

## DAFTAR PUSTAKA

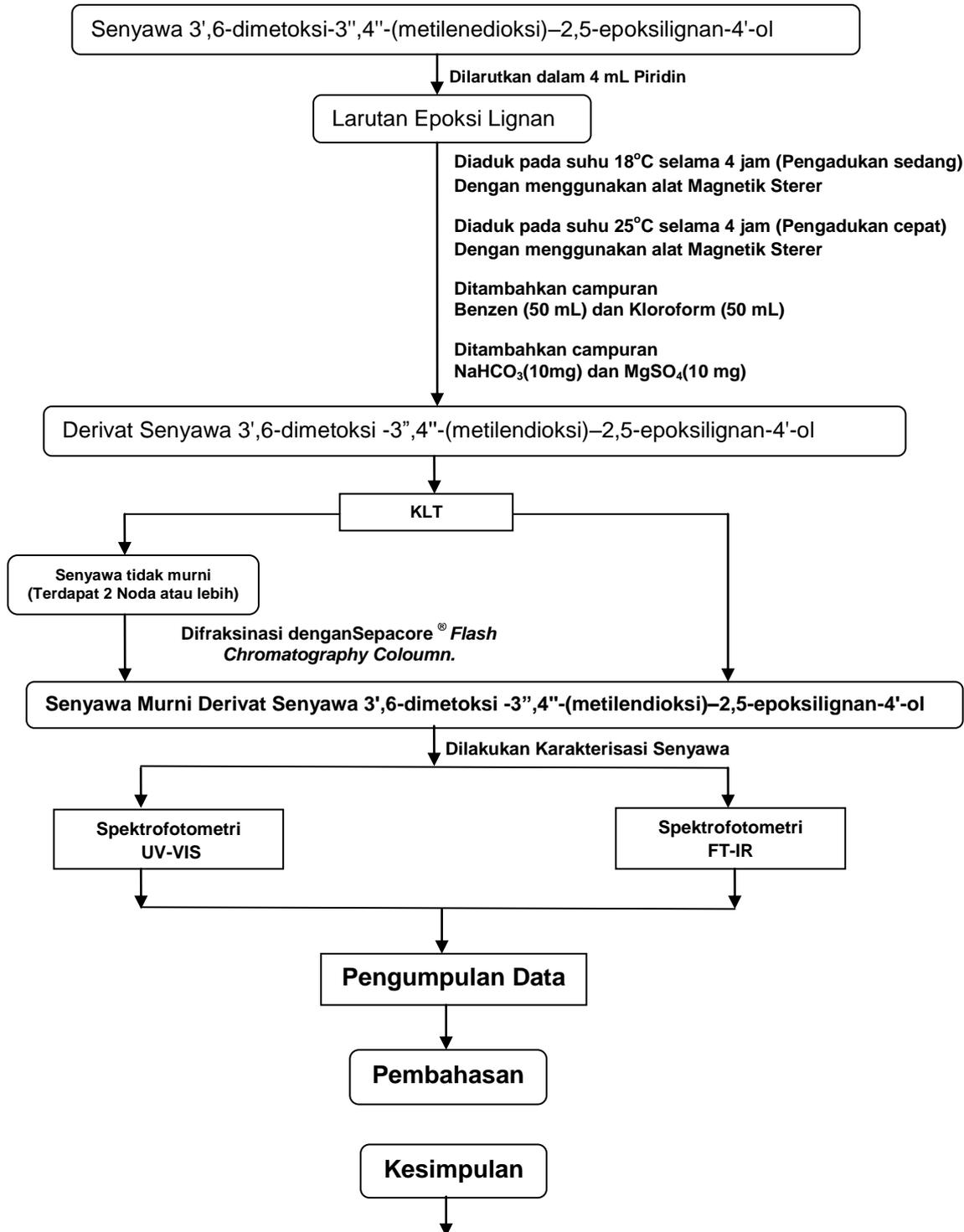
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## LAMPIRAN I

### SKEMA KERJA



## LAMPIRAN II

### PERHITUNGAN RENDAMEN SENYAWA TURUNAN EPOKSILIGNAN

$$\text{Bobot Epoksilignan-4'-ol} = 0,0319 \text{ gram} = 31,9 \text{ mg}$$

$$\text{Massa Molekul Relatif (Mr) Epoksilignan-4'-ol} = 372 \text{ gram / mol}$$

$$\begin{aligned} \text{Mol Epoksilignan 4'-ol} &= \frac{\text{Bobot Epoksilignan 4'-ol}}{\text{Mr Epoksilignan-4'-ol}} \\ &= \frac{0,0031 \text{ gram}}{372 \text{ gram/mol}} \\ &= 8,5753 \times 10^{-5} \text{ mol} \end{aligned}$$

1 mol Epoksilignan 4'-ol setara dengan 1 mol Epoksilignan-4'-ol-6-on.

$$\text{Massa Molekul Relatif (Mr) Epoksilignan-4'-ol-6-on} = 348 \text{ gram / mol}$$

$$\begin{aligned} \text{Bobot Teori Epoksilignan-4'-ol-6-on} &= \text{Mr Epoksilignan-4'-ol-6-on} \times \\ &\quad \text{Mol Epoksilignan-4'-ol-6-on} \\ &= 348 \text{ gram/mol} \times 8,333 \times \\ &\quad 10^{-5} \text{ mol} \\ &= 0,0298 \text{ gram} = 29,8 \text{ mg} \end{aligned}$$

$$\text{Bobot Praktek Epoksilignan-4'-ol-6-on} = 0,0256 \text{ gram} = 25.6 \text{ mg}$$

$$\begin{aligned} \text{Rendamen Epoksilignan-4'ol-6on} &= \frac{\text{Bobot Praktek}}{\text{Bobot Teori}} \times 100\% \\ &= \frac{25.6 \text{ mg}}{29,8 \text{ mg}} \times 100\% \\ &= 85,906 \% \end{aligned}$$