

## DAFTAR PUSTAKA

- Abali, Y., Bayca, S. U., Arisoy, K., & Vaizoğullar, A. İ. (2011). Optimization Of Dolomite Ore Leaching In Hydrochloric Acid Solutions. *Journal of Physicochem*, 46, 253-261.
- Alpionata, P., & Astuti. (2015). Sintesis Dan Karakterisasi Magnesium Oksida ( $MgO$ ) Dengan Variasi Massa PEG-6000. *Jurnal. Fis. Unand*, Vol. 4(2), pp. 167–172., 167-172.
- Arinaldo, P. (2016). *Pelindian Bijih Nikel Laterit Dengan Asam Klorida*. Banten: Skripsi Sarjana Universitas Islam Negeri Syarif Hidayatullah.
- Baba, A., Omipidan, A. O., Adekola, F., Obalowu, P., Alabi, A. G., Baran, A., & Samal, R. (2014). Optimization Study of a Nigerian Dolomite Ore Dissolution by Hydrochloric Acid. *Journal of Chemical Technology and Metallurgy*, 49(3), 280-287.
- Banerjee, A. (2016). Estimation Of Dolomite Formation: Dolomite Precipitation And Dolomitization. *Journal of Geological Society of India*, 87, 561-572.
- Basuki, & Sari, V. K. (2019). Efektifitas Dolomit Dalam Mempertahankan pH Tanah Inseptiol Perkebunan Tebu Blimbing Jatiroto. *Buletin Tanaman Tembakau, Serat & Minyak Industri*, 11(2), 58-64.
- Boggs, S. J. (2009). *Petrology of Sedimentary Rocks*. Cambridge University Press: Cambridge, UK.
- Chan, S. H., Parinov, I. A., & Topolov, V. Y. (2014). Advanced Materials. In V. Y. Topolov (Ed.), *Springer Proceedings in Physics*. 152. New York Dordrecht London: Springer International Publishing.
- Dewangga, B. S., Setyanugraha, A. M., & Edahwati, L. (2022). Uji Karakteristik Magnesium Fosfat dari Pelarutan Mineral Dolomit dengan Asam Fosfat. *Journal of Chemical Process Engineering*, 7(1), 1-8.
- Dong, B., Tian, Q., Xu, Z., Guo, X., Wang, Q., & Li, D. (2023). The Effect of Pre-Roasting on Atmospheric Sulfuric Acid Leaching of Saprolitic Laterites. *Journal Hydrometallurgy*, 1-8.
- Elias, M. (2002). Nickel laterite deposits – geological overview, resources and exploitation. *CODES Special Publication*.
- Febrini V., Ratnawulan, & Gusnaedi. (2014). Pengaruh Kalsinasi Terhadap Struktur Kristal Serpentin yang Terdapat di Jorong Sungai Padi Nagari Lubuak Gadang Kecematan Sangkir Kabupaten Solok Selatan. *Pillar of Physics*, 4, 97-104.



- Ginting, E. N., Pradiko, I., Farrasati, R., & Rahutomo, S. (2020). Pengaruh Rock Phosphate dan Dolomit Terhadap Distribusi Perakaran Tanaman Kelapa Sawit pada Tanah Ultisol. *Jurnal Agrikultura*, 31(1), 32-41.
- Gregg, J. M., Bish, D. L., Kaczmarek, S. E., & Machel, H. G. (2015). Mineralogy, Nucleation and Growth of Dolomite in the Laboratory and Sedimentary Environment: A Review. *Journal of Sedimentology*, 62(6), 1749-1769.
- Imen, Z., Borghaei, M. S., & Hassani, H. A. (2019). Comparison of the Effectiveness of Natural Dolomite and Modified Dolomite in the Removal of Heavy Metals From Aqueous Solutions. *Journal of Advances in Environmental Health Research*, 7, 61-74.
- Kulkarni, S. (2015). A Review on Studies and Research on Various Aspects of Leaching. *International Journal of Research & Review*, Vol.2, 579-583.
- Laksmanan, V., Sridhar, R., Chen, J., & Halim, M. (2016). Development of Mixed-Chloride Hydrometallurgical Processes for the Recovery of Value Metals from Various Resources. *Metallurgy Materials Engineering*, 69(1), 39-50.
- Li, G., Li, Z., & Ma, H. (2013). Comprehensive Use Of Dolomite-Talc Ore To Prepare Talc, Nano-MgO and Lightweight Caco3 Using An Acid Leaching Method. *Journal Applied Clay Science*, 86, 145-152.
- Mantilaka, M., Pitalawa, Karunaratne, & Rajapakse. (2014). Nanocrystalline Magnesium Oxide From Dolomite Via Poly (Acrylate) Stabilized Magnesium Hydroxide Colloids. *Colloids And Surfaces A: Physicochem. Eng*, 443, 201-208.
- Maulana, P. S., & Hidayat, T. M. (2015). Pengaruh Variasi Dolomit Material Lokal Kabupaten Bangkalan Sebagai Subsitusi Agregat Dalam Pembuatan Batako Terhadap Kuat Tekan Dan Absorbsi. *Jurnal Tek. Sipil*, 1(3), 1268.
- McCutcheon, J., Smith, A., & Johnson, B. (2015). The Effects of Magnesium on Human Health. *Journal of Health and Nutrition*, 28(4), 234-245.
- Moorkah, H. I., & Abolarin, M. S. (2005). Investigation Of The Properties Of Locally Available Dolomite For Refractory Applications. *Nigerian J. Tech*, 24(1), 79.
- Muntaha, M. (2007). Identifikasi Kekuatan Batu Kumbung (Batu Putih) Sebagai Salah Satu Alternatif Bahan Bangunan. *Jurnal Aplikasi*, 2(1).
- Ntengwe, F. W. (2010). The Leaching Of Dolomitic-Copper Ore Using Sulphuric Acid Under Controlled Conditions. *The Open Mineral Processing Journal*, 60.
- , N. (2017). *Studi Ekstraksi Mangan dari Bijih Oksida dengan Proses elindian Menggunakan Asam Sulfat*. Makassar: Skripsi Sarjana niversitas Hasanuddin.



- Olszak, H. M., & Jablonski, M. (2014). Thermal Behavior of Natural Dolomite. *Journal of Thermal Analysis and Calorimetry*, 119, 239 – 2248.
- Pichler, T., & Humphrey, J. D. (2001). Formation Of Dolomite In Recent Island-Arc Sediment Due To The Gas-Seawater-Sediment Interaction. *Journal of Sedimentary Research*, 71, 394-399.
- Prayudo, A. N., Novian , O., Setyadi, & Antaresti. (2015). Koefisien Transfer Massa Kurkumin dari Temulawak. *Jurnal Ilmiah Widya Teknik*, 14(01), 26-31.
- Raza, N., Zafar, I. Z., Haq, N. U., & Kumar, R. (2015). Leaching of natural magnesite ore in succinic acid solutions. *International Journal of Mineral Processing*, 25-30.
- Rohmawati, P. S., Sholica, P. S., & Seryatsih, W. (2019). Identification of Phase CaCO<sub>3</sub>/MgO in Bangkalan Dolomite Sand as An Antibacterial Substance. *J. Phys. Conf. Ser*, 1417, 120.
- Royani, A., & Sulistiyono, E. (2019). Ekstraksi Kalsium dari Bijih Dolomit Terkalsinasi Menggunakan Pelarutan Asam Klorida. *Jurnal Teknologi Mineral dan Batubara*, 13-22.
- Royani, A., Sulistiyono, E., & Sufiandi, D. (2016). Pengaruh Suhu Kalsinasi Pada Proses Dekomposisi Dolomit. *Jurnal Sains Materi Indonesia*, 18(1), 41-46.
- Royani, A., Sulistiyono, E., Prasetyo, A. B., & Subagja, R. (2018). Extraction of Magnesium from Calcined Dolomite Ore Using Hydrochoric Acid Leaching. *AIP Conference Proceedings*, 1-6.
- Saputri, D., & Rohmawati, L. (2021). Sintesis Magnesium Oksida (MgO) dari Dolomit Bangkalan dengan Metode Leaching. *Jurnal Teori dan Aplikasi Fisika*, 9(2), 203-210.
- Setiawan, I. (2016). Pengolahan Nikel Laterit Secara Pirometalurgi: Kini dan penelitian ke depan. *Seminar Nasional Sains dan Teknologi 2016 Fakultas Teknik Universitas Muhammadiyah Jakarta*.
- Sufriadin, Purwanto, Jihad, M. R., Aras, A., Santoso, A., & Hujanna, M. (2021). Analisis Mineralogi dan Kimia Dolomit Kabupaten Bone Bolango. *Jurnal Geomine*, 9(2), 95-102.
- Sulistiyono, E., Firdiyono, F., Natasha, N. C., & Amalia, Y. (2017). Comparison of Dolomite Crystal Structure, Calcination Dolomite, and Magnesium Hydroxide in Partial Calcination and Slaking Process. *Journal Materials Science and Engineering*.
- to, E., Firdiyono, F., Natasha, N. C., & Sufiandi, D. (2015). Pengaruh kuran Butiran Terhadap Struktur Kristal Pada Proses Kalsinasi Parsial olomit. *Jurnal Metalurgi*, 3, 25-132.



- Sulistiyono, E., Firdiyono, F., Natasha, N. C., Ramdhani, A., & Yunita, F. E. (2022). Percobaan Pendahuluan Proses Presipitasi Udara Panas Terhadap Bentuk Kristal Magnesium Karbonat. *Jurnal UMJ*, 1-6.
- Tang, Z., & B, L. (2014). Mgo Nanoparticles As Antibacterial Agent: Preparation And Activity. *Brazilian J. Chem. Eng.*, 31(3), 591-601.
- Ulinuha, M., Tyoso, A. W., & Kurniati, E. (2022). Ekstraksi Magnesium pada Dolomit Menggunakan Pelarut Asam Fosfat. *Journal of Chemical and Process Engineering*, 3(1), 69-74.
- Wahyudi, T., & Supriyatno, B. A. (2010). Uji Coba Pelarutan Dolomit Caro Dengan Asam Sulfat Menjadi Kiserit. *Jurnal Teknologi Mineral dan Batubara*, 6(4), 183-192.
- Wardhani, H. K., Samhadi, M., & Sari, N. K. (2020). Pemisahan Magnesium Dari Dolomit Dengan Penambahan Asam Nitrat Dengan Proses Ekstraksi. Seminar Nasional Teknik Kimia Soebardjo Brotohardjono XVI. UPN “Veteran” Jawa Timur. *Seminar Nasional Teknik Kimia Soebardjo Brotohardjono XVI. UPN “Veteran” Jawa Timur.*, 1-6.
- Warren, J. (2000). Dolomite: Occurrence, Evolution And Economically Important Associations. *Earth-Science Reviews*, 52, 1-81.
- Winarno, T., & Marin, J. (2020). *Buku Ajar Mineralogi*. Semarang: UNDIP PRESS Semarang.
- Zahrah, K. A., & Haider, A. A. (2019.). Influence of Different Concentrations of Nano-Magnesium Oxide on the Growth of Coelasrella Terrestris. *IOP Conf. Series: Journal of Physics*.



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## LAMPIRAN 1

### PERHITUNGAN PENGENCERAN ASAM KLORIDA (HCl)

#### **Pengenceran Asam Klorida (HCl)**

$$M_1 = 10.3$$

\*Larutan 1 M

$$M_1 V_1 = M_2 V_2$$

$$10.3 \times V_1 = 1 \times 100 \text{ mL}$$

$$V_1 = \frac{1 \times 100 \text{ mL}}{10.3}$$

$$= 9.71 \text{ mL}$$

\*Larutan 2 M

$$M_1 V_1 = M_2 V_2$$

$$10.3 \times V_1 = 2 \times 100 \text{ mL}$$

$$V_1 = \frac{2 \times 100 \text{ mL}}{10.3}$$

$$= 19.42 \text{ mL}$$

\*Larutan 3 M

$$M_1 V_1 = M_2 V_2$$

$$10.3 \times V_1 = 3 \times 100 \text{ mL}$$

$$V_1 = \frac{3 \times 100 \text{ mL}}{10.3}$$

$$= 29.13 \text{ mL}$$

\*Larutan 4 M

$$M_1 V_1 = M_2 V_2$$

$$10.3 \times V_1 = 4 \times 100 \text{ mL}$$

$$V_1 = \frac{4 \times 100 \text{ mL}}{10.3}$$

$$= 38.83 \text{ mL}$$



**LAMPIRAN 2**  
**HASIL ANALISIS ATOMIC ABSORPTION SPECTROMETER (AAS)**



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Telp. 0411-586016 • Fax. 0411-588551 • Email : [lpps.fmipa.unhas@gmail.com](mailto:lpps.fmipa.unhas@gmail.com)



**LAPORAN HASIL PENGUJIAN**

**CERTIFICATE OF ANALYSIS**

**Nomor Pekerjaan : LPPS.AJ-2312-11/9**

*Job Number*

**Dipersembahkan Kepada**

*Presented To*

Kepada Yth	: Amalia Ramadani	Jabatan	: Peneliti
Attention		Job Title	
Nama Pelanggan	: Amalia Ramadani	Tujuan Pengujian	: Analisis Logam
Customer Name		Purpose of analysis	
Alamat/Universitas	: Jl. Poros Malino, Kec. Bontomarannu	No. Faks/ Fax No.	: -
Address/University		No. Telp./ Phone No.	: 081258394308
Tanggal Sampel Diterima	: 19 Desember 2023	Tanggal Sampel	: 22 Desember 2023 –
Date of Sample Receipt		Dianalisis	11 Januari 2024
Email	: misamaliar10@gmail.com	Date of Sample Analysed	
Email		Total Halaman	: 2
Nama Pengujian	: Analisis Magnesium (Mg) pada Larutan	Total of pages	
Name of analysis	dan Padatan Dolomit Menggunakan AAS		

Hasil hanya berhubungan dengan contoh yang diuji dan laporan ini tidak boleh digandakan kecuali seluruhnya.  
*The result relate only to the samples tested and this report shall not be reproduced except in full*



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Telp. 0411-586016 • Fax. 0411-588551 • Email : [lpps.fmipa.unhas@gmail.com](mailto:lpps.fmipa.unhas@gmail.com)

**LAPORAN HASIL PENGUJIAN  
CERTIFICATE OF ANALYSIS**

**Nomor Pekerjaan**

**: LPPS.AJ-2312-11/9**

**I. Pelanggan / Principal**

1.1 Nama / Name : Amalia Ramadani  
1.2 Alamat / Address : Jln. Poros Malino, Kec. Bontomarannu  
1.3 Telepon / Phone : 081258394308  
1.4 Personil Penghubung / Contact Person : -  
1.5 Email / Email : misamaliar10@gmail.com

**II. Contoh Uji / Sample**

2.1 Kode Sampel / Sampel Code : LPPS.A-2312-11/9a – 9i  
2.2 Kemasan / Packaging : Botol Plastik dan Plastik Sampel  
2.3 Nama Sampel / Sample Name : Larutan (8) dan Padatan (1)  
2.4 Jumlah Sampel / Number of Sample : 9  
2.5 Tanggal Sampling / Date of Sampling : -  
2.6 Diterima / Date of Received : 19 Desember 2023  
2.7 Tanggal Uji / Date of Analysis : 12 Desember 2023 – 11 Januari 2024  
2.8 Jenis Uji / Type of Analysis : Logam Mg AAS

**III. Hasil Uji / Result**

Kode Sampel	Nama Sampel	Satuan	Konsentrasi Logam Mg
LPPS.A-2312-11/9a	AR-1	mg/L	3428.39
LPPS.A-2312-11/9b	AR-2	mg/L	31985.53
LPPS.A-2312-11/9c	AR-3	mg/L	20717.17
LPPS.A-2312-11/9d	AR-4	mg/L	41732.28
LPPS.A-2312-11/9e	AR-5	mg/L	1723.77
LPPS.A-2312-11/9f	AR-6	mg/L	19929.77
LPPS.A-2312-11/9g	AR-7	mg/L	30825.71
LPPS.A-2312-11/9h	AR-8	mg/L	25356.46
LPPS.A-2312-11/9i	Sampel awal	mg/kg	191583.26



Makassar, 16 Januari 2024  
Penanggung Jawab Mutu

Prof. Dr. Nunuk Hariani Soekamto, MS  
NIP. 19601215 198702 2 001



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maupun mengutip/menyalin sebagian isi hasil uji ini

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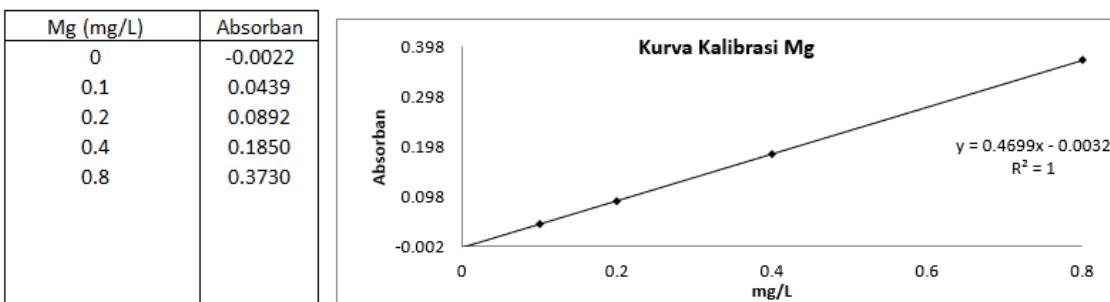
 <b>LAB. PPS FMIPA</b> <b>UNHAS</b>	<b>FORMULIR NO:</b> <b>FSOP-7.8-LPPS-FMIPAUH-01.4</b>	<b>Tanggal Berlaku : 1 April 2019</b>
	<b>REKAMAN HASIL ANALISIS</b>	
	<b>Edisi/Revisi Ke : 1/0</b> <b>Halaman : 1/3</b>	

### REKAMAN HASIL ANALISIS

Nomor Pekerjaan : LPPS.A-2312-11/9a – 9i  
 Tanggal Penerimaan : 19 Desember 2023  
 Tanggal Analisis : 22 Desember 2023 – 11 Januari 2024  
 Suhu Ruangan : 21,9 °C  
 Kelembapan Ruangan : 59% RH

#### 1. Analisis Logam Magnesium (Mg)

Optimasi Analisa Magnesium (Mg)			
Type Alat: AA 7000 Shimadzu : ASC-7000 HCL Mg Hamamatsu P. gelombang : 285.2 nm Lamp Current Low (Peak) (mA) : 8	Slit width : 0.7 nm Flame type : Air-C <sub>2</sub> H <sub>2</sub> Burner Height : 7,0 nm Burner angle : 0 degree	Fuel Gas Flow Rate (0.8-4.0) : 1.8 L/min Support gas flow rate (13.5-17.) : 15 L/min P. Gelombang Max : 285.20 nm	



Persyaratan	Hasil	Keterangan keberterimaan hasil	R <sup>2</sup> =	1
Linearitas ( r )	≥ 0.99	1.0000	Memenuhi	
%R	75%-120%	99.91	Memenuhi	
Kontrol sampel	Absorban [Mg] mg/L		Kontrol	Absorban [Mg] mg/L
Mg 0.2 mg/L	0.0913    0.20		Blanko	0.0533    0.12
	0.0901    0.20			0.0530    0.12
Rata-rata	0.0907    0.20		Rata-rata	1000    119.92

Kadar Air	B. Cawan Kosong (G)	Berat Sebelum Pemanasan (G)	B. Sampel (B. Basah) (G)	Berat setelah Pemanasan (G)	B. Sampel (B. Kering) (G)	Kadar Air (%)
LPPS.A.2312-11/9i	30.4746	30.9751	0.5005	30.9734	0.4988	0.34

Berat sampel peleburan	B. Cawan Kosong (G)	B. Cawan+sampel sblm di+pelebur (G)	B. Sampel (B. kering) (G)	B.cawan+sampel +pelebur sblm dilebur (G)	B.cawan+sampel +pelebur setelah dilebur (G)	Berat sampel setelah dilebur (G)
9i	30.4746	30.9734	0.4988	32.4784	31.5478	1.0732

 <b>LAB. PPS FMIPA</b> <b>UNHAS</b>	<b>FORMULIR NO:</b> <b>FSOP-7.8-LPPS-FMIPAUH-01.4</b>						<b>Tanggal Berlaku : 1 April 2019</b>		
							<b>Edisi/Revisi Ke : 1/0</b>		
	<b>REKAMAN HASIL ANALISIS</b>						<b>Halaman : 2/3</b>		

Kode Sampel	Absorban	[Mg] (mg/L)	fp (kali)	[Mg] x fp (mg/L)	([Mg] x fp) - [Blanko]	Berat sebelum peleburan (g)	V. Sampel (L)	Kadar Mg (mg/kg= ppm)	[Mg] dilaporkan (mg/kg= ppm)
LPPS.A.2312-11/9i	0.0918 0.0933	0.20 0.21	10000	2037.67	1917.7485	0.5005	0.05	191583.26	191583.26
<b>Rata-Rata</b>	<b>= 0.0926</b>	<b>0.2038</b>							

Kode Sampel	Absorban (mg/L)	[Mg] (kali)	fp (kali)	[Mg] x fp (mg/L)	[Mg] yang dilaporkan (mg/L=ppm)
LPPS.A.2312-11/9a	0.1580 0.1578	0.34 0.34	10000	3428.39	3428.39
<b>Rata-Rata</b>	<b>= 0.1579</b>	<b>0.3428</b>			

Kode Sampel	Absorban (mg/L)	[Mg] (kali)	fp (kali)	[Mg] x fp (mg/L)	[Mg] yang dilaporkan (mg/L=ppm)
LPPS.A.2312-11/9b	0.1478 0.1464	0.32 0.32	100000	31985.53	31985.53
<b>Rata-Rata</b>	<b>= 0.1471</b>	<b>0.3199</b>			
LPPS.A.2312-11/9c	0.0940 0.0943	0.21 0.21	100000	20717.17	20717.17
<b>Rata-Rata</b>	<b>= 0.0942</b>	<b>0.2072</b>			
LPPS.A.2312-11/9d	0.1930 0.1928	0.42 0.42	100000	41732.28	41732.28
<b>Rata-Rata</b>	<b>= 0.1929</b>	<b>0.4173</b>			
LPPS.A.2312-11/9e	0.0047 0.0051	0.02 0.02	100000	1723.77	1723.77
<b>Rata-Rata</b>	<b>= 0.0049</b>	<b>0.0172</b>			
LPPS.A.2312-11/9f	0.0907 0.0902	0.20 0.20	100000	19929.77	19929.77
<b>Rata-Rata</b>	<b>= 0.0905</b>	<b>0.1993</b>			
LPPS.A.2312-11/9g	0.1414 0.1419	0.31 0.31	100000	30825.71	30825.71
<b>Rata-Rata</b>	<b>= 0.1417</b>	<b>0.3083</b>			
	0.1166	0.25			
	0.1153	0.25	100000	25356.46	25356.46
	<b>= 0.1160</b>	<b>0.2536</b>			



 <b>LAB. PPS FMIPA UNHAS</b>	<b>FORMULIR NO:</b> <b>FSOP-7.8-LPPS-FMIPAUH-01.4</b>	<b>Tanggal Berlaku : 1 April 2019</b>
	<b>REKAMAN HASIL ANALISIS</b>	<b>Edisi/Revisi Ke : 1/0</b>
		<b>Halaman : 3/3</b>

**Perhitungan**

$$\text{mg/kg} = \frac{\text{konsentrasi (mg/L)} \times \text{Volume (mL)}}{\text{berat sampel (g)}}$$

$$\% = \frac{\text{konsentrasi (mg/L)} \times \text{Volume (L)}}{\text{berat sampel (mg)}} \times 100\%$$

$$\text{Kadar Air} = \frac{(\text{berat cawan kosong + sampel setelah pemanasan} - \text{berat cawan kosong + sampel sebelum pemanasan})}{\text{berat sampel (g)}} \times 100\%$$

Makassar, 16 Januari 2024

Analis



Fibiyanthy, S.Si, M.Si  
NIP. 19810202 200604 2 001



**LAMPIRAN 3  
PERHITUNGAN TINGKAT EKSTRAKSI Mg**



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Tingkat Ekstraksi Mg dihitung menggunakan rumus efisiensi pelindian dalam Dong *et al.*, (2023) pada persamaan 4 berikut:

$$\eta = \frac{C_i \times V}{m \times W_i} \times 100$$

Dimana:

$\eta$  = Tingkat Ekstraksi (%)

$C_i$  = Konsentrasi logam (mg/L) dalam pregnant leach solution (PLS)

$V$  = Volume PLS (L)

$m$  = Massa sampel (Kg)

$W_i$  = Kadar logam (mg/Kg)

### **Tingkat Ekstraksi Magnesium (Mg)**

Volume PLS (V) = 0,005 L

Mass sampel (m) = 0,01 Kg

Kadar Mg ( $W_i$ ) = 19.1583,26 mg/Kg

1. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 1 M ( $C_i = 3.428,39$  mg/L)

$$\begin{aligned}\text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{3.428,39 \times 0,005}{0,01 \times 19.1583,26} \times 100\% \\ &= 8,95\%\end{aligned}$$

2. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 2 M ( $C_i = 31.985,53$  mg/L)

$$\begin{aligned}\text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{31.985,53 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 83,48\%\end{aligned}$$

3. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 3 M ( $C_i = 20.717,17$  mg/L)

$$\text{Tingkat Ekstraksi Mg (\%)} = \frac{C_i \times V}{m \times W_i} \times 100\%$$



$$= \frac{20.717,17 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ = 54,07\%$$

4. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 4 M  
( $C_i = 41.732,28 \text{ mg/L}$ )

$$\begin{aligned} \text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{41.732,28 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 100\% \end{aligned}$$

5. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 1 M  
( $C_i = 1.723,77 \text{ mg/L}$ ) (Setelah Kalsinasi)

$$\begin{aligned} \text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{1.723,77 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 4,5\% \end{aligned}$$

6. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 2 M  
( $C_i = 19.929,77 \text{ mg/L}$ ) (Setelah Kalsinasi)

$$\begin{aligned} \text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{19.929,77 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 52,01\% \end{aligned}$$

7. Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 3 M  
( $C_i = 30.825,71 \text{ mg/L}$ ) (Setelah Kalsinasi)

$$\begin{aligned} \text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{30.825,71 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 80,45\% \end{aligned}$$



Tingkat Ekstraksi Magnesium pada kondisi pelindian 120 menit, 70°C, 4 M  
= 25.356,46 mg/L) (Setelah Kalsinasi)

$$\begin{aligned}\text{Tingkat Ekstraksi Mg (\%)} &= \frac{C_i \times V}{m \times W_i} \times 100\% \\ &= \frac{25.356,46 \times 0,05}{0,01 \times 19.1583,26} \times 100\% \\ &= 66,18\%\end{aligned}$$



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## **HASIL ANALISIS X-RAY DIFFRACTION (XRD) SAMPEL AWAL**



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# Match! Phase Analysis Report

Sample: DLM-Amalia



Sample D	File name	DLM-Amalia.txt
	File path	D:/OPPI/XRD AWAL/DLM-Amalia
	Data colle	Jun 1, 2024 20:32:39
	Data rang	5.000° - 70.000°
	Number o	3251
	Step size	0.020
	Rietveld refinement converged	No
	Alpha2 subtracted	No
	Background subtr.	No
	Data smoothed	Yes
	Radiation	X-rays
	Wavelength	1.541874 Å

## Matched Phases

Index	Amount (%)	Name	Formula sum
A	85.4	Dolomite	C2 Ca Mg O6
B	14.6	Calcite	C Ca O3
	13.6	Unidentified peak area	

### A: Dolomite (85.4 %)

Formula sum	C2 Ca Mg O6
Entry number	96-900-1007
Figure-of-Merit (FoM)	0.866905
Total number of peaks	106
Peaks in range	48
Peaks matched	32
Intensity scale factor	0.26
Space group	R -3
Crystal system	trigonal (hexagonal axes)
Unit cell	a=4.8162 Å c= 16.1320 Å
I/c	2.73
Calc. density	2.834 g/cm³
Reference	Reeder R. J., Markgraf S. A., "High-temperature crystal chemistry of dolomite Sample: Anisotropic refinement, Temperature = 400 deg C", American Mineralogist <b>71</b> , 795-804 (1986)

### B: Calcite (14.6 %)

Formula sum	C Ca O3
Entry number	96-900-9669
Figure-of-Merit (FoM)	0.747990
Total number of peaks	76
Peaks in range	36
Peaks matched	22
Intensity scale factor	0.06
Space group	R -3 c
Crystal system	trigonal (hexagonal axes)
Unit cell	a=4.9920 Å c= 17.0690 Å
I/c	3.58
Calc. density	2.705 g/cm³
Reference	Sitpu H., O'Connor B H, Li D., "Comparative evaluation of the March and generalized spherical harmonic preferred orientation models using X-ray diffraction data for molybdate and calcite powders Note: GSH model Locality: synthetic", Journal of Applied Crystallography <b>38</b> , 158-167 (2005)

## Candidates

Name	Formula	Entry No.	FoM
Tungsten	W	96-900-6510	0.6231
Tungsten	W	96-900-6514	0.6226
Tungsten	W	96-900-6509	0.6182
Iron	Fe	96-900-0658	0.6175
Iron	Fe	96-900-6589	0.6175
Iron	Fe	96-900-6588	0.6172
Iron-alpha	Fe	96-900-8537	0.6172
Iron	Fe	96-411-3937	0.6170
Cobalt iron arsenide (1.9/0.1/1) - \$-beta	As Co1.88 Fe0.1	96-100-8517	0.6151
hexagonal boron nitride	B N	96-201-6171	0.6117
Periclase	Mg O	96-901-3272	0.6115
Periclase	Mg O	96-901-3239	0.6109
Tungsten	W	96-900-6513	0.6088
Tungsten	Fe	96-411-3932	0.6047
Periclase	Mg O	96-901-3245	0.6015
Silicon oxide (Quartz low)	O2 Si	96-101-1160	0.5588
Quartz	O2 Si	96-901-2601	0.5584
Quartz	O2 Si	96-900-9667	0.5565
Silicon oxide \$-alpha (Quartz low)	O2 Si	96-101-1173	0.5524
Silicon oxide - \$-alpha (Quartz low)	O2 Si	96-101-1177	0.5524
Quartz	O2 Si	96-901-3222	0.5513
Silicon oxide (Quartz)	O2 Si	96-500-0036	0.5511
Silicon oxide - \$-alpha (Quartz low)	O2 Si	96-101-1098	0.5238
Calcite	C Ca O3	96-900-7688	0.3897
Calcite	C Ca O3	96-901-6023	0.3894
Calcite	C Ca O3	96-901-5391	0.3893
Calcite	C Ca O3	96-900-0096	0.3892
Calcite	C Ca O3	96-900-0966	0.3883
Calcite	C Ca O3	96-900-7690	0.3877
Calcite	C Ca O3	96-901-6707	0.3851
Calcite	C Ca O3	96-900-9669	0.3780
Calcite	C Ca O3	96-900-9668	0.3775
Calcite	C Ca O3	96-900-0967	0.3679
Calcite	C Ca O3	96-900-1299	0.3632
Calcium carbonate (Calcite)	Ca C O3	96-101-0929	0.3536
Calcite	C Ca O3	96-901-6706	0.3496
Calcite	C Ca O3	96-900-1298	0.3445
Calcite	C Ca O3	96-901-4892	0.3439
Calcium carbonate (Calcite)	Ca C O3	96-101-0963	0.3436
Calcite	C Ca O3	96-901-6180	0.3391
baryocalcite	C2 Ba Ca O6	96-591-0096	0.3380
Calcite	C Ca O3	96-900-0968	0.3369
Magnesium calcite	C Ca0.94 Mg0.06 O3	96-721-4218	0.3349
Calcite	C Ca O3	96-901-4878	0.3346
Magnesium Calcium Carbonate (.1/.9/1) (Calcite, magnesian)	C Ca0.9 Mg0.1 O3	96-721-4219	0.3300
Calcite	C Ca O3	96-900-0575	0.3286
Quartz	O2 Si	96-900-5019	0.3256
Calcite	C Ca O3	96-901-4416	0.3243
Calcite	C Ca O3	96-901-4773	0.3230
Calcite	C Ca O3	96-901-5836	0.3229

and 67 others...

## Search-Match

### Settings

Reference database used	COD-Inorg REV140301 2015.07.06
Automatic zero point adaptation	Yes
Minimum figure-of-merit (FoM)	0.60
2theta window for peak corr.	0.28 deg.
Minimum rel. int. for peak corr.	1
Parameter/influence 2theta	0.50
Parameter/influence intensities	0.50
Parameter multiple/single phase(s)	0.50

## Criteria for entries added by user

### Reference:

Entry number:	96-100-1742 96-100-1744 96-101-0918 96-101-0929 96-101-0963 96-591-0096 96-721-4218 96-721-4219 96-900-0096 96-900-0575 96-900-0966 96-900-0967 96-900-0968 96-900-0969 96-900-0970 96-900-0971 96-900-1298 96-900-1299 96-900-7287 96-900-7688 96-900-7690 96-900-9668 96-900-9669 96-900-9866 96-901-3466 96-901-4217 96-901-4345 96-901-4393 96-901-4416 96-901-4525 96-901-6201 96-901-6203 96-901-6180 96-901-6210 96-901-6465 96-901-6706 96-901-6707 96-101-1098 96-901-1160 96-101-1173 96-101-1177 96-101-1201 96-110-0020 96-500-0036 96-900-0778 96-900-0779 96-900-0780 96-900-0781 96-900-5019 96-900-5020 96-900-5021 96-900-5022 96-900-5023 96-900-5024 96-900-5025 96-900-5026 96-900-5027 96-900-5028 96-900-5029 96-900-5030 96-900-5031 96-900-5032 96-900-5033 96-900-5034
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### Peak List

No.	2theta [deg]	I/I0	FWHM	Matched	
1		6.33	0.2930	B	
2		20.43	0.3077	A	
3		5.05	0.2775		
4		81.28	0.3600	B	
5		388.67	0.2996	A	
6		16.61	0.3779	A	
7		11.78	0.3200	A	
8	36.06	2.4908	13.42	0.3600	B
9	37.24	2.4145	34.31	0.2704	A
10	39.52	2.2803	18.71	0.3445	B
11	40.96	2.2034	90.09	0.3012	A
12	43.28	2.0905	14.87	0.2800	B
13	43.66	2.0732	10.07	0.2800	A
14	44.74	2.0257	55.27	0.3011	A
15	47.62	1.9097	17.54	0.4000	B
16	48.64	1.8720	40.11	0.6000	B
17	49.04	1.8576	29.56	0.4800	A
18	50.10	1.8208	149.84	1.2705	A
19	50.78	1.7980	50.83	0.3304	A
20	57.52	1.6023	6.92	0.3145	B
21	58.66	1.5739	12.15	0.3600	A
22	59.58	1.5517	27.07	0.4124	A
23	63.12	1.4730	23.64	0.5200	AB
24	64.72	1.4404	29.41	1.3060	AB
25	65.76	1.4201	3.95	0.2332	AB
26	67.12	1.3946	20.31	0.4112	

### Rietveld Refinement using FullProf

Calculation was not run or did not converge.

### Integrated Profile Areas

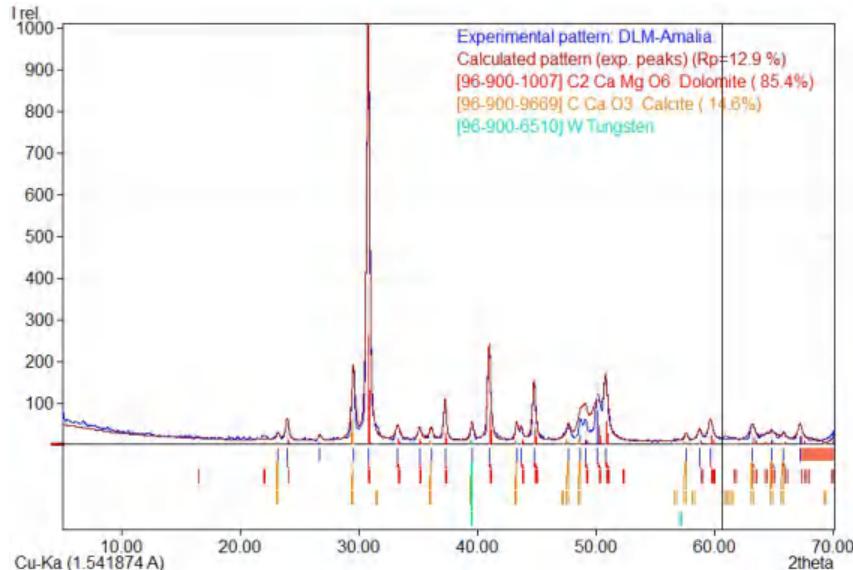
#### Based on calculated profile

Profile area	Counts	Amount
Overall diffraction profile	115750	100.00%
Background radiation	52987	45.78%
Diffraction peaks	62763	54.22%
Peak area belonging to selected phases	55176	47.67%
Peak area of phase A (Dolomite)	45165	39.02%
Peak area of phase B (Calcite)	10011	8.65%
Unidentified peak area	15791	13.64%

### Peak Residuals

Peak data	Counts	Amount
Overall peak intensity	1282	100.00%
Peak intensity belonging to selected phases	1083	84.46%
Unidentified peak intensity	199	15.54%

### Diffraction Pattern Graphics



# Match! Phase Analysis Report

Sample: AR-DOLOMIT-AK (5-70)



File name	AR-DOLOMIT-AK.RAW
File path	D:/OPPI/Semester 7/referensi Ta/laporan/referensi/AR-DOLOMIT-AK
Data colle	Jul 4, 2024 22:50:26
Data rang	5.000° - 70.000°
Number o	3251
Step size	0.020
Rietveld refinement converged	No
Alpha2 subtracted	No
Background subtr.	No
Data smoothed	Yes
Radiation	X-rays
Wavelength	1.540600 Å

## Matched Phases

Index	Amount (%)	Name
A	32.7	Calcium oxide
B	25.4	Brucite (deuterated)
C	25.3	Periclase
D	16.5	Calcium hydroxide Portlandite
	11.2	Unidentified peak area

### Formula sum

Ca O
D1.998 Mg O2
Mg O
Ca H2 O2

### A: Calcium oxide (32.7 %)

Formula sum	Ca O
Entry number	96-720-0687
Figure-of-Merit (FoM)	0.906520
Total number of peaks	13
Peaks in range	5
Peaks matched	5
Intensity scale factor	0.32
Space group	F m -3 m
Crystal system	cubic
Unit cell	a= 4.8107 Å
I/c	5.16
Calc. density	3.345 g/cm³
Reference	Verbraeken Maarten C., Suard Emmanuelle, Irvine John T. S., "Structural and electrical properties of calcium and strontium hydrides", Journal of Materials Chemistry <b>19</b> (18), 2766 (2009)

### B: Brucite (deuterated) (25.4 %)

Formula sum	D1.998 Mg O2
Entry number	96-900-1587
Figure-of-Merit (FoM)	0.249552
Total number of peaks	18
Peaks in range	8
Peaks matched	2
Intensity scale factor	0.12
Space group	P -3 m 1
Crystal system	trigonal (hexagonal axes)
Unit cell	a= 3.1382 Å c= 4.7130 Å
I/c	2.48
Calc. density	2.491 g/cm³
Reference	Parise J. B., Leinenweber K., Weidner D. J., Tan K., Von Dreele R. B., "Pressure-induced H bonding: Neutron diffraction study of brucite, Mg(OD)2, to 9.3 GPa P = 0.4 GPa", American Mineralogist <b>79</b> , 193-196 (1994)

### C: Periclase (25.3 %)

Formula sum	Mg O
Entry number	96-900-6791
Figure-of-Merit (FoM)	0.828755
Total number of peaks	10
Peaks in range	3
Peaks matched	2
Intensity scale factor	0.16
Space group	F m -3 m
Crystal system	cubic
Unit cell	a= 4.2081 Å
I/c	3.29
Calc. density	3.592 g/cm³
Reference	Zhang J., "Effect of pressure on the thermal expansion of MgO up to 8.2 GPa Sample: Run1 at T = 673 K, P = 2.66 GPa", Physics and Chemistry of Minerals <b>27</b> , 145-148 (2000)

### D: Calcium hydroxide Portlandite (16.5 %)

Formula sum	Ca H2 O2
Entry number	96-100-1770
Figure-of-Merit (FoM)	0.789147
Total number of peaks	23
Peaks in range	12
Peaks matched	7
Intensity scale factor	0.12
Space group	P -3 m 1
Crystal system	trigonal (hexagonal axes)
Unit cell	a= 3.5890 Å c= 4.9110 Å
I/c	3.84
Meas. density	2.240 g/cm³
Calc. density	2.245 g/cm³
Reference	Desgranges L., Grebille D., Calvarin g., Chevrier G., Floquet N., Niepce J-C., "Hydrogen thermal motion in calcium hydroxide: Ca (OH)2", Acta Crystallographica B (39,1983-) <b>49</b> , 812-817 (1993)

## Candidates

Name	Formula	Entry No.	FoM
Portlandite	Ca H2 O2	96-900-0113	0.6522
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-1770	0.6001
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-1769	0.5997
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-1789	0.5996
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-1788	0.5990
Ca(OH)2	Ca H2 O2	96-702-0139	0.5886
Portlandite	Ca H2 O2	96-900-6833	0.5869
Portlandite	Ca H2 O2	96-900-9099	0.5680
Calcium hydroxide	Ca H2 O2	96-100-0046	0.5663
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-8781	0.5663
Calcium hydroxide (Portlandite)	Ca H2 O2	96-100-8782	0.5626
Portlandite	Ca H2 O2	96-900-0114	0.5560
Calcite	C Ca O3	96-900-7287	0.4687
Quartz	O2 Si	96-901-2606	0.4494
Barium titanium magnesium oxide (1.2/6.8/1.2/16) (Hollandite)	Ba1.2 Mg1.2 O16 Ti6.8	96-100-8949	0.4098
Baryocalcite	C2 Ba Ca O6	96-900-9866	0.4068
Quartz	O2 Si	96-900-0780	0.3966
Baryocalcite	Ca H2 O2	96-210-1034	0.3702
Baryocalcite	C2 Ba Ca O6	96-901-3466	0.3243
Barium titanium magnesium oxide (1.2/6.8/1.2/16) (Hollandite (Ti, Mg))	Ba1.2 Mg1.2 O16 Ti6.8	96-100-8814	0.3227
Barium titanium magnesium oxide (1.2/6.8/1.2/16) (Hollandite)	Ba1.2 Mg1.2 O16 Ti6.8	96-100-8952	0.3227
Dolomite	C Ca0.5 Mg0.5 O3	96-900-3513	0.3084
Dolomite	C2 Ca1.13 Mg0.87 O6	96-900-4932	0.2992
Calcium nitrate tetrahydrate (Nitrocalcite)	Ca H8 N2 O10	96-100-1744	0.2977
Nitrocalcite	Ca N2 O10	96-901-5488	0.2977
Calcium barium carbonate (Baryocalcite)	Ba Ca O6	96-101-0918	0.2832
Dolomite	C2 Ca Mg O6	96-900-1007	0.2795
Dolomite	C2 Ca Mg O6	96-900-1011	0.2795
Dolomite	C2 Ca1.14 Mg0.86 O6	96-900-4934	0.2794
Dolomite	C2 Ca1.17 Mg0.83 O6	96-900-4930	0.2768
Dolomite	C2 Ca Mg O6	96-900-0084	0.2702
Magnesium hydroxide (Brucite)	C Ca0.5 Mg0.5 O3	96-900-3528	0.2490
Quartz	H2 Mg O2	96-100-0055	0.2471
Dolomite	O2 Si	96-900-0779	0.2441
Dolomite	C2 Ca Fe0.33 Mg0.67 O6	96-900-4935	0.2436
Dolomite	C Ca0.5 Mg0.5 O3	96-900-3512	0.2434
Dolomite	C Ca0.5 Mg0.5 O3	96-900-3514	0.2387
Dolomite	C2 Ca Mg O6	96-900-3509	0.2353
Dolomite	C Ca0.5 Mg0.5 O3	96-900-3529	0.2345
Calcium Magnesium Carbonate (Dolomite)	C2 Ca Mg O6	96-151-7798	0.2338
Dolomite	C2 Ca Mg O6	96-900-1005	0.2323
Dolomite	C2 Ca Mg O6	96-900-1009	0.2323

Dolomite	C Ca <sub>0.5</sub> Mg <sub>0.5</sub> O <sub>3</sub>	96-900-3526	0.2304
Dolomite	C Ca <sub>0.5</sub> Mg <sub>0.5</sub> O <sub>3</sub>	96-900-3530	0.2301
Dolomite	C Ca <sub>0.5</sub> Mg <sub>0.5</sub> O <sub>3</sub>	96-900-3527	0.2284
Quartz	O <sub>2</sub> Si	96-900-5026	0.2259
Quartz	O <sub>2</sub> Si	96-900-5027	0.2259
Quartz	O <sub>2</sub> Si	96-900-5028	0.2259
Quartz	O <sub>2</sub> Si	96-900-5029	0.2259
Quartz	O <sub>2</sub> Si	96-900-5030	0.2259
Quartz	O <sub>2</sub> Si	96-900-5031	0.2259
Quartz	O <sub>2</sub> Si	96-900-5033	0.2259

## Search-Match

## Settings

Reference database used	COD-Inorg REV140301 2015.07.06
Automatic zeroipnt adaptation	Yes
Minimum figure-of-merit (FOM)	0.60
2theta window for peak corr.	0.28 deg.
Minimum rel. int. for peak corr.	1
Parameter/influence 2theta	0.50
Parameter/influence intensities	0.50
Parameter multiple/single phase(s)	0.50

#### **Criteria for entries added by user**

## Reference

**Entry number:**

## Peak List

No.	2theta [°]	d [Å]	I/I₀	FWHM	Matched
1	18.02	4.9187	76.28	0.7812	D
2	28.76	3.1016	23.25	0.5235	D
3	32.26	2.7727	125.09	0.2784	A
4	33.18	2.6979	0.56	0.3640	B
5	34.16	2.6227	135.34	0.9097	D
6	37.40	2.4026	314.50	0.2525	A
7	42.96	2.1036	156.83	0.2759	C
8	46.94	1.9341	7.24	0.4547	D
9	47.68	1.9058	71.07	1.9305	
10	50.84	1.7945	36.37	0.6582	D
11	53.92	1.6991	184.04	0.2978	A
12	62.36	1.4878	95.36	0.3328	B,C
13	64.20	1.4496	58.57	0.3445	A,D
14	67.44	1.3876	67.65	0.4162	A

Rietveld Refinement using FullProf

Calculation was not run or did not converge

## Integrated Profile Area

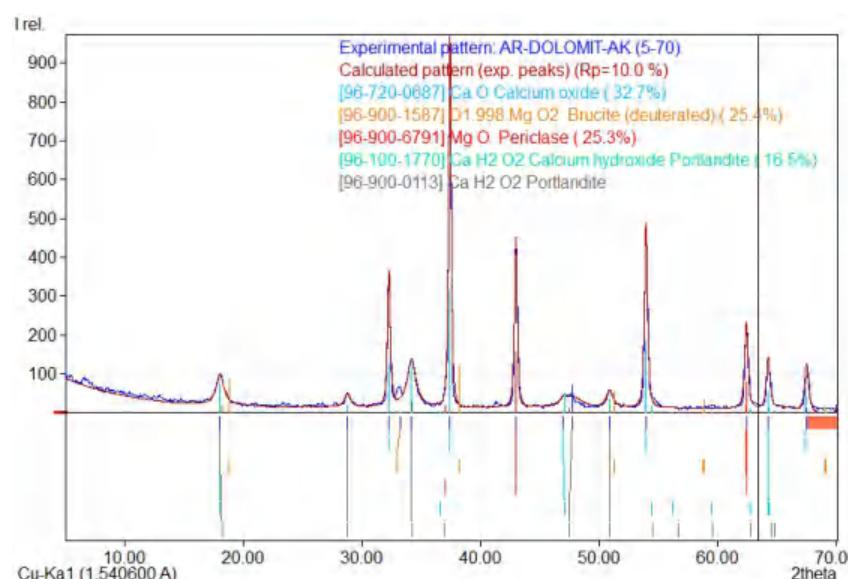
Based on calculated profile

<b>Profile area</b>	<b>Counts</b>	<b>Amount</b>
Overall diffraction profile	94031	100.00%
Background radiation	44950	47.80%
Diffraction peaks	49081	52.20%
Peak area belonging to selected phases	43336	46.09%
<i>Peak area of phase A (Calcium oxide)</i>	21367	22.72%
<i>Peak area of phase B (Brucite (deuterated))</i>	3046	3.24%
<i>Peak area of phase C (Periclase)</i>	8079	8.59%
<i>Peak area of phase D (Calcium hydroxide Portlandite)</i>	10844	11.53%
Unidentified peak area	10559	11.23%

## Peak Residuals

<b>Peak data</b>	<b>Counts</b>	<b>Amount</b>
Overall peak intensity	919	100.00%
Peak intensity belonging to selected phases	843	91.68%
Unidentified peak intensity	76	8.32%

## Diffraction Pattern Graphics



## HASIL ANALISIS X-RAY DIFFRACTION (XRD) SAMPEL RESIDU



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## Match! Phase Analysis Report

Sample: AMALIA-2-MOL-2-JAM

Sample Details	
File name	AMALIA-2-MOL-2-JAM.txt
File path	D:\OPPII\Semester 7\referensi Ta\laporan\referensi\AMALIA-2-MOL-2-JAM
Data colle	Jun 8, 2024 22:22:28
Data rang	5.000° - 70.000°
Number o	
Step size	0.020
Rietveld refinement converged	No
Alpha2 subtracted	No
Background subtr.	No
Data smoothed	Yes
Radiation	X-rays
Wavelength	1.541874 Å

## Matched Phases

<b>Index</b>	<b>Amount (%)</b>	<b>Name</b>	<b>Formula sum</b>
A	100.0	Magnesium calcite	Ca 0.94 Mg 0.06 O 100.00
	15.5	Unidentified peak area	

#### A: Magnesium calcite (100.0 %)

Formula sum	$\text{Ca Ca}0.94 \text{Mg}0.06 \text{O}_3$
Entry number	96-721-4218
Figure-of-Merit (FoM)	0.863146
Total number of peaks	86
Peaks in range	33
Peaks matched	24
Intensity scale factor	0.30
Space group	$R\bar{3} c$
Crystal system	trigonal (hexagonal axes)
Unit cell	$a = 4.9630 \text{ \AA}$ $c = 16.9570 \text{ \AA}$
$I/I_0$	3.42
Calc. density	2.729 g/cm <sup>3</sup>
Reference	Falini G., Fermani S., Gazzano M., Ripamonti A, "Structure and morphology of synthetic magnesium calcite", Journal of Materials Chemistry 8, 1061-1065 (1998)

## Candidates

Name	Formula	Entry No.	FoM
Calcite	Ca Ca O3	96-900-7690	0.539
Calcite	C Ca O3	96-900-0966	0.536
Calcite	C Ca O3	96-900-9668	0.534
Calcite	C Ca O3	96-901-6023	0.534
Calcite	C Ca O3	96-901-5391	0.534
Calcite	C Ca O3	96-900-0967	0.534
Magnesium dizinc cerium	Ce Mg Zn2	96-221-4424	0.531
Calcite	C Ca O3	96-900-0096	0.530
Calcite	C Ca O3	96-901-6707	0.529
Calcite	C Ca O3	96-900-7688	0.528
Calcite	C Ca O3	96-900-9669	0.527
Scandium magnesium gallide	Ga2 Mg Sc2	96-201-6996	0.525
Magnesium calcite	Ca0.94 Mg0.06 O3	96-721-4218	0.524
Calcite	Ca0.936 Mg0.064 O3	96-900-1298	0.522
Dicalcium magnesium disilicate	Ca2 Mg 07 Si2	96-101-0216	0.519
Calcite	C Ca O3	96-901-4217	0.510
Calcite	C Ca O3	96-901-5074	0.509
Calcite	C Ca O3	96-901-5461	0.507
Calcium trimagnesium carbonate (Huntite)	C4 Ca Mg3 O12	96-100-0053	0.507
Calcium carbonate (Calcite)	Ca C O3	96-101-0963	0.507
Magnesium	Mg	96-901-3061	0.507
Calcite	C Ca O3	96-901-6021	0.503
Calcite	C Ca O3	96-901-4525	0.503
Calcite	C Ca O3	96-901-5762	0.503
Magnesium	Mg	96-901-3062	0.502
Calcite	C Ca O3	96-901-4345	0.502
Calcite	C Ca O3	96-900-0968	0.501
Calcite	C Ca O3	96-901-5836	0.501
Magnesium	Mg	96-901-3063	0.501
Calcite	C Ca O3	96-901-6706	0.501
Calcite	C Ca O3	96-901-4892	0.500
Calcite	C Ca O3	96-901-6465	0.499
Calcite	C Ca O3	96-901-5482	0.499
Calcite	C Ca O3	96-901-5067	0.498
Trimagnesium Calcium Carbonate (Huntite)	C4 Ca Mg3 O12	96-231-0129	0.498
Calcite	C Ca O3	96-901-4612	0.498
Calcite	Ca0.9 Mg0.1 O3	96-900-0575	0.497
Calcite	C Ca O3	96-901-4745	0.496
Dilead magnesium tungstate - II	Mg O6 Pb2 W	96-100-1662	0.496
Calcite	C Ca O3	96-901-4773	0.496
Magnesium Calcium Carbonate (.1/.9/1) (Calcite, magnesian) <sup>c</sup>	Ca0.9 Mg0.1 O3	96-721-4219	0.495
Calcite	C Ca O3	96-901-5692	0.495
Calcite	C Ca O3	96-901-6201	0.495
Calcite	C Ca O3	96-901-4416	0.494
Calcite	C Ca O3	96-901-6180	0.494
Calcite	C Ca O3	96-901-4878	0.493
Calcite	C Ca O3	96-900-0968	0.493
Calcium carbonate (Calcite)	Ca C O3	96-101-0929	0.493
Calcite	C Ca O3	96-900-0970	0.491
Dilead magnesium tungstate - II	Mg O6 Pb2 W	96-100-1663	0.490
Dolomite	C2 Ca Mg O6	96-900-1418	0.489
Calcite	C Ca O3	96-900-0971	0.489

**and 150 others...**

## Search-Match

<b>Settings</b>				
Reference database used	COD-Ihng	REV140301	2015.07.06	
Automatic zero point adaptation	Yes			
Minimum figure-of-merit (FoM)	0.60			
2theta window for peak corr.	0.28 deg.			
Minimum rel. int. for peak corr.	1			
Parameter influence 2theta	0.50			
Parameter influence intensities	0.50			

#### **Criteria for entries added by user**

## Reference:

## Peak List

No.	$2t$		$I/I_0$	$FWHM$	$Matched$
1			82.44	0.7760	A
2			422.78	0.3534	A
3	31.56	2.8349	47.32	1.9692	A
4	36.16	2.4841	73.32	0.4263	A
5	39.60	2.2759	94.88	0.4345	A
6	43.38	2.0860	107.83	0.5230	A
7	47.56	1.9119	158.31	0.8000	A
8	48.66	1.8712	88.43	0.4255	A
9	57.70	1.5977	69.56	0.8354	A
10	59.96	1.5428	36.67	0.9200	
11	60.98	1.5194	57.51	0.8781	A
12	62.46	1.4869	22.04	0.8619	
13	63.48	1.4655	12.76	0.6646	A
14	65.08	1.4333	54.07	0.9733	A

Rietveld Refinement using FullProf

Calculation was not run or did not converge.

## Integrated Profile Areas

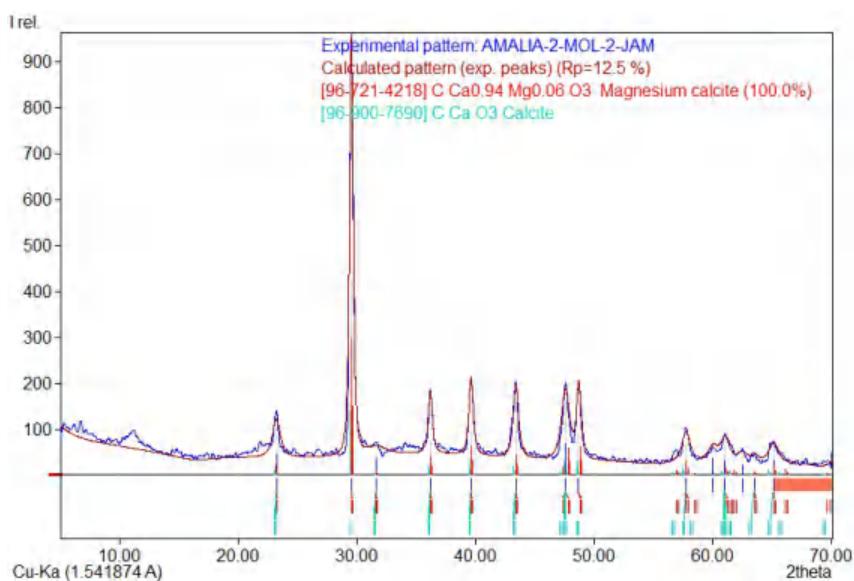
Based on calculated profile

<b>Profile area</b>	<b>Counts</b>	<b>Amount</b>
Overall diffraction profile	114206	100.00%
Background radiation	71802	62.87%
Diffraction peaks	42404	37.13%
Peak area belonging to selected phases	28265	24.75%
<i>Peak area of phase A (Magnesium calcite)</i>	28265	24.75%
Unidentified peak area	17692	15.49%

## Peak Residuals

<b>Peak data</b>	<b>Counts</b>	<b>Amount</b>
Overall peak intensity	725	100.00%
Peak intensity belonging to selected phases	524	72.33%
Unidentified peak intensity	201	27.67%

## Diffracted Pattern Graphics



Match! Copyright © 2003-2015 CRYSTAL IMPACT, Bonn, Germany

# Match! Phase Analysis Report

Sample: AMALIA-3-MOL-2-JAM



File name	AMALIA-3-MOL-2-JAM.txt
File path	D:/OPPI/Semester 7/referensi/Ta/laporan/referensi/AMALIA-3-MOL-2-JAM
Data colle	Jun 8, 2024 22:22:28
Data rang	5.000° - 70.000°
Number o	3251
Step size	0.020
Rietveld refinement converged	No
Alpha2 subtracted	No
Background subtr.	No
Data smoothed	Yes
Radiation	X-rays
Wavelength	1.541874 Å

## Matched Phases

Index	Amount (%)	Name	Formula sum
A	100.0	Magnesium chloride(VII) hexahydrate	C12 H12 Mg O14
	27.8	Unidentified peak area	

### A: Magnesium chloride(VII) hexahydrate (100.0 %)

Formula sum	C12 H12 Mg O14
Entry number	96-101-0116
Figure-of-Merit (FoM)	0.385626
Total number of peaks	656
Peaks in range	272
Peaks matched	36
Intensity scale factor	0.16
Space group	P m n 21 S
orthorhombic	
Unit cell	a=7.7600 Å b=13.4600 Å c=5.2600 Å
I/c	1.22
Calc. density	1.929 g/cm³
Reference	"Crystal Structures of Hydrated Compounds. II. Structure Type Mg (Cl O-4)-~(H-2-O)-6~". Zeitschrift fuer Kristallographie, Kristallgeometrie, Kristallphysik, Kristallchemie (-144,1977) 91, 480-493 (1935)

## Candidates

Name	Formula	Entry No.	FoM
Magnesium oxide (Periclase)	Mg O	96-101-1194	0.6504
Magnesium oxide (Periclase)	Mg O	96-101-1119	0.6493
Magnesium oxide (Periclase)	Mg O	96-101-1118	0.6452
Magnesium oxide (Periclase)	Mg O	96-101-1117	0.6448
Magnesium oxide (Periclase)	Mg O	96-500-0226	0.6448
Magnesium oxide (Periclase)	Mg O	96-100-0054	0.6447
Magnesium oxide (Periclase)	Mg O	96-101-1174	0.6443
Magnesium oxide (Periclase)	Mg O	96-101-1327	0.6443
Potassium iron(II)/magnesium iron(III) bis(orthophosphate)	Fe1.91 K Mg0.09 O8 P2	96-223-5287	0.6416
Magnesium chromate - S-beta	Cr Mg O4	96-100-8087	0.6267
Dicalcium sodium dimagnesium trivanadium tridecaoxide	Ca2 Mg2 Na O12 V3	96-201-3705	0.6023
Magnesium Molybdate	Mg Mo O4	96-230-0505	0.5971
Decacalcium pentamagnesium tricopper dodecavanadium oxide	Ca10 Cu3 Mg5 O48 V12	96-200-2865	0.5757
Pentacalcium trimagnesium zinc hexakis(vanadate)	Ca5 Mg3 O24 V6 Zn	96-200-2589	0.5668
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0073	0.5636
Dicalcium sodium dimagnesium arsenate (Berzeliiite)	As3 Ca2 Mg2 Na O12	96-101-0912	0.5552
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0075	0.5533
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0080	0.5522
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0079	0.5520
Magnesium iron oxide (1.6/1.6/4)	Fe1.6 Mg1.55 O4	96-100-6064	0.5431
Icosa Cerium nonadeca Magnesium heptaconta Zinc	Ce20 Mg19 Zn81	96-201-6697	0.5404
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0078	0.5333
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-210-4834	0.5330
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0074	0.5311
Dimagnesium titanium oxide (Qandilite)	Mg2 O4 Ti	96-101-1277	0.5293
Magnesium thiosulfate hexahydrate	H12 Mg O9 S2	96-110-0071	0.5227
Kalium trimagnesium palladium tris(arsenate)(V)	As3 K Mg3 O12 Pd	96-200-2680	0.5194
Trimagnesium bis(sulfate(VII)) dihydroxide	H2 Mg3 O10 S2	96-110-0100	0.5168
Silver trimagnesium phosphate bis(hydrogenphosphate)	Ag H2 Mg3 O12 P3	96-222-8708	0.5104
Dialuminium pentamagnesium trisilicate octahydroxide (Clinochlore 2M)	A12 H8 Mg5 O18 Si3	96-101-1016	0.5078
Neodymium magnesium aluminium oxide (1/1/19/19)	A11 Mg Nd O19	96-200-2338	0.5055
sodium magnesium decavanadate hydrate	H40 Mg2 Na2 O48 V10	96-201-3924	0.5053
Barium magnesium iron uranium oxide (2/0.5/0.7/0.8/6)	Ba2 Fe0.667 Mg0.5 O6 U0.833	96-100-1135	0.5022
magnesium hydroxide sulfate	H2 Mg3 O10 S2	96-210-1856	0.5002
Sodium magnesium copper trivanadate (1/1.6/4/1)	Cu0.36 Mg1.64 Na O10 V3	96-200-2664	0.5001
Magnesium diiron(III) oxide (Magnesiostefrite)	Fe2 Mg O4	96-101-1242	0.4938
tricerium hemimagnesium germanium heptasilicide	Ce3 Ge Mg0.5 S7	96-221-6156	0.4918
Calcium Magnesium Carbonate (Dolomite)	C2 Ca Mg O6	96-151-7798	0.4902
Cadmium copper magnesium lead vanadium oxide (1/1.5/1.5/0.5/3/12)	Cd Cu1.5 Mg1.5 O12 Pb0.5 V3	96-200-2720	0.4900
magnesium hydrogen phosphate	H4 Mg7 O24 P6	96-223-0980	0.4876
Magnesium Sodium Pyrovanadate	Mg2 Na6 O15 V4	96-201-4024	0.4870
Tetracalcium magnesium tetrairon(III) hexakis-phosphate tetrahydroxide dodecahydrate	Al1.07 Ca3.94 Fe2.93 H28 Mg1.01 O40 P6 Sr0.0696-223-9375	96-4855	
Potassium Magnesium imide-amide	D3 K Mg N2	96-151-3214	0.4852
Sodium dimagnesium trivanadate	Mg2 Na O10 V3	96-221-7730	0.4843
Potassium iron dimagnesium phyo-alumotrisilicate dihydroxide (Biotite)	Al Fe H2 K Mg2 O12 Si3	96-100-0039	0.4828
magnesium aluminum iron chromite	Al0.41 Cr1.42 Fe0.65 Mg0.4 O4	96-201-4617	0.4827
Zinc magnesium tantalum oxide (7.3/2/6)	Mg0.3 O6 Ta2 Zn0.7	96-200-2432	0.4820
Caesium magnesium orthophosphate	Cs Mg O4 P	96-222-2806	0.4815
calcium-magnesium silicide	Ca Mg Si	96-400-0955	0.4814
magnesium sulfate	Mg O4 S	96-230-0130	0.4813
heptasodium tridecamagnesium neodymium dodecakisphosphate	Mg13 Na7 Nd O48 P12	96-223-5282	0.4810
Trimagnesium tetralithium(III) hexakis(phosphate)	Mg3 O24 P6 Ti4	96-100-1441	0.4807
and 150 others...			

## Search-Match

Settings	
Reference database used	COD-Inorg REV140301 2015.07.06
Automatic zero point adaptation	Yes
Minimum figure-of-merit (FoM)	0.60
2theta window for peak corr.	0.28 deg.
Minimum rel. int. for peak corr.	1
Parameter/influence theta	0.50
Parameter/influence intensities	0.50
Parameter multiple/single phase(s)	0.50

## Criteria for entries added by user

### Reference:

Entry number:	96-100-0055; 96-900-1587; 96-900-1588; 96-900-1589; 96-900-1590; 96-900-2349; 96-900-2351; 96-900-2352; 96-900-2353; 96-900-2354; 96-900-2355; 96-900-3876; 96-900-3877; 96-900-3878; 96-900-6331; 96-900-6332; 96-900-6333; 96-900-6334; 96-100-0028; 96-100-0036; 96-100-0037; 96-100-0039; 96-100-0048; 96-100-0053; 96-100-0054; 96-100-0055; 96-100-1032; 96-100-1130; 96-100-1133; 96-100-1135; 96-100-1137; 96-100-1139; 96-100-1140; 96-100-1141; 96-100-1142; 96-100-1143; 96-100-1144; 96-100-1145; 96-100-1163; 96-100-1162; 96-100-1163; 96-100-1172; 96-100-1775; 96-100-1819; 96-100-1824; 96-100-5056; 96-100-5057; 96-100-5064; 96-100-7081; 96-100-7258; 96-100-7263; 96-100-8084; 96-100-8087; 96-100-8171; 96-100-8172; 96-100-8173; 96-100-8276; 96-100-8288; 96-100-8290; 96-100-8652; 96-100-8811; 96-100-8812; 96-100-8814; 96-100-8915; 96-100-8949; 96-100-9321; 96-100-9322; 96-100-9323; 96-100-9324; 96-100-9325; 96-100-9326; 96-100-9327; 96-100-9328; 96-100-9329; 96-100-9330; 96-100-9331; 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## Peak List

No.	2θ	I/I₀	FWHM	Matched	
1		39.86	0.1600	A	
2		495.97	0.5600	A	
3		171.92	0.1600		
4		119.00	0.1600		
5		274.28	1.2400		
6		42.75	0.1600	A	
7	35.64	2.5192	237.18	0.9200	A
8	37.50	2.3984	97.79	0.1600	A
9	42.96	2.1054	62.60	0.6800	
10	46.94	1.9357	45.11	0.5537	A
11	47.68	1.9074	15.94	1.7600	A
12	50.84	1.7960	111.47	0.8422	A
13	53.92	1.7005	134.31	1.1357	A
14	62.44	1.4874	286.70	1.8298	A
15	64.20	1.4508	76.18	0.4800	A
16	67.44	1.3887	91.44	0.1600	

## Rietveld Refinement using FullProf

Calculation was not run or did not converge.

## Integrated Profile Areas

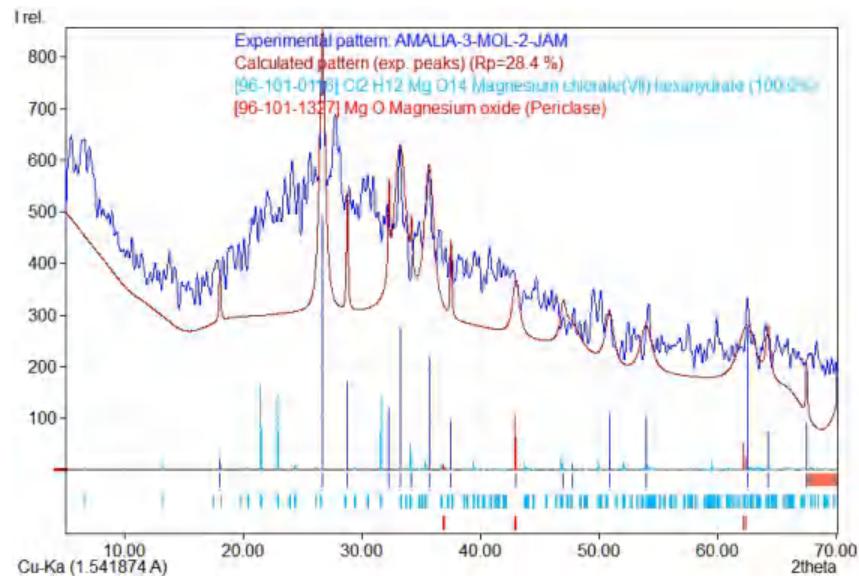
### Based on calculated profile

Profile area	Counts	Amount
Overall diffraction profile	130400	100.00%
Background radiation	89085	68.32%
Diffraction peaks	41315	31.68%
Peak area belonging to selected phases	7331	5.62%
Peak area of phase A (Magnesium chlorate(VII) hexahydrate)	7331	5.62%
Unidentified peak area	36199	27.76%

## Peak Residuals

Peak data	Counts	Amount
Overall peak intensity	244	100.00%
Peak intensity belonging to selected phases	18	7.32%
Unidentified peak intensity	226	92.68%

## Diffraction Pattern Graphics



## LAMPIRAN 5 KARTU KONSULTASI



Optimized using  
trial version  
[www.balesio.com](http://www.balesio.com)

**Lampiran B 10**  
**Kartu Konsultasi Tugas Akhir**

**JUDUL:** STUDI EKSTRAK MANGANEUM DARI DOLOMIT DENGAN MENGGUNAKAN PELARUT ASAM FLUORIDA

(Konsultasi minimal 8 kali)

TANGGAL	MATERI KONSULTASI	PARAF DOSEN
27/06/2024	<ul style="list-style-type: none"> <li>- Abstrak</li> <li>- Tujuan Penelitian</li> <li>- Hasil analisis XRD</li> </ul>	
30/06/2024	<ul style="list-style-type: none"> <li>- Hasil mikroskopis</li> <li>- Penulisan rumus kimia</li> <li>- Format penulisan</li> </ul>	
5/07/2024	<ul style="list-style-type: none"> <li>- Grafik analisis XRD</li> <li>- Penulisan bahasa Inggris</li> <li>- Grafik gabungan faktineru</li> </ul>	
8/07/2024	<ul style="list-style-type: none"> <li>- Analisis XRD</li> <li>- Daftar Pustaka</li> </ul>	
9/07/2024	<ul style="list-style-type: none"> <li>- Grafik lingkup Ekstraksi</li> <li>- Kesimpulan</li> </ul>	
10/07/2024	<ul style="list-style-type: none"> <li>- Format penulisan</li> <li>- Format artikel ilmiah</li> </ul>	



TANGGAL	MATERI KONSULTASI	PARAF DOSEN
15/07/2024	<ul style="list-style-type: none"> <li>- Artikel Ilmiah</li> <li>- Poster penelitian</li> </ul>	M
18/07/2024	<ul style="list-style-type: none"> <li>- Artikel Ilmiah</li> <li>- Poster penelitian</li> </ul>	M
30/07/2024	<ul style="list-style-type: none"> <li>- Penulisan tesis dan tesis</li> <li>- Tambahan gambar sampel</li> <li>- Tambahan penjelasan singkatan</li> </ul>	M
9/08/2024	<ul style="list-style-type: none"> <li>- Gambar sampel</li> <li>- Singkatan</li> </ul>	M Dede

