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## Lampiran 1. Perhitungan Energi Penyinaran

### 1. LED Merah

#### ❖ Transmitansi dan % serap

$$\lambda_{LED \text{ merah}} = 620 \text{ nm}$$

$$Abs = 0,112$$

$$T = 10^{-A}$$

$$= 10^{-(0,112)}$$

$$= 0,773$$

$$\% \text{ serap} = (1-T) \times 100\%$$

$$= (1-0,773) \times 100\%$$

$$= 22,7\%$$

#### ❖ Intensitas LED

$$I_{LED} = 1221 \text{ Lux}$$

$$= 1221 \times (1,464 \times 10^{-4} \text{ mW/cm}^2)$$

$$= 0,179 \text{ mW/cm}^2$$

#### ❖ Intensitas Serap

$$I_{serap} = \% \text{ serap} \times I_{LED}$$

$$= 22,7\% \times 0,179 \text{ mW/cm}^2$$

$$= 0,041 \text{ mW/cm}^2$$

##### a. Untuk t= 120 sekon

$$E = I_{serap} \times t$$

$$= 0,041 \times 120$$

$$= 4,92 \text{ mJ/cm}^2$$

##### b. Untuk t= 240 sekon

$$E = I_{serap} \times t$$

$$= 0,041 \times 240$$

$$= 9,84 \text{ mJ/cm}^2$$

##### c. Untuk t= 360 sekon

$$E = I_{serap} \times t$$

$$= 0,041 \times 360$$

$$= 14,76 \text{ mJ/cm}^2$$

##### d. Untuk t= 480 sekon

$$E = I_{serap} \times t$$

$$= 0,041 \times 480$$

$$= 19,68 \text{ mJ/cm}^2$$

##### e. Untuk t= 600 sekon

$$E = I_{serap} \times t$$

$$= 0,041 \times 600$$

$$= 24,60 \text{ mJ/cm}^2$$

### 2. LED Biru

#### ❖ Transmitansi dan % serap

$$\lambda_{LED \text{ biru}} = 450 \text{ nm}$$

$$\begin{aligned}
\text{Abs} &= 1,711 \\
T &= 10^{-A} \\
&= 10^{-(1,711)} \\
&= 0,019 \\
\% \text{ serap} &= (1-T) \times 100\% \\
&= (1-0,019) \times 100\% \\
&= 98,1\%
\end{aligned}$$

❖ Intensitas LED

$$\begin{aligned}
I_{\text{LED}} &= 961 \text{ Lux} \\
&= 961 \times (1,464 \times 10^{-4} \text{ mW/cm}^2) \\
&= 0,141 \text{ mW/cm}^2
\end{aligned}$$

❖ Intensitas Serap

$$\begin{aligned}
I_{\text{serap}} &= \% \text{ serap} \times I_{\text{LED}} \\
&= 98,1\% \times 0,141 \text{ mW/cm}^2 \\
&= 0,138 \text{ mW/cm}^2
\end{aligned}$$

- a. Untuk  $t = 120$  sekon
- $$\begin{aligned}
E &= I_{\text{serap}} \times t \\
&= 0,138 \times 120 \\
&= 16,56 \text{ mJ/cm}^2
\end{aligned}$$
- b. Untuk  $t = 240$  sekon
- $$\begin{aligned}
E &= I_{\text{serap}} \times t \\
&= 0,138 \times 240 \\
&= 33,12 \text{ mJ/cm}^2
\end{aligned}$$
- c. Untuk  $t = 360$  sekon
- $$\begin{aligned}
E &= I_{\text{serap}} \times t \\
&= 0,138 \times 360 \\
&= 49,68 \text{ mJ/cm}^2
\end{aligned}$$
- d. Untuk  $t = 480$  sekon
- $$\begin{aligned}
E &= I_{\text{serap}} \times t \\
&= 0,138 \times 480 \\
&= 66,24 \text{ mJ/cm}^2
\end{aligned}$$
- e. Untuk  $t = 600$  sekon
- $$\begin{aligned}
E &= I_{\text{serap}} \times t \\
&= 0,138 \times 600 \\
&= 82,80 \text{ mJ/cm}^2
\end{aligned}$$

Lampiran 2. Data Mentah *Optical density* (OD) dengan Kode Perlakuan

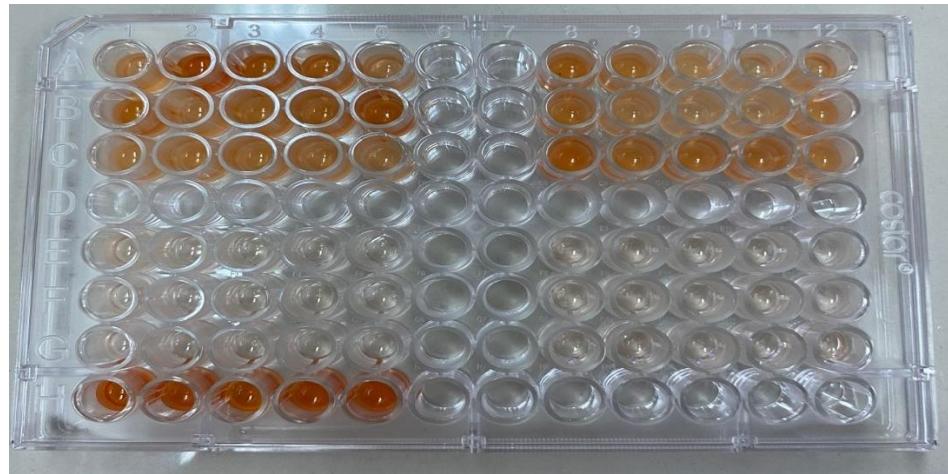
❖ Nilai Optical Density (OD)

	1	2	3	4	5	6	7	8	9	10	11	
A	1.101	1.364	1.133	0.785	1.049	X	X	1.47	1.223	1.002	0.76	0.64
B	1.203	0.859	0.9009	0.1.169	0.912	X	X	1.1372	0.934	0.742	1.123	1.0346
C	1.201	0.9925	0.9523	0.793	0.6601	X	X	1.6451	0.958	1.087	0.881	1.0645
D	X	X	X	X	X	X	X	X	X	X	X	X
E	0.1962	0.1278	0.1553	0.0836	0.0707	X	X	0.186	0.1763	0.1645	0.1429	0.1174
F	0.1451	0.1185	0.0478	0.1164	0.0774	X	X	0.1911	0.1764	0.1623	0.1761	0.1201
G	0.1331	0.1357	0.1286	0.0785	0.0815	X	X	0.2014	0.1992	0.1785	0.1552	0.1557
H	1.439	1.354	1.526	1.455	1.148	X	X	X	X	X	X	X

❖ Kode Perlakuan

1	2	3	4	5	6	7	8	9	10	11	
L1B	L2B	L3B	L4B	L5B	X	X	L1M	L2M	L3M	L4M	L5M
L1B	L2B	L3B	L4B	L5B	X	X	L1M	L2M	L3M	L4M	L5M
L1B	L2B	L3B	L4B	L5B	X	X	L1M	L2M	L3M	L4M	L5M
X	X	X	X	X	X	X	X	X	X	X	X
CL1B	CL2B	CL3B	CL4B	CL5B	X	X	CL1M	CL2M	CL3M	CL4M	CL5M
CL1B	CL2B	CL3B	CL4B	CL5B	X	X	CL1M	CL2M	CL3M	CL4M	CL5M
CL1B	CL2B	CL3B	CL4B	CL5B	X	X	CL1M	CL2M	CL3M	CL4M	CL5M
C-	C-	C-	C-	C-	X	X	X	X	X	X	X

❖ Gambar Sampel



Lampiran 3. Perhitungan % Inhibition.

Kelompok Perlakuan	OD ( $\lambda=490$ )									
	LED Biru					LED Merah				
	1	2	3	Rerata	SD	1	2	3	Rerata	SD
C-	1.439	1.354	1.526	1.440	0.086	1.439	1.354	1.526	1.440	0.086
C+	0.312	0.313	0.414	0.346	0.059	0.312	0.313	0.414	0.346	0.059
L <sub>1</sub>	1.101	1.203	1.201	1.168	0.058	1.470	1.137	1.6451	1.417	0.257
L <sub>2</sub>	1.364	0.859	0.992	1.072	0.262	1.223	0.934	0.958	1.038	0.161
L <sub>3</sub>	1.133	0.9	0.952	0.995	0.122	1.002	0.742	1.087	0.943	0.180
L <sub>4</sub>	0.785	1.169	0.793	0.916	0.219	0.760	1.123	0.881	0.921	0.184
L <sub>5</sub>	1.049	0.912	0.66	0.874	0.197	0.640	1.035	1.064	0.913	0.236
CL <sub>1</sub>	0.196	0.145	0.133	0.158	0.033	0.186	0.191	0.201	0.193	0.008
CL <sub>2</sub>	0.127	0.118	0.135	0.127	0.008	0.176	0.176	0.199	0.184	0.013
CL <sub>3</sub>	0.155	0.047	0.128	0.11	0.055	0.164	0.162	0.179	0.168	0.009
CL <sub>4</sub>	0.083	0.116	0.078	0.092	0.020	0.143	0.176	0.155	0.158	0.017
CL <sub>5</sub>	0.070	0.077	0.081	0.077	0.005	0.117	0.120	0.156	0.131	0.021

$$\% \text{ inaktivasi} = \left| \frac{\text{OD}_{\text{kontrol}} - \text{OD}_{\text{perlakuan}}}{\text{OD}_{\text{kontrol}}} \right| \times 100\%$$

### ❖ Untuk LED Biru

- a. Fotosensitizer + LED 2 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.158}{1.440} \right| \times 100\%$$

$$= 89,03 \%$$

- b. Fotosensitizer + LED 4 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.127}{1.440} \right| \times 100\%$$

$$= 91,18\%$$

c. Fotosensitizer + LED 6 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.110}{1.440} \right| \times 100\%$$

$$= 92,36 \%$$

d. Fotosensitizer + LED 8 menit

$$\begin{aligned}\% \text{ inaktivasi} &= \left| \frac{1.440 - 0.092}{1.440} \right| \times 100\% \\ &= 93,61 \%\end{aligned}$$

e. Fotosensitizer + LED 10 menit

$$\begin{aligned}\% \text{ inaktivasi} &= \left| \frac{1.440 - 0.077}{1.440} \right| \times 100\% \\ &= 94,65\%\end{aligned}$$

#### ❖ Untuk LED Merah

a. Fotosensitizer + LED 2 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.193}{1.440} \right| \times 100\%$$

$$= 86,60 \%$$

b. Fotosensitizer + LED 4 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.184}{1.440} \right| \times 100\%$$

$$= 87,20 \%$$

c. Fotosensitizer + LED 6 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.168}{1.440} \right| \times 100\%$$

$$= 88,30 \%$$

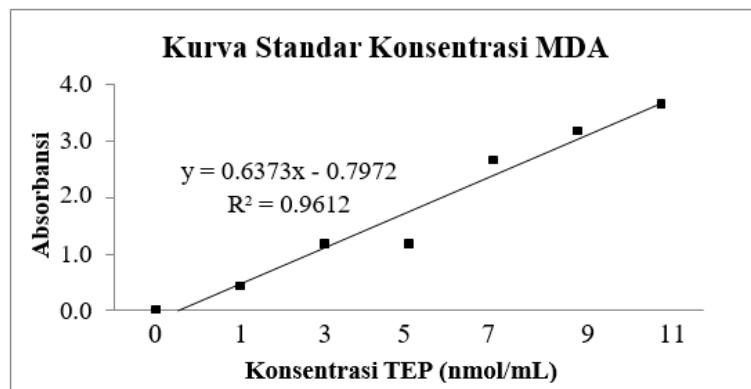
d. Fotosensitizer + LED 8 menit

$$\begin{aligned}\% \text{ inaktivasi} &= \left| \frac{1.440 - 0.158}{1.440} \right| \times 100\% \\ &= 89,03\%\end{aligned}$$

e. Fotosensitizer + LED 10 menit

$$\% \text{ inaktivasi} = \left| \frac{1.440 - 0.131}{1.440} \right| \times 100\% \\ = 90,90 \%$$

Lampiran 4. Perhitungan Kadar MDA



Persamaan yang diperoleh:

$$y = 0,6373x - 0,7972$$

$$x = \frac{y + 0,7972}{0,6373}$$

$$\text{Kadar MDA } (\frac{\text{nmol}}{\text{mL}}) = \frac{y + 0,7972}{0,6373}$$

❖ Untuk Perlakuan Kontrol (C-)

$$\text{Nilai absorbansi} = 0,014$$

$$\text{Kadar MDA } (\frac{\text{nmol}}{\text{mL}}) = \frac{0,014 + 0,7972}{0,6373}$$

$$= 1,272 \text{ nmol/mL}$$

❖ Untuk Perlakuan Kontrol (C+)

$$\text{Nilai absorbansi} = 0,048$$

$$\text{Kadar MDA } (\frac{\text{nmol}}{\text{mL}}) = \frac{0,048 + 0,7972}{0,6373}$$

$$= 1,326 \text{ nmol/mL}$$

A. LED Merah

❖ Untuk LED saja 2 Menit

Nilai absorbansi = 0,055

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,055 + 0,7972}{0,6373}$$
$$= 1,337 \text{ nmol/mL}$$

❖ Untuk LED saja 4 Menit

Nilai absorbansi = 0,057

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,057 + 0,7972}{0,6373}$$
$$= 1,340 \text{ nmol/mL}$$

❖ Untuk LED saja 6 Menit

Nilai absorbansi = 0,060

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,060 + 0,7972}{0,6373}$$
$$= 1,345 \text{ nmol/mL}$$

❖ Untuk LED saja 8 Menit

Nilai absorbansi = 0,062

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,062 + 0,7972}{0,6373}$$
$$= 1,348 \text{ nmol/mL}$$

❖ Untuk LED saja 10 Menit

Nilai absorbansi = 0,064

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,064 + 0,7972}{0,6373}$$
$$= 1,351 \text{ nmol/mL}$$

❖ Untuk LED saja 2 Menit Kombinasi AgNPs

Nilai absorbansi = 0,203

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,203 + 0,7972}{0,6373}$$

$$= 1,569 \text{ nmol/mL}$$

- ❖ Untuk LED saja 4 Menit Kombinasi AgNPs  
Nilai absorbansi = 0,230

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,230 + 0,7972}{0,6373}$$

$$= 1,611 \text{ nmol/mL}$$

- ❖ Untuk LED saja 6 Menit Kombinasi AgNPs  
Nilai absorbansi = 0,278

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,278 + 0,7972}{0,6373}$$

$$= 1,687 \text{ nmol/mL}$$

- ❖ Untuk LED saja 8 Menit Kombinasi AgNPs  
Nilai absorbansi = 0,309

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,309 + 0,7972}{0,6373}$$

$$= 1,735 \text{ nmol/mL}$$

- ❖ Untuk LED saja 10 Menit Kombinasi AgNPs  
Nilai absorbansi = 0,353

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,353 + 0,7972}{0,6373}$$

$$= 1,804 \text{ nmol/mL}$$

## B. LED Biru

- ❖ Untuk LED saja 2 Menit  
Nilai absorbansi = 0,061

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,061 + 0,7972}{0,6373}$$

$$= 1,346 \text{ nmol/mL}$$

❖ Untuk LED saja 4 Menit

Nilai absorbansi = 0,063

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,063 + 0,7972}{0,6373}$$

$$= 1,350 \text{ nmol/mL}$$

❖ Untuk LED saja 6 Menit

Nilai absorbansi = 0,064

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,064 + 0,7972}{0,6373}$$

$$= 1,351 \text{ nmol/mL}$$

❖ Untuk LED saja 8 Menit

Nilai absorbansi = 0,067

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,067 + 0,7972}{0,6373}$$

$$= 1,356 \text{ nmol/mL}$$

❖ Untuk LED saja 10 Menit

Nilai absorbansi = 0,069

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,064 + 0,7972}{0,6373}$$

$$= 1,359 \text{ nmol/mL}$$

❖ Untuk LED saja 2 Menit Kombinasi AgNPs

Nilai absorbansi = 0,281

$$\text{Kadar MDA} \left( \frac{\text{nmol}}{\text{mL}} \right) = \frac{0,281 + 0,7972}{0,6373}$$

$$= 1,692 \text{ nmol/mL}$$

❖ Untuk LED saja 4 Menit Kombinasi AgNPs

Nilai absorbansi = 0,333

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,333 + 0,7972}{0,6373}$$

$$= 1,773 \text{ nmol/mL}$$

❖ Untuk LED saja 6 Menit Kombinasi AgNPs

Nilai absorbansi = 0,365

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,365 + 0,7972}{0,6373}$$

$$= 1,823 \text{ nmol/mL}$$

❖ Untuk LED saja 8 Menit Kombinasi AgNPs

Nilai absorbansi = 0,377

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,377 + 0,7972}{0,6373}$$

$$= 1,842 \text{ nmol/mL}$$

C. Untuk LED saja 10 Menit Kombinasi AgNPs

Nilai absorbansi = 0,380

$$Kadar MDA \left( \frac{nmol}{mL} \right) = \frac{0,380 + 0,7972}{0,6373}$$

$$= 1,847 \text{ nmol/mL}$$

Lampiran 5. Dokumentasi Penelitian

