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BAB V

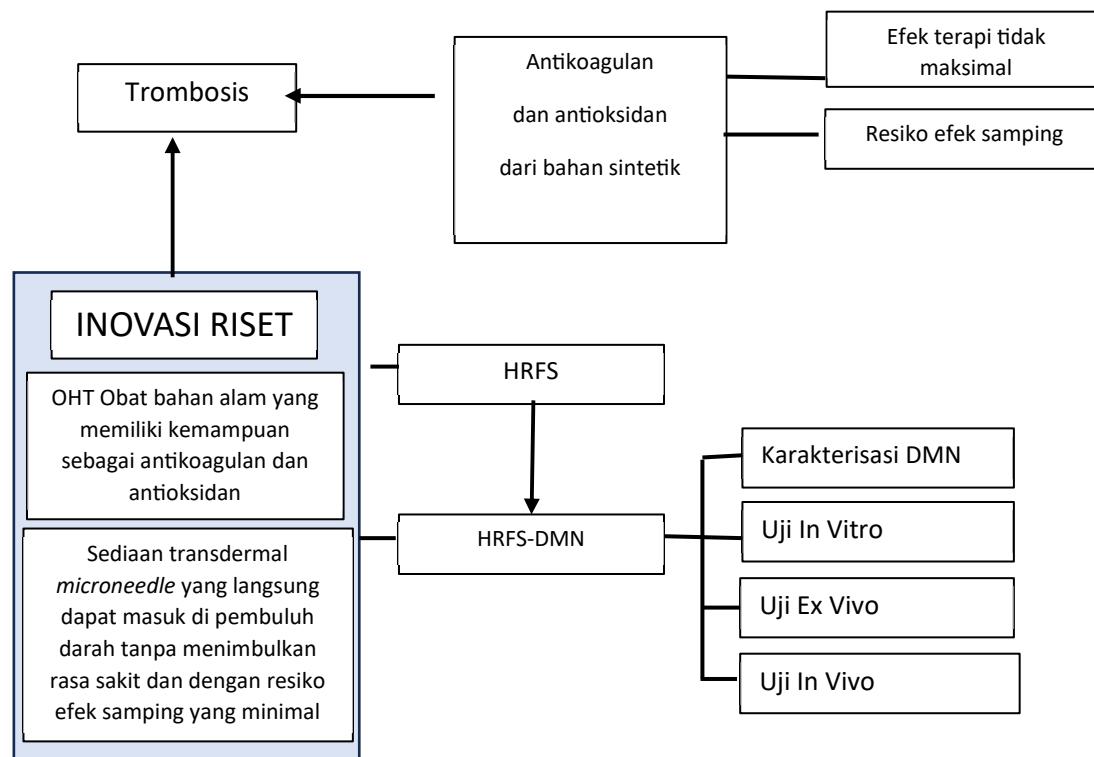
KESIMPULAN UMUM

Penelitian ini menyimpulkan bahwa:

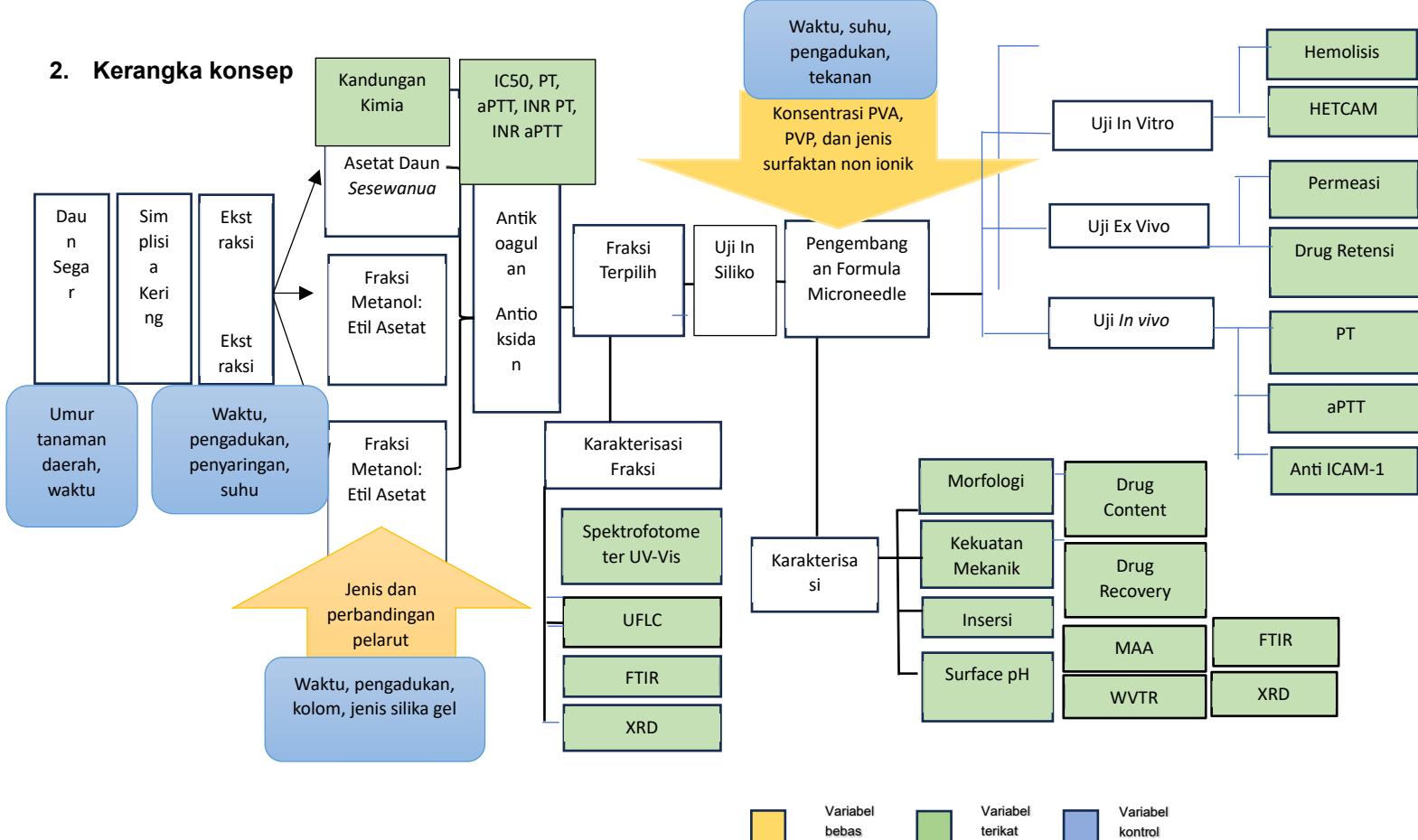
1. Fraksi Toluen-Etil Asetat Sesewanua (*Clerodendrum fragrans* Willd.) dengan perbandingan pelarut 10:0 dan 9:1 yang memiliki kadar hispidulin yang tinggi menunjukkan aktivitas antioksidan yang sangat kuat dan antikoagulan.
2. Sediaan *Dissolving Microneedle* dari Fraksi Toluen-Etil Asetat Sesewanua merupakan kandidat antitrombosis yang bekerja melalui 2 mekanisme, yaitu: antikoagulan dan antioksidan

LAMPIRAN

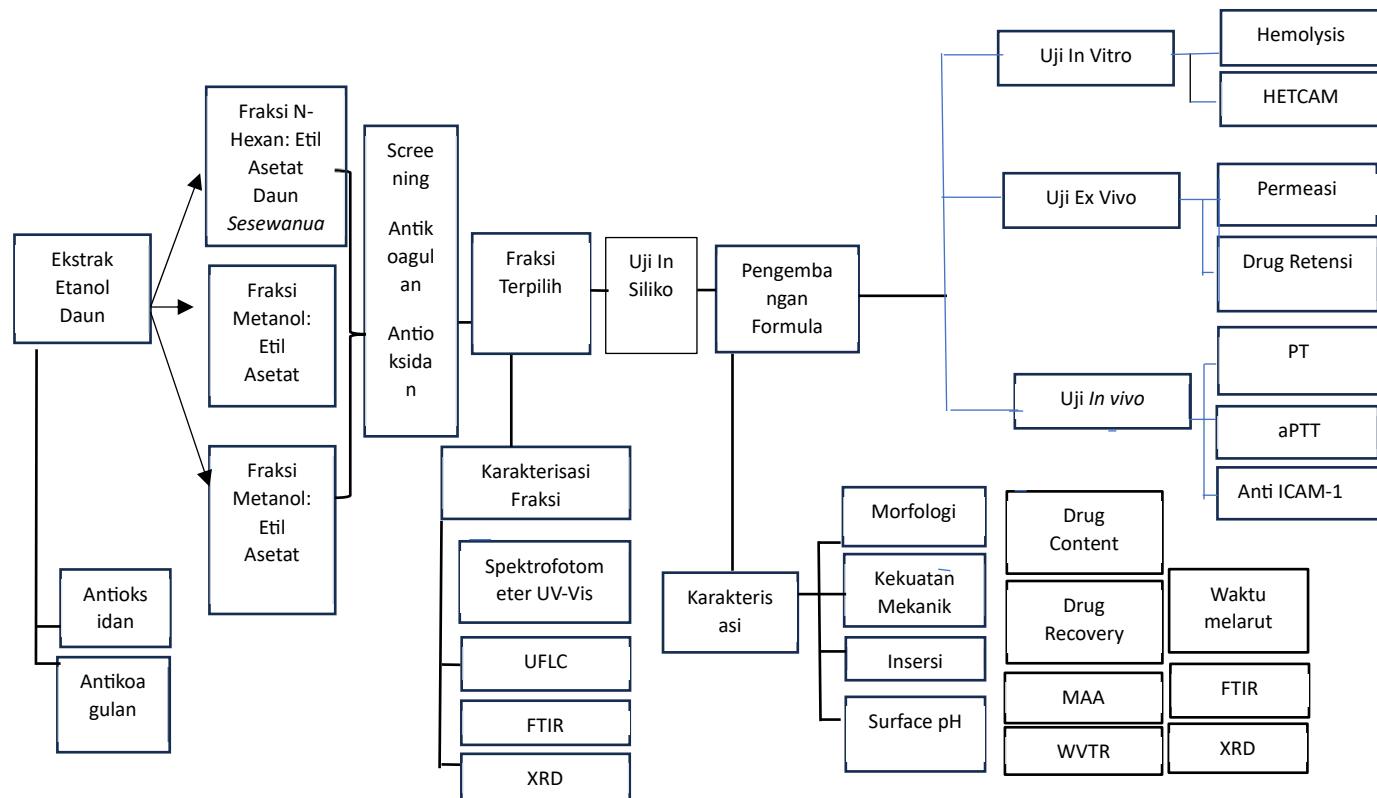
Lampiran 1. Kerangka teori



2. Kerangka konsep



3. Skema Kerja



Lampiran 2. Panjang gelombang maksimum dan kurva baku

1. Panjang gelombang maksimum dan Kurva baku Hispidulin dalam PBS 7,4

Konsentrasi Baku Hispidulin 1.000 bpj selanjutnya diencerkan menjadi 500 bpj dengan cara, mengambil 80 μ l diencerkan dengan PBS hingga 2 ml. Selanjutnya Baku Hispidulin Konsentrasi 500 bpj diencerkan dengan memipet 120 μ l larutan 500 bpj dan diencerkan hingga 750 μ l sehingga diperoleh larutan dengan konsentrasi 80 μ l dan dimasukkan kedalam Spektro UV VIS dan dilakukan penetapan karakteristik profil scan lamda senyawa baku hispidulin menggunakan spektrofotometri UV VIS pada panjang gelombang yang telah disetting dari 200 nm sampai dengan 800 nm.

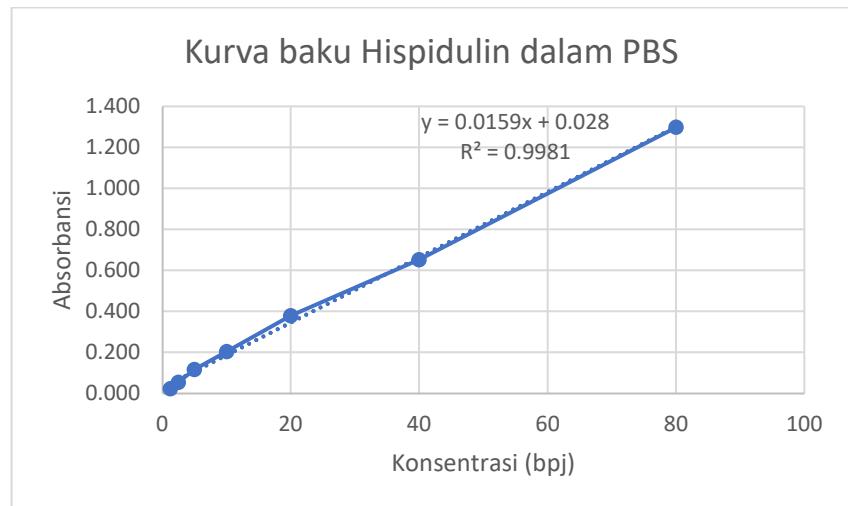


Konsentrasi (bpj)	Rata-rata
80	1.298
40	0,651
20	0,378
10	0,203
5	0,116
2.5	0,054
1,25	0,022

Persamaan garis kurva baku *Hispidulin* dalam PBS pH 7,4

$$Y=0,0159x+0,028$$

$$R^2 = 0,9981$$



Lampiran 3. Dokumentasi penelitian

Sesewanua (*Clerodendrum
fragrans* Willd.)



Simplisia



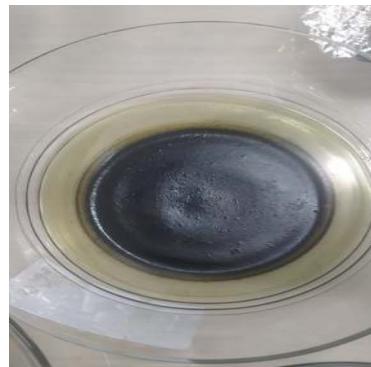
Proses maserasi 3x24 jam



Ekstrak etanol Sesewanua



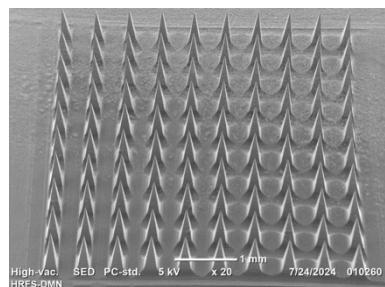
Kromatografi Kolom



Fraksi Sesewanua



Skrining fitokimia



HRFS-DMN



Parafilm setelah insersi



Uji permeasi



Preparasi Kulit Tikus



Proses penempelan HRFS-DMN



Tikus yang telah di tempelkan 4 buah HFRS-DMN



Pengukuran kekentalan darah HRFs-DMN menggunakan alat *coagulometer*



Alat coagulometer



Reagen ICAM-1



Proses penggeraan pengukuran ICAM-
1



Alat ELISA

Lampiran 4. Perhitungan
Perhitungan

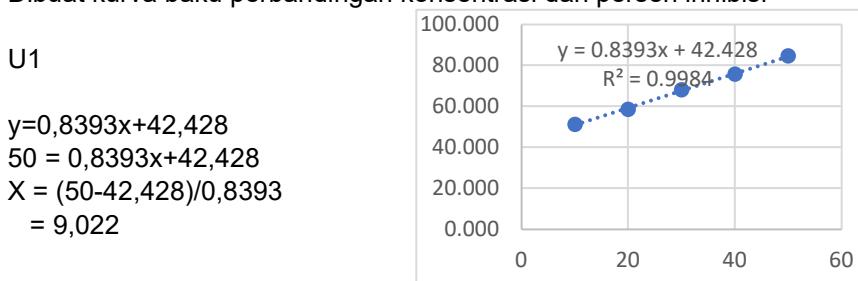
a. Perhitungan IC 50

Fraksi	IC50			Rata-Rata	SD
	U1	U2	U3		
TEFS1	9,02180	10,08125	11,70587	10,26964	1,351913
TEFS2	9,11468	10,06671	10,60240	9,92793	0,753504
TEFS3	1,70920	2,21647	4,54435	2,82334	1,511867
TEFS4	6,83376	7,61380	10,06504	8,17086	1,686131
HEFS1	9,29987	9,49732	7,42264	8,73994	1,145082
HEFS2	12,02175	10,24212	8,46746	10,24378	1,777145
HEFS3	8,94570	8,63641	6,63758	8,07323	1,252891
MEFS1	7,81905	9,03636	9,95679	8,93740	1,072298
MEFS2	1,54595	3,26294	4,28784	3,03225	1,385427
MEFS3	9,92323	11,51832	12,83589	11,42581	1,458531
EES	24,66915	17,15466	6,9100595	16,24462	8,914452

Data inhibisi TEFS1

Konsentrasi	Inhibition			Persen Inhibisi		
	U1	U2	U3	U1	U2	U3
10	0,422	0,419	0,427	51,214	50,822	47,990
20	0,359	0,365	0,344	58,497	57,160	58,100
30	0,276	0,278	0,278	68,092	67,371	66,139
40	0,211	0,205	0,201	75,607	75,939	75,518
50	0,133	0,132	0,132	84,624	84,507	83,922
Kontrol						
DPPH	0,865	0,852	0,821	0,000	0,000	0,000

Dibuat kurva baku perbandingan konsentrasi dan persen inhibisi

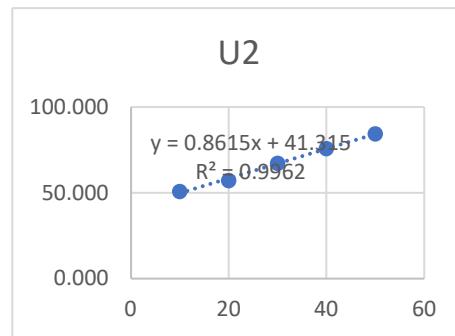


U2

$$Y = 0,8615x + 41,315$$

$$50 = 0,8615x + 41,315$$

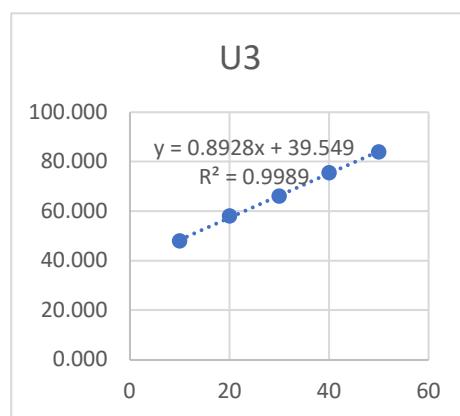
$$X = (50 - 41,315) / 0,8615 \\ = 10,081$$



$$Y = 0,8928x + 39,549$$

$$50 = 0,8928x + 39,549$$

$$X = (50 - 39,549) / 0,8928 \\ = 11,706$$



Jadi, IC50 TEFS1 = $10,27 \pm 1,35$

b. Perhitungan bahan HRFS-DMN

Formula :

Formula	HRFS (%)	PVP (%)	PVA (%)	Tween 80 0,5% (%)	Pluronic F-127 0,5% (%)	Akuades
F5	10	35	5	40	-	10
F8	10	25	10	-	40	15
F9	10	30	5	-	40	15
F11	10	40	5	40	-	5
F12	10	40	5	-	40	5

Perhitungan Bahan (5 g):

Formula	HRFS (mg)	PVP 60% (g)	PVA 30% (g)	Tween 80 8% (g)	Pluronic F-127 8% (g))	Akuades (g)
F5	0,500	2,917	0,833	0,125	-	0,625
F8	0,500	2,083	1,667	-	0,125	0,625
F9	0,500	2,500	0,833	-	0,125	1,042

F11	0,500	3,333	0,833	0,125	-	0,208
F12	0,500	3,333	0,833	-	0,125	0,208

Pembuatan stok 50 g:

- PVP 60% : $\frac{60}{100} \times 50 \text{ g} = 30 \text{ g}$, akuades ad 50 g
- PVA 30% : $\frac{30}{100} \times 50 \text{ g} = 15 \text{ g}$, akuades ad 50 g
- Tween 80 8% : $\frac{8}{100} \times 50 \text{ g} = 4 \text{ g}$, akuades ad 50 g
- Pluronic-127 : $\frac{8}{100} \times 50 \text{ g} = 4 \text{ g}$, akuades ad 50 g

Untuk pembuatan 5 g, berikut perhitungan untuk mengetahui jumlah bahan menggunakan stok, contoh F9:

- HRFS 10% : $\frac{10}{100} \times 5 \text{ g} = 0,500 \text{ g}$
- PVP 30% : $\frac{30}{60} \times 5 \text{ g} = 2,500 \text{ g}$
- PVA 5% : $\frac{5}{30} \times 5 \text{ g} = 0,833 \text{ g}$
- Pluronic F-127 0,5% : $\frac{0,5}{8} \times 40\% \times 5 \text{ g} = 0,125 \text{ g}$
- Akuades ad : $5 - (0,500 + 2,500 + 0,833 + 0,125) = 1,042 \text{ g}$

c. Hasil penentuan volume dan densitas balok HFRS-DMN

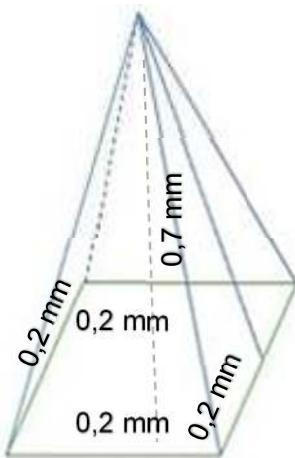
Formula	Berat (mg)	Panjang (mm)	Lebar (mm)	Tinggi (mm)	Densitas (mg/mm ³)	Rata-rata	SD
F5	57,3	4,8	4,6	2,0	1,298	1,224	0,076
	48,5	4,6	4,6	2,0	1,146		
	54,2	4,8	4,6	2,0	1,227		
F8	43,1	4,7	4,6	1,7	1,173	1,073	0,223
	43,2	4,5	4,8	1,7	1,228		
	33,2	4,7	4,8	1,8	0,818		
F9	57,7	4,7	4,7	1,7	1,535	1,451	0,118
	51,9	4,7	4,6	1,6	1,500		
	46,3	4,5	4,6	1,7	1,316		
F11	41,5	4,69	4,67	1,77	1,070	1,207	0,118
	50,0	4,58	4,85	1,79	1,270		
	52,8	4,53	4,74	1,92	1,281		
F12	65,8	4,76	4,81	2,04	1,409	1,190	0,191
	49,4	4,61	4,88	1,99	1,103		
	49,3	4,63	4,73	2,13	1,057		

$$\rho = \frac{\text{massa}}{\text{volume}}$$

$$= \frac{57,3 \text{ mg}}{(4,8 \times 4,6 \times 2,0) \text{ mm}^3}$$

$$= 1,298 \text{ mg/mm}^3$$

d. Data Dimensi dan Volume HRFS-DMN



Panjang (mm)	Lebar (mm)	Tinggi (mm)	Volume (mm ³)	Volume 100 jarum (mm ³)
0,2	0,2	0,7	0,00933	0,933

$$\text{Volume} = \frac{1}{3} \times \text{Luas alas} \times \text{tinggi}$$

$$\text{Volume DMN} = \frac{1}{3} \times 0,2 \times 0,2 \times 0,7$$

$$= 0,00933 \text{ mm}^3$$

Bobot 100 jarum :

Bobot = Densitas x Volume

Formula	Densitas (mg/mm ³)	Volume (mm ³)	Bobot 100 jarum (mm)	Rata-rata	SD
F5	1,298	0,993	1,211		
	1,146	0,993	1,070	1,142	0,071
	1,227	0,993	1,145		
F8	1,173	0,993	1,095		
	1,228	0,993	1,146	1,001	0,208
	0,818	0,993	0,763		
F9	1,535	0,993	1,434		
	1,500	0,993	1,400	1,345	0,110
	1,316	0,993	1,228		
F11	1,070	0,993	0,999		
	1,270	0,993	0,185	1,126	0,111
	1,281	0,993	1,196		
F12	1,409	0,993	1,315		
	1,103	0,993	1,029	1,110	0,179
	1,057	0,993	0,986		

e. Perhitungan LOD (*Loss on Drying*) HRFS-DMN

Formula	Bobot basah (g)	Bobot kering (g))	LOD (%)	Rata-rata	SD
F5	0,537	0,217	59,590	55,439	4,709
	0,465	0,231	50,323		
	0,523	0,228	56,405		
F8	0,952	0,342	64,076	60,783	4,278
	0,929	0,350	62,325		
	0,765	0,337	55,948		
F9	0,666	0,241	63,814	60,736	2,958
	0,706	0,279	60,482		
	0,556	0,234	57,914		
F11	0,375	0,175	53,333	55,359	1,776
	0,369	0,162	56,098		
	0,150	0,150	56,647		
F12	0,497	0,210	57,746	57,946	1,655
	0,539	0,235	56,401		
	0,454	0,183	59,692		

$$\% \text{ LOD} = \frac{\text{bobot basah} - \text{bobot kering}}{\text{bobot basah}} \times 100\%$$

$$= \frac{0,537 - 0,217}{0,537} \times 100\% \\ = 59,590\%$$

f. Perhitungan persentasi HRFS dalam massa kering

Formula	HRFS (%)	Rata-rata	SD
F5	24,75	22,61	2,33
	20,13		
	22,94		
F8	27,84	25,69	2,67
	26,54		
	22,70		
F9	27,63	25,57	1,95
	25,30		
	23,76		
F11	21,43	22,42	0,87
	22,78		
	23,07		
F12	23,67	23,80	0,94
	22,94		
	24,81		

Diketahui berat HRFS awal dalam formula adalah 10% b/b dan LOD 59,590%.

g. Penentuan bobot teoritis HRFS dan HIS dalam DMN

Formula	HRFS dalam massa kering (mg)	Bobot 100 jarum (mg)	Bobot teoritis HRFS dlm sediaan (mg)	Bobot teoritis HIS dlm sediaan (mg)	Rata-rata	SD
F5	0,247	1,211	0,300	0,147	0,127	0,021
	0,201	1,070	0,215	0,106		
	0,229	1,145	0,263	0,129		
F8	0,278	1,095	0,305	0,150	0,128	0,037
	0,265	1,146	0,304	0,149		
	0,227	0,763	0,173	0,085		
F9	0,276	1,434	0,396	0,195	0,171	0,026
	0,253	1,400	0,354	0,174		
	0,238	1,228	0,292	0,143		
F11	0,214	0,999	0,214	0,105	0,124	0,017
	0,228	0,185	0,270	0,133		
	0,231	1,196	0,276	0,135		
F12	0,237	1,315	0,311	0,153	0,130	0,020
	0,229	1,029	0,236	0,116		
	0,248	0,986	0,245	0,120		

Diketahui HRFS mengandung 49,1% HIS

Berat hispidulin dalam sediaan = bobot HRFS dalam massa kering x bobot 100

jarumx %bobot HIS dalam HRFS

$$= 0,274 \times 1,211 \times 0,491$$

$$= 0,147 \text{ mg}$$

h. Penentuan % drug recovery

Formula	Absorban si	Konsentrasi (bpj)	Konsentrasi awal (bpj)	Drug recover y (%)	Rata-rata	SD
F5	0,068	2,516	2,546	98,81	97,99	1,426
	0,068	2,516	2,546	98,81		
	0,067	2,453	2,561	96,34		
F8	0,068	2,516	2,561	98,23	96,59	1,418
	0,067	2,453	2,561	95,78		
	0,067	2,453	2,561	98,23		
F9	0,082	3,396	3,412	99,57	98,95	1,064
	0,081	3,333	3,412	97,72		
	0,082	3,396	3,412	99,57		
F11	0,067	2,453	2,487	98,63	97,78	1,460
	0,066	2,390	2,487	96,10		
	0,067	2,453	2,487	98,63		
F12	0,069	2,579	2,593	99,45	98,64	1,400
	0,068	2,516	2,593	97,02		
	0,069	2,579	2,593	99,45		

Diketahui bobot teoritis HIS dalam F5, F8, F9, F11, dan F12 adalah 0,127, 0,128, 0,171, 0,125 dan 0,126

Persamaan kurva baku = $0,0159x + 0,028$

Berat teoritis dari HRFS-DMN ($F_5 = 0,127$) dilarutkan dalam 5 ml PBS sehingga konsentrasi larutan stok menjadi 25,46 kemudian dicuplik 100 μl dan dicukupkan hingga 1000 μL .

$$\begin{aligned}\text{Konsentrasi akhir larutan} &= \frac{100 \text{ mikroliter}}{1000 \text{ mikroliter}} \times 25,46 \text{ bpj} \\ &= 2,546\end{aligned}$$

$$\begin{aligned}\text{Konsentrasi HIS dalam sampel} &= \frac{0,068 - 0,028}{0,0159} \\ &= 2,516 \text{ bpj}\end{aligned}$$

$$\begin{aligned}\% \text{ drug recovery} &= \frac{\text{konsentrasi sampel}}{\text{konsentrasi larutan}} \times 100\% \\ &= \frac{2,516}{2,546} \times 100\% \\ &= 98,81\%\end{aligned}$$

i. Permeasi

Contoh perhitungan permeasi obat secara *ex vivo*

Diketahui persamaan garis linear *hispidulin* dalam PBS pH 7,4 adalah $y = 0,0159x + 0,028$

Ket:

y = serapan / absorbansi

x = konsentrasi

F5 – Replikasi 1, jam ke-60 diperoleh serapan 0,115.

Sehingga perhitungan konsentrasi adalah sebagai berikut:

$$\begin{aligned}x &= \frac{y-b}{a} \\ x &= \frac{0,115-0,028}{0,0159} = 5,472 \mu\text{g/ml}\end{aligned}$$

$$\begin{aligned}\text{Konsentrasi dalam 1 ml} &= 5,472 \mu\text{g/ml} \times 1 \text{ ml} \\ &= 5,472 \mu\text{g}\end{aligned}$$

Faktor pengenceran 4x

$$\begin{aligned}\text{Konsentrasi dalam 13 ml} &= 71,226 \mu\text{g/ml} \times 13 \times 1 \text{ ml} \times 4 \\ &= 284,528 \mu\text{g}\end{aligned}$$

$$\begin{aligned}\text{Faktor koreksi} &= \text{Konsentrasi jam sebelumnya} + \text{faktor koreksi jam sebelumnya} \\ &= 5,472 \mu\text{g} + 25,660 \mu\text{g} \\ &= 31,132 \mu\text{g}\end{aligned}$$

$$\text{Jumlah permeasi obat (mg)} = \text{Konsentrasi dalam 13 ml} + \text{faktor koreksi}$$

$$\begin{aligned}&= \frac{(284,528 \mu\text{g} + 31,132 \mu\text{g})}{1000} \\ &= 0,316 \text{ mg}\end{aligned}$$

$$\begin{aligned}\text{Jumlah permeasi obat perluas area kulit} &= \frac{0,316}{1,9856} = 0,159 \text{ mg/cm}^2 \\ &= 159 \mu\text{g/cm}^2\end{aligned}$$

j. Perhitungan dosis untuk tikus

Heparin Injeksi

Faktor konversi manusia dengan berat 70 kg ke tikus dengan berat 200 g adalah 0,018.

Diketahui :

Dosis heparin untuk manusia adalah 200 IU/kg BB.

Untuk manusia dengan bobot 70 kg = 200 IU x 70 kg = 14000 IU

Untuk tikus dengan bobot 200 g = 1400 IU x 0,018 = 250 IU

Volume yang akan diebrikan pada tikus dengan bobot 200 g adalah 0,5 ml

Larutan stok yang akan dibuat 5 ml

Bobot etiket heparin injeksi = 5000 IU/ml

Perhitungan :

Jumlah heparin yang dibutuhkan untuk membuat larutan stok 5 ml

250 IU untuk 0,5 ml dan untuk 5 ml = 2500 IU

Jumlah heparin yang dicuplik untuk membuat larutan stok 5 ml

$$= \frac{(2500)}{5000}$$

$$= 0,5 \text{ ml}$$

Pemberian menyesuaikan dengan berat tikus.

$$\text{Tikus I} = \frac{290 \text{ g}}{200 \text{ g}} \times 0,5 \text{ ml} = 0,73 \text{ ml}$$

$$\text{Tikus II} = \frac{310 \text{ g}}{200 \text{ g}} \times 0,5 \text{ ml} = 0,78 \text{ ml}$$

$$\text{Tikus III} = \frac{290 \text{ g}}{200 \text{ g}} \times 0,5 \text{ ml} = 0,73 \text{ ml}$$

Quersetin oral

Dosis manusia = 120 mg/hari, terbagi dalam 3 kali pemberian (Chekalina et al., 2018)

Konversi dosis untuk tikus 200 g = 120 mg x 0,018 = 2,16 mg (dalam 2 ml)

Dibuat suspensi CMC 0,5% sebanyak 10 ml sebagai pembawa, disuspensikan 10,8 mg. Pemberian menyesuaikan dengan berat tikus.

$$\text{Tikus I} = \frac{290 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,9 \text{ ml}$$

$$\text{Tikus II} = \frac{310 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 3,1 \text{ ml}$$

$$\text{Tikus III} = \frac{290 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,9 \text{ ml}$$

HRFS-O1

Dosis tikus oral 1 diberikan setara dengan drug content 1 HRFS-DMN, yaitu 0,347 mg HRFS yang didispersikan dalam suspensi NaCMC 0,5% sebanyak 2 ml. Dibuat suspensi CMC 0,5% sebanyak 10 ml sebagai pembawa, disuspensikan 1,735 mg. Pemberian menyesuaikan dengan berat tikus.

$$\text{Tikus I} = \frac{253 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,5 \text{ ml}$$

$$\text{Tikus II} = \frac{245 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,5 \text{ ml}$$

$$\text{Tikus III} = \frac{290 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,9 \text{ ml}$$

HRFS-O2

Dosis tikus oral 1 diberikan setara dengan drug content 2 HRFS-DMN, yaitu 0,694 mg HRFS yang didispersikan dalam suspensi NaCMC 0,5% sebanyak 2 ml. Dibuat suspensi CMC 0,5% sebanyak 10 ml sebagai pembawa, disuspensikan 3,47 mg. Pemberian menyesuaikan dengan berat tikus.

$$\text{Tikus I} = \frac{283 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,83 \text{ ml}$$

$$\text{Tikus II} = \frac{343 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 3,43 \text{ ml}$$

$$\text{Tikus III} = \frac{344 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 3,44 \text{ ml}$$

HRFS-O4

Dosis tikus oral 1 diberikan setaradengan drug content 4 HRFS-DMN, yaitu 1,388 mg HRFS yang didispersikan dalam suspensi NaCMC 0,5% sebanyak 2 ml. Dibuat suspensi CMC 0,5% sebanyak 10 ml sebagai pembawa, disuspensikan 6,94 mg. Pemberian menyesuaikan dengan berat tikus.

$$\text{Tikus I} = \frac{231 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,3 \text{ ml}$$

$$\text{Tikus II} = \frac{235 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,4 \text{ ml}$$

$$\text{Tikus III} = \frac{225 \text{ g}}{200 \text{ g}} \times 2 \text{ ml} = 2,3 \text{ ml}$$

Lampiran 5. Data *In vivo***A. Data *In vivo* PT**

Sediaan	AE	KB	KD	Rata-rata	SD
HEP	17,3	17,1	17,2	17,2	0,1
QUE	13,1	14,0	11,8	13,0	1,1
HRFSFree-DMN	9,5	8,0	7,1	8,2	1,2
HRFS-DMN1	14,3	13,8	13,4	13,8	0,5
HRFS-DMN2	17,1	17,0	16,8	17,0	0,1
HRFS-DMN4	17,2	17,0	17,1	17,1	0,1
HRFS-O1	9,9	6,5	11,0	9,1	2,3
HRFS-O2	11,5	12,4	10,0	11,3	1,2
HRFS-O4	12,7	13,2	14,7	13,5	1,0

B. Data *in vivo* aPTT

Sediaan	AE	KB	KD	Rata-rata	SD
HEP	1,3	33,0	32,2	22,2	18,1
QUE	23,2	23,8	17,1	21,4	3,7
HRFSFree-DMN	12,1	12,0	15,8	13,3	2,2
HRFS-DMN1	23,5	23,8	22,7	23,3	0,6
HRFS-DMN2	29,8	28,8	29,3	29,3	0,5
HRFS-DMN4	32,0	32,6	31,6	32,1	0,5
HRFS-O1	16,6	16,8	16,3	16,6	0,3
HRFS-O2	21,4	19,7	18,4	19,8	1,5
HRFS-O4	22,8	25,5	26,0	24,8	1,7

C. Data *in vivo* ICAM

Sediaan	AE	KB	KD	Rata-rata	SD
HEP	2,9070	3,7053	3,6798	3,4307	0,4537
QUE	3,6334	3,3885	3,848	3,6233	0,2299
HRFSFree-DMN	3,1496	3,1063	3,2043	3,1534	0,0491
HRFS-DMN1	3,0312	3,4469	3,1423	3,2068	0,2152
HRFS-DMN2	2,3751	2,3610	2,3481	2,3614	0,0135
HRFS-DMN4	1,861	2,4383	2,1362	2,1452	0,2888
HRFS-O1	3,1473	4,0559	3,3678	3,5237	0,4739
HRFS-O2	3,468	2,7148	2,5095	2,8974	0,5047
HRFS-O4	2,2216	2,7757	2,9188	2,6387	0,3682

Lampiran 6. Etik Penelitian



LEMBAR KEPUTUSAN ETIK

Nomor : 265/UN4.17.8/KP.06.07/2024
 Judul Penelitian : Pengembangan Formula *Microneedle Fraksi Daun Sesewamua (Clerodendrum fragrans Wild.)* sebagai Kandidat Antithrombosis
 Nama Peneliti : Zulfiayu
 Nomor Registrasi :

U	H	0	1	2	4	0	2	0	3	9
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A	Rangkuman penilaian oleh <i>reviewers</i>									
B	Perlu <i>full board</i> : <input type="checkbox"/> Ya <input checked="" type="checkbox"/> Tidak a. Ya (terus ke C) b. Tidak (terus ke D)									
C	Catatan Rapat Etik (<i>Full Board</i>) — Tgl/bulan/tahun _____ Tindak lanjut/catatan rapat etik Dikirimkan kembali ke yang bersangkutan dengan tembusan kepimpinan instansi									
D	Hasil Penilaian <input type="checkbox"/> a. Disetujui <input checked="" type="checkbox"/> b. Disetujui dengan revisi minor (lihat lembaran pertimbangan/saran/petunjuk) <input type="checkbox"/> c. Disetujui dengan revisi mayor (lihat lembaran pertimbangan/saran/petunjuk) <input type="checkbox"/> d. Ditunda untuk beberapa alasan (lihat lembaran pertimbangan/saran/petunjuk) <input type="checkbox"/> e. Ditolak/tidak dapat disetujui (lihat lembaran pertimbangan/saran/petunjuk)									
E	Penugasan pengawasan jalannya penelitian di lapangan untuk yang berisiko sedang – berat, mengobservasi apakah ada penyimpangan etik (tulis nama anggota komisi etik yang ditunjuk oleh rapat): —									

Makassar, 4 Maret 2024

Sekretaris

Nurhasni Hasan, M.Si., M.Pharm.Sc., Ph.D., Apt
 NIP. 19860116 201012 2 009



CURRICULUM VITAE

A. Data Pribadi

1. Nama : Zulfiayu
2. Tempat, Tanggal Lahir : Selong Lombok, 08-08-1975
3. Alamat : Jalan Brigjen Piola Isa Blok Tatudi No 14
Griya Fitrah Mandiri, Gorontalo
4. Kewarganegaraan : Indonesia

B. Riwayat Pendidikan

1. Tamat SMA tahun 1994 di SMAN 1 Mataram
2. Sarjana Farmasi tahun 2000 di Universitas Hasanuddin, Makassar
3. Profesi Apoteker tahun 2002 di Universitas Hasanuddin, Makassar
4. Magister Farmasi tahun 2006 di Universitas Gadjah Mada, Yogyakarta

C. Pekerjaan dan Riwayat Pekerjaan

1. Jenis Pekerjaan : Dosen
2. NIDN : 4008087501
3. Pangkat/Golongan : Pembina/IVA
4. Jabatan : Lektor

D. Karya Ilmiah yang Telah Dipublikasikan

1. Determination of total flavonoid levels of ethanol extract *Sesewanua* leaf (*Clerodendrum fragrans* Willd.) with maceration method using UV-vis spectrophotometry. *Pharmacognosy Journal*, 12(2), 356–360. <https://doi.org/10.5530/pj.2020.12.56>
2. *Hispidulin-rich fraction of Clerodendrum fragrans* Willd. (*Sesewanua*) dissolving microneedle as antithrombosis candidate: A proof of concept study, *International Journal of Pharmaceutics*, Vol Volume 666 pp 1-17, 5 December 2024, <https://doi.org/10.1016/j.ijpharm.2024.124766>.
3. Anti-inflammatory activities of flavonoid derivates, *ADMET & DMPK*, 11(3), 2023, 331-359, <https://doi.org/10.5599/admet.1918>.

E. Makalah pada Seminar/Konferensi Ilmiah Nasional dan Internasional

1. Fractionation of Ethanol Extract of *Sesewanua* Leaves (*Clerodendrum fragrans* Willd.) in n-Hexane: Ethyl Acetate and Their Antioxidant Activity, International Conference on Science, FMIPA Unhas, 2022
2. Cytotoxic Activity of *Sesewanua* (*Clerodendrum fragrans* Willd) Leaf Ethanol Extract on Breast Cancer Cell. Proceeding 1 st International Conference on Clinical Laboratory and Environmental Health (ICOCLEH) (p. 24). Health Polytechnic of the Ministry of Health, Surabaya