

DAFTAR PUSTAKA

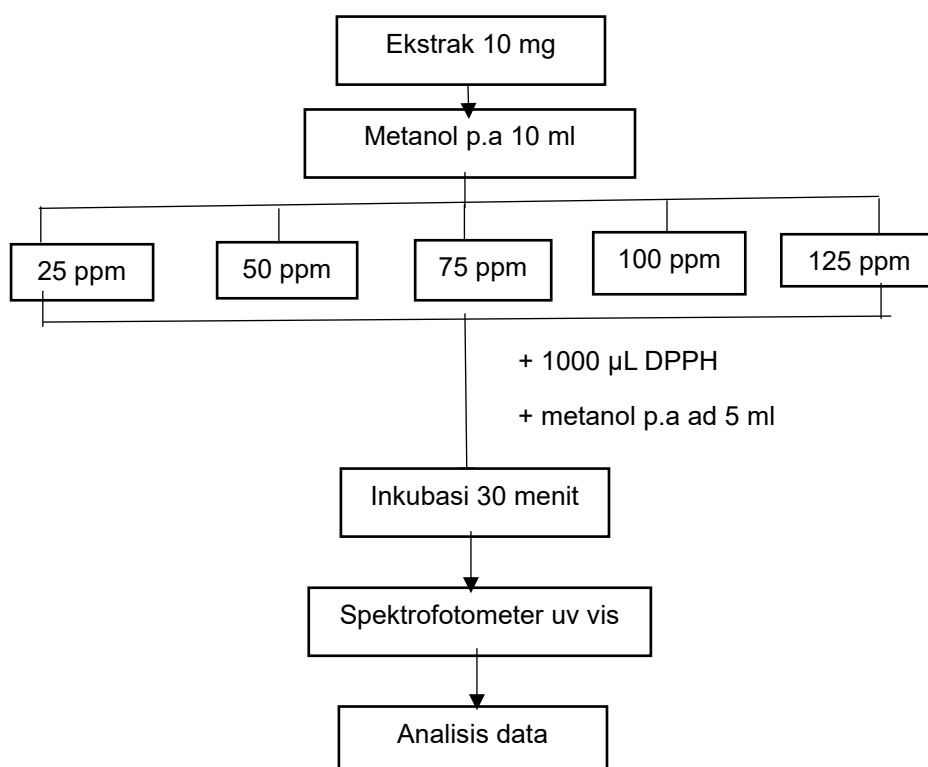
- Arung, E. T., Kusuma, I. W., Kim, Y. U., Shimizu, K., & Kondo, R. (2012). Antioxidative compounds from leaves of Tahongai (*Kleinhovia hospita*). *Journal of wood science*, 58(1), 77-80.
- Akhlaghi, M., & Bandy, B. (2009). Mechanisms of flavonoid protection against myocardial ischemia–
- Domitrović, R., & Potočnjak, I. (2016). A comprehensive overview of hepatoprotective natural compounds: mechanism of action and clinical perspectives. *Archives of toxicology*, 90(1), 39-79.
- DepKes, R. 2007. Pharmaceutical care untuk penyakit hati. Departemen Kesehatan.47.
- Dolores Lopes-Terrada et al. 2009. HepG2 is a hepatoblastoma-derived cell line. *Hum Pathol* 40:1512-1515.
- Doyle A, Griffiths JB. 2000. Cell and Tissue Culture for Medical Research. New
- Enos. Mengilmiahkan Obat Alamiah. 2012: diambil dari <http://www.ristek.go.id>. diakses 1 Maret, 2015.
- Gan, L. S., Ren, G., Mo, J. X., Zhang, X. Y., Yao, W., & Zhou, C. X. (2009). Cycloartane triterpenoids from
- Haryoto, H., & Frista, A. (2019). Aktivitas antioksidan ekstrak etanol, fraksi polar, semipolar dan non polar dari daun mangrove kacang (*Rhizophora apiculata*) dengan Metode DPPH dan FRAP. *Jurnal Sains dan Kesehatan*, 2(2), 131-138.
- Hanifa, D. D., & Hendriani, R. (2016). Tanaman herbal yang memiliki aktivitas hepatoprotektor. *Farmaka*, 14(4), 43-51.
- Kim, O.S., 2005, Radical Scavenging Capacity and Antioxidant Activity of The Vitamin Fraction In rice bran. *J Food Sci.* (3): 208- 213
- Maesaroh, K., Kurnia, D., & Al Anshori, J. (2018). Perbandingan metode uji aktivitas antioksidan DPPH, FRAP dan FIC terhadap asam askorbat, asam galat dan kuersetin. *Chimica et natura acta*, 6(2), 93-100.
- Morilla, L.J.G., Nuñez, O.M. & Uy, M.M. 2015, Brine shrimp lethality test of *Kleinhovia hospita* stem and bark from agusan del sur, *ELBA Bioflux*, 7(1): 61 – 66
- Muthia, R., Saputri, R., & Verawati, S. A. (2019). Uji Aktivitas Antioksidan Ekstrak Etanol Kulit Buah Mundar (*Garcinia forbesii* King.) Menggunakan Metode DPPH (2, 2-Diphenyl-1-Picrylhydrazil). *Jurnal Pharmascience*, 6(1), 74-82.
- Mutmainnah, P. A., Hakim, A., & Savalas, L. R. T. (2017). Identifikasi Senyawa Turunan Hasil Fraksinasi Kayu Akar *Artocarpus odoratissimus*. *Jurnal Penelitian Pendidikan IPA*, 3(2).
- Nuralih et al. 2018. Pengaruh ekstrak etanol daun murbei (*Morus alba* L.) dengan glibenklamid terhadap ekspresi gen CYP3A4 pada kultur sel HepG2. *Jurnal Farmasi Indonesia* 15:29-36
- Paramita, S. (2016). Tahongai (*Kleinhovia hospita* L.): Review Sebuah Tumbuhan Obat dari Kalimantan Timur. *Jurnal Tumbuhan Obat Indonesia*, 9(1), 29–36.
- Rahim, A., Saito, Y., Miyake, K., Goto, M., Chen, C. H., Alam, G., ... & Nakagawa-Goto, K. (2018). Kleinhospitine E and cycloartane triterpenoids from *Kleinhovia hospita*. *Journal of natural products*, 81(7), 1619-1627.
- Rahman, A. 2014. *Studies In Natural Products Chemistry*. Elsevier. United Kingdom.
- Saputra, T. R., Ngatin, A., & Sarungu, Y. T. (2018). Penggunaan metode ekstraksi maserasi dan partisi pada tumbuhan cocor bebek (*kalanchoe pinnata*) dengan kepolaran berbeda. *Fullerene Journal of Chemistry*, 3(1), 5-8.

- Senthilraja, P., & Kathiresan, K. (2015). In vitro cytotoxicity MTT assay in Vero, HepG2 and MCF-7 cell lines study of Marine Yeast. *Journal of applied pharmaceutical science*, 5(3), 080-084.
- Setiawan, F., Yunita, O., & Kurniawan, A. (2018). Uji aktivitas antioksidan ekstrak etanol kayu secang (*Caesalpinia sappan*) menggunakan metode DPPH, ABTS, dan FRAP. *Media Pharmaceutica Indonesiana*, 2(2), 82-89.
- Vakele, Y., Odun-Ayo, F., & Reddy, L. (2022). In vitro antioxidant and cytotoxicity activities of selected indigenous South African medicinal plants. *African Health Sciences*, 22(1), 395-403.
- Vijayalakshmi, M., and Ruckmani, K. 2016. Ferric Reducing Anti-oxidant Power assay in Plant Extract. ISI Impact Factor. (11). pp.570-572
- Wang, C. C., Chu, C. Y., Chu, K.O., Choy, K. W., Khaw, K.S., Rogers, M. S., Pang, C. P. 2004. Trolox-Equivalent Antioxidant Capacity Assay Versus Oxygen Radical Absorbance Capacity Assay in Plasma, *Clinical Chemistry* Vol. 50 (5): 952-954
- Wahyono, A., Elly K., Kasutjaningati, Kang, H.P., Won, W.K. 2018. Optimasi Proses Pembuatan Tepung Labu Kuning Menggunakan Response Surface Methodology Untuk Meningkatkan Aktivitas Antioksidannya. *Jurnal Teknologi dan Industri Pangan*. Vol. 29(1): 29-38. ISSN: 1979-778
- Winarsi, H., 2007, *Antioksidan Alami dan Radikal Bebas*, Kanisius, Yogyakarta.
- Zhou, C. X., Zou, L., Gan, L. S., & Cao, Y. L. (2013). Kleinhospitines a–d, new cycloartane triterpenoid alkaloids from *kleinhovia hospita*. *Organic letters*, 15(11), 2734-2737.

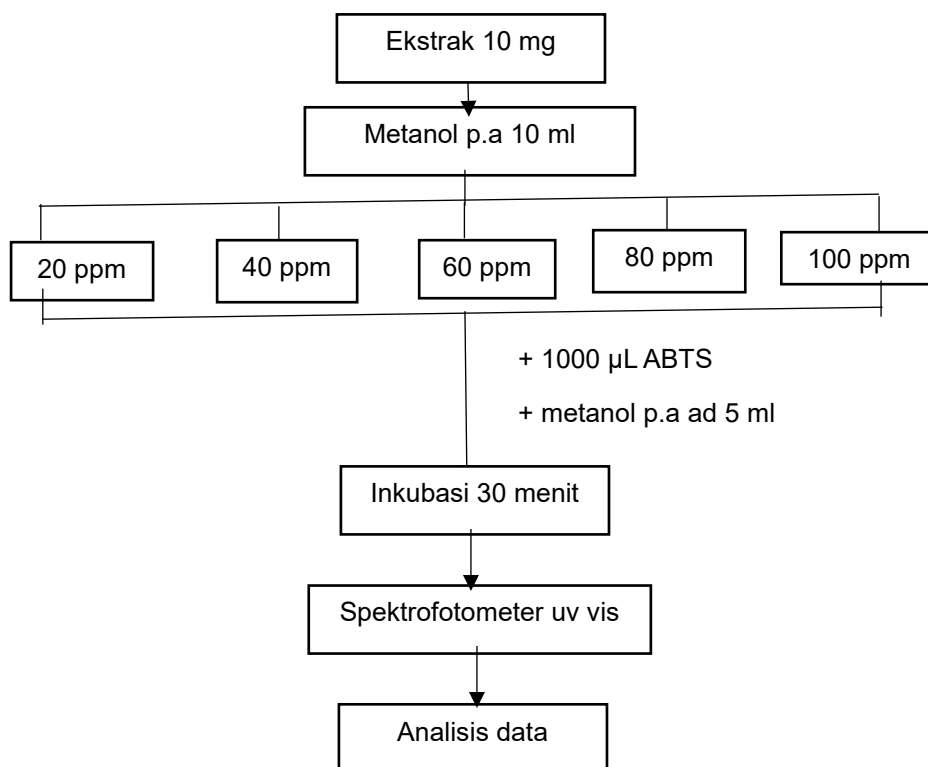
Lampiran

Lampiran 1. Skema kerja penelitian

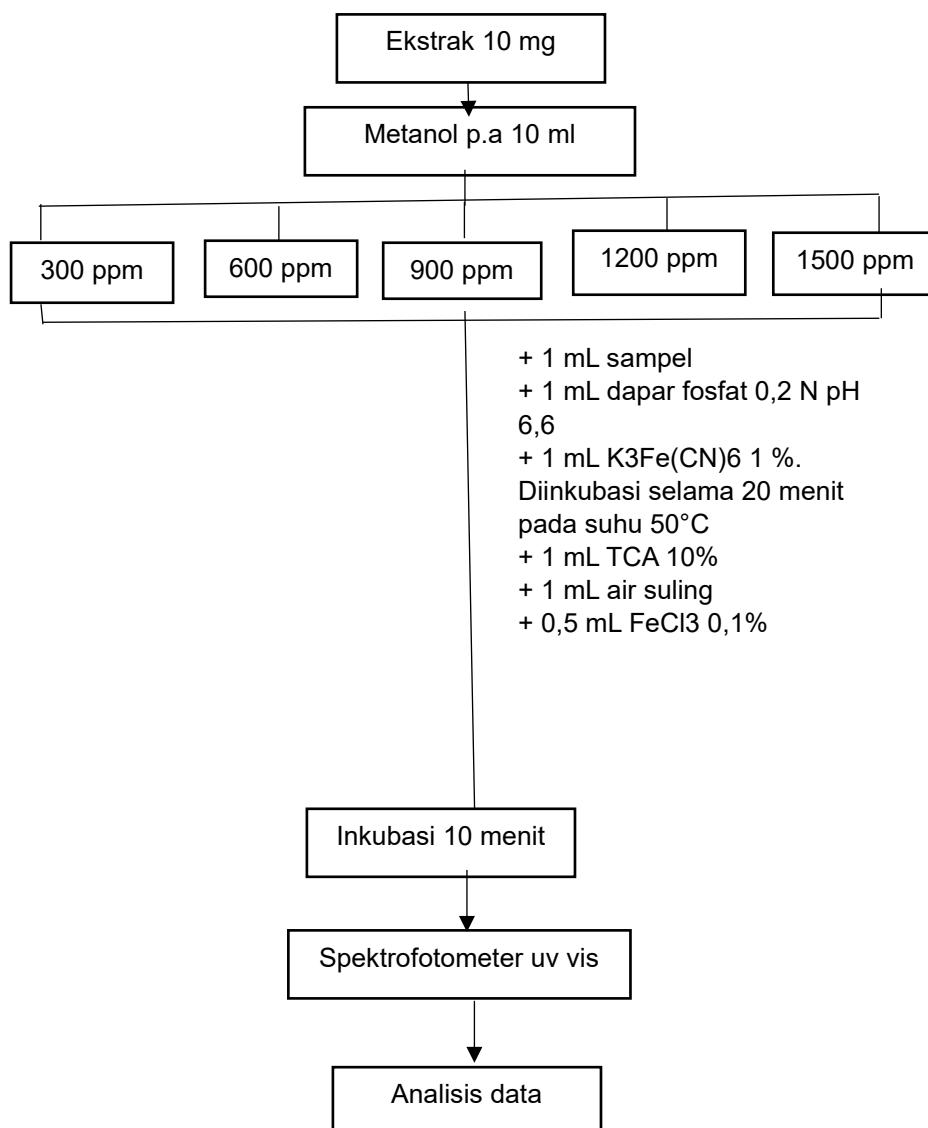
1.1 skema kerja uji antioksidan dengan metode DPPH



1.2 skema kerja uji antioksidan dengan metode ABTS



1.3 skema kerja uji antioksidan dengan metode FRAP



Lampiran 2. Perhitungan

2.1 perhitungan DPPH

Tabel 5. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* Asam Askorbat metode DPPH

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko DPPH	Blanko	0.902	0.910		
			0.915			
			0.915			
2	Asam askorbat	1	0.901	0.896	1.5	4.727
			0.894			
			0.894			
		2	0.786	0.779	14	
			0.764			
			0.787			
		3	0.652	0.643	29	
			0.631			
			0.646			
		4	0.561	0.569	37	
0.586						
0.560						
5	0.415	0.410	54			
	0.408					
	0.408					

Persentase inhibisi radikal bebas asam askorbat

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

$$\text{Konsentrasi 1 ppm} = \frac{0.910 - 0.896}{0.910} \times 100\% = 1.5 \%$$

$$\text{Konsentrasi 2 ppm} = \frac{0.910 - 0.779}{0.910} \times 100\% = 14 \%$$

$$\text{Konsentrasi 3 ppm} = \frac{0.935 - 0.643}{0.935} \times 100\% = 29 \%$$

$$\text{Konsentrasi 4 ppm} = \frac{0.935 - 0.569}{0.935} \times 100\% = 37 \%$$

$$\text{Konsentrasi 5 ppm} = \frac{0.935 - 0.410}{0.935} \times 100\% = 54 \%$$

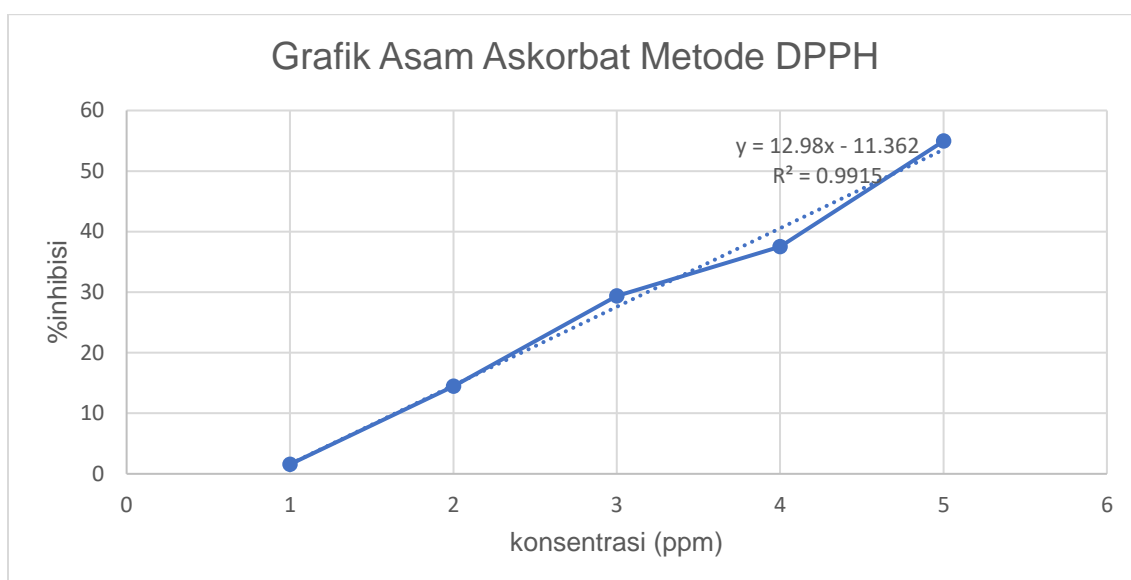
Perhitungan IC50

$$Y = ax + b$$

$$Y = 12.98x + (-11.32)$$

$$50 = 12.98x + 11.32$$

$$X = 4.727$$



Gambar 3. Grafik asam askorbat metode DPPH

Tabel 6. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode DPPH Fraksi D

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko DPPH	Blanko	1.087	1.083		
			1.087			
		25	0.844 0.851 0.852	0.849	21.60	
		50	0.678 0.682 0.672	0.677	37.48	
2	Fraksi D	75	0.526	0.532	50.87	80.90
			0.551 0.521			
		100	0.433 0.446 0.447	0.442	59.18	
		125	0.344 0.353 0.358	0.351	67.59	

Persentase inhibisi radikal bebas fraksi D

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

$$\text{Konsentrasi 25 ppm} = \frac{1.083 - 0.849}{1.083} \times 100\% = 21.60 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{1.083 - 0.677}{1.083} \times 100\% = 37.48 \%$$

$$\text{Konsentrasi 75 ppm} = \frac{1.083 - 0.532}{1.083} \times 100\% = 50.87 \%$$

$$\text{Konsentrasi 100 ppm} = \frac{1.083 - 0.442}{1.083} \times 100\% = 59.18 \%$$

$$\text{Konsentrasi } 125 \text{ ppm} = \frac{1.083-0.351}{1.083} \times 100\% = 67.59\%$$

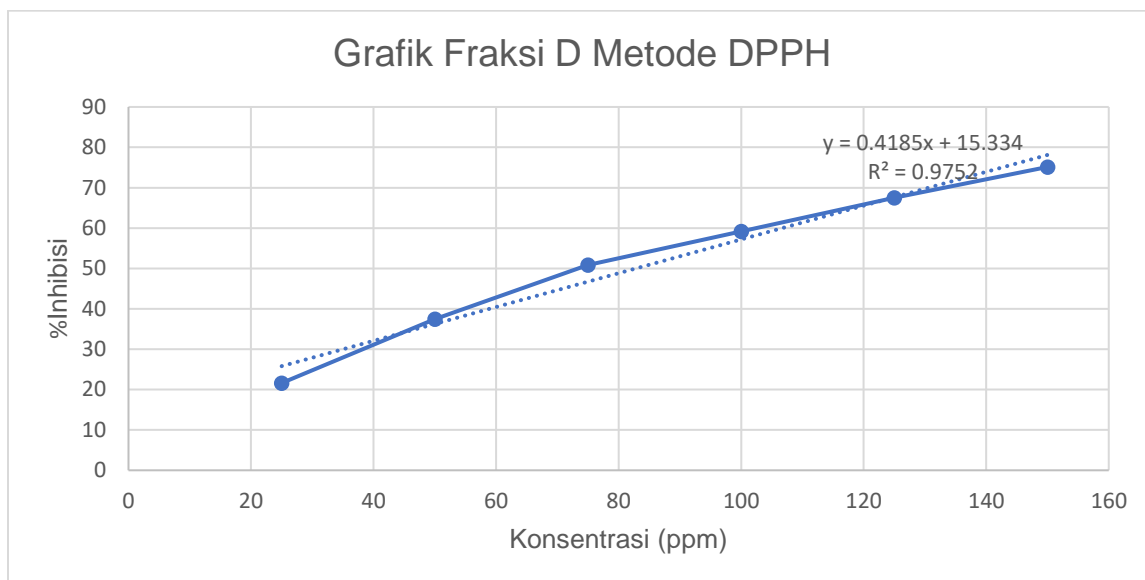
Perhitungan IC50

$$Y = ax+b$$

$$Y = 0.4185x+15.334$$

$$50 = 0.4185x+15.334$$

$$X = 82.83$$



Gambar 4. Grafik fraksi D metode DPPH

Table 7. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode DPPH Fraksi N

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko DPPH	Blanko	1.112	1.098	18.21	
			1.092			
			1.091			
2	Fraksi N	25	0.912	0.898	27.50	110.33
			0.924			
			0.860			
		50	0.810	0.796	36.06	
			0.809			
			0.769			
75	0.721	0.702	46.72			
	0.704					
	0.681					
100	0.594	0.753	55.37			
	0.599					
	0.564					
125	0.494	0.718				
	0.481					
	0.495					

Persentase inhibisi radikal bebas fraksi N

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

$$\text{Konsentrasi 25 ppm} = \frac{1.098 - 0.898}{1.098} \times 100\% = 18.21 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{1.098 - 0.796}{1.098} \times 100\% = 27.50 \%$$

$$\text{Konsentrasi 75 ppm} = \frac{1.098 - 0.702}{1.098} \times 100\% = 36.06 \%$$

$$\text{Konsentrasi 100 ppm} = \frac{1.098 - 0.585}{1.098} \times 100\% = 46.72 \%$$

$$\text{Konsentrasi 125 ppm} = \frac{1.098 - 0.49}{1.098} \times 100\% = 55.37 \%$$

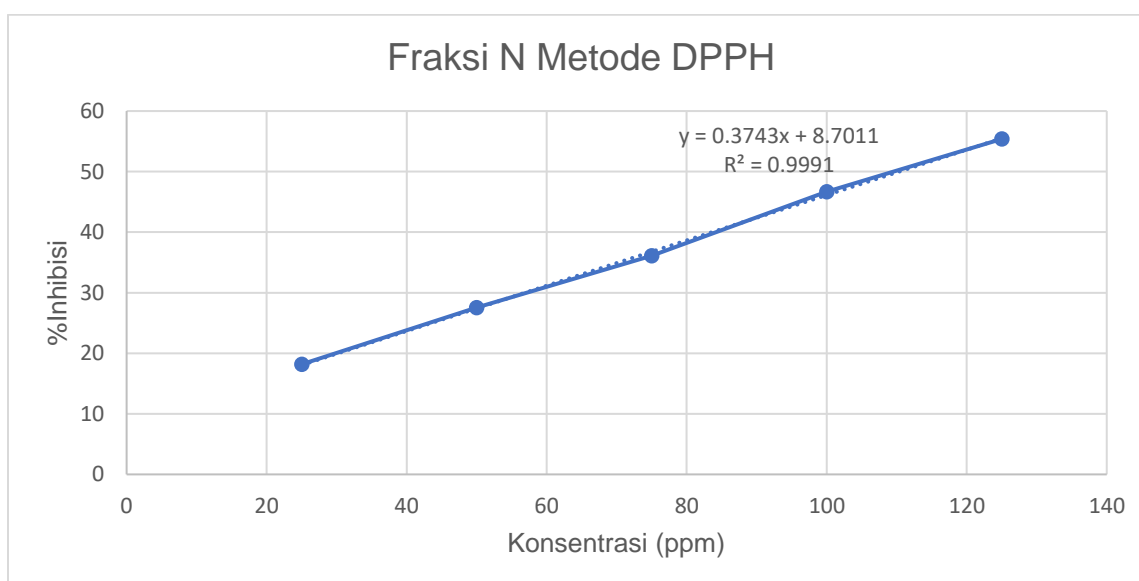
Perhitungan IC50

$$Y = ax + b$$

$$Y = 0.3743x + 8.7011$$

$$50 = 0.3743x + 8.7011$$

$$X = 110.33$$



Gambar 5. Grafik fraksi N metode DPPH

Tabel 8. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode DPPH Fraksi O

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko DPPH	Blanko	1.087 1.087 1.075	1.083		
2	Faksi O	25	0.955 0.976 0.956	0.962	11.17	120.56

	0.819		
50	0.817	0.817	24.56
	0.816		
	0.729		
75	0.720	0.734	32.22
	0.755		
	0.636		
100	0.638	0.627	42.10
	0.608		
	0.556		
125	0.521	0.529	51.15
	0.510		

Persentase inhibisi radikal bebas fraksi O

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

$$\text{Konsentrasi 25 ppm} = \frac{1.083 - 0.962}{1.083} \times 100\% = 11.17 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{1.083 - 0.817}{1.083} \times 100\% = 24.56 \%$$

$$\text{Konsentrasi 75 ppm} = \frac{1.083 - 0.734}{1.083} \times 100\% = 32.22 \%$$

$$\text{Konsentrasi 100 ppm} = \frac{1.083 - 0.627}{1.083} \times 100\% = 42.10 \%$$

$$\text{Konsentrasi 125 ppm} = \frac{1.083 - 0.529}{1.083} \times 100\% = 51.15 \%$$

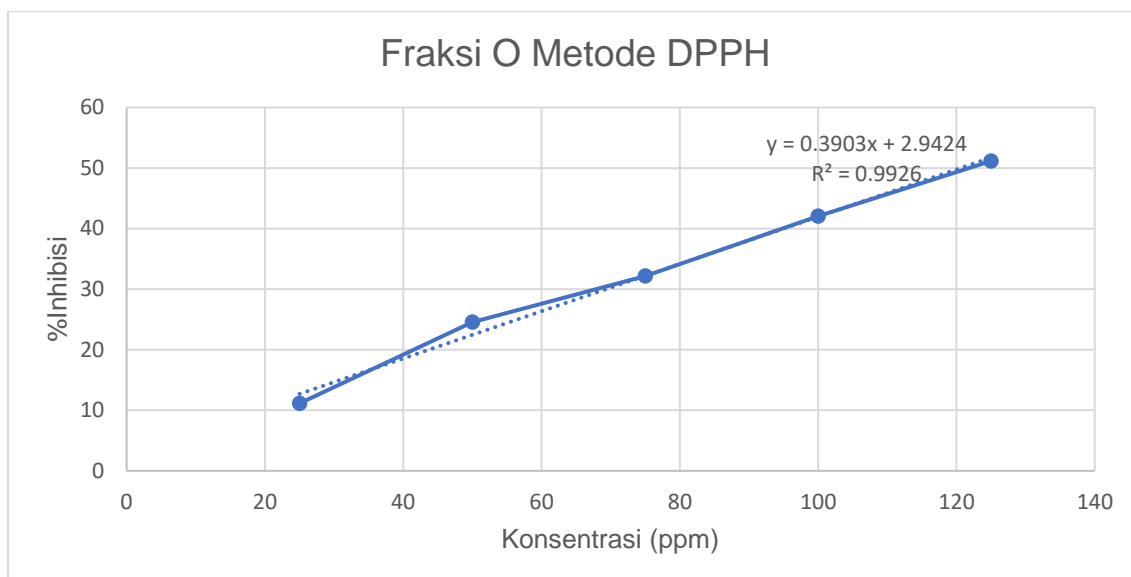
Perhitungan IC50

$$Y = ax + b$$

$$Y = 0.3903x + 2.9424$$

$$50 = 0.3903x + 2.9424$$

$$X = 120.56$$



Gambar 6. Grafik fraksi O metode DPPH

2.2 perhitungan ABTS

Tabel 9. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* Asam Askorbat metode ABTS

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas
1	Blanko ABTS	Blanko	0.846	0.870	
			0.882		
			0.883		
2	Asam askorbat	0.5	0.82	0.809	12.79
			0.815		
			0.793		
		1	0.744	0.734	20.68
			0.73		
			0.729		
2	1.5	0.625	0.64	26.84	
		0.642			
		0.653			
2	2	0.551	0.557	36	
		0.563			
		0.557			
2.5	2.5	0.494	0.493	43.35	
		0.493			
		0.492			

Persentase inhibisi radikal bebas asam askorbat

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

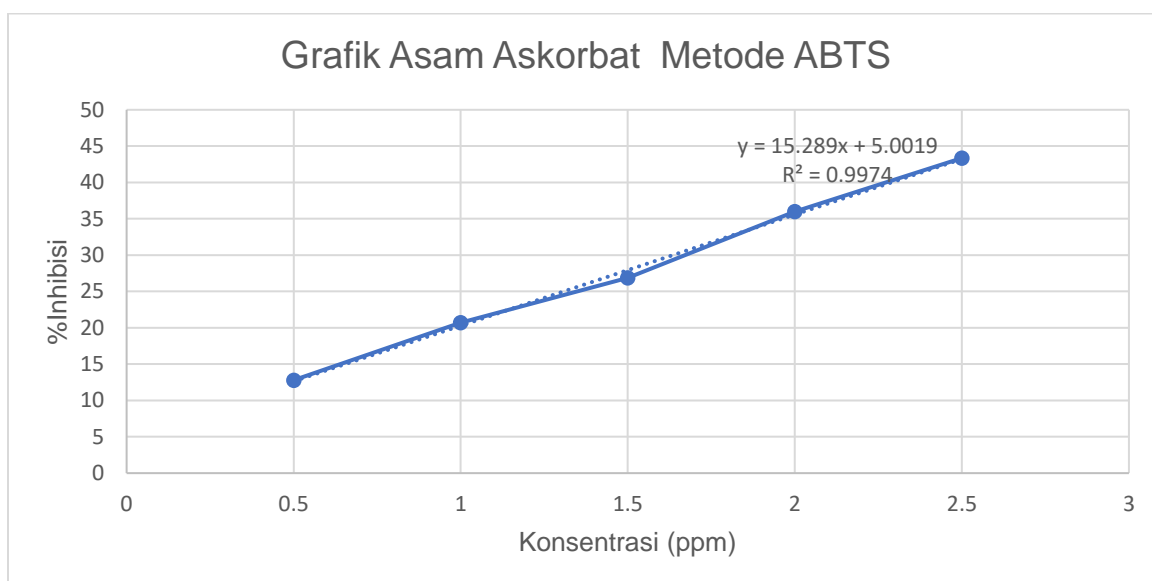
$$\text{Konsentrasi 1 ppm} = \frac{0.870 - 0.809}{0.870} \times 100\% = 12.79 \%$$

$$\text{Konsentrasi 2 ppm} = \frac{0.870 - 0.734}{0.870} \times 100\% = 20.68 \%$$

$$\text{Konsentrasi 3 ppm} = \frac{0.870 - 0.64}{0.870} \times 100\% = 26.84 \%$$

$$\text{Konsentrasi 4 ppm} = \frac{0.870-0.557}{0.870} \times 100\% = 36 \%$$

$$\text{Konsentrasi 5 ppm} = \frac{0.870-0.493}{0.870} \times 100\% = 43.35 \%$$



Gambar 7. Grafik asam askorbat metode ABTS

Tabel 10. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode ABTS Fraksi D

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko ABTS	Blanko	0.950	0.983		
			1.000			
			1.001			
		20 ppm	0.740	0.727	26.04	
			0.734			
			0.708			
		30 ppm	0.542	0.537	45.37	
			0.536			
			0.534			
2	Fraksi D	40 ppm	0.429	0.426	56.66	36.50
			0.424			
			0.425			
		50 ppm	0.346	0.335	65.92	
			0.348			
			0.313			
		60 ppm	0.235	0.22	77.61	
			0.224			
			0.201			

Persentase inhibisi radikal bebas fraksi D

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

$$\text{Konsentrasi 20 ppm} = \frac{0.983-0.727}{0.983} \times 100\% = 26.04 \%$$

$$\text{Konsentrasi 30 ppm} = \frac{0.983-0.537}{0.983} \times 100\% = 45.37 \%$$

$$\text{Konsentrasi 40 ppm} = \frac{0.983-0.426}{0.983} \times 100\% = 56.66 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{0.983-0.335}{0.983} \times 100\% = 65.92 \%$$

$$\text{Konsentrasi 60 ppm} = \frac{0.983-0.22}{0.983} \times 100\% = 77.61\%$$

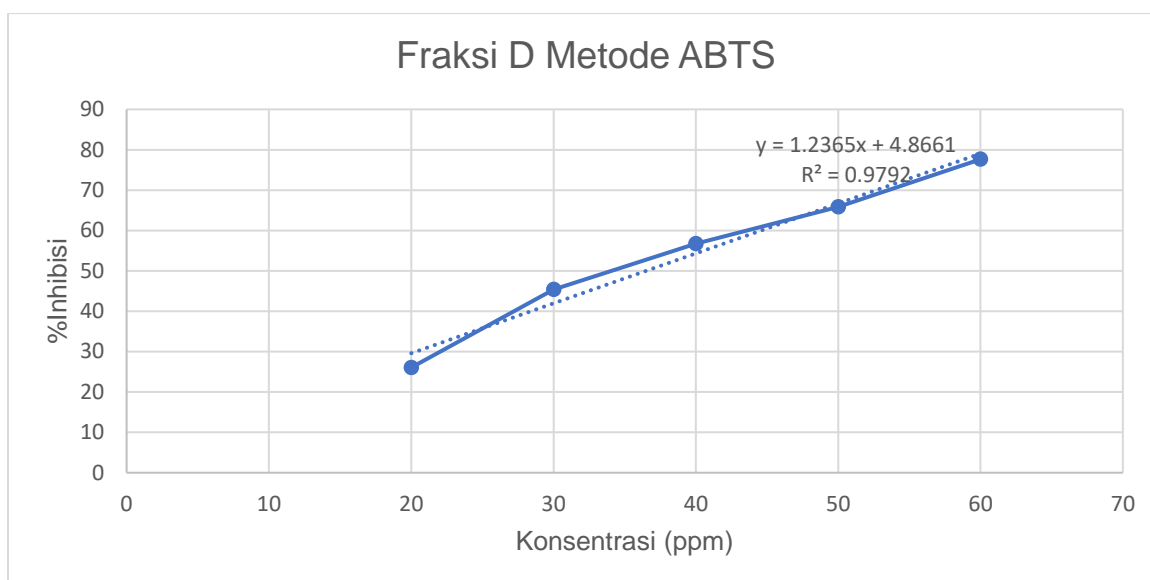
Perhitungan IC50

$$Y = ax+b$$

$$Y = 1.2365x+4.8661$$

$$50 = 36.50133441$$

$$X = 36.50$$



Gambar 8. Grafik fraksi D metode ABTS

Tabel 11. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode ABTS Fraksi N

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas
1	Blanko ABTS	Blanko	0.950	0.983	
			1.000		
			1.001		
2	Fraksi N	20 ppm	0.862	0.878	10.68
			0.881		
			0.892		
2	Fraksi N	30 ppm	0.756	0.759	22.78
			0.764		
			0.757		

40 ppm	0.691 0.690 0.690	0.690	29.80
50 ppm	0.638 0.636 0.636	0.636	35.30
60 ppm	0.568 0.580 0.550	0.566	42.40

Persentase inhibisi radikal bebas fraksi N

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

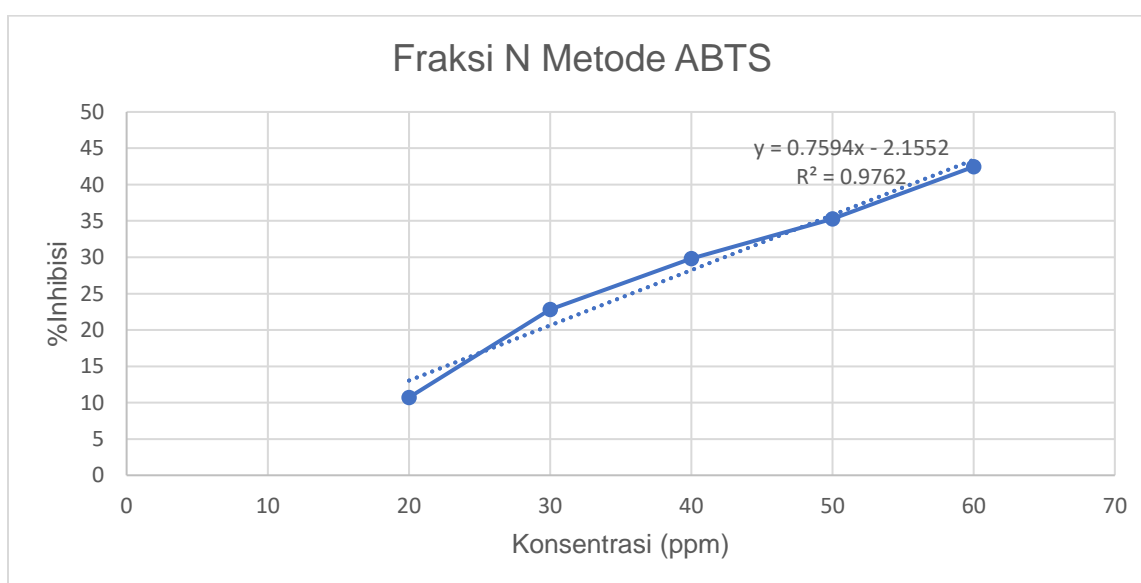
$$\text{Konsentrasi 20 ppm} = \frac{0.983 - 0.878}{0.983} \times 100\% = 10.68 \%$$

$$\text{Konsentrasi 30 ppm} = \frac{0.983 - 0.759}{0.983} \times 100\% = 22.78 \%$$

$$\text{Konsentrasi 40 ppm} = \frac{0.983 - 0.690}{0.983} \times 100\% = 29.80 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{0.983 - 0.636}{0.983} \times 100\% = 35.30 \%$$

$$\text{Konsentrasi 60 ppm} = \frac{0.983 - 0.566}{0.983} \times 100\% = 42.42 \%$$



Gambar 9. Grafik fraksi N metode ABTS

Tabel 12. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode ABTS Fraksi O

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas
1	Blanko ABTS	Blanko	0.950 1.000 1.001	0.983	

2	Fraksi O	20 ppm	0.926 0.918 0.896	0.913	7.12
		30 ppm	0.837 0.827 0.807	0.823	16.27
		40 ppm	0.781 0.773 0.777	0.777	20.95
		50 ppm	0.700 0.698 0.699	0.699	28.89
		60 ppm	0.639 0.628 0.613	0.626	36.31

Persentase inhibisi radikal bebas fraksi O

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan blanko}) - (\text{rata-rata serapan sampel})}{\text{rata-rata serapan blanko}} \times 100 \%$$

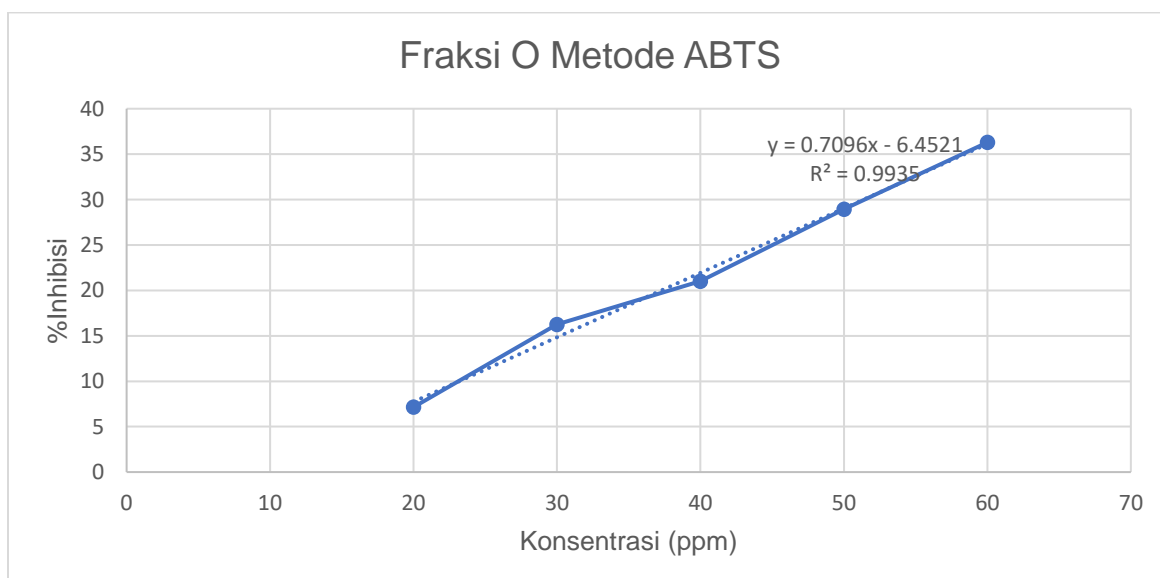
$$\text{Konsentrasi 20 ppm} = \frac{0.983 - 0.913}{0.983} \times 100\% = 7.12 \%$$

$$\text{Konsentrasi 30 ppm} = \frac{0.983 - 0.823}{0.983} \times 100\% = 16.27 \%$$

$$\text{Konsentrasi 40 ppm} = \frac{0.983 - 0.777}{0.983} \times 100\% = 20.95 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{0.983 - 0.699}{0.983} \times 100\% = 28.89 \%$$

$$\text{Konsentrasi 60 ppm} = \frac{0.983 - 0.626}{0.983} \times 100\% = 36.31 \%$$



Gambar 10. Grafik fraksi O metode ABTS

Tabel 13. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* Asam Askorbat metode FRAP

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50		
1	Blanko FRAP	Blanko	0.127	0.127				
			0.136					
			10				0.809	6.61
			0.286					
			30				0.734	55.59
2	Asam askorbat	50	0.564	0.64	77.48	36.57		
			0.986					
			70				0.557	87.11
			1.179					
		90	0.493	89.22				

Persentase inhibisi radikal bebas asam askorbat

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan sampel}) - (\text{rata-rata serapan blanko})}{\text{rata-rata serapan sampel}} \times 100 \%$$

$$\text{Konsentrasi 1 ppm} = \frac{0.136 - 0.127}{0.136} \times 100\% = 6.61 \%$$

$$\text{Konsentrasi 2 ppm} = \frac{0.286 - 0.127}{0.286} \times 100\% = 55.59 \%$$

$$\text{Konsentrasi 3 ppm} = \frac{0.564 - 0.127}{0.564} \times 100\% = 77.48 \%$$

$$\text{Konsentrasi 4 ppm} = \frac{0.984 - 0.127}{0.984} \times 100\% = 87.11 \%$$

$$\text{Konsentrasi 5 ppm} = \frac{1.179 - 0.127}{1.179} \times 100\% = 89.22 \%$$

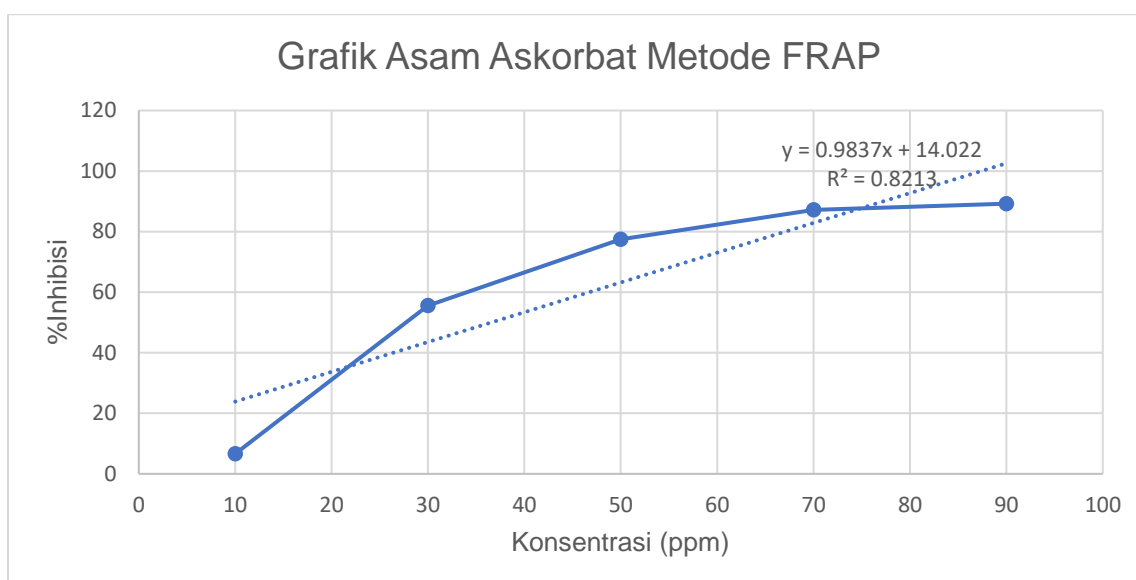
Perhitungan IC50

$$Y = ax + b$$

$$Y = 0.9837x + 14.022$$

$$50 = 0.9837x + 14.022$$

$$X = 36.57$$



Gambar 11. Grafik asam askorbat metode FRAP

Tabel 14. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode FRAP fraksi D

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko FRAP	Blanko	0.197	0.197		
		300 ppm	0.378 0.396 0.398	0.390	49.57	
2	FRAP	600 ppm	0.673 0.693 0.673	0.679	71.01	
		900 ppm	0.995 0.974 0.97	0.979	79.89	80.33
		1200 ppm	1.347 1.372 1.396	1.371	86.63	
		1500 ppm	1.565 1.579 1.59	1.578	87.51	

Persentase inhibisi radikal bebas Fraksi D

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan sampel}) - (\text{rata-rata serapan blanko})}{\text{rata-rata serapan sampel}} \times 100 \%$$

$$\text{Konsentrasi 300 ppm} = \frac{0.390 - 0.197}{0.197} \times 100\% = 49.57 \%$$

$$\text{Konsentrasi 600 ppm} = \frac{0.679 - 0.197}{0.197} \times 100\% = 71.01 \%$$

$$\text{Konsentrasi 900 ppm} = \frac{0.979 - 0.197}{0.197} \times 100\% = 79.89 \%$$

$$\text{Konsentrasi 1200 ppm} = \frac{1.371-0.197}{0.197} \times 100\% = 86.63 \%$$

$$\text{Konsentrasi 1500 ppm} = \frac{1.578-0.197}{0.197} \times 100\% = 87.51 \%$$

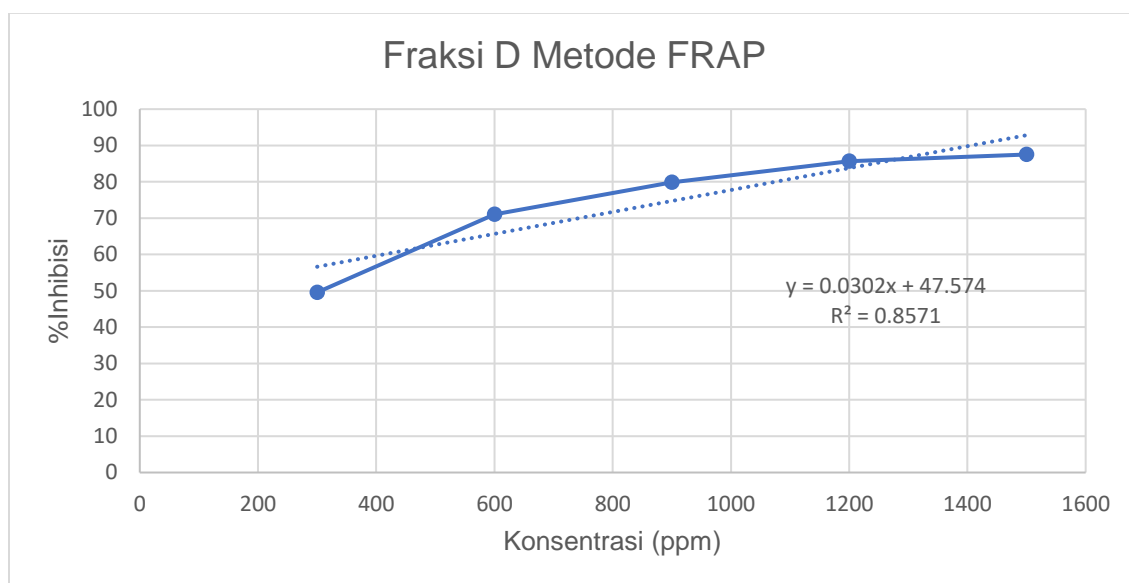
Perhitungan IC50

$$Y = ax+b$$

$$Y = 0.0302x+47.574$$

$$50 = 0.0302x+47.574$$

$$X = 80.33$$



Gambar 12. Grafik fraksi D metode ABTS

Tabel 15. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode FRAP fraksi N

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko FRAP	Blanko	0.117	0.117		
2	Fraksi N	300 ppm	0.161	0.169	30.90	442.641
			0.171			
			0.176			
		600 ppm	0.342	0.352	66.76	
			0.35			
			0.364			
900 ppm	0.568	0.584	79.98			
	0.592					
	0.594					
1200 ppm	0.802	0.815	85.64			
	0.803					
	0.84					

	0.947		
1500 ppm	0.957	0.964	87.86
	0.989		

Persentase inhibisi radikal bebas fraksi N

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan sampel}) - (\text{rata-rata serapan blanko})}{\text{rata-rata serapan sampel}} \times 100 \%$$

$$\text{Konsentrasi 20 ppm} = \frac{0.169 - 0.117}{0.169} \times 100\% = 30.90 \%$$

$$\text{Konsentrasi 30 ppm} = \frac{0.352 - 0.117}{0.352} \times 100\% = 66.76 \%$$

$$\text{Konsentrasi 40 ppm} = \frac{0.584 - 0.117}{0.584} \times 100\% = 79.98 \%$$

$$\text{Konsentrasi 50 ppm} = \frac{0.815 - 0.117}{0.815} \times 100\% = 85.64 \%$$

$$\text{Konsentrasi 60 ppm} = \frac{0.964 - 0.117}{0.964} \times 100\% = 87.86 \%$$

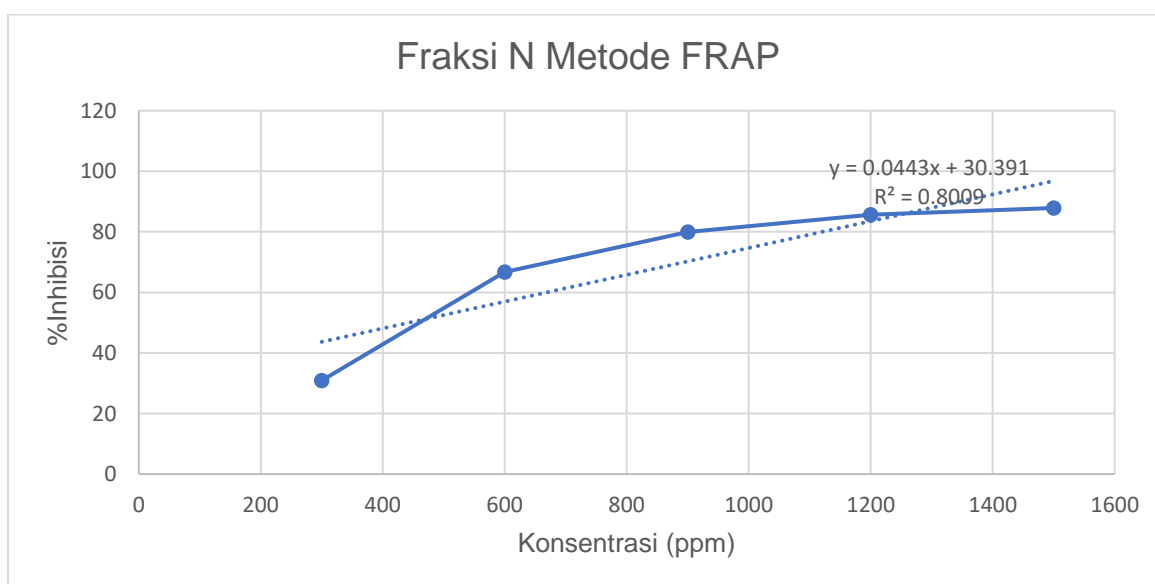
Perhitungan IC50

$$Y = ax + b$$

$$Y = 0.0443x + 30.391$$

$$50 = 0.0443x + 30.391$$

$$X = 442.64$$



Gambar 13. Grafik fraksi N metode FRAP

Tabel 16. Hasil Pengukuran Aktivitas Antioksidan *K. hospita* metode FRAP fraksi O

No	Zat uji	Konsentrasi (ppm)	Serapan	Rata-rata serapan	% inhibisi radikal bebas	IC50
1	Blanko FRAP	Blanko	0.218	0.218		
			0.338 0.339 0.331			
		300 ppm		0.336	35.11	
		600 ppm	0.421 0.42 0.421	0.420	48.17	
2	FRAP	900 ppm	0.543 0.568 0.574	0.561	61.18	686.644
			0.641 0.654 0.654			
			0.763 0.773 0.769			
		1200 ppm		0.649	66.44	
		1500 ppm		0.768	71.62	

Persentase inhibisi radikal bebas Fraksi O

$$\text{Konsentrasi} = \frac{(\text{rata-rata serapan sampel}) - (\text{rata-rata serapan blanko})}{\text{rata-rata serapan sampel}} \times 100 \%$$

$$\text{Konsentrasi 300 ppm} = \frac{0.336 - 0.218}{0.336} \times 100\% = 35.11 \%$$

$$\text{Konsentrasi 600 ppm} = \frac{0.420 - 0.218}{0.420} \times 100\% = 48.17 \%$$

$$\text{Konsentrasi 900 ppm} = \frac{0.561 - 0.218}{0.561} \times 100\% = 61.18 \%$$

$$\text{Konsentrasi 1200 ppm} = \frac{0.649 - 0.218}{0.649} \times 100\% = 66.44 \%$$

$$\text{Konsentrasi 1500 ppm} = \frac{0.768 - 0.218}{0.768} \times 100\% = 71.62 \%$$

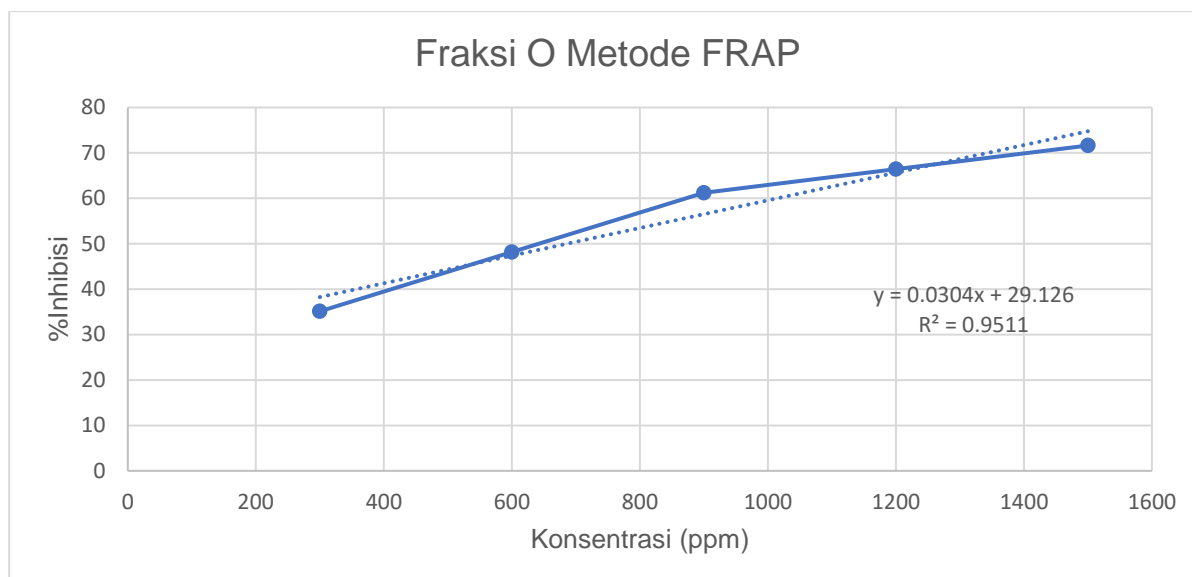
Perhitungan IC50

$$Y = ax + b$$

$$Y = 0.0304x + 29.126$$

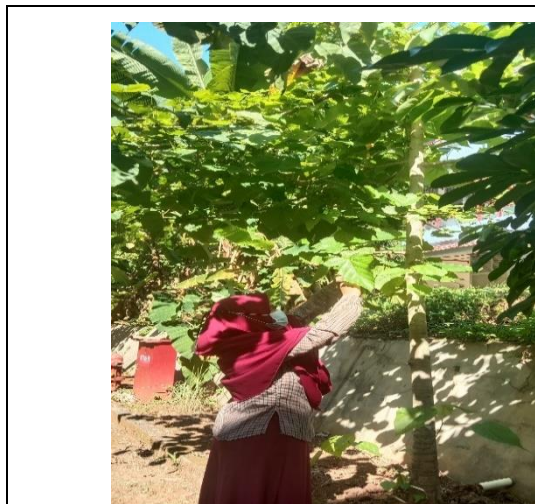
$$50 = 0.0304x + 29.126$$

$$X = 686.644$$



Gambar 14. Grafik fraksi O metode FRAP

Lampiran 3. Foto penelitian

Gambar 15 : pengambilan sampel daun paliasa (*Kleinhovia hospita* L)

Gambar 16 : proses pengeringan daun paliasa



Gambar 17 : sampel yang telah kering



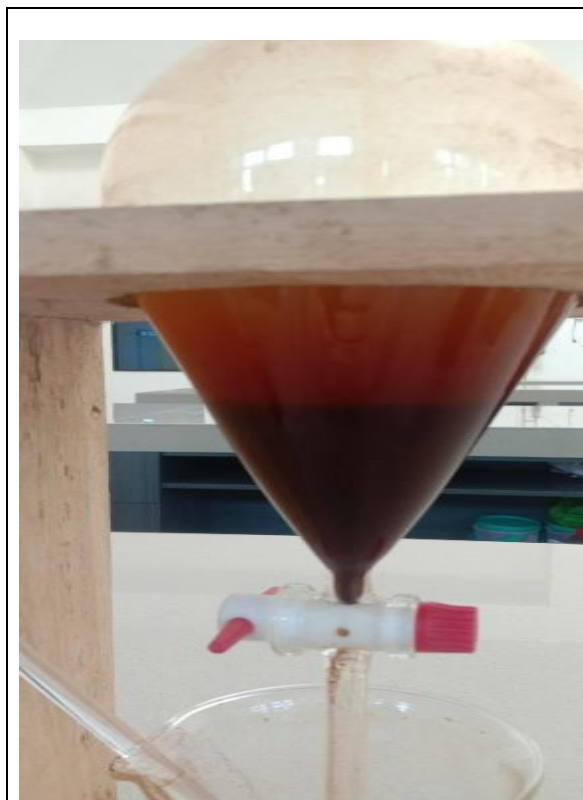
Gambar 18 : perendaman sampel metode maserasi



Gambar 19 : proses penyaringan



Gambar 20 : proses rotarievaporator



Gambar 21 : proses partisi



Gambar 22 : rotavapor partisi



Gambar 23 : proses fraksinasi



Gambar 24 : hasil fraksinasi



Gambar 25 : penotolan fraksi



Gambar. 26 lempeng klt seelum di semprot DPPH diawah sinar UV 366 nm



Gambar 27 : penyemprotan pada klt menggunakan DPPH



Gambar. 28 Hasil setelah di semprot DPPH



Gambar 29 : hasil pengujian aktivitas antioksidan metode DPPH



Gambar 30 : hasil pengujian aktivitas antioksidan metode DPPH



Gambar 31 : hasil pengujian aktivitas antioksidan metode DPPH