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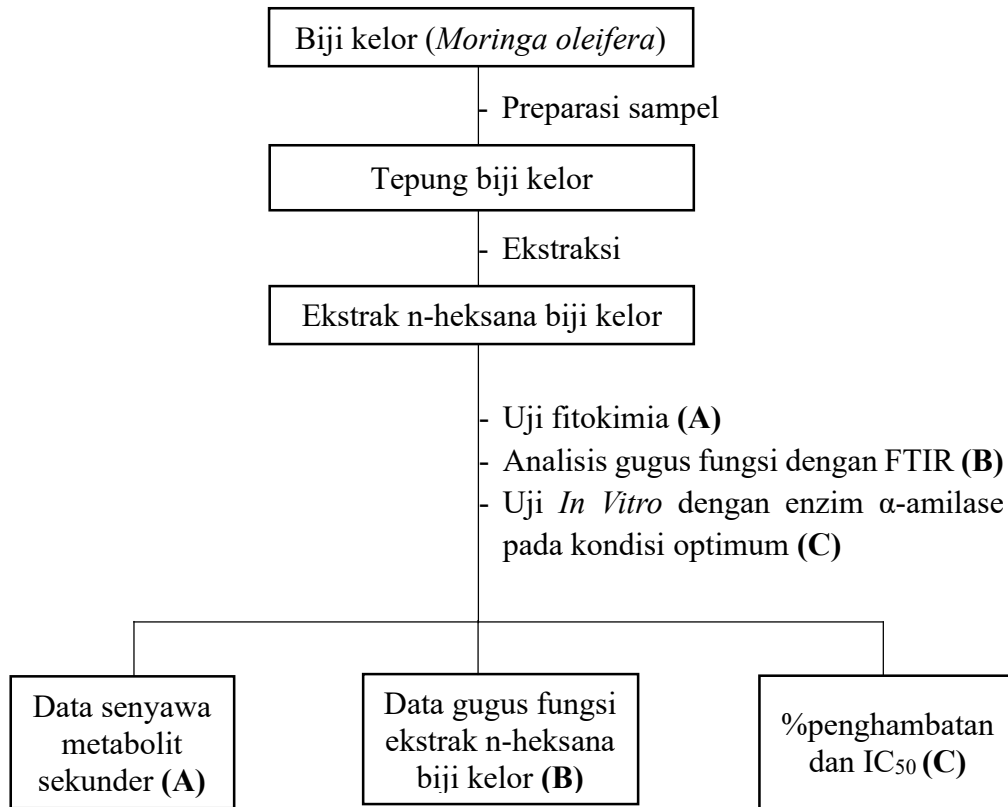
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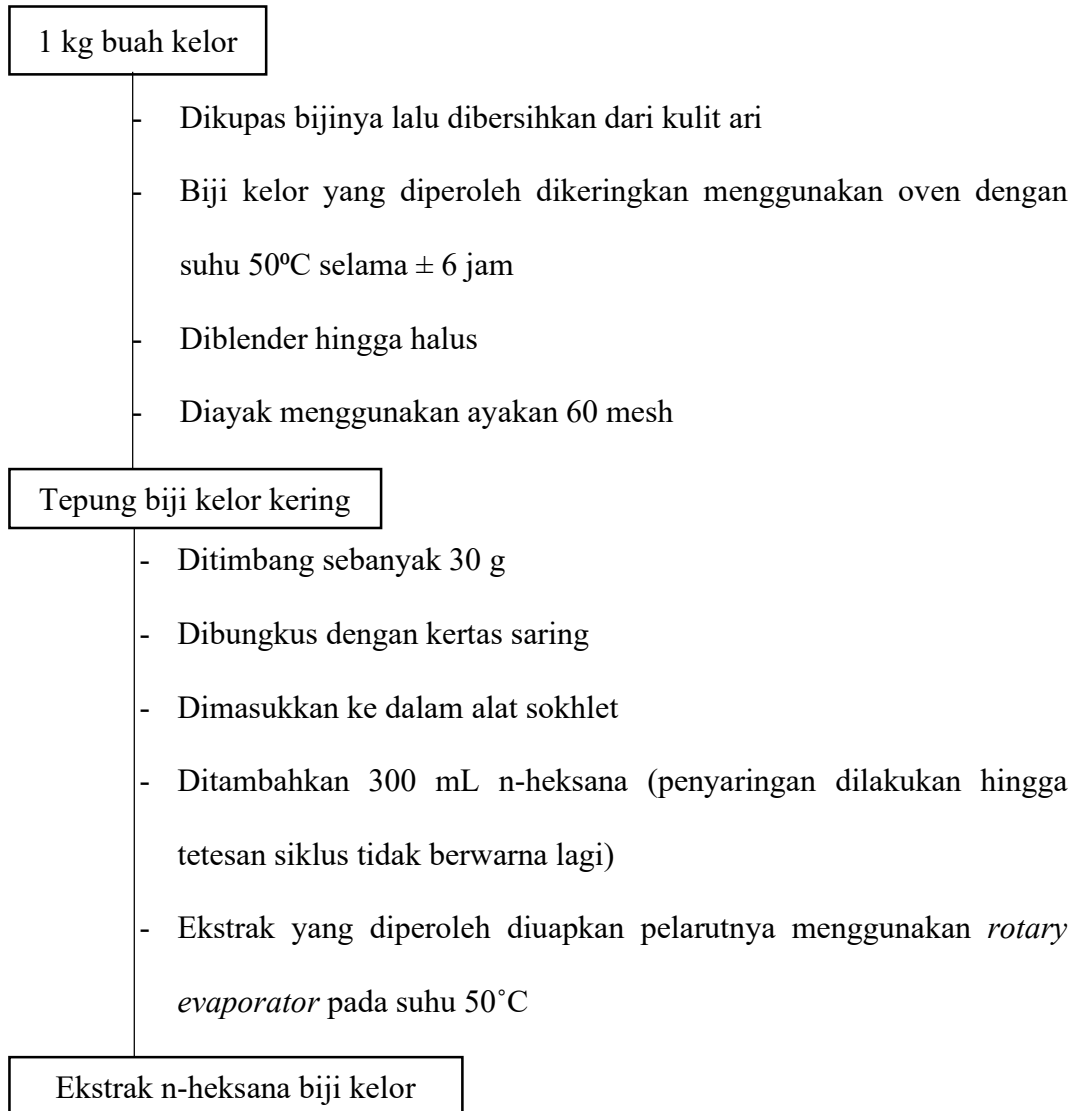
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Lampiran 1. Diagram Alir Penelitian



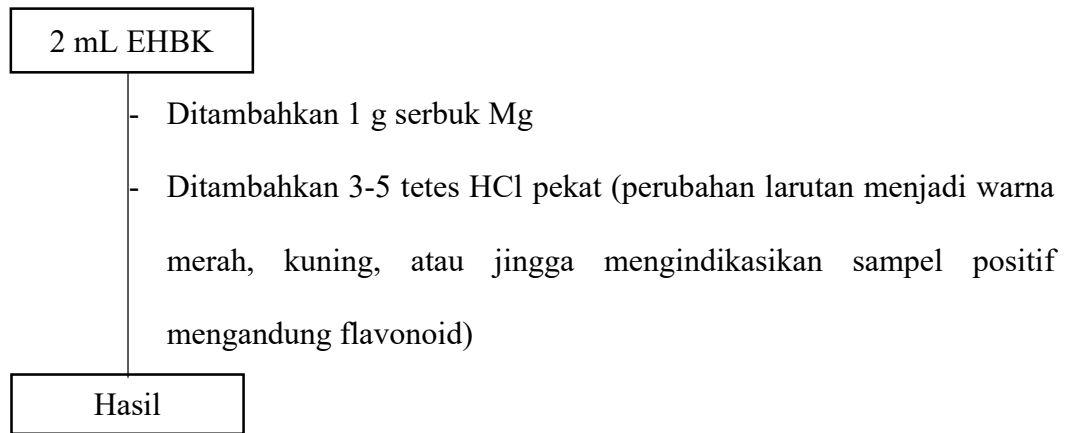
Lampiran 2. Bagan Kerja

1. Preparasi dan Ekstraksi Biji Kelor

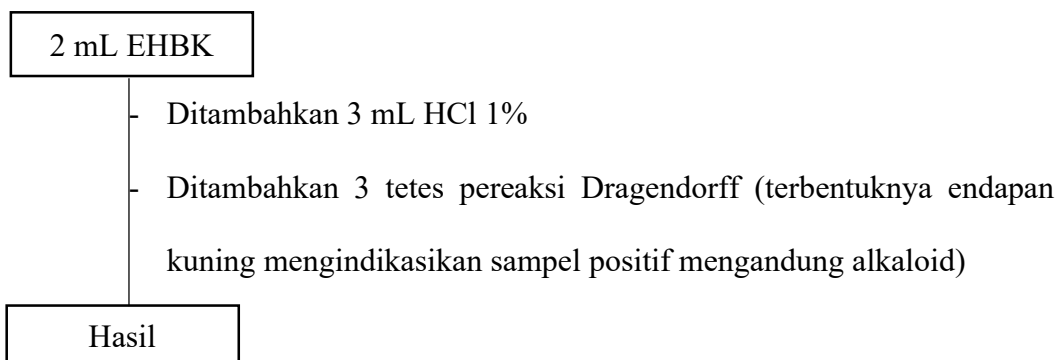


2. Uji Fitokimia

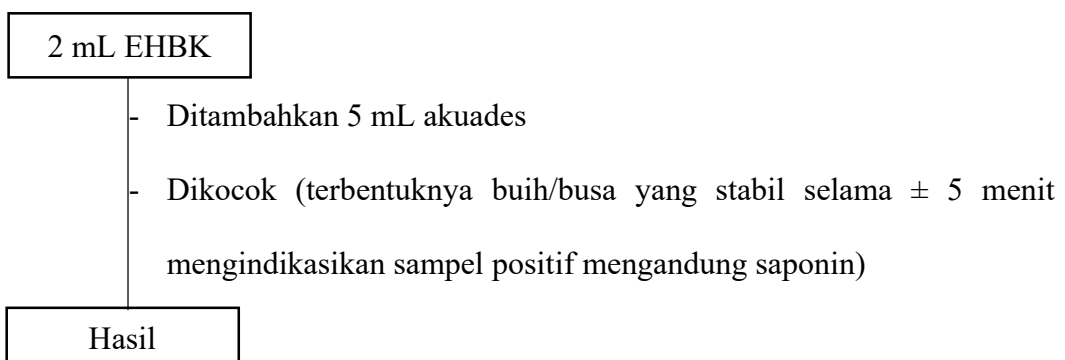
a) Flavonoid



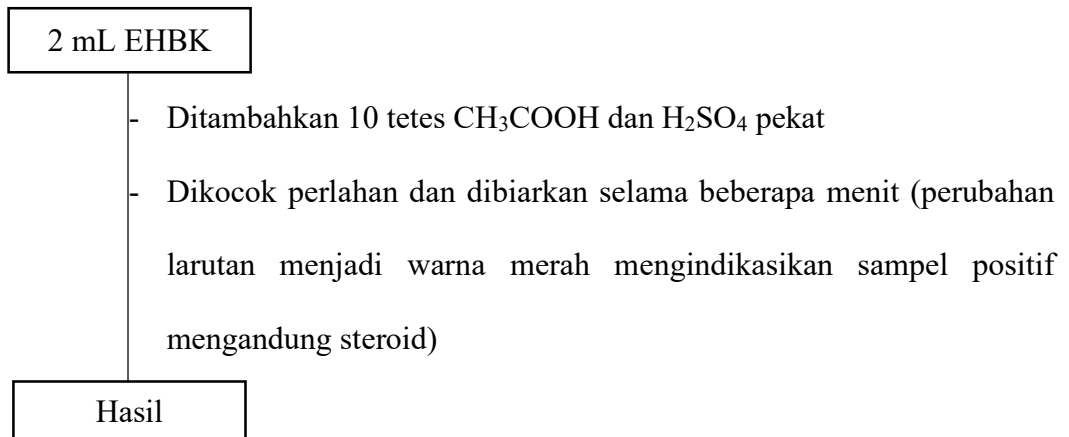
b) Uji Alkaloid



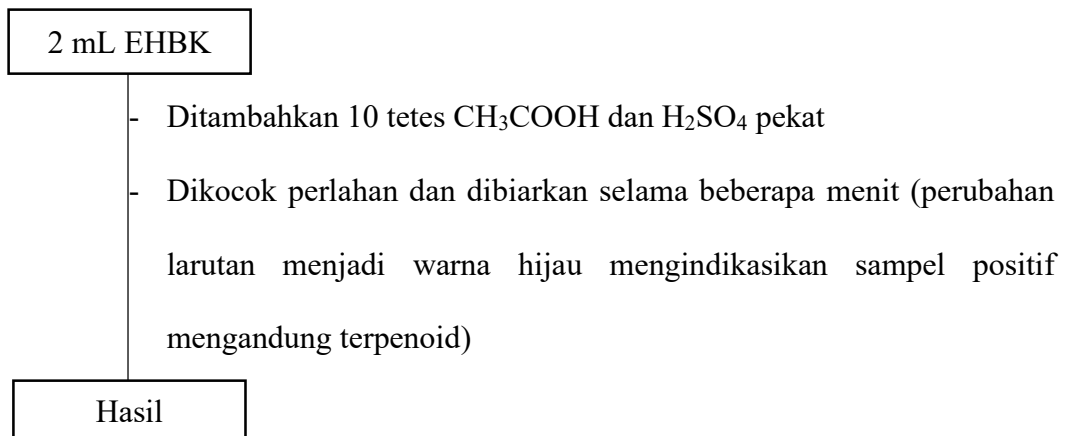
c) Uji Saponin



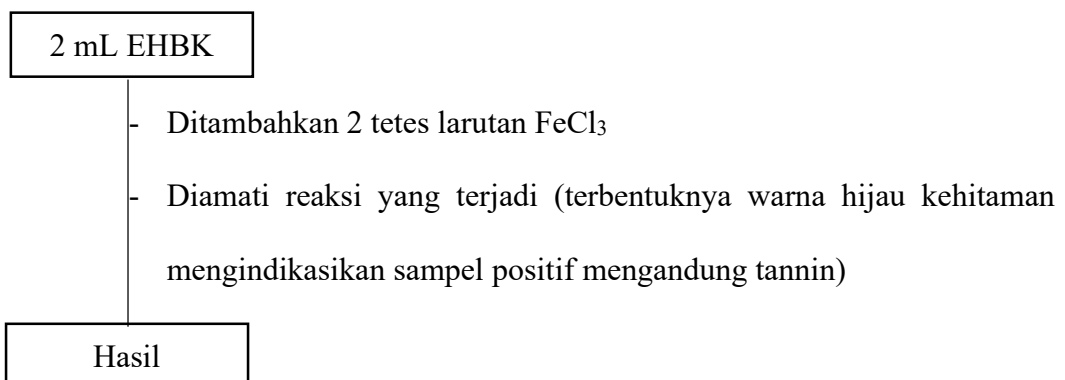
d) Uji Steroid



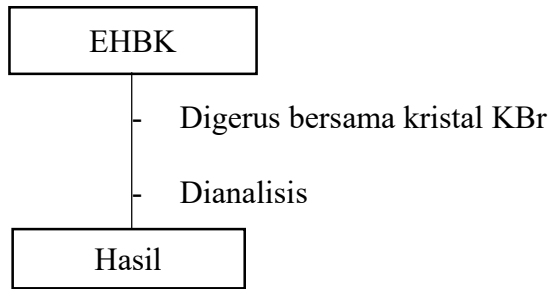
e) Uji Terpenoid



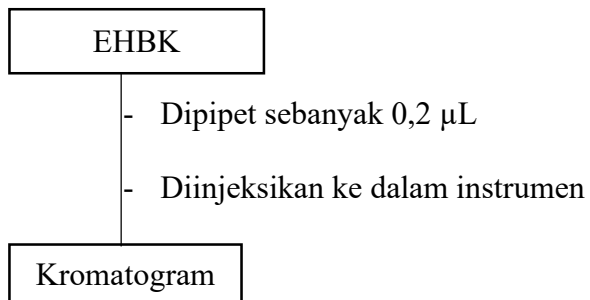
f) Uji Tanin



3. Uji FTIR

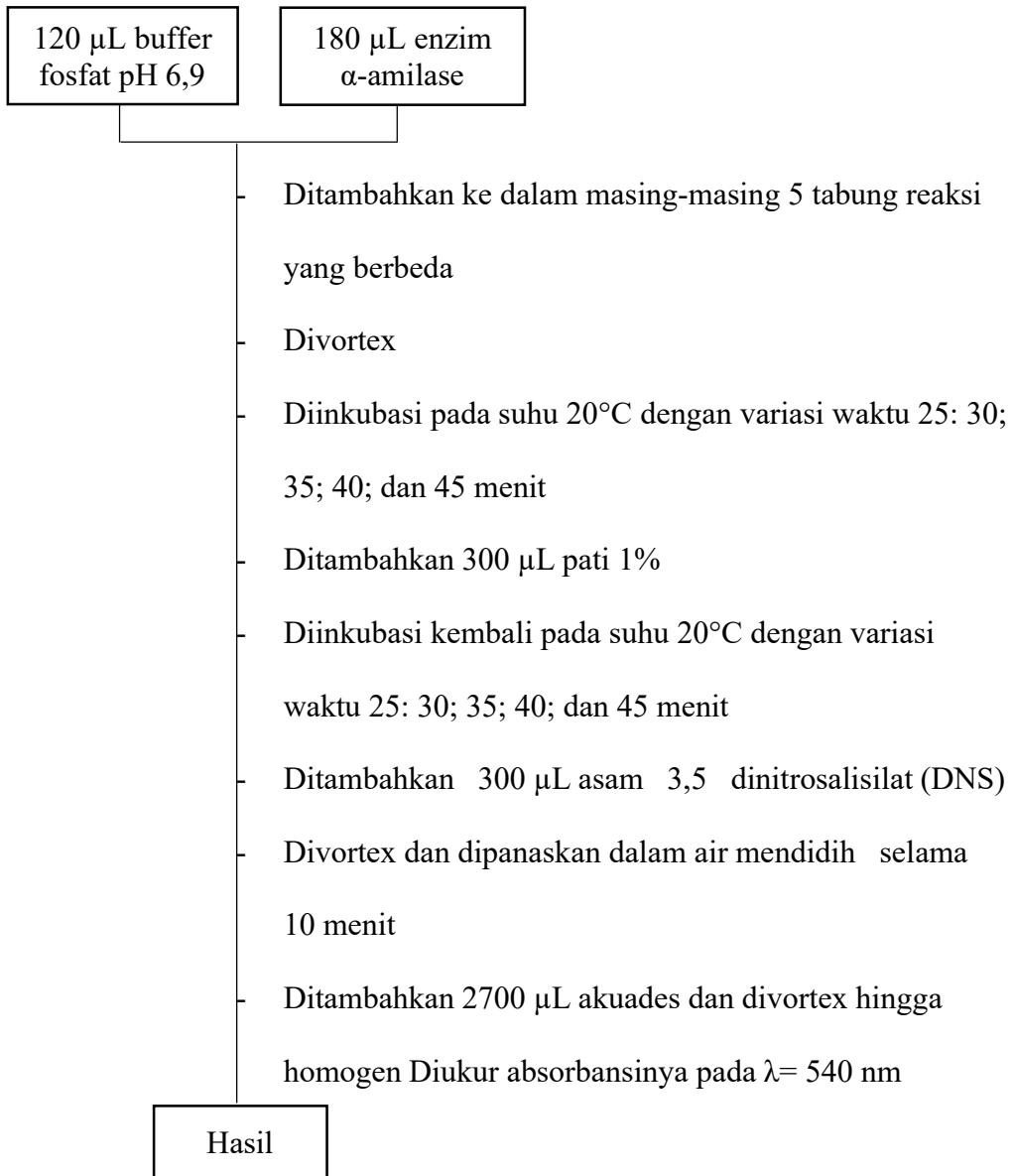


4. Uji GC-MS



5. Uji Aktivitas Enzim

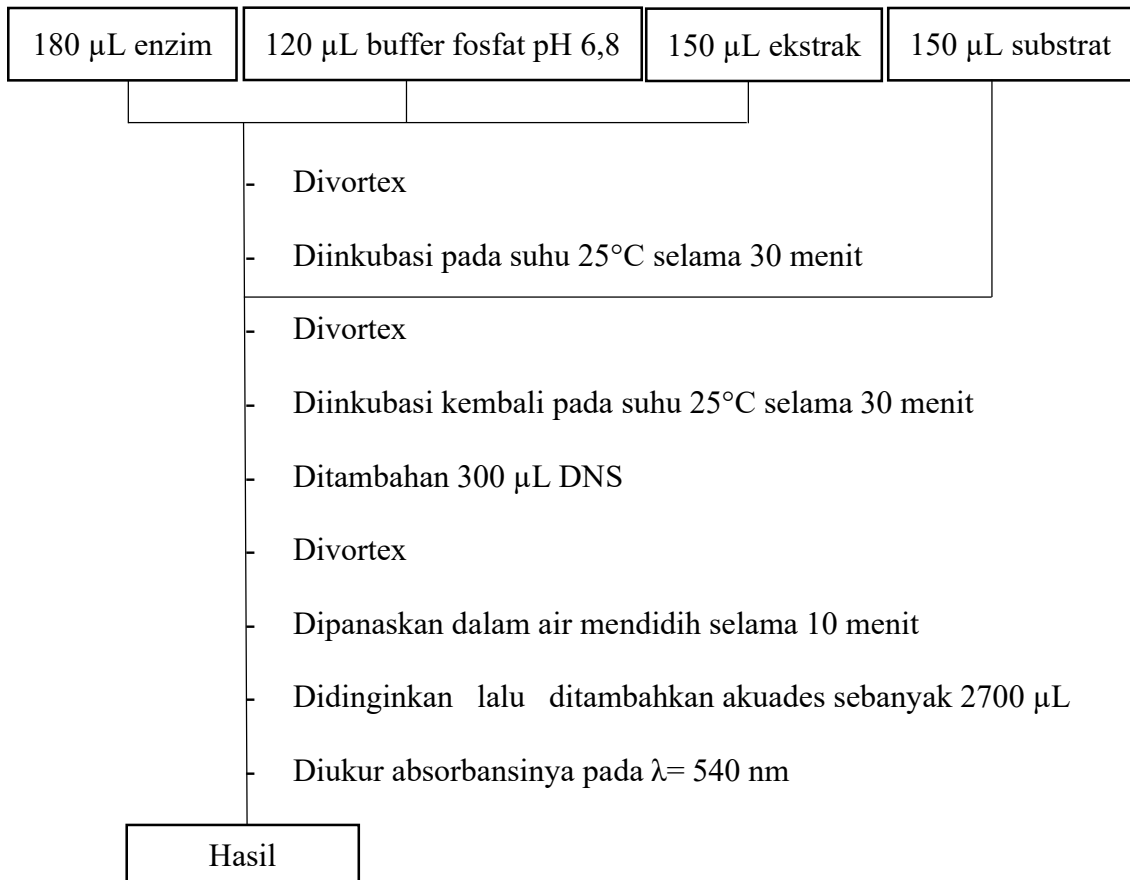
- Penentuan Waktu Optimum Enzim



Catatan:

- Dilakukan prosedur yang sama pada penentuan pH optimum (variasi pH pH 6,4; 6,8; 6,9; 7; dan 7,4) dan suhu optimum (variasi suhu 20°C, 25°C, 30°C, 37°C, dan 40°C).
- Blanko dibuat tanpa penambahan enzim

6. Uji Aktivitas Penghambatan Enzim α -amilase secara *In Vitro*



Lampiran 3. Tempat Pengambilan Sampel



Lampiran 4. Perhitungan Data Penelitian

1. Perhitungan Rendemen Ekstrak n-Heksana Biji Kelor

$$\begin{aligned}\% \text{ rendemen} &= \frac{\text{berat ekstrak (g)}}{\text{berat sampel (g)}} \times 100\% \\ &= \frac{5,09 \text{ g}}{30,01 \text{ g}} \times 100\% \\ &= 16,96\%\end{aligned}$$

2. Pembuatan Larutan

a. Buffer Fosfat

1) Larutan stok

- Ditimbang NaH_2PO_4 sebanyak 0,68995 g lalu dilarutkan ke dalam 25 mL akuades (A)
- Ditimbang Na_2HPO_4 sebanyak 0,44497 g lalu dilarutkan ke dalam 25 mL akuades (B)

x mL larutan stok A + y mL larutan stok B, lalu dicukupkan volumenya hingga 10 mL

x (mL)	y (mL)	pH
3,675	1,325	6,4
2,55	2,75	6,8
2,25	1,325	6,9
1,95	3,05	7
0,95	4,05	7,4

b. Larutan NaOH 2 M dalam 50 mL

$$M = \frac{\text{massa}}{\text{MR}} \times \frac{1000}{V}$$

$$2 \text{ M} = \frac{\text{massa}}{40 \text{ g/mol}} \times \frac{1000}{50 \text{ mL}}$$

$$\text{massa} = 4 \text{ g}$$

$$w = \frac{2,5\%}{100\%} \times 10 \text{ mL}$$
$$= 0,25 \text{ g}$$

c. Larutan enzim α -amilase

- Larutan induk

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100,28 \text{ U/mL} = 10 \text{ mL} \times 20 \text{ U/mL}$$

$$V_1 = 1,994 \text{ mL}$$

$$= 1994 \mu\text{L}$$

- Larutan enzim 0,5 U/mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 20 \text{ U/mL} = 10 \text{ mL} \times 0,5 \text{ U/mL}$$

$$V_1 = 0,4 \text{ mL}$$

$$= 400 \mu\text{L}$$

3. Perhitungan %inhibisi

a. Ekstrak n-heksana biji kelor 5%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,167}{0,340} \times 100\%$$

$$= 55,4\%$$

b. Ekstrak n-heksana biji kelor 10%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,154}{0,340} \times 100\%$$

$$= 58,4\%$$

c. Ekstrak n-heksana biji kelor 15%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,121}{0,121} \times 100\%$$

$$= 67,6\%$$

d. *Acarbose* 5%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,293}{0,340} \times 100\%$$

$$= 13,82\%$$

e. *Acarbose* 10%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,251}{0,340} \times 100\%$$

$$= 26,17\%$$

f. *Acarbose* 15%

$$\%inhibisi = \frac{\text{Absorbansi Kontrol}-\text{Absorbansi Sampel}}{\text{Absorbansi Kontrol}} \times 100\%$$

$$\%inhibisi = \frac{0,340-0,198}{0,340} \times 100\%$$

$$= 41,76\%$$

Lampiran 5. Dokumentasi Penelitian



Sampel biji kelor kering



Proses penghalusan sampel



Biji kelor yang telah halus



Penimbangan biji kelor



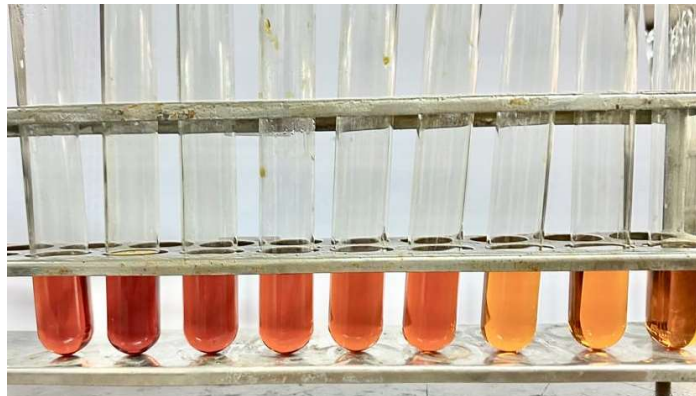
Proses sokhletasi



Proses evaporasi



Deret standar glukosa



Uji aktivitas inhibisi enzim α -amilase dengan ekstrak n-heksana biji kelor