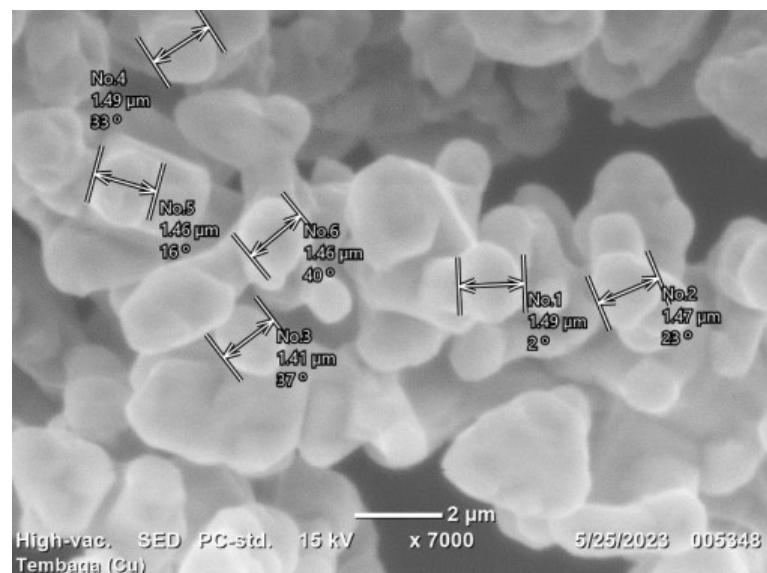
Alumunium

No.	Length	[unit]	Angle[deg]
1	50.6	um	85
2	53.2	um	41
3	54	um	50
4	59.3	um	56
5	59.8	um	3
Rata-rata	55.38	um	
Standar Deviasi	4.01273		



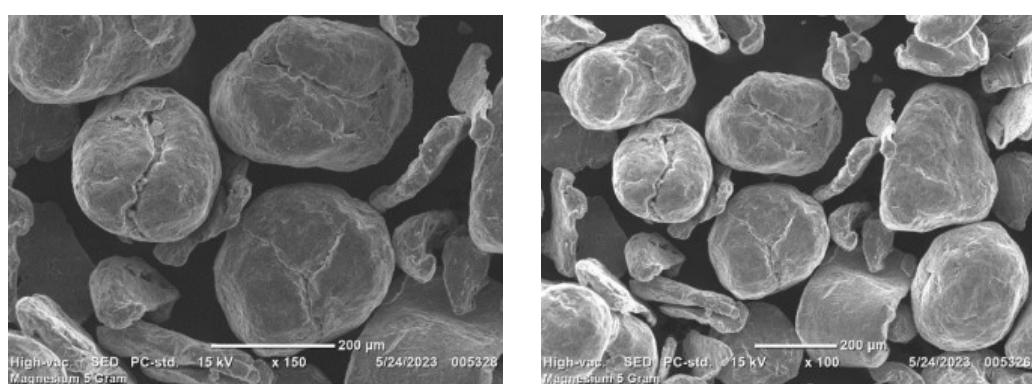
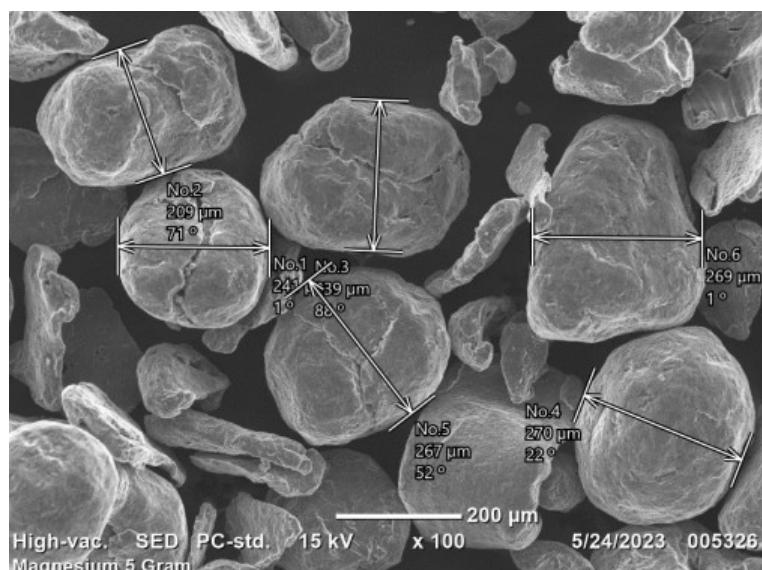
Tembaga

No.	Length	[unit]	Angle[deg]
1	1.49	um	2
2	1.47	um	23
3	1.41	um	37
4	1.49	um	33
5	1.46	um	16
6	1.46	um	40
Rata-rata	1.463333	um	
Standar deviasi	0.029439		



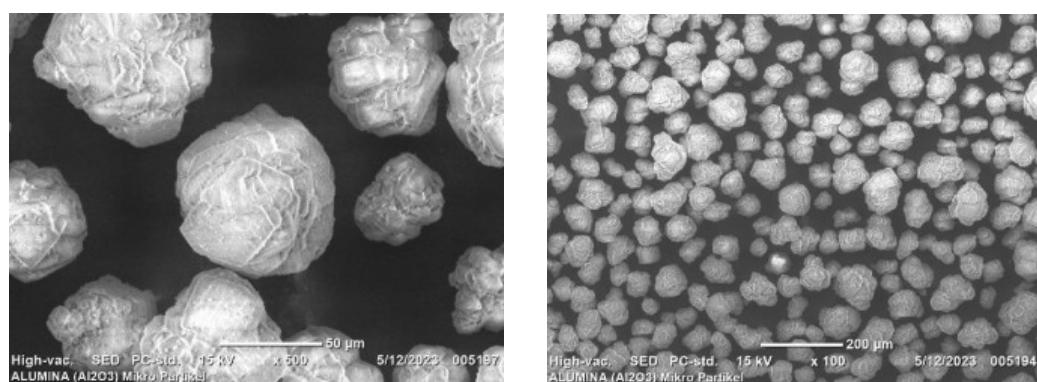
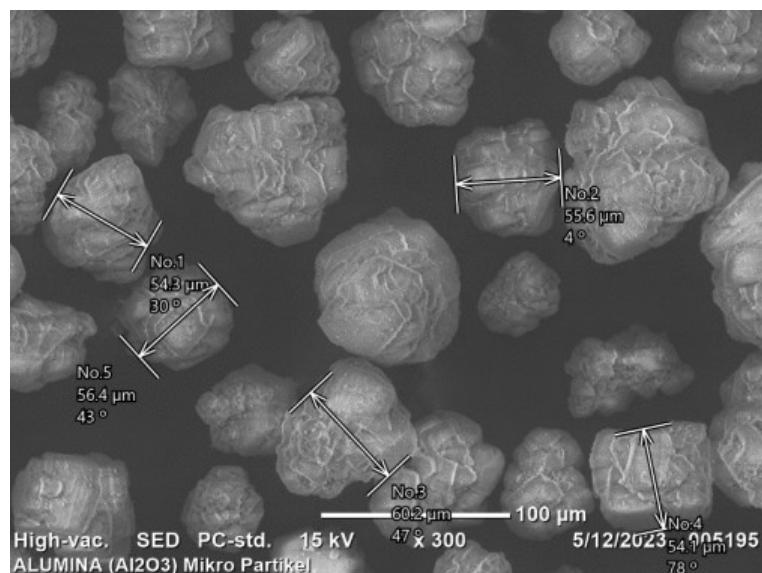
Magnesium

No.	Length	[unit]	Angle[deg]
1	241	um	1
2	209	um	71
3	239	um	88
4	270	um	22
5	267	um	52
6	269	um	1
Rata-rata	249.1667	um	
Standar deviasi	24.20262		



mikro-Alumina

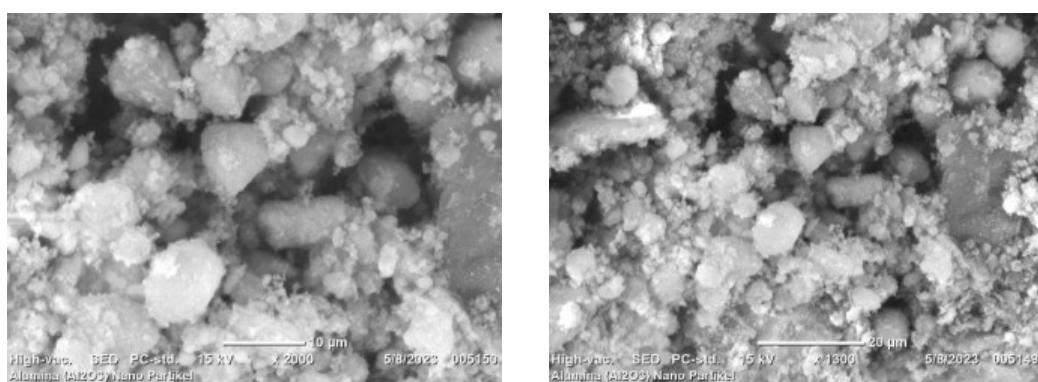
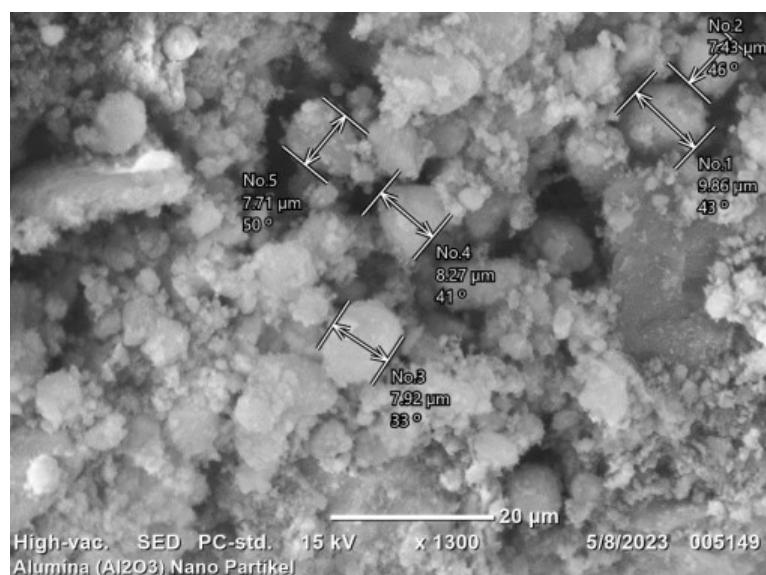
No.	Length	[unit]	Angle[deg]
1	54.3	um	30
2	55.6	um	4
3	60.2	um	47
4	54.1	um	78
5	56.4	um	43
Rata-rata	56.12	um	
Standar deviasi	2.46921		



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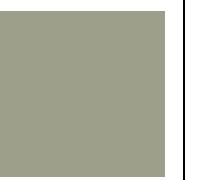
nano-Alumina

No.	Length	[unit]	Angle[deg]
1	9.86	um	43
2	7.43	um	46
3	7.92	um	33
4	8.27	um	41
5	7.71	um	50
Rata-rata	8.238	um	
Standar deviasi	0.957168		



LAMPIRAN 5

Fotografi Sampel *Green Compact* dan *Sintered Compact*

Fotografi Spesimen Green Compact				
SP 1	SP 2	SP 3	SP 4	SP 5
				

Fotografi Spesimen Sintered Compact				
SP 1	SP 2	SP 3	SP 4	SP 5
				
SP 6	SP 7	SP 8	SP 9	SP 10
				
SP 11	SP 12	SP 13	SP 14	SP 15
				



LAMPIRAN 6

Tabulasi Nilai Densitas dan Densitas Relatif *Green Compact*

Kode	% Fraksi Komposisi				Ukuran Butir Al_2O_3	$\rho_{teoritis}$ (g/cm ³)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	$\bar{x} \rho$ (g/cm ³)	$\bar{x} \rho_{relatif}$ (%)
	Al	Cu	Mg	Al_2O_3			Sampel 1	Sampel 2	Sampel 3					
SP 1	100	0	0	0	-	2.700	1.723	63.81	1.724	63.86	1.724	63.84	1.724	63.83
SP 2	93.5	4.5	1	1	56 μm	2.983	1.661	55.69	1.661	55.67	1.660	55.64	1.661	55.67
SP 3	93.5	4.5	1	1	20 nm	2.983	1.609	53.94	1.612	54.03	1.610	53.97	1.610	53.98
SP 4	93	4.5	1.5	1	56 μm	2.980	1.705	57.23	1.700	57.04	1.706	57.25	1.704	57.17
SP 5	93	4.5	1.5	1	20 nm	2.980	1.624	54.49	1.617	54.25	1.615	54.19	1.618	54.31
SP 6	100	0	0	0	-	2.700	1.723	63.80	1.724	63.86	1.724	63.85	1.724	63.84
SP 7	93.5	4.5	1	1	56 μm	2.983	1.661	55.68	1.660	55.65	1.661	55.67	1.661	55.67
SP 8	93.5	4.5	1	1	20 nm	2.983	1.610	53.96	1.611	54.02	1.610	53.97	1.610	53.98
SP 9	93	4.5	1.5	1	56 μm	2.980	1.705	57.20	1.700	57.03	1.707	57.29	1.704	57.17
SP 10	93	4.5	1.5	1	20 nm	2.980	1.624	54.48	1.618	54.28	1.615	54.20	1.619	54.32
SP 11	100	0	0	0	-	2.700	1.722	63.79	1.725	63.87	1.724	63.85	1.724	63.84
SP 12	93.5	4.5	1	1	56 μm	2.983	1.661	55.69	1.660	55.66	1.660	55.65	1.660	55.66
SP 13	93.5	4.5	1	1	20 nm	2.983	1.609	53.94	1.611	54.01	1.611	54.00	1.610	53.98
SP 14	93	4.5	1.5	1	56 μm	2.980	1.707	57.29	1.701	57.09	1.708	57.31	1.705	57.23
		4.5	1.5	1	20 nm	2.980	1.624	54.51	1.623	54.46	1.616	54.22	1.621	54.40



LAMPIRAN 7
Tabulasi Nilai Densitas & Densitas Relatif Sintered Compact

Kode	% Fraksi Komposisi				Ukuran Butir Al ₂ O ₃	Temp. Sinter (°C)	$\rho_{teoritis}$ (g/cm ³)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	ρ (g/cm ³)	$\rho_{relatif}$ (%)	$\bar{x} \rho$ (g/cm ³)	$\bar{x} \rho_{relatif}$ (%)	STDV
	Al	Cu	Mg	Al ₂ O ₃				Sampel 1	Sampel 2	Sampel 3						
SP 1	100	0	0	0	-	500	2.700	1.718	63.6	1.729	64.0	1.740	64.4	1.729	64.04	0.41
SP 2	93.5	4.5	1	1	56 μm		2.983	1.690	56.7	1.679	56.3	1.678	56.3	1.682	56.39	0.23
SP 3	93.5	4.5	1	1	20 nm		2.983	1.621	54.3	1.625	54.5	1.635	54.8	1.627	54.55	0.25
SP 4	93	4.5	1.5	1	56 μm		2.980	1.704	57.2	1.705	57.2	1.706	57.2	1.705	57.21	0.03
SP 5	93	4.5	1.5	1	20 nm		2.980	1.640	55.0	1.635	54.9	1.645	55.2	1.640	55.03	0.16
SP 6	100	0	0	0	-	550	2.700	1.749	64.8	1.742	64.5	1.747	64.7	1.746	64.66	0.14
SP 7	93.5	4.5	1	1	56 μm		2.983	1.694	56.8	1.699	56.9	1.696	56.8	1.696	56.85	0.09
SP 8	93.5	4.5	1	1	20 nm		2.983	1.640	55.0	1.639	54.9	1.639	54.9	1.639	54.96	0.03
SP 9	93	4.5	1.5	1	56 μm		2.980	1.706	57.2	1.704	57.2	1.708	57.3	1.706	57.25	0.08
SP 10	93	4.5	1.5	1	20 nm		2.980	1.648	55.3	1.637	54.9	1.646	55.2	1.644	55.15	0.21
SP 11	100	0	0	0	-	600	2.700	1.757	65.1	1.758	65.1	1.756	65.0	1.757	65.06	0.03
SP 12	93.5	4.5	1	1	56 μm		2.983	1.699	57.0	1.702	57.1	1.705	57.2	1.702	57.07	0.10
SP 13	93.5	4.5	1	1	20 nm		2.983	1.646	55.2	1.644	55.1	1.645	55.1	1.645	55.15	0.03
SP 14	93	4.5	1.5	1	56 μm		2.980	1.714	57.5	1.709	57.3	1.712	57.5	1.712	57.45	0.10
SP 15	93	4.5	1.5	1	20 nm		2.980	1.674	56.2	1.639	55.0	1.659	55.7	1.657	55.61	0.59



LAMPIRAN 8
Tabulasi Nilai Porositas *Sintered Compact*

Kode	% Fraksi Komposisi				Ukuran Butir Al ₂ O ₃	Temp. Sinter (°C)	m_1 (g/cm ³)	m_2 (g/cm ³)	ε (%)	m_1 (g/cm ³)	m_2 (g/cm ³)	ε (%)	m_1 (g/cm ³)	m_2 (g/cm ³)	ε (%)	$\bar{x} \varepsilon$ (%)	STDV
	Al	Cu	Mg	Al ₂ O ₃						Sampel 1			Sampel 2			Sampel 3	
SP 1	100	0	0	0	-	500	4.864	5.097	4.79	5.029	5.268	4.75	5.053	5.289	4.67	4.74	0.06
SP 2	93.5	4.5	1	1	56 µm		5.195	5.686	9.45	5.187	5.681	9.52	5.269	5.777	9.64	9.54	0.10
SP 3	93.5	4.5	1	1	20 nm		5.170	5.514	6.65	5.169	5.519	6.77	5.245	5.590	6.58	6.67	0.10
SP 4	93	4.5	1.5	1	56 µm		5.045	5.455	8.13	5.005	5.431	8.51	5.046	5.509	9.18	8.60	0.53
SP 5	93	4.5	1.5	1	20 nm		5.036	5.358	6.39	5.023	5.354	6.59	5.057	5.372	6.23	6.40	0.18
SP 6	100	0	0	0	-	550	5.070	5.307	4.67	5.070	5.306	4.65	5.067	5.293	4.46	4.60	0.12
SP 7	93.5	4.5	1	1	56 µm		5.357	5.834	8.90	5.290	5.765	8.98	5.251	5.728	9.08	8.99	0.09
SP 8	93.5	4.5	1	1	20 nm		5.236	5.566	6.30	5.264	5.610	6.57	5.333	5.673	6.38	6.42	0.14
SP 9	93	4.5	1.5	1	56 µm		4.103	4.409	7.46	4.104	4.423	7.77	4.102	4.444	8.34	7.86	0.45
SP 10	93	4.5	1.5	1	20 nm		5.186	5.495	5.96	5.111	5.435	6.34	5.171	5.467	5.72	6.01	0.31
SP 11	100	0	0	0	-	600	5.100	5.335	4.61	5.118	5.338	4.30	5.095	5.316	4.34	4.41	0.17
SP 12	93.5	4.5	1	1	56 µm		5.125	5.571	8.70	5.078	5.524	8.78	5.126	5.577	8.80	8.76	0.05
SP 13	93.5	4.5	1	1	20 nm		5.272	5.596	6.15	5.227	5.551	6.20	5.181	5.503	6.22	6.19	0.04
SP 14	93	4.5	1.5	1	56 µm		5.037	5.404	7.29	5.053	5.424	7.34	5.050	5.444	7.80	7.48	0.28
			1.5	1	20 nm		5.170	5.462	5.65	5.177	5.453	5.33	5.133	5.411	5.42	5.47	0.16



LAMPIRAN 9
Tabulasi Nilai Penyusutan *Sintered Compact*

Kode	% Fraksi Komposisi				Ukuran Butir Al₂O₃	Temp. Sinter (°C)	V_{GC}	V_{SC}	ΔV (%)	V_{GC}	V_{SC}	ΔV (%)	V_{GC}	V_{SC}	ΔV (%)	Σ ΔV (%)	STDV
	Al	Cu	Mg	Al₂O₃			Sampel 1			Sampel 2			Sampel 3				
SP 1	100	0	0	0	-	500	2.065	2.033	1.55	2.155	2.120	1.62	2.187	2.149	1.74	1.64	0.09
SP 2	93.5	4.5	1	1	56 μm		2.205	2.121	3.81	2.181	2.097	3.85	2.215	2.129	3.88	3.85	0.04
SP 3	93.5	4.5	1	1	20 nm		2.030	1.981	2.41	2.038	1.988	2.45	2.090	2.038	2.49	2.45	0.04
SP 4	93	4.5	1.5	1	56 μm		2.152	2.085	3.11	2.134	2.069	3.05	2.153	2.088	3.02	3.06	0.05
SP 5	93	4.5	1.5	1	20 nm		2.001	1.966	1.75	1.987	1.951	1.81	2.019	1.982	1.83	1.80	0.04
SP 6	100	0	0	0	-	550	2.215	2.171	1.99	2.202	2.159	1.95	2.210	2.166	1.99	1.98	0.02
SP 7	93.5	4.5	1	1	56 μm		2.284	2.194	3.94	2.267	2.176	4.01	2.242	2.154	3.93	3.96	0.05
SP 8	93.5	4.5	1	1	20 nm		2.097	2.044	2.53	2.106	2.052	2.56	2.133	2.079	2.53	2.54	0.02
SP 9	93	4.5	1.5	1	56 μm		1.760	1.698	3.52	1.751	1.695	3.20	1.756	1.701	3.13	3.28	0.21
SP 10	93	4.5	1.5	1	20 nm		2.087	2.039	2.30	2.029	1.988	2.02	2.073	2.030	2.07	2.13	0.15
SP 11	100	0	0	0	-	600	2.249	2.197	2.31	2.257	2.206	2.26	2.245	2.193	2.32	2.30	0.03
SP 12	93.5	4.5	1	1	56 μm		2.199	2.109	4.09	2.184	2.095	4.08	2.210	2.120	4.07	4.08	0.01
SP 13	93.5	4.5	1	1	20 nm		2.124	2.069	2.59	2.103	2.048	2.62	2.085	2.031	2.59	2.60	0.01
SP 14	93	4.5	1.5	1	56 μm		2.180	2.099	3.72	2.172	2.096	3.50	2.175	2.101	3.40	3.54	0.16
	5	1.5	1	20 nm			2.133	2.081	2.44	2.069	2.018	2.46	2.091	2.039	2.49	2.46	0.02



LAMPIRAN 10

TABULASI DATA SIFAT FISIS KOMPOSIT

Tabulasi Data Hasil Penelitian **Sifat Fisis** Komposit Paduan Al – Cu – Mg - Al₂O₃

Kode	Komposisi (%)				Ukuran Butir Al ₂ O ₃	Densitas Teoritis (g/cm ³)	Densitas GC (g/cm ³)	Densitas SC (g/cm ³)	Densitas Relatif GC (%)	Densitas Relatif SC (%)	Porositas (%)	Shrinkage (%)
	Al	Cu	Mg	Al ₂ O ₃								
500°C												
SP1	100	0	0	0	-	2.700	1.724	1.729	63.83	64.04	4.74	1.64
SP2	93.5	4.5	1	1	μ	2.983	1.661	1.682	55.67	56.39	9.54	3.85
SP3	93.5	4.5	1	1	n	2.983	1.610	1.627	53.98	54.55	6.67	2.45
SP4	93	4.5	1.5	1	μ	2.980	1.704	1.696	57.17	57.21	8.60	3.06
SP5	93	4.5	1.5	1	n	2.980	1.618	1.640	54.31	55.03	6.40	1.80
550°C												
SP6	100	0	0	0	-	2.700	1.724	1.746	63.84	64.66	4.60	1.98
SP7	93.5	4.5	1	1	μ	2.983	1.661	1.696	55.67	56.85	8.99	3.96
SP8	93.5	4.5	1	1	n	2.983	1.610	1.639	53.98	54.96	6.42	2.54
SP9	93	4.5	1.5	1	μ	2.980	1.704	1.706	57.17	57.25	7.86	3.28
SP10	93	4.5	1.5	1	n	2.980	1.619	1.644	54.32	55.15	6.01	2.13
600°C												
SP11	100	0	0	0	-	2.700	1.724	1.757	63.84	65.06	4.41	2.30
SP12	93.5	4.5	1	1	μ	2.983	1.660	1.702	55.66	57.07	8.76	4.08
		1	1	n		2.983	1.610	1.645	53.98	55.15	6.19	2.60
		1.5	1	μ		2.980	1.705	1.712	57.23	57.45	7.48	3.54
		1.5	1	n		2.980	1.621	1.657	54.40	55.61	5.47	2.46



LAMPIRAN 11
Tabulasi Nilai Kekerasan Mikro *Sintered Compact*

Kode	% Fraksi Komposisi				Ukuran Butir Al_2O_3	Temp. Sinter (°C)		Kiri (HV)	Tengah (HV)	Kanan (HV)	\bar{x} HV (%)
	Al	Cu	Mg	Al_2O_3							
SP 1	100	0	0	0	-	500	Atas	23.5	28.5	23.6	27.5
							Tengah	31.2	30.1	29.5	
							Bawah	32.2	23.0	25.6	
SP 2	93.5	4.5	1	1	56 μm	500	Atas	47.3	50.5	50.8	48.1
							Tengah	50.6	46.7	44.6	
							Bawah	48.1	47.8	46.3	
SP 3	93.5	4.5	1	1	20 nm	500	Atas	52.5	55.6	59.8	52.1
							Tengah	50.6	46.7	44.8	
							Bawah	58.1	47.8	52.8	
SP 4	93	4.5	1.5	1	56 μm	500	Atas	55.0	51.4	37.7	49.6
							Tengah	49.9	51.8	50.2	
							Bawah	45.4	50.6	54.8	
SP 5	93	4.5	1.5	1	20 nm	500	Atas	55.3	51.3	56.0	53.5
							Tengah	52.2	52.9	52.1	
							Bawah	48.1	57.7	55.7	



Kode	% Fraksi Komposisi				Ukuran Butir Al₂O₃	Temp. Sinter (°C)		Kiri (HV)	Tengah (HV)	Kanan (HV)	\bar{x} HV (%)
	Al	Cu	Mg	Al₂O₃				Atas	37.4	23.6	29.4
SP 6	100	0	0	0	-	550	Tengah	33.7	30.1	29.5	49.0
SP 7	93.5	4.5	1	1	56 μm		Bawah	32.2	29.0	21.1	
SP 8	93.5	4.5	1	1	20 nm		Atas	46.1	46.9	45.8	
SP 9	93	4.5	1.5	1	56 μm		Tengah	53.9	58.8	56.9	53.6
SP 10	93	4.5	1.5	1	20 nm		Bawah	44.3	43.9	44.7	
							Atas	55.6	54.9	66.4	
							Tengah	50.3	61.4	43.3	51.3
							Bawah	57.6	46.2	47.0	
							Atas	56.1	57.0	56.9	
							Tengah	47.3	57.9	42.2	54.6
							Bawah	34.5	57.9	51.8	
							Atas	55.1	49.6	58.0	
							Tengah	54.7	58.8	51.9	54.6
							Bawah	57.6	53.8	52.3	



Kode	% Fraksi Komposisi				Ukuran Butir Al₂O₃	Temp. Sinter (°C)		Kiri (HV)	Tengah (HV)	Kanan (HV)	\bar{x} HV (%)
	Al	Cu	Mg	Al₂O₃				Atas	37.4	33.1	31.1
SP 11	100	0	0	0	-	600	Tengah	33.7	31.1	29.5	31.1
SP 12	93.5	4.5	1	1	56 μm		Bawah	32.2	29.0	25.6	
SP 13	93.5	4.5	1	1	20 nm		Atas	47.3	58.5	49.8	51.3
SP 14	93	4.5	1.5	1	56 μm		Tengah	50.6	51.1	54.6	
SP 15	93	4.5	1.5	1	20 nm		Bawah	48.1	55.5	46.3	
							Atas	64.2	66.3	58.2	54.8
							Tengah	48.0	56.7	44.8	
							Bawah	54.2	47.8	52.8	
							Atas	58.2	49.3	57.7	52.1
							Tengah	49.9	51.8	51.2	
							Bawah	45.4	50.6	54.8	
							Atas	65.3	51.3	46.0	55.7
							Tengah	62.2	52.9	52.1	
							Bawah	48.1	57.7	65.7	



LAMPIRAN 12
Tabulasi Nilai Laju Keausan *Sintered Compact*

Kode	Al	Cu	Mg	Al ₂ O ₃	Ukuran Butir Al ₂ O ₃	ρ (g/cm ³)	Beban Uji (N)	Panjang Lintasan (meter)	Waktu Uji (detik)	Temp. Sinter (°C)	M loss (gram)	vol. Keausan (mm ³)	Laju Keausan (mm ³ /m)	Laju Keausan (mm ³ /m-N)
SP 1	100	0	0	0	-	1.730	2.5	16.14	60	500	0.025	14.449	0.895	0.358
SP 2	93.5	4.5	1	1	56 μm	1.682					0.021	12.484	0.773	0.309
SP 3	93.5	4.5	1	1	20 nm	1.626					0.020	12.300	0.762	0.305
SP 4	93	4.5	1.5	1	56 μm	1.705					0.021	12.318	0.763	0.305
SP 5	93	4.5	1.5	1	20 nm	1.636					0.019	11.614	0.720	0.288
SP 6	100	0	0	0	-	1.745			60	550	0.023	13.183	0.817	0.327
SP 7	93.5	4.5	1	1	56 μm	1.694					0.019	11.217	0.695	0.278
SP 8	93.5	4.5	1	1	20 nm	1.639					0.018	10.979	0.680	0.272
SP 9	93	4.5	1.5	1	56 μm	1.705					0.019	11.144	0.690	0.276
SP 10	93	4.5	1.5	1	20 nm	1.649					0.016	9.702	0.601	0.240
SP 11	100	0	0	0	-	1.757	600	16.14	60	600	0.020	11.381	0.705	0.282
SP 12	93.5	4.5	1	1	56 μm	1.703					0.017	9.980	0.618	0.247
SP 13	93.5	4.5	1	1	20 nm	1.646					0.015	9.113	0.565	0.226
SP 14	93	4.5	1.5	1	56 μm	1.709					0.016	9.360	0.580	0.232
				4.5	1.5	1.659					0.012	7.235	0.448	0.179



LAMPIRAN 13
TABULASI DATA PENELITIAN SIFAT MEKANIS KOMPOSIT

Tabulasi Data Hasil Penelitian **Sifat Mekanis Komposit Paduan Al – Cu – Mg - Al₂O₃**

Kode	Komposisi (%)				Ukuran Butir Al₂O₃	Densitas SC (g/cm³)	Hardness (HV)	Laju Aus (mm³/m)
	Al	Cu	Mg	Al₂O₃				
500°C								
SP1	100	0	0	0	-	1.729	27.5	0.895
SP2	93.5	4.5	1	1	μ	1.682	48.1	0.773
SP3	93.5	4.5	1	1	n	1.627	52.1	0.762
SP4	93	4.5	1.5	1	μ	1.696	49.6	0.763
SP5	93	4.5	1.5	1	n	1.640	53.5	0.720
550°C								
SP6	100	0	0	0	-	1.746	29.4	0.817
SP7	93.5	4.5	1	1	μ	1.696	49.0	0.695
SP8	93.5	4.5	1	1	n	1.639	53.6	0.680
SP9	93	4.5	1.5	1	μ	1.706	51.3	0.690
SP10	93	4.5	1.5	1	n	1.644	54.6	0.601
600°C								
SP11	100	0	0	0	-	1.757	31.1	0.705
SP12	93.5	4.5	1	1	μ	1.702	51.3	0.618
	5	4.5	1	1	n	1.645	54.8	0.565
		4.5	1.5	1	μ	1.712	52.1	0.580
		4.5	1.5	1	n	1.657	55.7	0.448



**EFEK DARI SUHU SINTERING, UKURAN BUTIR ALUMINA DAN
KOMPOSISI MAGNESIUM PADA KOMPOSIT Al-Cu-Mg-Al₂O₃
TERHADAP SIFAT FISIS, MEKANIS, DAN STRUKTUR MIKRO
DIPRODUKSI DENGAN METALURGI SERBUK**

**Effects of Sintering Temperature, Grain Size of Alumina and Magnesium Composition of
Al-Cu-Mg-Al₂O₃ on Physical, Mechanical and Microstructural Properties Produced by
Powder Metallurgy**

**TOMMY TARMINSYAH
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Tgl. 21 - 8 - 2023 .



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**EFEK DARI SUHU SINTERING, UKURAN BUTIR ALUMINA DAN
KOMPOSISI MAGNESIUM PADA KOMPOSIT Al-Cu-Mg-Al₂O₃
TERHADAP SIFAT FISIS, MEKANIS, DAN STRUKTUR MIKRO
DIPRODUKSI DENGAN METALURGI SERBUK**

**Effects of Sintering Temperature, Grain Size of Alumina and Magnesium Composition of
Al-Cu-Mg-Al₂O₃ on Physical, Mechanical and Microstructural Properties Produced by
Powder Metallurgy**

TOMMY TARMINSYAH

D022202005



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Arie
Jawat
S. Lot 2023

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