

DAFTAR PUSTAKA

- Amin, I., & Setyorini, D. (2022). PENGARUH VARIASI SOLVENT ORGANIK DALAM PROSES MASERASI KARAGENAN EUCHEUMA COTTONII DI PT BIOTA LAUT GANGGANG PINRANG. *Prosiding Seminar Nasional Teknologi Industri IX, 2022*, 171–175.
- Arikalang TG, Sudewi S, & Rorong JA. (2018). Optimasi Dan Validasi Metode Analisis Dalam Penentuan Kandungan Total Fenolik Pada Ekstrak Daun Gedi Hijau (*Abelmoschus Manihot L.*) Yang Diukur Dengan Spektrofotometer Uv-Vis. *PHARMACONJ Urnal Ilmiah Farmasi-USN RAT*, 7(3), 14–21.
- Bartlová, M., Ziolkowska, D., Pospiech, M., Shyichuk, A., & Tremlová, B. (2021). Determination of carrageenan in jellies with new methylene blue dye using spectrophotometry, smartphone-based colorimetry and spectrophotometric titration. *Food Science and Technology (Brazil)*, 41, 81–90. <https://doi.org/10.1590/fst.01220>
- Blakemore, W. R., Brant, A. F., Bissland, J. G., & Bissland, N. D. (2014). Food Additives & Contaminants : Part A Carrageenan analysis . Part 3 : Quantification in swine plasma. *Food Additives & Contaminants*, 31(10), 1673–1677. <https://doi.org/10.1080/19440049.2014.955538>
- Damayanti, E. T., & Kurniawati, P. (2017). Perbandingan Metode Penentuan Vitamin C pada Minuman Kemasan Menggunakan Metode Spektrofotometer UV-Vis dan Iodimetri. *Universitas Islam Indonesia Journal*, 4(2), 258–266.

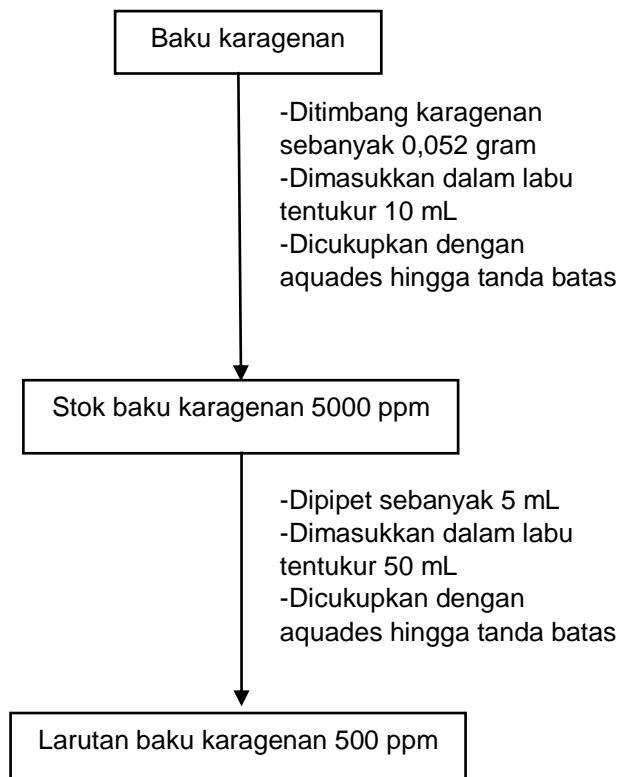
- Fatimah, S. F., Aisyah, V., Nurani, L. H., & Edityaningrum, C. A. (2018). Validasi Metode Analisis β-Karoten Dalam Ekstrak Etanol 96% Spirulina maxima Dengan Spektrofotometri Visibel. *Media Farmasi: Jurnal Ilmu Farmasi*, 15(1), 1. <https://doi.org/10.12928/mf.v15i1.12354>
- Harmita. (2004). Petunjuk Pelaksanaan Validasi dan Cara Penggunaannya. *Majalah Ilmu Kefarmasian*, 1(3), 117.
- Harmono, H. D. (2020). *Validasi Metode Analisis Logam Merkuri (Hg) Terlarut pada Air Permukaan dengan Automatic Mercury Analyzer*. 2(3), 11–16.
- Isrul, M., Dewi, C., & Wahdini, V. (2020). Uji Efek Antiinflamasi Infusa Daun Bayam Merah (*Amaranthus tricolor L.*) Terhadap Tikus Putih (*Rattus norvegicus*) Yang Diinduksi Karagenan. *Jurnal Mandala Pharmacon Indonesia*, 6(2), 97–103. <https://doi.org/10.35311/jmpi.v6i1.61>
- Jia, E., & Bartlett, M. G. (2020). Recent advances in liquid chromatographic methods for the determination of selective serotonin reuptake inhibitors and serotonin norepinephrine reuptake inhibitors. *Biomedical Chromatography*, 34(3). <https://doi.org/10.1002/bmc.4760>
- Khaldun, I. (2018). *Kimia Analisa Instrumen*. Syiah Kuala University Press.
- Lapwanit, S., Sooksimuang, T., & Trakulsujaritchok, T. (2018). Adsorptive removal of cationic methylene blue dye by kappa-carrageenan/poly(glycidyl methacrylate) hydrogel beads: Preparation and characterization. *Journal of Environmental Chemical Engineering*,

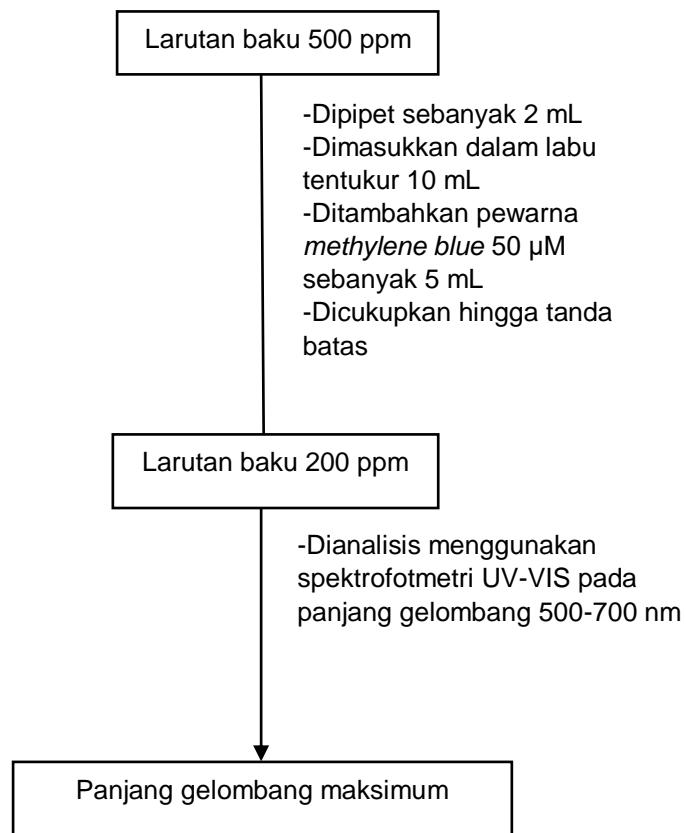
- 6(5), 6221–6230. <https://doi.org/10.1016/j.jece.2018.09.050>
- Ling, Y. P., & Heng, L. Y. (2013). Complexation between carrageenan and methylene blue for sensor design. *AIP Conference Proceedings*, 1571, 717–724. <https://doi.org/10.1063/1.4858739>
- Mittal, H., Al Alili, A., & Alhassan, S. M. (2020). High efficiency removal of methylene blue dye using κ-carrageenan-poly(acrylamide-co-methacrylic acid)/AQSOA-Z05 zeolite hydrogel composites. *Cellulose*, 27(14), 8269–8285. <https://doi.org/10.1007/s10570-020-03365-6>
- Moelyono, M. (2016). *Farmasi Bahari*. Deepublish.
- Mulyati, A. H., Sutanto, & Apriyani, D. (2011). VALIDASI METODE ANALISIS KADAR AMBROKSOL HIDROKLORIDA DALAM SEDIAAN TABLET CYSTELIS® SECARA KROMATOGRAFI CAIR KINERJA TINGGI. *International Journal*, 66(2), 137–144.
- Musiam, S., & Alfian, R. (2017). Validasi Metode Spektrofotometri UV pada Analisis Penetapan Kadar Asam Mefenamat dalam Sediaan Tablet Generik. *Jurnal Ilmiah Ibnu Sina*, 2(1), 31–43.
- Nazar, M. (2018). *Spektroskopi Molekul*. Syiah Kuala University Press.
- Rohman, A. (2014). *Validasi dan Penjaminan Mutu Metode Analisis Kimia*. Gadjah Mada University Press.
- Sahriwati, Sumarlin, & Wahyuni, S. (2020). Validasi Metode dan Penetapan Kadar Kolesterol Ayam Broiler. *Lutjanus*, 31–40.
- Sastrohamidjojo, H. (2013). *Dasar-Dasar Spektroskopi*. Gadjah Mada University Press.

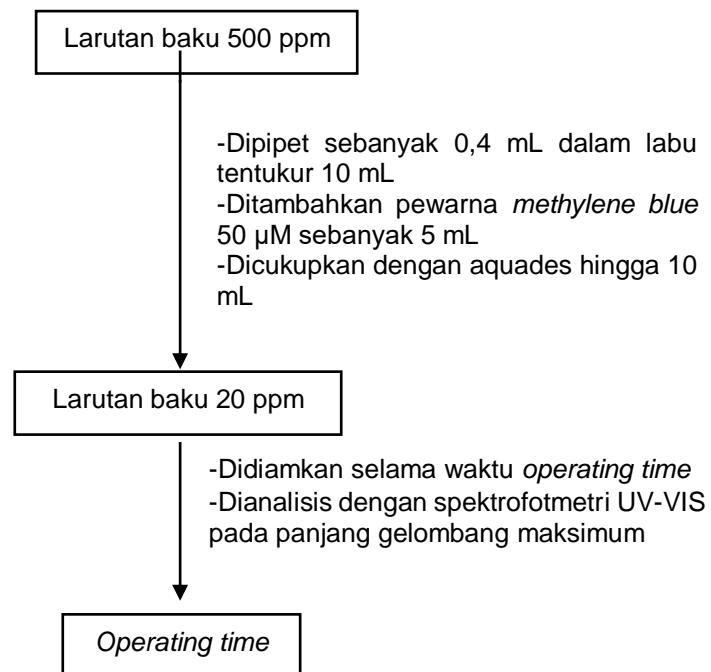
- Sinurat, E., Murdinah, & Fransiska, D. (2010). Karakterisasi Permen Jeli yang Dibuat dari Hasil F.pdf. In *Jurnal Pascapanen dan Bioteknologi Kelautan dan Perikanan* (Vol. 5, Issue 1, pp. 57–64).
- Soedjak, H. S. (1994). Colorimetric Determination of Carrageenans and Other Anionic Hydrocolloids with Methylene Blue. *Analytical Chemistry*, 66(24), 4514–4518. <https://doi.org/10.1021/ac00096a018>
- Sukmawati, Sudewi, S., & Pontoh, J. (2018). *PENENTUAN KANDUNGAN TOTAL FLAVONOID PADA EKSTRAK DAUN GEDI HIJAU (Abelmoscus manihot L.) YANG DIUKUR MENGGUNAKAN SPEKTROFOTOMETER UV-VIS.* 7(3), 32–41.
- Sun, Y., Zhu, X., Shen, X., & Wang, W. (2021). Determination of Carrageenan in Livestock and Poultry Meat by Ultrahigh-Performance Liquid Chromatography-Tandem Mass Spectrometry. *International Journal of Analytical Chemistry*, 2021. <https://doi.org/10.1155/2021/5277453>
- Suseno, D. (2021). *Validasi Metode Analisis Formalin dan Aplikasinya Pada Ikan Asin.* 7(1), 173–182.
- Wikanta'r, T., & Kurniawan, R. (2005). PE NGARUH PEMERIAN-KARAG I NAN DAN T.KARAG I NAN TERHADAP PENURUNAN KADAR GLUKOSA DARAH DAN HISTOPATOLOGI USUS KELI NCI. *Jurnal Penelitian Perikanan Indonesia*, 11, 57–68.
- Winarno, F. G., & Winarno, S. A. A. (2017). *Gastronomi Molekuler.*

Kompas Gramedia.

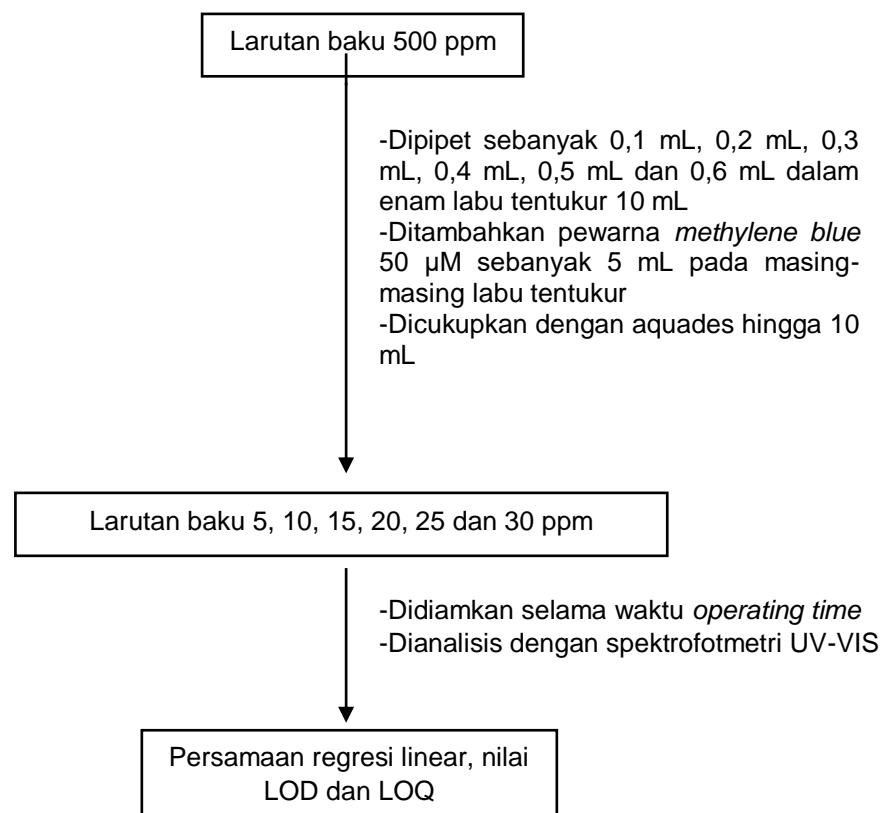
Younes, M., Aggett, P., Aguilar, F., Crebelli, R., Filipič, M., Frutos, M. J., Galtier, P., Gott, D., Gundert-Remy, U., Kuhnle, G. G., Lambré, C., Leblanc, J. C., Lillegaard, I. T., Moldeus, P., Mortensen, A., Oskarsson, A., Stankovic, I., Waalkens-Berendsen, I., Woutersen, R. A., ... Dusemund, B. (2018). Re-evaluation of carrageenan (E 407) and processed Eucheuma seaweed (E 407a) as food additives. *EFSA Journal*, 16(4). <https://doi.org/10.2903/j.efsa.2018.5238>

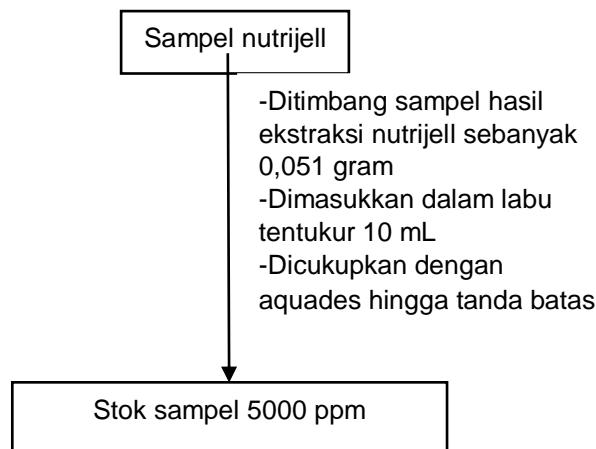
Lampiran 1. Skema kerja pembuatan larutan baku karagenan

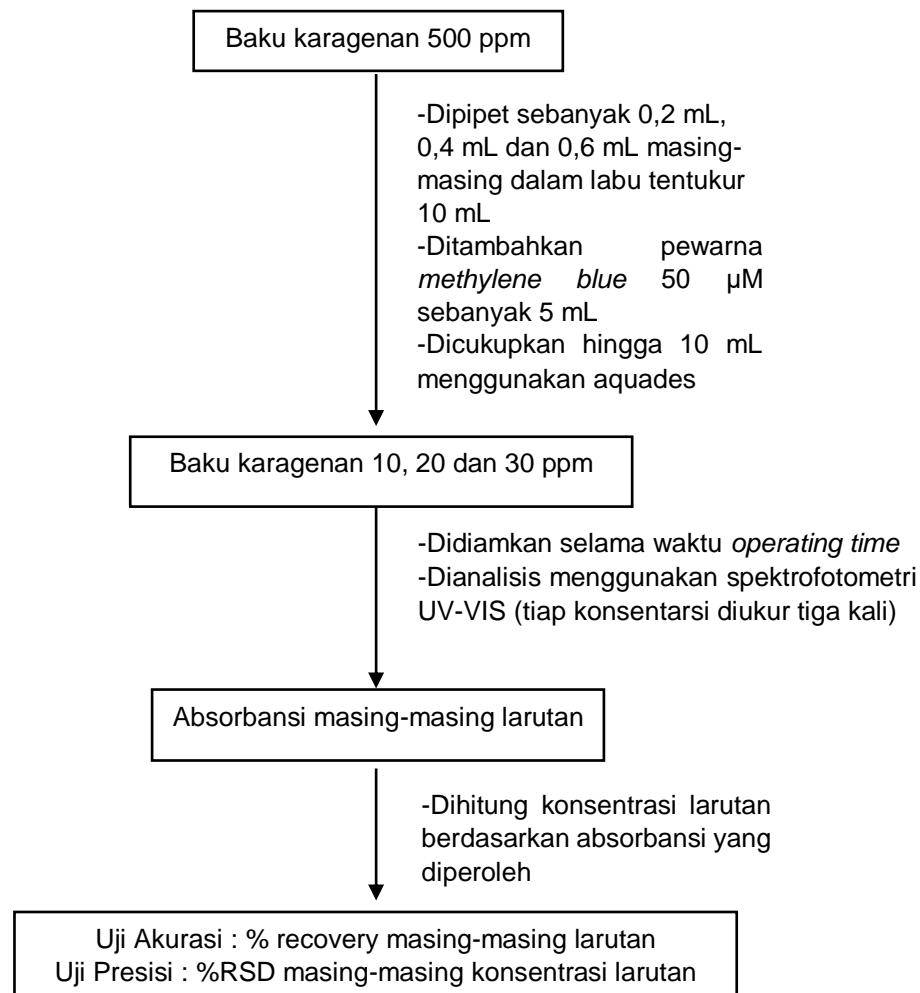
Lampiran 2. Skema kerja penentuan panjang gelombang maksimum

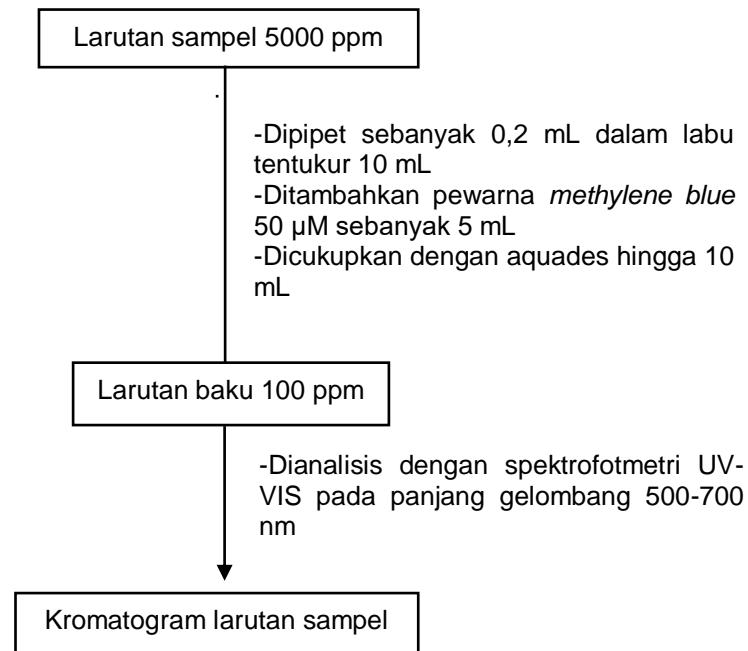
Lampiran 3. Skema kerja penetapan *operating time*

Lampiran 4. Skema kerja pembuatan kurva baku, uji linearitas, LOD & LOQ

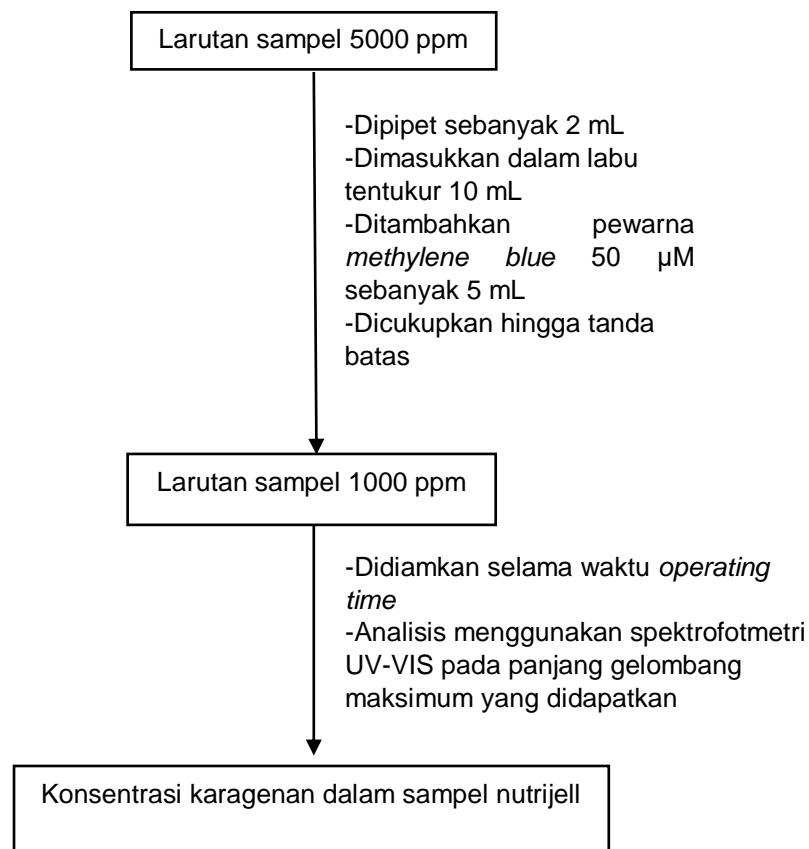


Lampiran 5. Skema kerja pembuatan larutan sampel

Lampiran 6. Skema kerja uji akurasi dan presisi

Lampiran 7. Skema kerja uji selektivitas

Lampiran 8. Skema kerja penetapan kadar karagenan dalam produk nutrijell



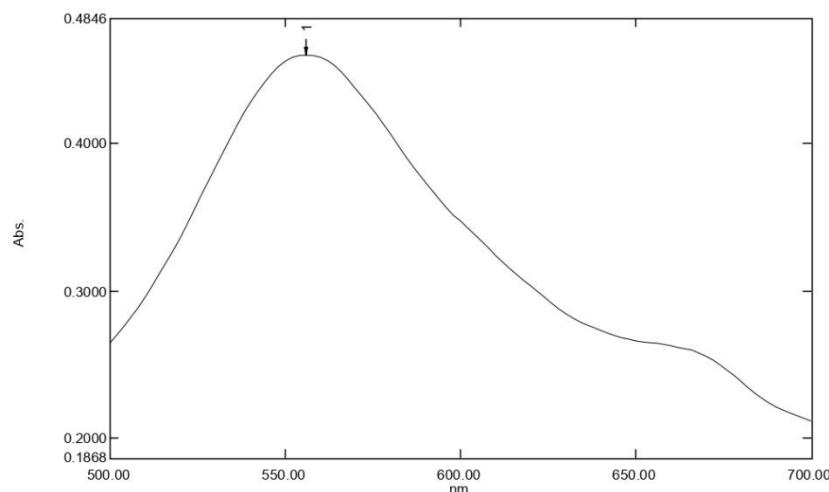
Lampiran 9. Hasil analisis spektrofotometri UV-VIS

a. Panjang gelombang maksimum

Spectrum Peak Pick Report

13/11/2023 16:07:26

Data Set: kur pip_160627 - RawData



No.	P/V	Wavelength	Abs.	Description
1	●	556.00	0.4598	

[Measurement Properties]
Wavelength Range (nm.): 500.00 to 700.00
Scan Speed: Medium
Sampling Interval: 2.0
Auto Sampling Interval: Disabled
Scan Mode: Auto

[Instrument Properties]
Instrument Type: UV-1780 Series
Measuring Mode: Absorbance
Slit Width: 1.0 nm
Light Source Change Wavelength: 340.8 nm
S/R Exchange: Normal

[Attachment Properties]
Attachment: None

[Operation]
Threshold: 0.0010000
Points: 4
Interpolate: Disabled
Average: Disabled

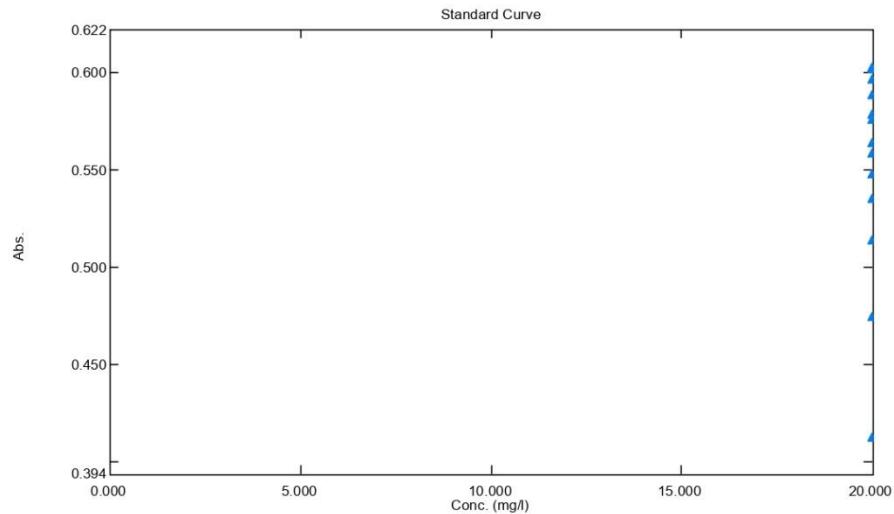
[Sample Preparation Properties]
Weight:
Volume:
Dilution:
Path Length:
Additional Information: DNA

b. Penetapan *operating time*

Standard Table Report

27/11/2023 17:58:21

File Name: C:\Users\user\Documents\PENELITIAN S1\001-KF-FFUH\operating time 6.pho



Standard Table

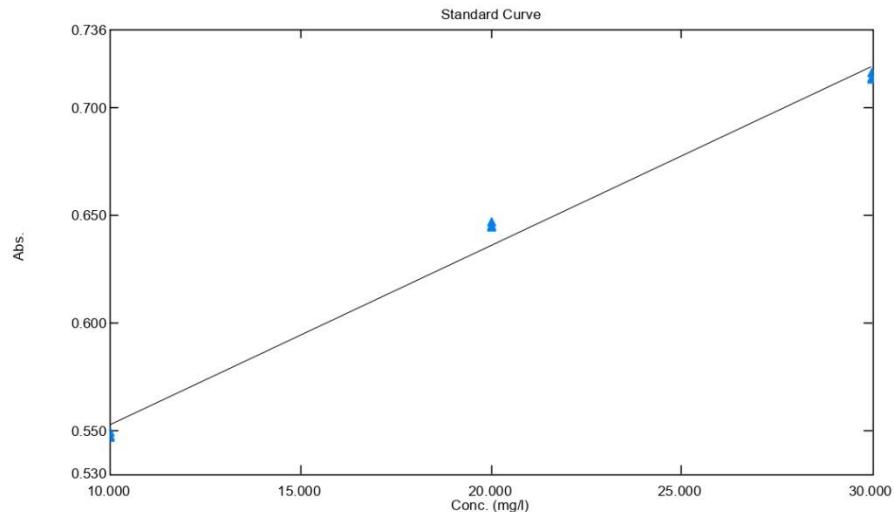
	Sample ID	Type	Ex	Conc	WL556.0	Wgt.Factor	Comments
1	10min	Standard		20.000	0.413	1.000	
2	20min	Standard		20.000	0.475	1.000	
3	30min	Standard		20.000	0.515	1.000	
4	40	Standard		20.000	0.535	1.000	
5	50	Standard		20.000	0.548	1.000	
6	60	Standard		20.000	0.559	1.000	
7	70	Standard		20.000	0.564	1.000	
8	80	Standard		20.000	0.576	1.000	
9	90	Standard		20.000	0.579	1.000	
10	100	Standard		20.000	0.589	1.000	
11	110	Standard		20.000	0.597	1.000	
12	120	Standard		20.000	0.603	1.000	
13	130	Standard		20.000	0.603	1.000	
14	140	Standard		20.000	0.602	1.000	
15							

c. Akurasi dan presisi

Standard Table Report

01/12/2023 15:38:34

File Name: C:\Users\user\Documents\PENELITIAN S1\001-KF-FFUH\akurasi presisi.pho



Standard Table

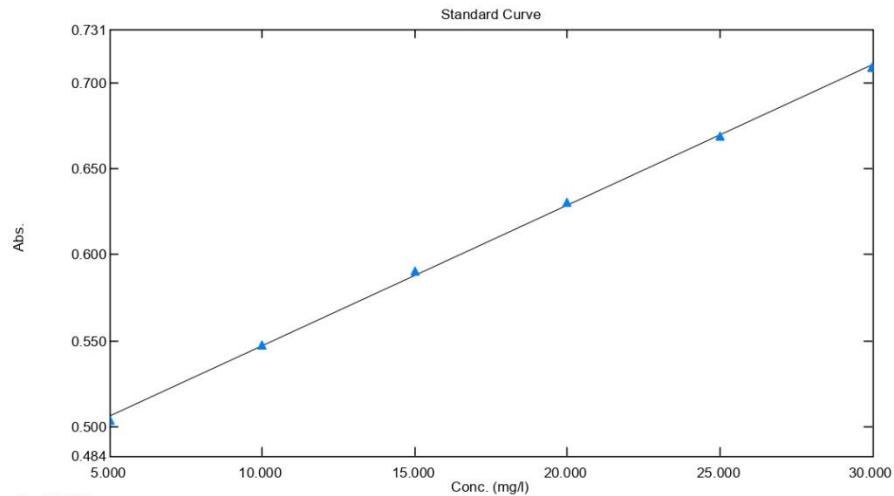
	Sample ID	Type	Ex	Conc	WL556.0	Wgt.Factor	Comments
1	baku11	Standard		10.000	0.548	1.000	
2	baku13	Standard		10.000	0.548	1.000	
3	baku12.	Standard		10.000	0.550	1.000	
4	baku23.	Standard		20.000	0.644	1.000	
5	baku22.	Standard		20.000	0.645	1.000	
6	baku21.	Standard		20.000	0.647	1.000	
7	baku32.	Standard		30.000	0.716	1.000	
8	baku33.	Standard		30.000	0.713	1.000	
9	baku31.	Standard		30.000	0.713	1.000	
10							

d. Linearitas, LOD dan LOQ

Standard Table Report

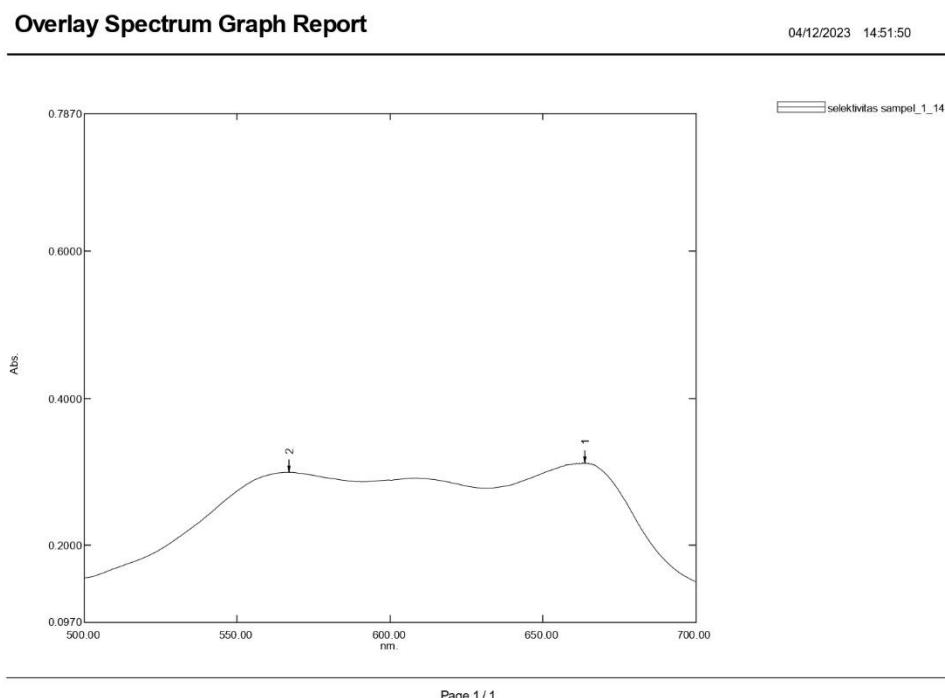
30/11/2023 13:38:15

File Name: C:\Users\user\Documents\PENELITIAN S1\001-KF-FFUH\kurva baku3.pho



Standard Table

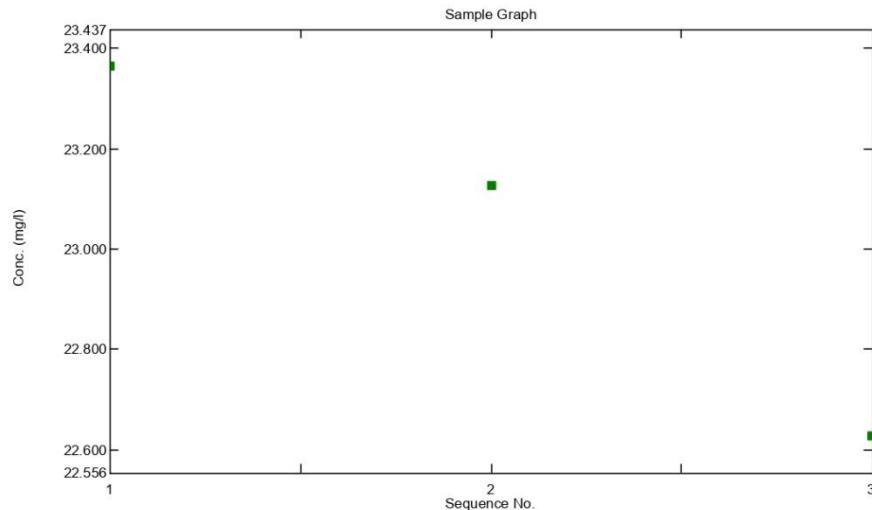
	Sample ID	Type	Ex	Conc	WL556.0	Wgt.Factor	Comments
1	baku2	Standard		10.000	0.548	1.000	
2	baku1	Standard		5.000	0.504	1.000	
3	baku5..	Standard		25.000	0.669	1.000	
4	baku6..	Standard		30.000	0.709	1.000	
5	baku3..	Standard		15.000	0.591	1.000	
6	baku4..	Standard		20.000	0.631	1.000	
7							

e. Selektivitas sampel

f. Penetapan kadar karagenan dalam produk nutrijell**Sample Table Report**

04/12/2023 14:52:51

File Name: C:\Users\user\Documents\PENELITIAN S1\001-KF-FFUH\kurva baku3.pho



Sample Table

	Sample ID	Type	Ex	Conc	WL556.0	Comments
1	sampel5	Unknown		23.364	0.656	
2	52	Unknown		23.128	0.654	
3	53	Unknown		22.629	0.650	
4						

Lampiran 10. Perhitungan

a. Akurasi

Konsentrasi 10 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,548 - 0,4659)}{0,0082} = 10,01 \text{ ppm}$$

$$\% \text{recovery} = \frac{10,01}{10} \times 100\% = 100\%$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,550 - 0,4659)}{0,0082} = 10,26 \text{ ppm}$$

$$\% \text{recovery} = \frac{10,26}{10} \times 100\% = 103\%$$

Replikasi 3

$$\text{Konsentrasi} = \frac{(0,548 - 0,4659)}{0,0082} = 10,01 \text{ ppm}$$

$$\% \text{recovery} = \frac{10,01}{10} \times 100\% = 100\%$$

$$\text{Rata - rata \% recovery} = \frac{100\% + 103\% + 100\%}{3} = 101\%$$

Konsentrasi 20 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,644 - 0,4659)}{0,0082} = 21,72 \text{ ppm}$$

$$\% \text{recovery} = \frac{21,72}{20} \times 100\% = 109\%$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,647 - 0,4659)}{0,0082} = 22,09 \text{ ppm}$$

$$\% \text{recovery} = \frac{22,09}{20} \times 100\% = 110\%$$

Replikasi 3

$$\text{Konsentrasi} = \frac{(0,645 - 0,4659)}{0,0082} = 21,84 \text{ ppm}$$

$$\% \text{recovery} = \frac{21,84}{20} \times 100\% = 109\%$$

$$\text{Rata - rata \% recovery} = \frac{109\% + 110\% + 109\%}{3} = 109\%$$

Konsentrasi 30 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,713-0,4659)}{0,0082} = 30,13 \text{ ppm}$$

$$\% \text{recovery} = \frac{30,13}{30} \times 100\% = 100\%$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,716-0,4659)}{0,0082} = 30,50 \text{ ppm}$$

$$\% \text{recovery} = \frac{30,50}{30} \times 100\% = 102\%$$

Replikasi 3

$$\text{Konsentrasi} = \frac{(0,713-0,4659)}{0,0082} = 30,13 \text{ ppm}$$

$$\% \text{recovery} = \frac{30,13}{30} \times 100\% = 100\%$$

$$\text{Rata - rata \% recovery} = \frac{100\% + 102\% + 100\%}{3} = 101\%$$

b. Presisi

Konsentrasi 10 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,548-0,4659)}{0,0082} = 10,01 \text{ ppm}$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,550-0,4659)}{0,0082} = 10,26 \text{ ppm}$$

Replikasi 3

$$\text{Konsentrasi} = \frac{(0,548-0,4659)}{0,0082} = 10,01 \text{ ppm}$$

$$\text{Rata - rata konsentrasi} = \frac{10,01 + 10,26 + 10,01}{3} = 10,09$$

$$\text{SD} = 0,14$$

$$\% \text{RSD} = \frac{0,14}{10,09} \times 100\% = 1,40\%$$

Konsentrasi 20 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,644-0,4659)}{0,0082} = 21,72 \text{ ppm}$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,647-0,4659)}{0,0082} = 22,09 \text{ ppm}$$

Replikasi 3

$$\text{Konsentrasi} = \frac{(0,645-0,4659)}{0,0082} = 21,84 \text{ ppm}$$

$$\text{Rata - rata konsentrasi} = \frac{21,72+22,09+21,84}{3} = 21,88$$

$$\text{SD} = 0,19$$

$$\%RSD = \frac{0,19}{21,88} \times 100\% = 0,85\%$$

Konsentrasi 30 ppm

Replikasi 1

$$\text{Konsentrasi} = \frac{(0,713-0,4659)}{0,0082} = 30,13 \text{ ppm}$$

Replikasi 2

$$\text{Konsentrasi} = \frac{(0,716-0,4659)}{0,0082} = 30,50 \text{ ppm}$$

Replikasi 3

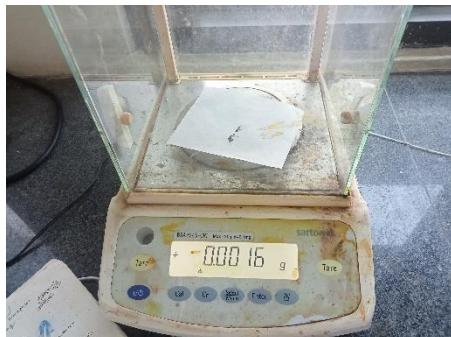
$$\text{Konsentrasi} = \frac{(0,713-0,4659)}{0,0082} = 30,13 \text{ ppm}$$

$$\text{Rata - rata konsentrasi} = \frac{30,13+30,50+30,13}{3} = 30,26$$

$$\text{SD} = 0,21$$

$$\%RSD = \frac{0,21}{30,26} \times 100\% = 0,70\%$$

Lampiran 11. Dokumentasi



Gambar 13. Penimbangan *methylene blue*



Gambar 14. Pembuatan larutan baku



Gambar 15. Uji akurasi & presisi



Gambar 16. Penyiapan larutan sampel



Gambar 17. Ekstraksi sampel



Gambar 18. Penguapan pelarut