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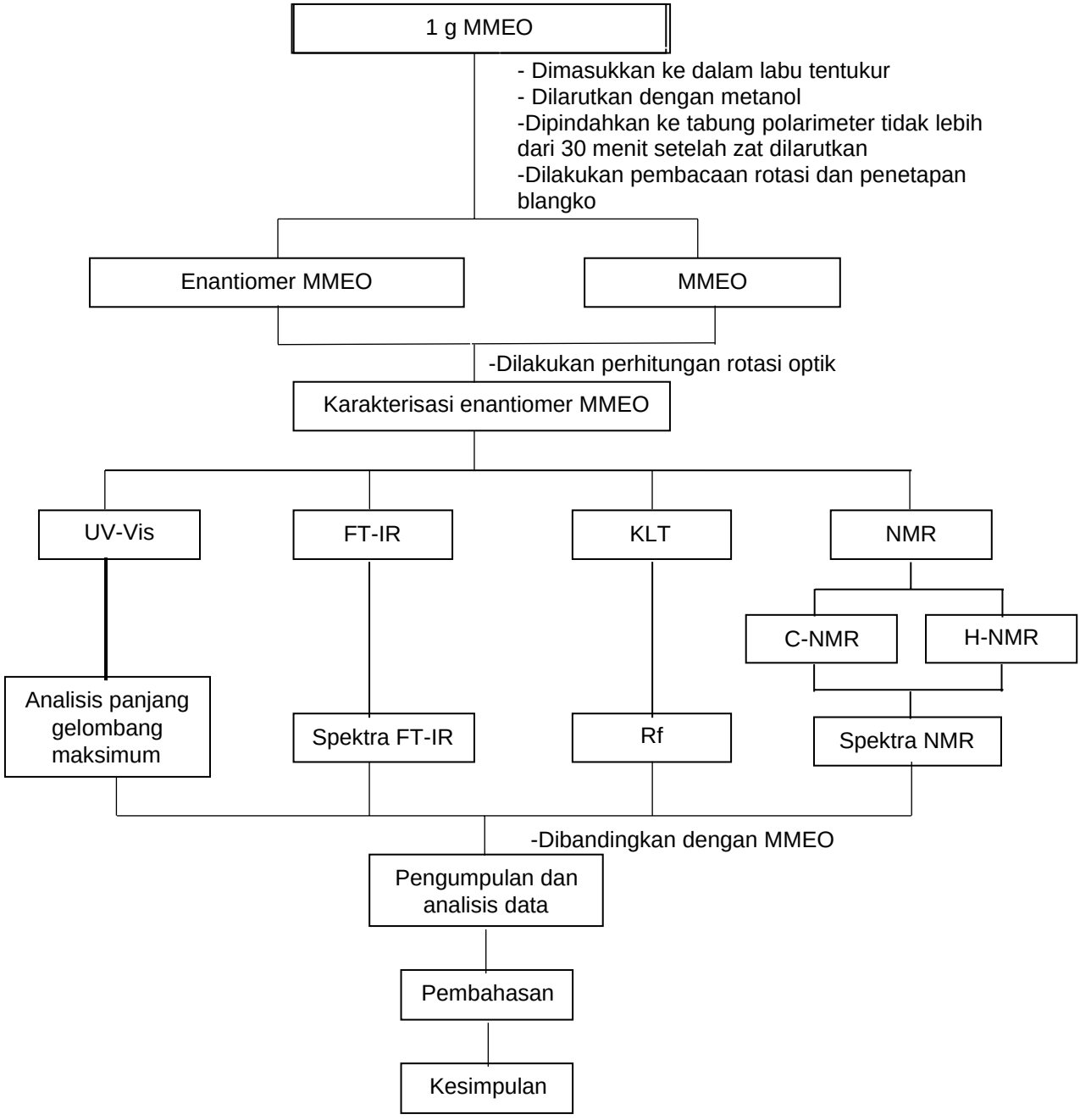
LAMPIRAN

Lampiran 1. Skema Kerja

Enantiomerisasi MMEO



Lampiran 2. Perhitungan Konsentrasi Larutan untuk Spektrofotometri UV-VIS

1. Larutan Stok (500 ppm) dalam labu tentukur 5 mL

1 ppm: 0,001 mg/mL

500 ppm: 0,5 mg/mL: 2,5 mg dalam labu tentukur 5 mL

2. Perhitungan Larutan Standar Internal

25 ppm dalam labu tentukur 10 mL

V1 x N1 = V2 x N2

10 mL x 25 ppm = V2 x 500 ppm

V2 = 0,5 mL = 500 µL

50 ppm dalam labu tentukur 10 mL

V1 x N1 = V2 x N2

10 mL x 50 ppm = V2 x 500 ppm

V2 = 1 mL = 1.000 µL

75 ppm dalam labu tentukur 10 mL

V1 x N1 = V2 x N2

10 mL x 75 ppm = V2 x 500 ppm

V2 = 1,5 mL = 1.500 µL

3. Perhitungan larutan sampel (50 ppm) dalam labu tentukur 5 mL

50 ppm = 0,05 mg/mL

= 0,25 mg pada labu tentukur 5 mL

Lampiran 3. Perhitungan Konstanta Dielektrik KLT

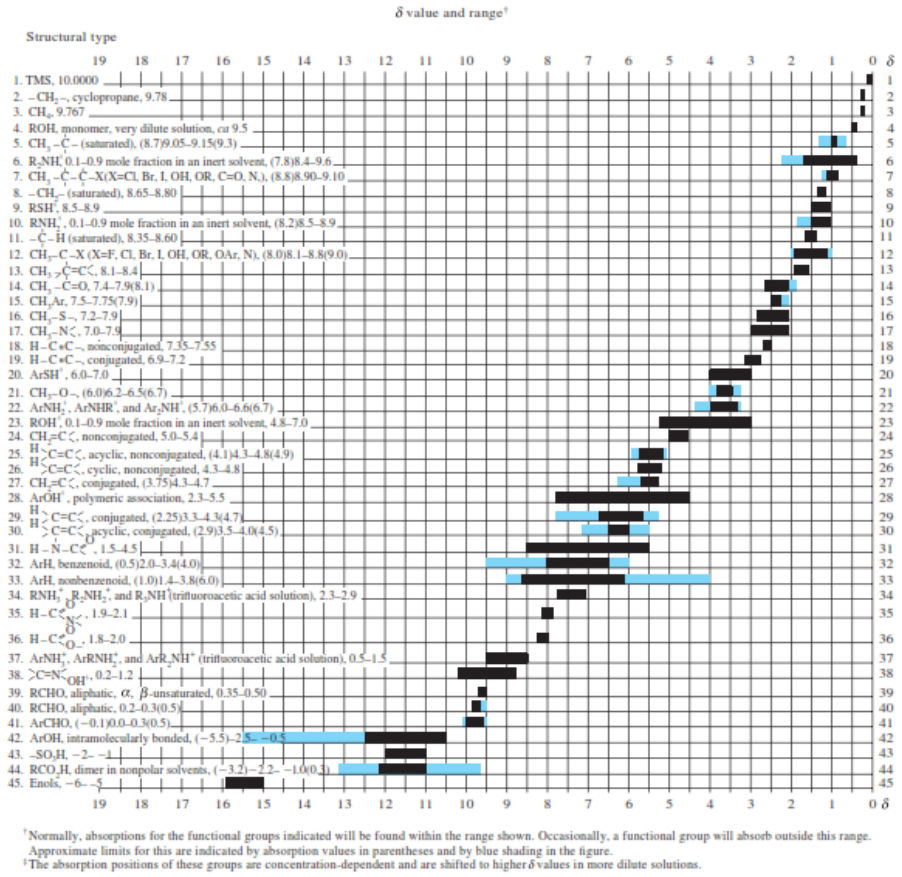
Heksan:Etil asetat = 4:1

$$K_d = \frac{(4 \times 2)+(1 \times 6)}{4+1} = 14:5 = 2,8$$

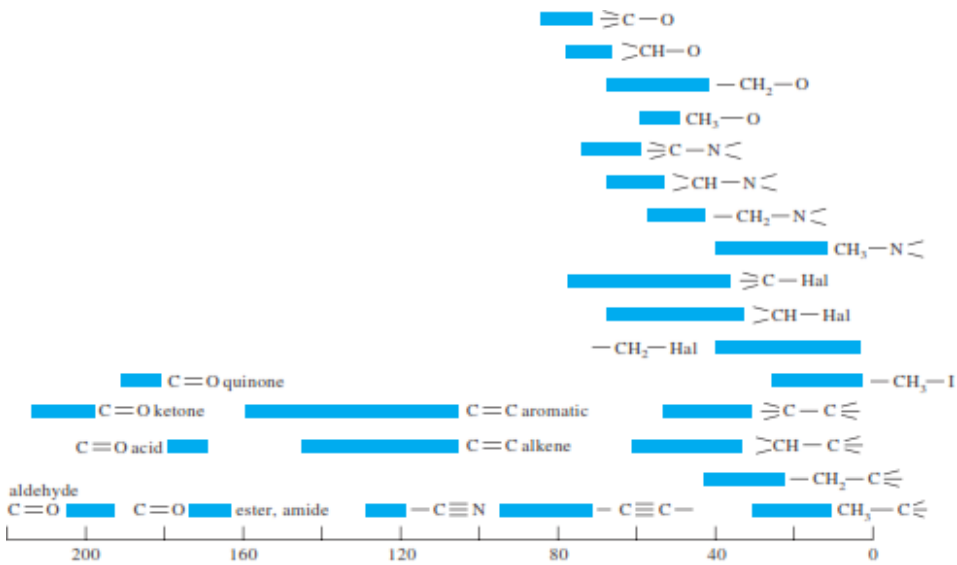
Heksan:Etil asetat 3:5

$$K_d = \frac{(3 \times 2)+(5 \times 6)}{4+1} = 36:5 = 7,2$$

Lampiran 4. Tabel Absorpsi H-NMR dan C-NMR

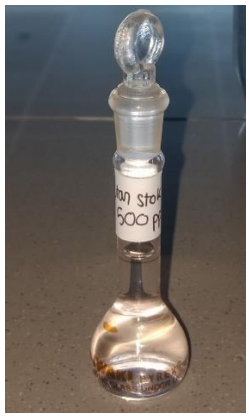


Gambar 16. Tabel pergeseran kimia pada H-NMR (Skoog *et al.*, 2018)



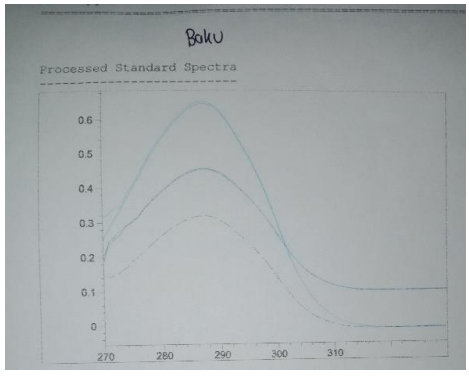
Gambar 17. Skema pergeseran kimia pada C-NMR (Skoog *et al.*, 2018)

Lampiran 6. Dokumentasi Penelitian

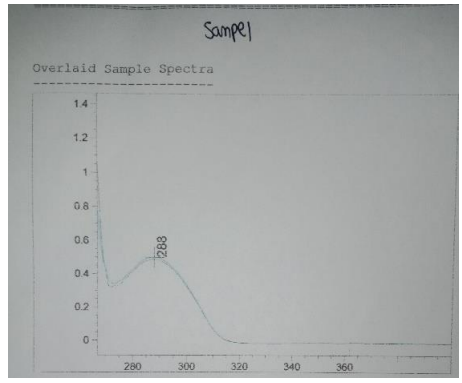


Gambar 18. Larutan stok MMEO sebanyak 500 ppm

Gambar 19. Penimbangan enantiomer MMEO



Gambar 20. Kurva baku MMEO pada spektrofotometri UV-Vis



Gambar 21. Kurva spektrofotometri UV-Vis enantiomer MMEO

Gambar 22. Hasil pengamatan MMEO dan enantiomer MMEO pada KLT dengan perbandingan heksan:etil asetat sebanyak 4:1. (A) MMEO, (B) Enantiomer MMEO

Gambar 23. Hasil pengamatan MMEO dan enantiomer MMEO pada KLT dengan perbandingan heksan:etil asetat sebanyak 3:5. (A) MMEO, (B) Enantiomer MMEO