

## DAFTAR PUSTAKA

- Adijuwana, N. M., 1989. Teknik Spektroskopi dalam Analisis Biologi. Pusat Antar Universitas IPB: Bogor.
- Adetayo, A., & Runsewe, D., 2019. Synthesis and Fabrication of Graphene and Graphene Oxide: A Review. *Journal of Composite Materials*, **9**: 207-229.
- Ait Salah, A., Mauger, A., Zaghib, K., Goodenough, J. B., Ravet, N., Gauthier, M., & Julien, C.M., 2006. Reduction Fe<sup>3+</sup> of Impurities in LiFePO<sub>4</sub> from Pyrolysis of Organic Precursor Used for Carbon Deposition. *Journal of The Electrochemical Society*, **153**(9): A1692-A1701.
- Al Hassan, M. R., Sen, A., Zaman, T., & Mostari, M. S., 2019. Emergence of Graphene as a Promising Anode Material for Rechargeable Batteries: A review. *Materials Today Chemistry*, **11**: 225-243.
- Alpha, M., Uno, U. E., Isah, K. U., & Ahmadu, U., 2019. Structural and Electrochemical Properties of Reduced Graphene Oxide (RGO) Synthesised using an Improved Modified Hummers Method as Electrode Material for Electronics Applications. *Nigerian Journal of Technology*, **38**(4): 997-1002.
- Aliyev, E., Filiz, V., Khan, M. M., Lee, Y. J., Abetz, C., & Abetz, V., 2019. Structural Characterization of Graphene Oxide: Surface Functional Groups and Fractionated Oxidative Debris. *Nanomaterials*, **9**(8): 1-15.
- Alshamkhani, M. T., Teong, L. K., Putri, L. K., Mohamed, A. R., Lahijani, P., & Mohammadi, M., 2021. Effect of Graphite Exfoliation Routes on The Properties of Exfoliated Graphene and its Photocatalytic Applications. *Journals Environmental Chemical Engineering*, **9**(6):106506.
- Alver, U., Tanrıverdi, A., & Akgul, O., 2016. Hydrothermal Preparation of ZnO Electrodes Synthesized from Different Precursors for Electrochemical Supercapacitors. *Synthetic Metals*, **211**: 30-34.
- Amaliah, F., Sahara, & Fuadi, N., 2018. Pengaruh Temperatur terhadap Nilai Kapasitansi Elektroda Superkapasitor Graphene Berbasis Bambu Betung (Dendrocalamus Asper). *JFT.*, **5**(2): 90-100.
- Aragaw, B. E., 2020. Reduced Graphene Oxide-Intercalated Graphene Oxide Nano-Hybrid for Enhanced Photoelectrochemical Water Reduction. *Journal of Nanostructure in Chemistry*, **10**(1): 9-18.
- Aryanto, D., Hastuti, E., Taspika, M., Anam, K., Isnaeni, I., Widyanto, W.B., dkk., 2020. Characteristics and Photocatalytic Activity of Highly C-Axis-Oriented ZnO thin Films. *Journal of Sol-Gel Science and Technology*, **96**: 226-235.

- Augustyn, V., Simon, P., & Dunn, B., 2014. Pseudocapacitive Oxide Materials for High-Rate Electrochemical Energy Storage. *Energy Environmental Science*, 7(5): 1597-1614.
- Azis, T., Muzakkar, M. Z., Nurwahida, A. T., Dali, N., Kadir, La Ode A., Lestari, D.A., Salim, La Ode A. 2023. ZnO-Enhanced Reduced Graphene Oxide Electrodes from Cocoa Shell: Nanoarchitectonics Platform for Photoelectrocatalytic Detection of Methylene Blue. *Journal of Oleo Science*, 72:(12): 1133-1140.
- Badan Pusat Statistik. 2023. Luas Panen dan Produksi Jagung di Indonesia 2022 (Angka Sementara). *Berita Resmi Statistik*, 2023(74): 1-16.
- Baig, Z., Mamat, O., Mustapha, M., Mumtaz, A., Munir, K. S., & Sarfraz, M., 2018. Investigation of Tip Sonication Effects on Structural Quality of Graphene Nanoplatelets (GNPs) for Superior Solvent Dispersion. *Ultrasonics Sonochemistry*, 45: 133-149.
- Bera, M., Chandravati, Gupta, P., & Maji, P. K., 2018. Facile One-Pot Synthesis of Graphene Oxide by Sonication Assisted Mechanochemical Approach and Its Surface Chemistry. *Journal of Nanoscience and Nanotechnology*, 18(2): 902-912.
- Buldu-Aktruk, M., Toufani, M., Tufanic, A., & Erdem, E., 2022. ZnO and Reduced Graphene Oxide Electrodes for All-In-One Supercapacitor Devices. *Nanoscale*, 14: 3269-3278.
- Bychko, I., Abakumov, A., Didenko, O., Chen, M., Tang, J., & Strizhak, P., 2022. Differences in The Structure and Functionalities of Graphene Oxide and Reduced Graphene Oxide Obtained from Graphite with Various Degrees of Graphitization. *Journal of Physics and Chemistry of Solids*, 164(5): 110614.
- Cahyana, A. H., Liandi, A. R., Yunarti, R. T., Febriantini, D., & Ardiansah, B., 2019. Green Synthesis of Dihydropyrimidine based on Cinnamaldehyde Compound Under Solvent-Free using Graphene Oxide as Catalyst. *AIP Conference Proceedings*, 2168: 1-7.
- Callister, Jr., & William, D., 2009. Materials Science and Engineering an Introduction. 8th Edition. New Jersey: John Wiley & Sons, Inc, Hoboken.
- Chasanah, U., Trisunaryanti, W., Oktaviano, H. S., Triyono, Santoso, I., & Fatmawati, D. A., 2022. Study of Green Reductant Effects of Highly Reduced Graphene Oxide Production and Their Characteristics. *Communications in Science and Technology*, 7(2): 103-111.
- Chauhan, D. S., Quraishi, M. A., Ansari, K. R., & Saleh, T. A., 2020. Graphene and Graphene Oxide as New Class of Materials for Corrosion Control and Protection: Present Status and Future Scenario. *Progress in Organic Coatings*, 147: 1-23.

- Chaudhary, S., James, L. S., Kiran, K. A. B. V., Ramana, C. V. V., Mishra, D. K., Thomas, S., & Kim, D., 2019. Reduced Graphene Oxide/ZnO Nanorods Nanocomposite: Structural, Electrical and Electrochemical Properties. *Journal of Inorganic and Organometallic Polymers and Materials*, **29**(6): 2282-2290.
- Christica, I. S., & Julia, R., 2018. Activated Carbon Utilization from Corn Cob (*Zea Mays*) as a Heavy Metal Adsorbent in Industrial Waste. *Asian Journal of Pharmaceutical Research and Development*, **6**(5): 1-4.
- Cronqvist, P., 2019. Chemistry of Ascorbic Acid Reduction of Graphene Oxide Reduction of Graphene Oxide in Solution and Film.
- Day, R. A., & Underwood, A. L., 1988. Quantitative Analysis. Four Edition. Prentice-Hall, Inc. London. Terjemahan Drs. R. Soendoro. Analisis Kimia Kuantitatif. Edisi Keempat. Erlangga: Jakarta.
- Ding, J., Zhu, S., Zhu, T., Sun, W., Li, Q., Wei, G., dkk., 2015. Hydrothermal Synthesis of Zinc Oxide-Reduced Graphene Oxide Nanocomposites for an Electrochemical Hydrazine Sensor. *RSC Advances*. **5**(29):35-42.
- Ding, R., W. Li, X. Wang, T. Gui, B. Li, P. Han, H. Tian, A. Liu, X. Wang, X. & Liu, 2018. A brief Review Of Corrosion Protective Films and Coatings based on Graphene and Graphene Oxide. *Journal of Alloys and Compounds*, **764**: 1039-1055.
- Dywili, N., Ntziouni, A., Ndipingwi, M. M., Ikpo, C., Nwanya, A. C., Kordatos, K., & Iwuoha, E., 2023. High Power Asymmetric Supercapacitor Based on Activated Carbon/Reduced Graphene Oxide Electrode System. *Materials Today Communications*, **35**: 1-10.
- Ebnalwaled, A. A., Abu El-Fadl, A., Tuhamy, M.A, 2019. Characterization Studies of Reduced Graphene Oxide/Zinc Oxide Nanocomposites Synthesized by Hydrothermal Method. *Journal of Materials and Applications*, **8**(2):80-90.
- Elbasuney, S., El-Sayyad, G. S., Tantawy, H., & Hashem, A. H., 2021. Promising Antimicrobial and Antibiofilm Activities of Reduced Graphene Oxide-Metal Oxide (RGO-Nio, RGO-AgO, and RGO-ZnO) Nanocomposites. *RSC Advances*, **11**: 25961-25975.
- Ezeigwe, E. R., Tan, M. T. T., Khiew, P. S., Siong, C. W., 2015. One-Step Green Synthesis of Graphene/ZnO Nanocomposites for Electrochemical Capacitors. *Ceramics International*, **41**(1): 715-724.
- Fan, X., Peng, W., Li, Y., Li, X., Wang, S., Zhang, G., & Zhang, F. 2008. Deoxygenation of Exfoliated Graphite Oxide under Alkaline Conditions: A Green Route to Graphene Preparation. *Advanced Materials (FRG)*, **20**(23): 4490-4493.

- Feng, Y., Feng, N., Wei, Y., & Zhang, G., 2014. An In Situ Gelatin-Assisted Hydrothermal Synthesis of ZnO-Reduced Graphene Oxide Composites with Enhanced Photocatalytic Performance Under Ultraviolet and Visible Light. *RSC Advances*, **4**: 7933-7943.
- Fernandez-Merino M., Guardia, L., Parades, J., Villar-Rodil, S., Solis-Fernandez, P., Marteniz-Alonso, A., & Tascon, J. 2010. Vitamin C is an Ideal Substitute for Hydrazine in the Reduction of Graphene Oxide Suspensions. *Journal Of Physical Chemistry C*, **114**(14): 6426-6432,
- Fikri, A. A., Dwandaru, W.S., & Brams, 2016. Pengaruh Variasi Konsentrasi Surfaktan dan Waktu Ultrasonik terhadap Sintesis Material Graphene dengan Metode Liquid Sonification Exfoliation menggunakan Tweeter Ultrasonication Graphite Oxide Generator. *Jurnal Fisika*, **5**(8): 188-197.
- Ghorbanzadeh, M., Allahyari, E., Riahifar, R., & Hadavi, S. M. M., 2017. Effect of Al and Zr Co-doping on Electrochemical Performance of Cathode  $\text{Li}[\text{Li}_{0.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}]\text{O}_2$  for Li-ion Battery. *Journal of Solid State Electrochemistry*, **22**(4): 1155-1163.
- Girao, A. V., Caputo, G., & Ferro, M. C., 2017. Application of Scanning Electron Microscopy-Energy Dispersive X-Ray Spectroscopy (SEM-EDS). *Comprehensive Analytical Chemistry*, **75**: 153-168.
- Go, S.H., Kim, H., Yu, J., You, N. H., Ku, B. C., & Kim, Y. K., 2018. Synergistic Effect of UV and L-Ascorbic Acid on the Reduction of Graphene Oxide: Reduction Kinetics and Quantum Chemical Simulations. *Solid State Sciences*, **84**: 120-125.
- Grace, A. S., & Malar, G. S. P., 2020. Synthesis and Characterization of Graphene Oxide from Coconut Husk Ash. *Oriental Journal Of Chemistry*. **36**(2): 348-352.
- Gupta, S. P., Gosavi, S. W., Late, D. J., Qiao, Q., & Walke, P. S., 2020. Temperature Driven High-Performance Pseudocapacitor of Carbon Nano-Onions Supported Urchin Like Structures of A- $\text{MnO}_2$  Nanorods. *Electrochimica Acta*, **354**: 136626.
- Han, Z. J., Pineda, S., Murdock, A. T., Seo, D. H., Ostrikov, K., & Bendavid, A., 2017. RuO<sub>2</sub>-Coated Vertical Graphene Hybrid Electrodes for High-Performance Solid-State Supercapacitors. *Journal of Materials Chemistry A*, **5**(33): 17293-17301.
- Han, W., Ren, L., Qi, X., Liu, Y., Wei, X., Huang, Z., & Zhong, J., 2014. Synthesis of Cds/Zno/Graphene Composite with High-Efficiency Photoelectrochemical Activities Under Solar Radiation. *Applied Surface Science*, **299**: 12-8.
- Hashemi, M., Omidi, M., Muralidharan, B., Tayebi, L., Herpin, M. J., Mohagheghi, M. A., & Milner, T. E. 2018. Layer-by-Layer Assembly of Graphene Oxide on

- Thermosensitive Liposomes for Photo-Chemotherapy. *Acta Biomaterialia*, **65**: 376-392.
- Hassinen, J., Kauppila, J., Leiro, J., Maatanen, A., Ihlainen, P., Peltonen, J., & Lukkari, J. 2013. Low-Cost Reduced Graphene Oxide-Based Conductometric Nitrogen Dioxide-Sensitive Sensor on Paper. *Analytical and Bioanalytical Chemistry*, **405**(11), 3611-3617.
- Hidayat, A., Setiadji, S., & Hadiantoso, E. P., 2019. Sintesis Oksida Grafena Tereduksi (RGO) dari Arang Tempurung Kelapa (Cocos Nucifera). *Al-Kimiya*, **5**(2): 68-73.
- Honorisal, M. B. P., Huda N., Partuti T., & Sholehah A., 2020. Sintesis dan Karakterisasi Grafena Oksida dari Tempurung Kelapa dengan Metode Sonikasi dan Hidrotermal. *Jurnal Sains dan Teknologi*, **16**(1): 1-11.
- Ishaq, S., Moussa, M., Kanwal, F., Ehsan, M., Saleem, M., Van, T. N., & Losic, D., 2019. Facile Synthesis of Ternary Graphene Nanocomposites with Doped Metal Oxide and Conductive Polymers as Electrode Materials for High Performance Supercapacitors. *Scientific Reports*, **9**(1): 1-11.
- Jayachandiran, J., Yesuraj, M., Arivanandhan, A., Raja, S. A., Suthanthiraraj, R., Jayavel, D., & Nedumaran., 2018. Synthesis and Electrochemical Studies of RGO/ZnO Nanocomposite for Supercapacitor Application. *Journal of Inorganic and Organometallic Polymers and Materials*, **28**(5): 2046-2055.
- Kachere, A. R., Kakade, P. M., Kandawe, A. R., Dani, P., Mandlik, N. T., Rondiya, S. C., dkk., 2022. Zinc Oxide/Graphene Oxide Nanocomposites: Synthesis, Characterization and Their Optical Properties. *ES Materials and Manufacturing*, **16**: 19-29.
- Kamel, S., Mohamed El-Sakhawy, M., Anis B., & Tohamy, H. A. S., 2019. Graphene's Structure, Synthesis and Characterization; A Brief review. *Egyptian Journal of Chemistry*, **63**: 593-608.
- Kanta, U-a., Thongpool, V., Sangkhun, W., Wongyao, N., & Wootthikanokkhan, J., 2017. Preparations, Characterizations, and a Comparative Study on Photovoltaic Performance of Two Different Types of Graphene/TiO<sub>2</sub> Nanocomposites Photoelectrodes. *Journal of Nanomaterials*, **2017**: 1-13.
- Kim, D., & Leem, J.Y., 2021. Optimal Temperature of The Sol-Gel Solution Used to Fabricate High Quality ZnO thin Films Via The Dip Coating Method for Highly Sensitive UV Photodetectors. *Journal of the Korean Physical Society*, **78**: 504-509.
- Krey, H., Haris, A., Widodo, D. S., & Khabibi K., 2021. Sintesis Lapis Tipis Semikonduktor Zno-Cds/Fto dan Aplikasinya sebagai Fotodegradasi Limbah Cair Methyl Orange. *Open Journal Systems*, **14**(12): 3729-3739.

- Kulandaivalu, S., & Sulaiman, Y., 2019. Recent Advances in Layer-by-Layer Assembled Conducting Polymer Based Composites for Supercapacitors. *Energies*, **12**(11): 2107.
- Kusrini, E., Suhrowati, A., Usman, A., Khalil, M., & Degirmenci, V., 2019. Synthesis and Characterization of Graphite Oxide, Graphene Oxide, and Reduced Graphene Oxide from Graphite Waste using Modified Hummers' Method and Zinc as Reducing Agent. *International Journal of Technology*, **10**(6): 1093-1104.
- Lee, K. S., Park, C. W., & Kim, J. D., 2018. Synthesis Of ZnO/Activated Carbon with High Surface Area for Supercapacitor Electrodes. *Colloids and Syrfaces A Physicochemical and Engineering Aspects*, **555**: 482-490.
- Li, B., Dai, F., Xiao, Q., Yang, L., Shen, J., Zhang, C., & Cai, M., 2016. Nitrogen-Doped Activated Carbon for A High Energy Hybrid Supercapacitor. *Energy & Environmental Science*, **9**(1): 102-106.
- Li, X., Wang, Z., Qiu, Y., Pan, Q., & Hu, P., 2015. 3D Graphene/ZnO Nanorods Composite Networks as Supercapacitor Electrodes. *Journal of Alloys and Compounds*, **620**: 31-37.
- Li, Y., Li, H., Du, A., Wang, M., Zeng, J., 2017. Morphology and Isothermal Crystallization of Graphene Oxide Reinforced Biodegradable Poly (Butylene Succinate). *Polymer Testing*, **59**: 1-9.
- Ling, L. J., Yee, C. S., & Jaafar, M., 2017. Effect of Sonication Time on the Properties of Multilayer Graphene. *AIP Conference Proceedings*. **1865**(1):1-6.
- Luo, Q., Xu, P., Qiu, Y., Cheng, Z., Chang, H., Fan, H., 2017. Synthesis of ZnO Tetrapods for High-Performance Supercapacitor Applications. *Material Letters*, **198**: 192-195.
- Luo, Q. P., Yu, X. Y., Lei. B. X., Chen, H. Y., Kuang, D. B., & Su, C.Y., 2012. Reduced Graphene Oxide-Hierarchical ZnO Hollow Sphere Composites with Enhanced Photocurrent and Photocatalytic Activity. *The Journal of Physical Chemistry*, **116**(14): 8111-8117.
- Mahmood, K., Khalid, A., & Mehran, M. T., 2018. Nanostructured ZnO Electron Transporting Materials for Hysteresis-free Perovskite Solar Cells. *Solar Energy*, **173**: 496-503.
- Majeed, A., Ullah, W., Anwar, A. W., Nasreen, F., Sharif, A., Mustafa, G., & Khan, A., 2016. Graphene-Metal Oxide/Hydroxide Nanocomposite Materials: Fabrication Advancements and Supercapacitive Performance. *Journal of Alloys and Compounds*, **671**: 1-10.

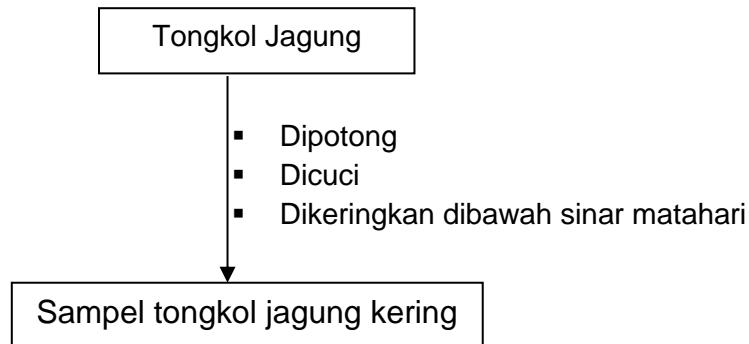
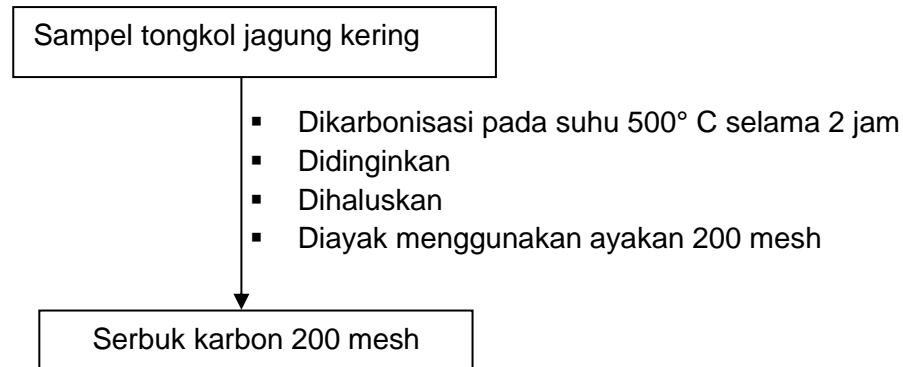
- Manjunatha, C., Chirag, V., Shivaraj, B.W., Srinivasa, N., Ashoka, S., 2020. One Pot Green Synthesis of Novel Rgo@Zno Nanocomposite and Fabrication of Electrochemical Sensor for Ascorbic Acid using Screen-Printed Electrode. *Journal Nanostructur*, **10**(3): 531-539.
- Mas'udah, K.W., 2022. Material Grafena dalam Tongkol Jagung. November. Available: <https://www.researchgate.net/publication/365131007>.
- Miah, M., Mondal, T. K., Ghosh, A., & Saha, S. K., 2020. Study of Highly Porous ZnO Nanospheres Embedded Reduced Graphene Oxide for High Performance Supercapacitor Application. *Electrochimica Acta*, **354**: 136675.
- Muhammad, W., Ullah, N., Haroon, M., & Abbasi, B. H., 2019. Optical, morphological and biological analysis of zinc oxide nanoparticles (ZnO NPs) using *Papaver somniferum L.* *RSC Advance*. **9**: 26541-29548
- Mulyani, R., Buchari, Noviandri, & Ciptati, 2012. Studi Voltametri Siklik Sodium Dodecyl Benzen Sulfonat dalam Berbagai Elektroda dan Elektrolit Pendukung. *Teknologi Pengelolaan Limbah*, **15**(1): 51-56.
- Nam, S., Jeong, Y. J., Park, C. E., & Jang, J., 2017. Enhanced Gas Barrier Properties of Graphene TiO<sub>2</sub> Nanocomposites on Plastic Substrates Assisted by Uv Photoreduction of Graphene Oxide. *Organic Electronics*, **48**: 323-329.
- Naraprawatphong, R., Chokradjaroen, C., Thiangtham, S., Yang, L., & Saito, N., 2022. Nanoscale advanced carbons as an anode for lithium-ion battery. *Materials Today Advances* **16**:1-31.
- Oliveira, A. E. F., Braga, G. B., Tarley, C. R. T., & Pereira, A. C., 2018. Thermally Reduced Graphene Oxide: Synthesis, Studies and Characterization. *Journal Materials Science*, **53**: 12005-12015.
- Osman, A., Elhakeem, A., Kaytbay, S., & Ahmed, A., 2021. Thermal, Electrical and Mechanical Properties of Graphene/Nano-Alumina/Epoxy Composites. *Materials Chemistry and Physics*, **257**: 123809.
- Otun, K. O., Xaba, M. S., Zong, S., Liu, X., Hildebrandt, D., El-Bahy, S. M., Alotaibi, M. T., El- & Bahy, Z. M., 2022. ZIF-8-derived ZnO/C Decorated Hydroxyl-Functionalized Multi-Walled Carbon Nanotube as a New Composite Electrode for Supercapacitor Application. *Colloid and Interface Science Communications*, **47**: 100589.
- Paeru, R. H., & Dewi, T. Q., 2017. Panduan Praktis Budidaya Jagung. Cetak 1. Jakarta Penebar Swadaya: Jakarta.
- Pandolfo, A. G., & Hollenkamp, A. F., 2006. Carbon Properties and Their Role in Supercapacitors. *Journal of Power Sources*, **157**(1): 11-27.

- Pandia, A. Br., Sumarni, W., & Izzania, R. A., 2021. Pengembangan Alat Peraga Uji Daya Hantar Listrik Berbasis Stem dan Pengaruhnya terhadap Literasi Kimia Peserta Didik. *Journal of Chemistry In Education*, **10**(1): 30-37.
- Pei, S., & Cheng, M. H., 2012. The Reduction of Graphene Oxide. *Carbon*, **50**(9): 3210- 3228.
- Priya, D. S., Kennedy, L. J., & Anand, G. T., 2023. Effective Conversion of Waste Banana Bract Into Porous Carbon Electrode for Supercapacitor Energy Storage Applications. *Results in Surfaces and Interfaces*. **10**: 100096.
- Qu, G., Fan, G., Zhou, M., Rong, X., Li, T., Zhang, R., Sun, J., & Chen, D., 2019. Graphene-Modified ZnO Nanostructures for LowTemperature NO<sub>2</sub> Sensing. *ACS Omega*, **4**: 4221-4232.
- Rahayu, P., Putri N.P., & Rohmawati L., 2017. Karakteristik Reduced Graphene Oxide (RGO) Berbahan Dasar Limbah Batang Padi. *Sains & Matematika*, **6**(1): 26-31.
- Rai, S., Bhujel, R., Khadka, M., Chetry, R. L., Swain, B. P., & Biswas, J., 2021. Synthesis, Characterizations, and Electrochemical Studies of ZnO/Reduced Graphene Oxide Nanohybrids for Supercapacitor Application. *Materials Today Chemistry*, **20**: 100472.
- Ramadan, R., & Abdel-Aal, S. K., 2021. Facile Synthesis of Nanostructured ZnO-RGO Based Graphene and Its Application in Wastewater Treatment. *Journal of Materials Science: Materials in Electronics*, **32**(14): 19667-19675.
- Rattanaveeranon, S., & Jiamwattanapong, K., 2022. Effect of CuO/rGO and ZnO/rGO Hybrid Additioinal Layers on Supercapacitor Performance. *Trends in Science*, **19**(15): 3-9.
- Roberts, M. W., Clemons, C. B., Wilber, J. P., Young, G. W. Buldum, A., & Quinn D. D., 2010. Continuum Plate Theory and Atomistic Modeling to Find the Flexural Rigidity of a Graphene Sheet Interacting with a Substrate. *Journal of Nanotechnology*, **2010**: 1-9.
- Rouf, T. B., & Kokini, J. L., 2016. Biodegradable Biopolymer-Graphene Nanocomposites. *Journal of Materials Science*, **51**(22): 9915-9945.
- Sari, D. R., 2018. Studi Terhadap Kinerja Graphite dan Grafena sebagai Elektroda pada Sel Baterai Primer. Tesis. Medan: Fakultas Matematika dan Ilmu Pengetahuan Alam Universitas Sumatera Utara.
- Saravanan, T., Shanmugam, M., Anandan, P., Azhagurajan, M., Pazhanivel, K., Arivanandhan, M., & Jayavel, R., 2015. Facile Synthesis Of Graphene-CeO<sub>2</sub> Nanocomposites with Enhanced Electrochemical Properties for Supercapacitors. *Dalton Transactions*, **44**(21): 9901-9908.

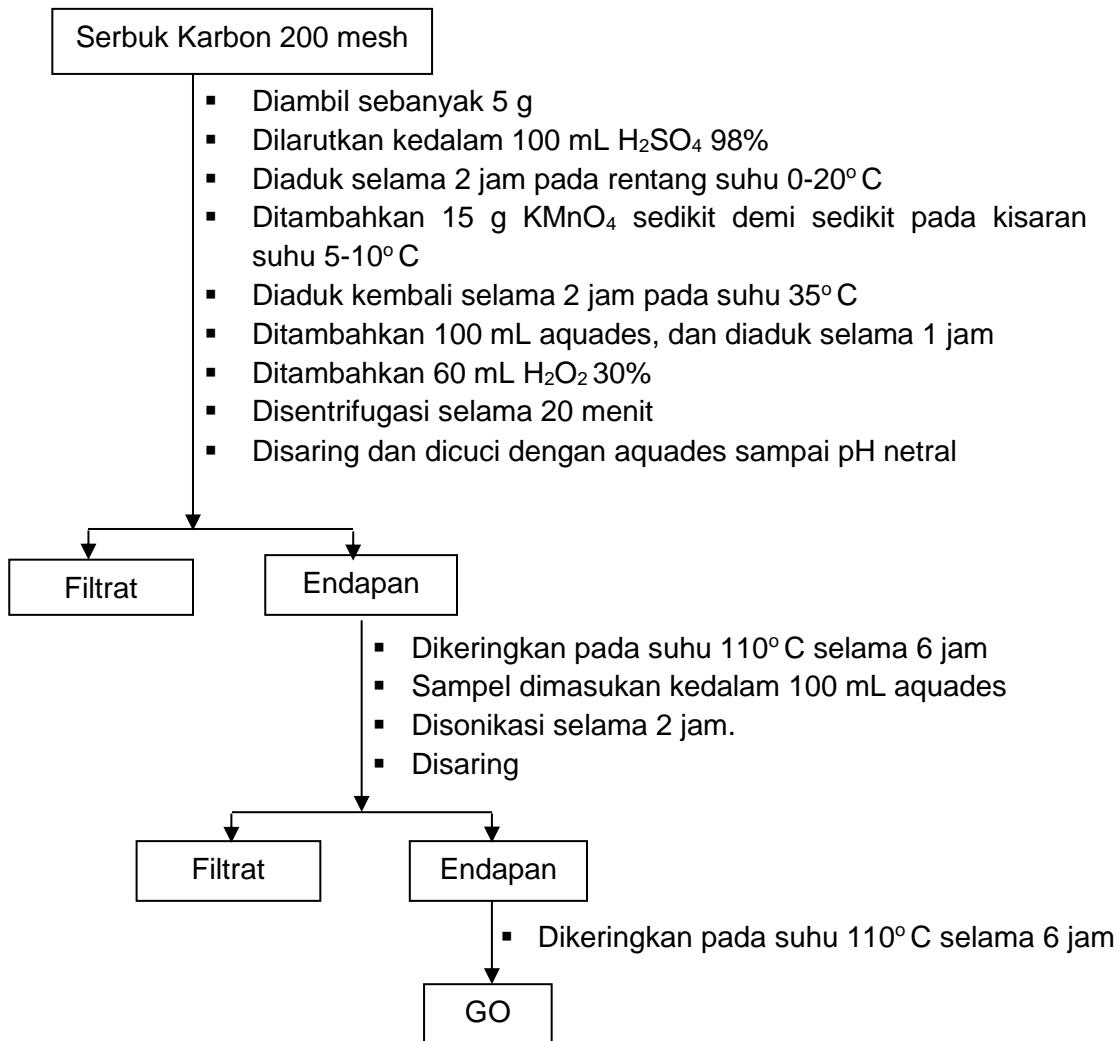
- Shahdeo, D., Roberts, A., Abbineni, N., & Gandhi, S., 2020. Graphene Based Sensors. *Comprehensive Analytical Chemistry*, **91**:175-199.
- Shao, Y., El-Kady, M. F., Sun, J., Li, Y., Zhang, Q., Zhu, M., & Kaner, R. B., 2018. Design and Mechanisms of Asymmetric Supercapacitors. *Chemical Reviews*, **118**(18): 9233-9280.
- Shao, X., Wang, J., Liu, Z., Hu, N., Liu, M., & Xu, Y. 2020. Preparation and Characterization of Porous Microcrystalline Cellulose from Corncob. *Industrial Crops and Products*, **151**: 1-6.
- Shen, J., Shi, M., Li, N., Yan, B., Ma, H., Hu, Y., & Ye, M., 2010. Facile Synthesis and Application of Ag-Chemically Converted Graphene Nanocomposite. *Nano Research*, **3**(5): 339-349.
- Shen, J., Yan, B., Shi, M., Ma, H., Li, N., & Ye, M., 2011. One Step Hydrothermal Synthesis of  $TiO_2$ -Reduced Graphene Oxide Sheets. *Journal of Materials Chemistry*, **21**(10): 3415-3421.
- Shetti, N. P., Bukkitgar, S. D., Reddy, K. R., Reddy, C. V., & Aminabhavi, T. M., 2019. ZnO-Based Nanostructured Electrodes for Electrochemical Sensors and Biosensors in Biomedical Applications. *Biosensors and Bioelectronics*, **141**: 111417.
- Sinha, R., Roy, N., Mandal, T.K., 2022. SWCNT/ZnO Nanocomposite Decorated with Carbon Dots for Photoresponsive Supercapacitor Applications. *Chemical Engineering Journal*, **431**: 133915.
- Siti, I. N., 2020. Isolasi dan Karakterisasi alfa-Selulosa dari Tumbuhan Alang-Alang (*Imperata Cylindrical*) Sebagai Bahan Mikrofilter. Yogyakarta: Universitas Islam Indonesia.
- Skoog, D. A., Holler, F. J., West, D. M., Crouch, S. R., 2013. Fundamental of Analytical Chemistry Ninth Edition, Brooks Cole: USA.
- Song, P., Zhang, X. Y., Sun, M., Ciu, X. L. & Lin, Y., 2012. Synthesis of Graphene Nanosheets Via Oxalic Acid-Induced Chemical Reduction of Exfoliated Graphite Oxide. *RSC Advances*, **2**(3): 1168-1173.
- Sorokina, N. E., Khaskov, M. A., Avdev, V. V., & Nikol'skaya, I. V., 2005. Reaction of Graphite with Sulfuric Acid in the Presence of  $KMnO_4$ . *Russian Journal of General Chemistry*, **75**(2): 162-168
- Sreejesh, M., Dhanush, S., Rossignol, F., & Nagaraja, H. S., 2017. Microwave Assisted Synthesis of RGO/ZnO Composites for Non-Enzymatic Glucose Sensing and Supercapacitor Applications. *Ceramics International*, **43**(6): 4895-4903.

- Sugianto, S., Putra, N. M. D., Rahayu, E. F., Widayatno, W. B., Firdharini, C., Priyono, S., & Aryanto, D., 2023. Synthesis, Characterization, and Electrochemical Performance of Reduced Graphene Oxide-Metal (Cu, Zn)-Oxide Materials. *Indonesian Journal of Science and Technology*, **8**(2): 329-344.
- Su, Y., Kravets, V.G., Wong, S.L., Waters, J., Geim, A.K., & Nair, R.R. 2014. Impermeable Barrier Films and Protective Coatings Based on Reduced Graphene Oxide. *Nature Communications*, **5**:2014.
- Syed, N., Sharma, N., & Kumar, L., 2017. Synthesis of Graphene Oxide (GO) by Modified Hummers Method and its Thermal Reduction to Obtain Reduced Graphene Oxide (RGO). *Graphene*, **6**(01): 1-18.
- Taer, E., Pratiwi, L., Apriwandi, A., Mustika, W.S. Taslim, R., & Agustino, A. 2020. Three-Dimensional Pore Structure of Activated Carbon Monolithic Derived from Hierarchically Bamboo Stem for Supercapacitor Application. *Communication Science Technology*, **5**(1): 22-30.
- Thangavel, S., & Venugopal, G., 2014, Understanding The Adsorption Property of Grapheneoxide with Different Degrees of Oxidation Levels, *Powder Technology*, **257**: 141-148.
- Thallah, S. I. M., Fitrialawati, F., Joni, I. M., Syakir, N., 2022. Karakteristik Lapisan Komposit RGO/CeO<sub>2</sub> sebagai Elektroda pada Model Superkapasitor Simetris Dengan Elektrolit Potassium Klorida. *Jurnal Material dan Energi Indonesia*, **12**(02): 50-56.
- Thebora, M. E. S., Ningsih, K. N., dan Shalihin, M. I., 2019. Sintesis Grafena dari Limbah Pelepas Sawit (Elaeis Sp.) dengan Metode Reduksi Graphite Oksida menggunakan Pereduksi Zn. *Khazanah Intelektual*, **3**(2): 462-476.
- Tiwari, S. K., Sahoo, S., Wang, N., & Huczko, A., 2020. Graphene Research and their Outputs: Status and Prospect. *Journal of Science: Advanced Materials and Devices*, **5**(1): 10-29.
- Tolentino-Hernandez, R. V., Jimenez-Melero, E., Espinosa-Faller, F. J., Guarneros-Aguilar, C., & Caballero-Briones, F., 2021. One-Step, Low Temperature Synthesis of Reduced Graphene Oxide Decorated with ZnO Nanocrystals using Galvanized Iron Steel Scrap. *Material Research Express*, **8**(6): 1-12.
- Wahyuningsih, S., Ramelan, A.H., Fuad, M., & Hanif, A. Q., 2020. Sintesis Grafena Oksida Tereduksi Terdoping Nitrogen dan Sulfur dari Amonium Tiosianat Sebagai Elektroda Lawan Pada Sistem Dye Sensitized Solar Cell (DSSC). *ALCHEMY Jurnal Penelitian Kimia*, **16**(1):126-139.
- Wang, J., Gao, Z., Li, Z., Wang, Bin., Yan, Y., Liu, Qi., dkk., 2011. Green Synthesis of Graphene Nanosheets/ZnO Composites and Electrochemical Properties. *Journal of Solid State Chemistry*, **184**(6): 1421-1427.

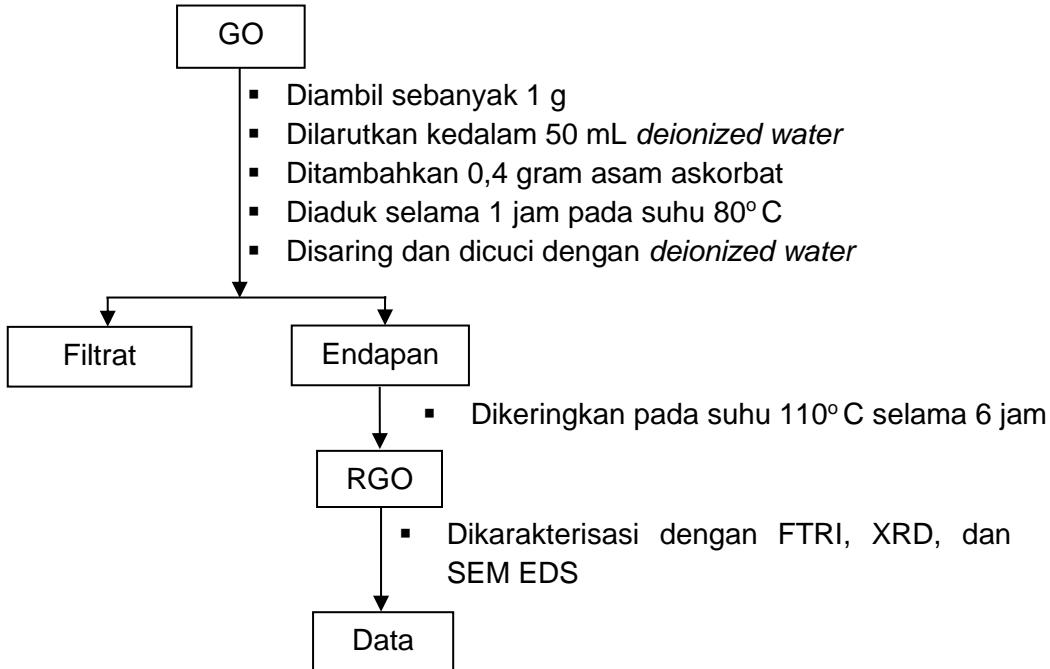
- Wang, J., 2006. Analytical Electrochemistry, 3ed. A John Wiley & Son, Inc., Publication: 29-32.
- Wang, X., Li, W., Wang, X., Zhang, J., Sun, L., Gao, C., Shang, J., Hu, Y., & Zhu, Q. 2017. Electrochemical properties of NiCoO<sub>2</sub> synthesized by hydrothermal method. *RSC Advances*, **7**(80): 50753-50759.
- Wang, Y., Xiao, X., Xue, H., & Pang, H., 2018. Zinc Oxide Based Composite Materials for Advanced Supercapacitors. *Chemistry Select*, **3**(2): 550-565.
- Wang, Z., Ma, C., Wang, H., Liu, Z., & Hao, Z., 2013. Facilely synthesized Fe<sub>2</sub>O<sub>3</sub>-Graphene Nanocomposite as Novel Electrode Materials for Supercapacitors with High Performance. *Journal of Alloys Compounds*, **552**: 486-491.
- Williams, G., Seger, B., & Kamat, P. V., 2008. TiO<sub>2</sub>-Graphene Nanocomposites Uv-Assisted Photocatalytic Reduction of Graphene Oxide. *ACS nano*, **2**(7): 1487-1491.
- Xie, X., Wu, D., Wu, H., Hou, C., Sun, X., Zhang, Y., dkk., 2020. Dielectric Parameters of Activated Carbon Derived from Rosewood and Corncob. *Journal of Materials Science: Materials in Electronics*, **31**: 18077-18084.
- Yunita, Nurlina, N., & Syahbanu, I., 2020. Sintesis Nanopartikel Zink Oksida (ZnO) dengan Penambahan Ekstrak Klorofil sebagai Capping Agent. *Positron*, **10**(2): 123-130.
- Yurestira, I., Aji, A. P., Desfri, M. F., Rini A. S., & Yolanda, R., 2021. Potential of ZnO/ZnS as Electron Transport Materials on Perovskite Solar Cells. *Journal of Aceh Physics Society*, **10**(2): 41-47.
- Yusoff, F., Khing, N. T., Hao, C. C., Sang, L.P., Muhamad, N.B., & Saleh, Md. N. 2018. The Electrochemical Behavior of Zinc Oxide/Reduced Graphene Oxide Composite Electrode in Dopamine. *Malaysian Journal of Analytical Sciences*, **22**(2): 227-237.
- Zaaba, N. I., Foo, K. L., Hashim, U., Tan, S. J., Liu, W. W., & Voon, C. H., 2017. Synthesis of Graphene Oxide using Modified Hummers Method: Solvent Influence, *Procedia Eng.*, **184**: 469-477.
- Zhang LL ., & Zhao XS., 2009. Carbon-Based Materials as Supercapacitor Electrodes. *Chemical Society Reviews*, **38**(9):2520-2531.
- Zhai, Z., Zhang, L., Du, T., Ren, B., Xu, Y., Wang, S., Miao, J., & Liu, Z., 2022. A Review Of Carbon Materials For Supercapacitors. *Materials & Design*, **221**: 111017.
- Zhu, Y., Kong, G., Pan, Y., Liu, L., Yang, B., Zhang, S., Lai, D., & Che, C., 2022. An Improved Hummers Method to Synthesize Graphene Oxide using Much Less Concentrated Sulfuric Acid. *Chinese Chemical Letter*, **33**(10): 4541-4544.

**Lampiran 1. Preparasi Sampel****Lampiran 2. Proses Karbonisasi**

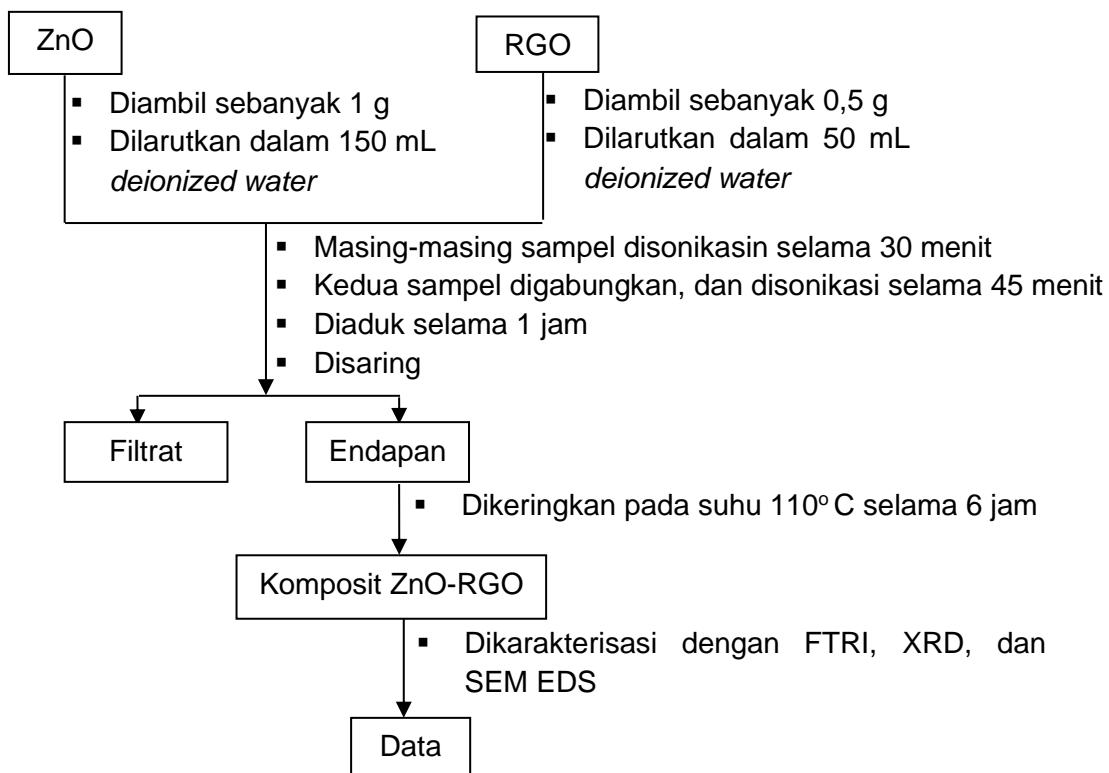
### Lampiran 3. Sintesis GO



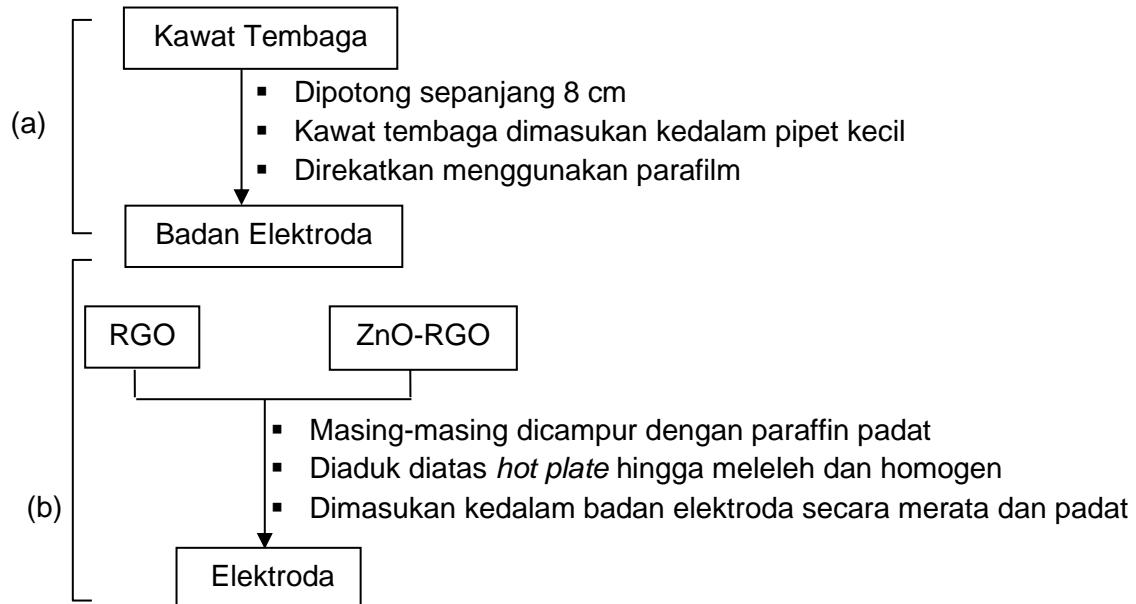
## **Lampiran 4. Sintesis RGO**



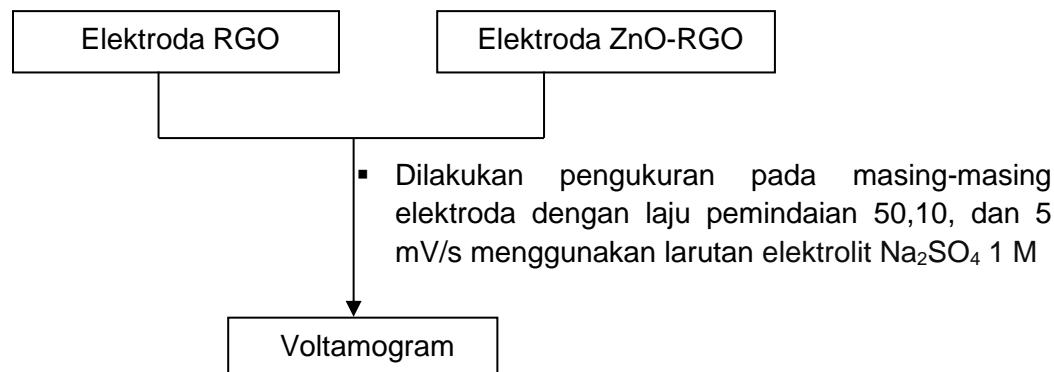
## **Lampiran 5. Komposit ZnO-RGO**



**Lampiran 6.** Pembuatan Elektroda: (a) Badan Elektroda, dan (b) Pasta Elektroda



**Lampiran 7.** Pengukuran Elektrokimia



**Lampiran 8. Dokumentasi Penelitian****1. Preparasi Sampel**

Limbah Tongkol jagung

Proses Pengeringan Sampel

**2. Proses Karbonisasi**

Karbonisasi

Serbuk Karbon 200  
mesh**3. Proses Sintesis RGO**Proses oksida  
menggunakan  
 $H_2SO_4 + KMnO_4$ Pengadukan pada  
suhu 35 °C setelah  
penambahan  $KMnO_4$ Penambahan  
 $H_2O_2$  30%



Setelah disentrifugasi pada 10.000 rpm selama 20 menit



Proses Pencucian dan Penyaringan



Setelah dikeringkan dalam oven



Proses Sonikasi selama 2 jam



Penyaringan



Setelah dikeringkan dalam oven



Proses reduksi menggunakan asam askorbat



Proses Pencucian dan Penyaringan



Setelah dikeringkan dalam oven

#### 4. Komposit ZnO-RGO



Proses Sonikasi  
RGO dan ZnO



Proses Sonikasi larutan  
komposit ZnO-RGO



Pengadukan selama 1 jam



Setelah dikeringkan  
dalam oven

#### 5. Elektroda RGO dan Elektroda Komposit

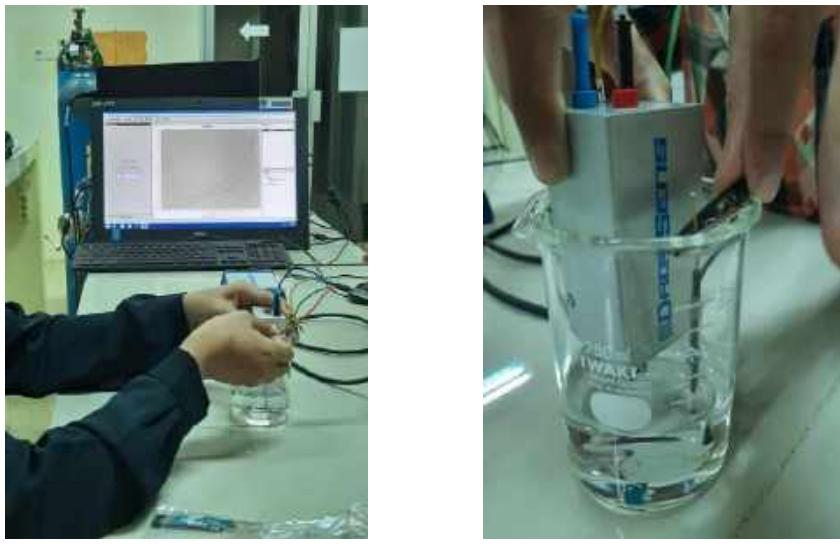


Kawat tembaga

Pipet kecil (sebagai  
badan elektroda)

Campuran sampel dan  
paraffin (berbentuk pasta)

## 6. Pengukuran Elektrokimia



Pengukuran kapasitansi spesifik dengan  
metode siklik voltametri menggunakan  
Potentiostat.

### Lampiran 9. Perhitungan Jarak Lapisan RGO

Jarak antar lembaran RGO dapat dihitung dengan persamaan Bragg:

$$\lambda = 2d \sin (\theta)$$

dimana  $\lambda$  adalah panjang gelombang, dimana  $d$  adalah jarak antar lapisan,  $\theta$  adalah setengah dari sudut difraksi.

Diketahui:

$$\lambda = 0,154 \text{ nm}$$

$$\theta = 23,30^\circ / 2$$

$$= 11,65^\circ$$

Ditanya:

Jarak antar lapisan ( $d$ )?

Sehingga:

$$\begin{aligned} d &= \frac{\lambda}{2 \sin \theta} \\ &= \frac{0,154}{2 \sin(11,65^\circ)} \\ &= \frac{0,154}{2 \times 0,202} \end{aligned}$$

$$d = 0,381 \text{ nm}$$

### Lampiran 10. Penentuan Kapasitansi Spesifik

$$C_{sp} = \frac{A}{m \times k \times (\Delta V)}$$

Keterangan:

- $C_{sp}$  = Kapasitansi Spesifik (F/g)
- $A$  = Luas area kurva (Ampere.Volt)
- $m$  = Massa material (g)
- $k$  = Scan rate (V/s)
- $\Delta V$  =  $(V_2 - V_1)$ , Tegangan (V)

Sampel GO:

1. Scan rate 50 mV/s

$$\begin{aligned} C_{sp} &= \frac{0,00310 \text{ Ampere.Volt}}{0,1162 \text{ g} \times 0,05 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}} \\ &= \frac{0,00310}{0,1162 \times 0,05 \times 1,1} \\ C_{sp} &= 0,4851 \text{ F/g} \end{aligned}$$

2. Scan rate 10 mV/s

$$\begin{aligned} C_{sp} &= \frac{0,00269 \text{ Ampere.Volt}}{0,1162 \text{ g} \times 0,01 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}} \\ &= \frac{0,00296}{0,1162 \times 0,01 \times 1,1} \\ C_{sp} &= 2,1045 \text{ F/g} \end{aligned}$$

3. Scan rate 5 mV/s

$$\begin{aligned} C_{sp} &= \frac{0,00242 \text{ Ampere.Volt}}{0,1162 \text{ g} \times 0,005 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}} \\ &= \frac{0,00242}{0,1162 \times 0,005 \times 1,1} \\ C_{sp} &= 3,7866 \text{ F/g} \end{aligned}$$

### Sampel RGO

#### 1. Scan rate 50 mV/s

$$C_{sp} = \frac{0,03997 \text{ Ampere.Volt}}{0,1164 \text{ g} \times 0,05 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}}$$

$$= \frac{0,03997}{0,1164 \times 0,05 \times 1,1}$$

$$C_{sp} = 6,2433 \text{ F/g}$$

#### 2. Scan rate 10 mV/s

$$C_{sp} = \frac{0,03367 \text{ Ampere.Volt}}{0,1164 \text{ g} \times 0,01 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}}$$

$$= \frac{0,03367}{0,1164 \times 0,01 \times 1,1}$$

$$C_{sp} = 26,2964 \text{ F/g}$$

#### 3. Scan rate 5 mV/s

$$C_{sp} = \frac{0,01834 \text{ Ampere.Volt}}{0,1164 \text{ g} \times 0,005 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}}$$

$$= \frac{0,01834}{0,1164 \times 0,005 \times 1,1}$$

$$C_{sp} = 28,4672 \text{ F/g}$$

### Sampel komposit ZnO-RGO

#### 1. Scan rate 50 mV/s

$$C_{sp} = \frac{0,27595 \text{ Ampere.Volt}}{0,1168 \text{ g} \times 0,05 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}}$$

$$= \frac{0,27595}{0,1168 \times 0,05 \times 1,1}$$

$$C_{sp} = 42,9561 \text{ F/g}$$

2. Scan rate 10 mV/s

$$\begin{aligned}
 C_{sp} &= \frac{0,13817 \text{ Ampere.Volt}}{0,1168 \text{ g} \times 0,01 \frac{\text{V}}{\text{s}} \times ((1) - (-0,1)) \text{ Volt}} \\
 &= \frac{0,13817}{0,1168 \times 0,01 \times 1,1} \\
 C_{sp} &= 107,5420 \text{ F/g}
 \end{aligned}$$

3. Scan rate 5 mV/s

$$\begin{aligned}
 C_{sp} &= \frac{0,13755 \text{ Ampere.Volt}}{0,1168 \times 0,005 \times ((1) - (-0,1))} \\
 &= \frac{0,13755}{0,1168 \times 0,005 \times 1,1} \\
 C_{sp} &= 214,1189 \text{ F/g}
 \end{aligned}$$

**Lampiran 11.** Penentuan Energi Spesifik

$$E_{sp} = \frac{1}{2} \times C_{sp} \times (\Delta V)^2$$

Keterangan:

- $E_{sp}$  = Energi spesifik (Wh/kg)
- $C_{sp}$  = Kapasitansi spesifik (F/g)
- $\Delta V$  = Tegangan (volt)

Sampel GO

1. Scan rate 50 mV/s

$$\begin{aligned}
 E_{sp} &= \frac{1}{2} \times 0,4851 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt} \\
 &= \frac{1}{2} \times 0,4851 \times (1,1)^2 \\
 &= \frac{1}{2} \times 0,586971
 \end{aligned}$$

$$E_{sp} = 0,2934 \text{ Wh/kg}$$

2. Scan rate 10 mV/s

$$\begin{aligned}
 E_{sp} &= \frac{1}{2} \times 2,1045 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt} \\
 &= \frac{1}{2} \times 2,1045 \times (1,1)^2
 \end{aligned}$$

$$= \frac{1}{2} \times 2,546445$$

$$E_{sp} = 1,2732 \text{ Wh/kg}$$

3. Scan rate 5 mV/s

$$E_{sp} = \frac{1}{2} \times 3,7866 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt}$$

$$= \frac{1}{2} \times 3,7866 \times (1,1)^2$$

$$= \frac{1}{2} \times 4,581786$$

$$E_{sp} = 2,2908 \text{ Wh/kg}$$

Sampel RGO

1. Scan rate 50 mV/s

$$E_{sp} = \frac{1}{2} \times 6,2433 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt}$$

$$= \frac{1}{2} \times 6,2433 \times (1,1)^2$$

$$= \frac{1}{2} \times 7,554393$$

$$E_{sp} = 3,7771 \text{ Wh/kg}$$

2. Scan rate 10 mV/s

$$E_{sp} = \frac{1}{2} \times 26,2964 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt}$$

$$= \frac{1}{2} \times 26,2964 \times (1,1)^2$$

$$= \frac{1}{2} \times 31,818644$$

$$E_{sp} = 15,9093 \text{ Wh/kg}$$

3. Scan rate 5 mV/s

$$E_{sp} = \frac{1}{2} \times 28,6472 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt}$$

$$= \frac{1}{2} \times 28,6472 \times (1,1)^2$$

$$= \frac{1}{2} \times 34,663112$$

$$E_{sp} = 17,3315 \text{ Wh/kg}$$

### Sampel komposit ZnO-RGO

#### 1. Scan rate 50 mV/s

$$\begin{aligned} E_{sp} &= \frac{1}{2} \times 42,9561 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt} \\ &= \frac{1}{2} \times 42,9561 \times (1,1)^2 \\ &= \frac{1}{2} \times 51,976881 \end{aligned}$$

$$E_{sp} = 25,9884 \text{ Wh/kg}$$

#### 2. Scan rate 10 mV/s

$$\begin{aligned} E_{sp} &= \frac{1}{2} \times 107,5420 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt} \\ &= \frac{1}{2} \times 107,5420 \times (1,1)^2 \\ &= \frac{1}{2} \times 130,12582 \end{aligned}$$

$$E_{sp} = 65,0629 \text{ Wh/kg}$$

#### 3. Scan rate 5 mV/s

$$\begin{aligned} E_{sp} &= \frac{1}{2} \times 214,1189 \text{ F/g} \times ((1) - (-0,1))^2 \text{ Volt} \\ &= \frac{1}{2} \times 214,1189 \times (1,1)^2 \\ &= \frac{1}{2} \times 259,083869 \end{aligned}$$

$$E_{sp} = 129,5419 \text{ Wh/kg}$$

### Lampiran 12. Penentuan Daya Spesifik

$$P_{sp} = \frac{A}{m}$$

Keterangan:

- $P_{sp}$  = Daya spesifik (kW/kg)
- A = Luas area kurva (Ampere.Volt)
- M = Massa material (g)

### Sampel GO

#### 1. Scan rate 50 mV/s

$$P_{sp} = \frac{0,0031 \text{ Ampere.Volt}}{0,1162 \text{ g}}$$

$$= 0,0266 \text{ kW/kg}$$

2. Scan rate 10 mV/s

$$P_{sp} = \frac{0,00269 \text{ Ampere.Volt}}{0,1162 \text{ g}}$$

$$= 0,0231 \text{ kW/kg}$$

3. Scan rate 5 mV/s

$$P_{sp} = \frac{0,00242 \text{ Ampere.Volt}}{0,1162 \text{ g}}$$

$$= 0,0208 \text{ kW/kg}$$

Sampel RGO

1. Scan rate 50 mV/s

$$P_{sp} = \frac{0,03997 \text{ Ampere.Volt}}{0,1164 \text{ g}}$$

$$= 0,3433 \text{ kW/kg}$$

2. Scan rate 10 mV/s

$$P_{sp} = \frac{0,03367 \text{ Ampere.Volt}}{0,1164 \text{ g}}$$

$$= 0,2892 \text{ kW/kg}$$

3. Scan rate 5 mV/s

$$P_{sp} = \frac{0,01834 \text{ Ampere.Volt}}{0,1162 \text{ g}}$$

$$= 0,1575 \text{ kW/kg}$$

Sampel ZnO-RGO

1. Scan rate 50 mV/s

$$P_{sp} = \frac{0,27595 \text{ Ampere.Volt}}{0,1168 \text{ g}}$$

$$= 2,3625 \text{ kW/kg}$$

2. Scan rate 10 mV/s

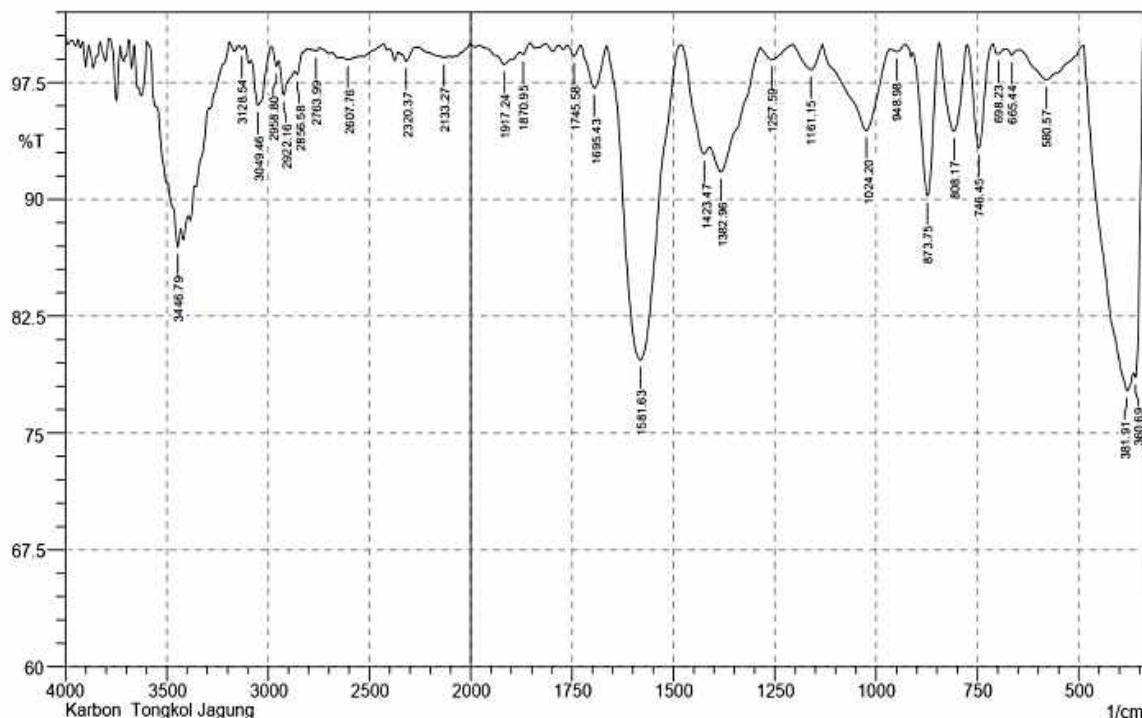
$$P_{sp} = \frac{0,13817 \text{ Ampere.Volt}}{0,1164 \text{ g}}$$
$$= 1,1829 \text{ kW/kg}$$

3. Scan rate 5 mV/s

$$P_{sp} = \frac{0,13755 \text{ Ampere.Volt}}{0,1164 \text{ g}}$$
$$= 1,1776 \text{ kW/kg}$$

### Lampiran 13. Hasil Analisis FTIR

 SHIMADZU



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	360.69	78.57	3.876	364.55	343.33	1.483	0.425
2	381.91	77.705	3.786	487.99	366.48	8.681	2.266
3	580.57	97.074	2.04	648.08	489.92	0.957	0.774
4	665.44	99.236	0.402	680.87	648.08	0.073	0.021
5	698.23	99.33	0.514	711.73	680.87	0.064	0.041
6	746.45	93.293	6.651	775.38	711.73	0.875	0.86
7	808.17	94.362	5.586	842.89	777.31	1.026	1.012
8	873.75	90.225	9.518	908.47	844.82	1.443	1.366
9	948.98	99.465	0.307	966.34	927.76	0.062	0.026
10	1024.2	94.417	5.322	1132.21	968.27	2.16	2.008
11	1161.15	98.322	1.814	1203.58	1132.21	0.313	0.293
12	1257.59	98.987	0.806	1286.52	1203.58	0.215	0.151
13	1382.96	91.761	2.969	1409.96	1286.52	2.831	0.934
14	1423.47	92.93	1.529	1479.4	1411.89	1.372	0.323
15	1581.63	79.694	20.175	1664.57	1481.33	8.771	8.669
16	1695.43	97.145	2.726	1730.15	1664.57	0.454	0.418
17	1745.58	99.201	0.693	1762.94	1730.15	0.064	0.049
18	1870.95	99.302	0.314	1878.67	1853.59	0.045	0.016
19	1917.24	98.659	0.654	1938.46	1878.67	0.258	0.086
20	2133.27	99.102	0.187	2280.57	2106.27	0.444	0.095
21	2320.37	98.876	0.551	2339.65	2260.57	0.225	0.068
22	2607.76	98.981	0.129	2638.62	2574.97	0.267	0.02
23	2763.99	99.517	0.118	2775.57	2744.71	0.055	0.009
24	2856.58	98.014	0.6	2870.08	2827.64	0.261	0.046
25	2922.16	96.711	1.964	2947.23	2870.08	0.794	0.306
26	2958.8	98.524	0.658	2983.88	2947.23	0.15	0.045
27	3049.46	96.035	3.195	3084.18	2985.81	1.027	0.755
28	3128.54	99.641	0.22	3142.04	3115.04	0.031	0.014
29	3446.79	86.946	1.785	3496.94	3433.29	3.262	0.208

Comment;

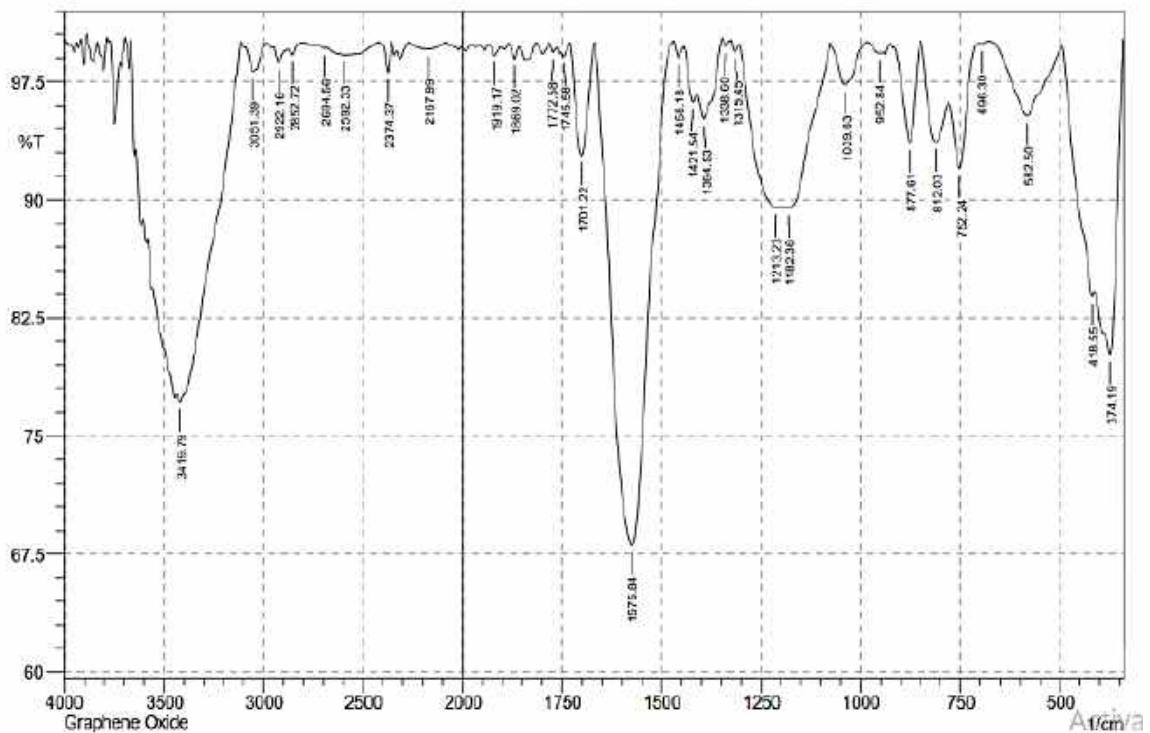
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No. of Scans;

Resolution;

Ac  
5



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	374.18	80.155	6.808	387.69	343.33	2.674	0.932
2	418.55	83.873	0.974	493.78	414.7	3.522	0.436
3	582.5	95.326	4.58	677.01	485.71	1.644	1.774
4	696.3	99.908	0.054	702.09	677.01	0.002	0.002
5	752.24	92.001	5.504	779.24	704.02	1.379	0.757
6	812.03	93.069	4.25	848.68	781.17	1.365	0.776
7	877.61	93.602	6.347	916.19	850.61	0.937	0.913
8	952.84	99.308	0.187	983.7	943.19	0.077	0.021
9	1039.63	97.3	2.666	1076.28	983.7	0.548	0.539
10	1182.36	89.458	1.564	1199.72	1076.28	3.677	0.749
11	1213.23	89.509	1.217	1303.88	1201.65	3.498	0.965
12	1315.45	99.423	0.543	1327.03	1303.88	0.033	0.03
13	1338.8	99.747	0.396	1348.24	1330.88	0.004	0.016
14	1394.53	95.151	2.402	1409.96	1348.24	0.866	0.443
15	1421.54	98.206	1.472	1444.68	1409.96	0.393	0.116
16	1458.18	98.992	0.927	1469.76	1444.68	0.059	0.049
17	1575.84	68.078	31.935	1686.5	1469.76	16.158	16.174
18	1701.22	92.816	7.168	1730.15	1668.43	1.087	1.083
19	1745.58	99.105	0.49	1755.22	1730.15	0.061	0.025
20	1772.58	99.373	0.448	1782.23	1762.94	0.032	0.017
21	1889.02	98.963	0.853	1884.45	1857.45	0.064	0.044
22	1919.17	99.176	0.619	1932.67	1903.74	0.065	0.038
23	2167.99	99.56	0.451	2272.15	2085.05	0.227	0.234
24	2374.37	98.091	1.951	2420.66	2357.01	0.178	0.188
25	2592.33	99.194	0.042	2681.05	2584.61	0.273	0.035
26	2694.56	99.611	0.082	2719.63	2681.05	0.055	0.007
27	2852.72	99.212	0.462	2873.94	2806.43	0.106	0.035
28	2922.16	98.67	0.968	2949.16	2873.94	0.249	0.123
29	3051.39	98.165	1.614	3089.96	2983.88	0.472	0.387
30	3419.79	77.165	1.464	3433.29	3116.97	20.578	3.324

Comment;

Graphene Oxide

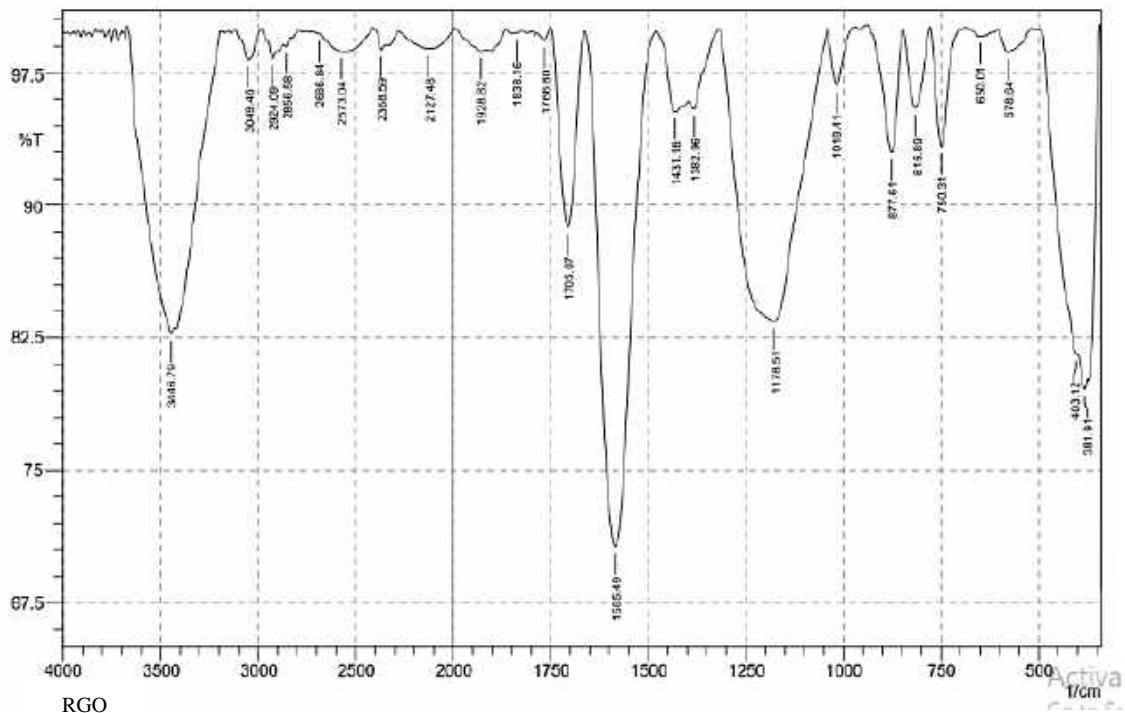
Date/Time; 3/24/2023 9:58:39 AM

No. of Scans;

Activa

Go to Se

Resolution;

**SHIMADZU**


No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	381.91	79.59	1.429	505.35	376.12	7.133	1.013
2	403.12	81.615	0.341	505.35	401.19	4.668	0.067
3	578.64	98.653	1.306	605.65	516.92	0.306	0.297
4	650.01	99.559	0.005	651.94	636.51	0.029	0.001
5	750.31	93.315	6.739	777.31	668.59	0.992	1.01
6	815.89	95.506	4.47	846.75	777.31	0.779	0.775
7	877.61	92.965	8.843	925.83	848.68	1.247	1.175
8	1018.41	96.778	3.211	1041.56	872.12	0.441	0.444
9	1176.51	63.425	16.396	1315.45	1043.49	13.097	12.885
10	1382.96	95.441	1.132	1396.46	1317.38	0.807	0.072
11	1431.18	95.285	2.188	1479.4	1398.39	1.122	0.344
12	1585.49	70.69	20.144	1684.57	1481.33	12.611	12.478
13	1705.07	68.762	11.108	1740.44	1660.5	2.337	2.314
14	1766.5	99.327	0.558	1784.15	1749.44	0.06	0.043
15	1838.16	99.726	0.044	1842.02	1824.66	0.017	0.002
16	1928.82	98.711	0.268	1998.25	1917.24	0.273	0.067
17	2127.48	98.652	1.101	2270.86	1998.25	0.962	0.894
18	2306.59	98.61	0.501	2411.02	2349.3	0.160	0.047
19	2573.04	98.714	0.052	2681.05	2567.25	0.46	0.063
20	2886.84	99.685	0.026	2719.63	2681.05	0.043	0.004
21	2856.58	98.98	0.253	2870.06	2833.43	0.133	0.019
22	2924.09	98.306	0.921	2951.09	2870.06	0.439	0.152
23	3049.46	98.231	1.363	3086.03	2983.88	0.481	0.331
24	3446.79	82.724	1.447	3670.54	3431.36	12.243	2.603

Comment:

RGO

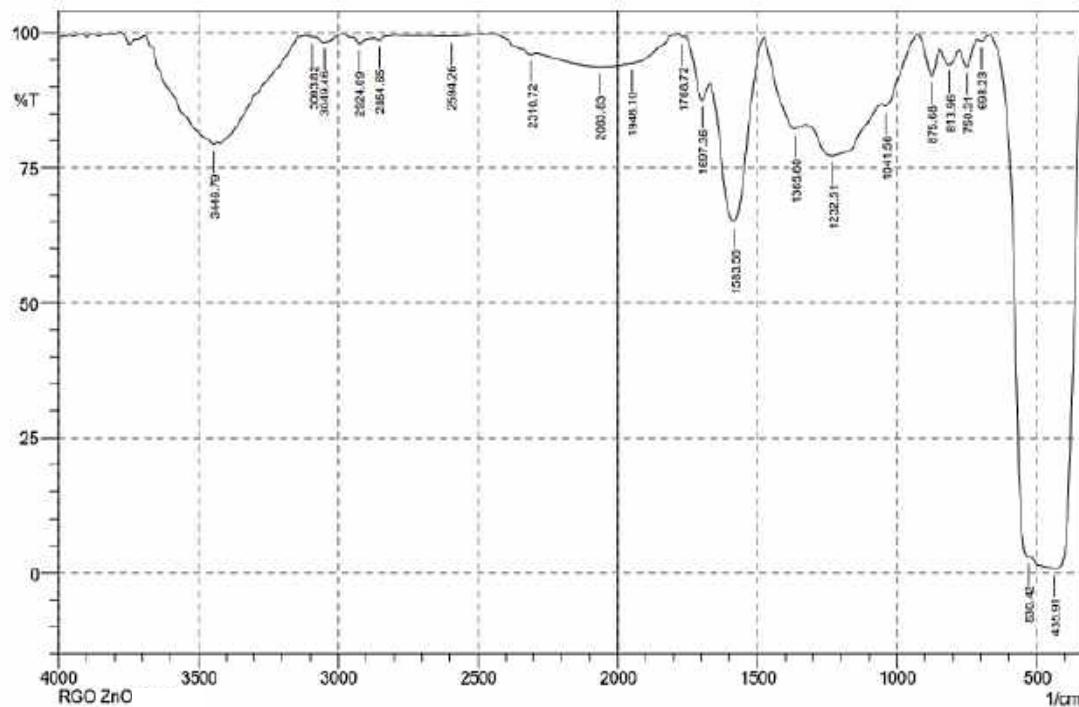
Date/Time: 4/17/2023 10:58:13 AM

No. of Scans:

Activat

Go to Se

Resolution:

 SHIMADZU


No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	435.91	0.811	48.92	524.64	345.26	260.971	125.081
2	530.42	2.98	2.715	665.44	526.57	61.344	1.258
3	698.23	98.383	0.6	711.73	667.37	0.217	0.053
4	750.31	93.719	3.819	777.31	711.73	1.197	0.546
5	813.96	94.06	2.748	846.75	779.24	1.442	0.489
6	875.68	91.963	5.871	925.83	848.68	1.565	0.959
7	1041.56	86.386	1.642	1055.06	927.76	4.411	0.515
8	1232.51	77.183	7.063	1327.03	1056.99	25.388	6.045
9	1365.6	82.181	4.778	1475.54	1328.95	8.849	2.504
10	1583.56	65.043	29.211	1668.43	1477.47	19.671	15.131
11	1697.36	87.537	5.619	1759.08	1670.35	3.111	1.026
12	1768.72	99.209	0.192	1780.3	1761.01	0.057	0.009
13	1948.1	94.323	0.137	1951.96	1805.37	2.359	0.374
14	2063.83	93.682	0.174	2287.58	2050.33	5.687	0.353
15	2310.72	96.002	0.735	2447.67	2289.5	1.578	0.245
16	2594.26	99.301	0.19	2634.76	2447.67	0.385	0.122
17	2854.65	98.603	0.493	2873.94	2794.85	0.299	0.036
18	2924.09	97.959	1.121	2949.16	2873.94	0.458	0.146
19	3049.46	98.261	1.113	3086.11	2983.88	0.515	0.266
20	3093.82	99.172	0.107	3115.04	3086.11	0.09	0.007
21	3446.79	79.344	1.351	3691.75	3433.29	16.032	2.823

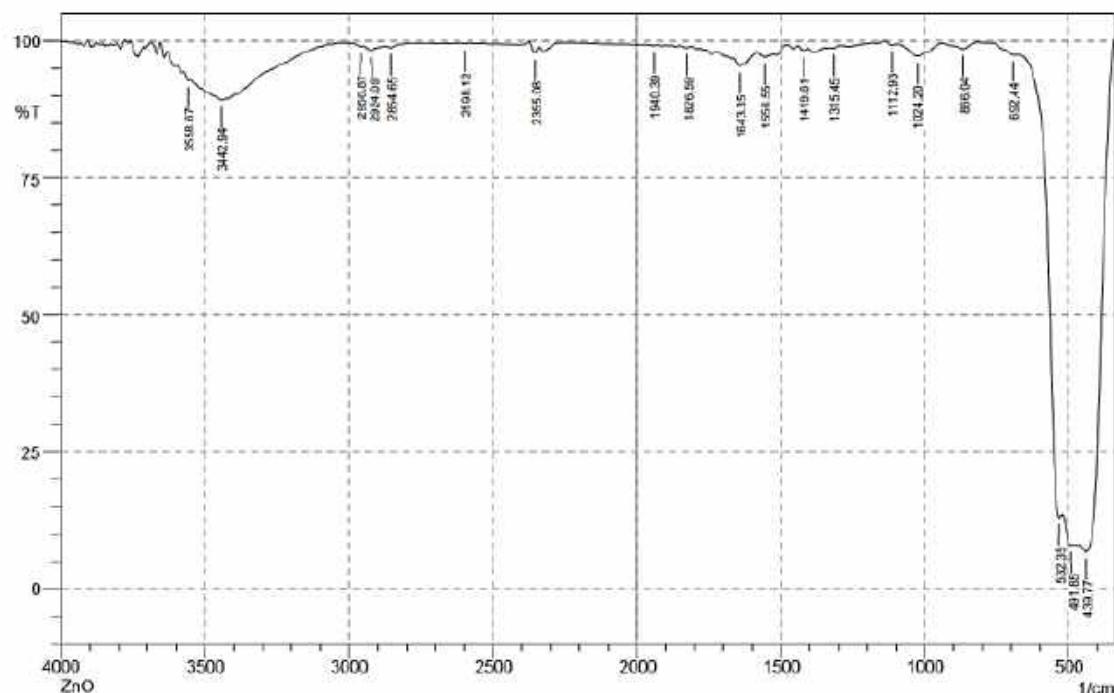
Comment;

RGO ZnO

Date/Time; 12/4/2023 11:24:45 AM

No. of Scans;

Resolution;



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	439.77	6.826	19.254	462.92	345.26	73.773	14.26
2	491.85	8.024	1.453	520.78	462.2	39.459	1.501
3	532.35	13.036	5.712	680.87	522.71	38.774	1.132
4	692.44	97.49	0.303	758.02	682.8	0.54	0.072
5	866.04	98.619	1.09	941.26	813.96	0.455	0.279
6	1024.2	97.235	2.241	1093.64	941.26	1.066	0.722
7	1112.93	99.339	0.316	1141.86	1093.64	0.096	0.035
8	1315.45	98.731	0.232	1327.03	1294.24	0.154	0.013
9	1419.61	98.241	0.551	1440.83	1406.11	0.223	0.048
10	1556.55	97.06	0.421	1585.49	1544.98	0.458	0.03
11	1643.35	65.443	1.683	1662.64	1587.42	1.10	0.32
12	1826.59	98.792	0.273	1853.59	1811.16	0.201	0.031
13	1940.39	99.19	0.115	1955.82	1928.82	0.087	0.005
14	2365.08	97.824	1.655	2374.37	2339.65	0.2	0.114
15	2598.12	99.535	0.005	2630.91	2586.54	0.089	0.001
16	2854.65	98.821	0.293	2872.01	2769.07	0.272	0.011
17	2924.09	98.261	0.751	2951.09	2872.01	0.439	0.097
18	2956.87	99	0.079	3014.74	2951.09	0.171	-0.01
19	3442.94	69.281	0.659	3550.95	3427.51	5.253	0.273
20	3558.67	92.778	0.454	3576.02	3552.88	0.679	0.017

Comment;

ZnO

Date/Time; 12/4/2023 11:14:22 AM

No. of Scans;

Resolution;

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## Lampiran 14. Hasil Analisis XRD

### a. Karbon Tongkol Jagung

#### Peak List

No.	2theta [°]	d [Å]	I/I0 (peak height)	Counts (peak area)	FWHM	Matched
1	17.82	4.9734	186.68	508.59	5.2800	
2	20.62	4.3040	197.22	651.28	6.4000	
3	22.84	3.8904	201.68	878.31	8.4400	
4	25.68	3.4662	298.00	43.11	0.2400	
5	28.38	3.1423	195.11	56.38	0.5600	
6	33.28	2.6900	201.68	14.04	0.1350	
7	38.06	2.3624	252.04	36.41	0.2800	
8	40.56	2.2224	129.66	28.11	0.4202	
9	44.30	2.0431	1000.00	123.84	0.2400	A
10	50.28	1.8132	62.44	8.00	0.2481	
11	64.62	1.4412	393.25	40.58	0.2000	
12	77.68	1.2293	304.47	31.42	0.2000	

#### Integrated Profile Areas

Based on calculated profile

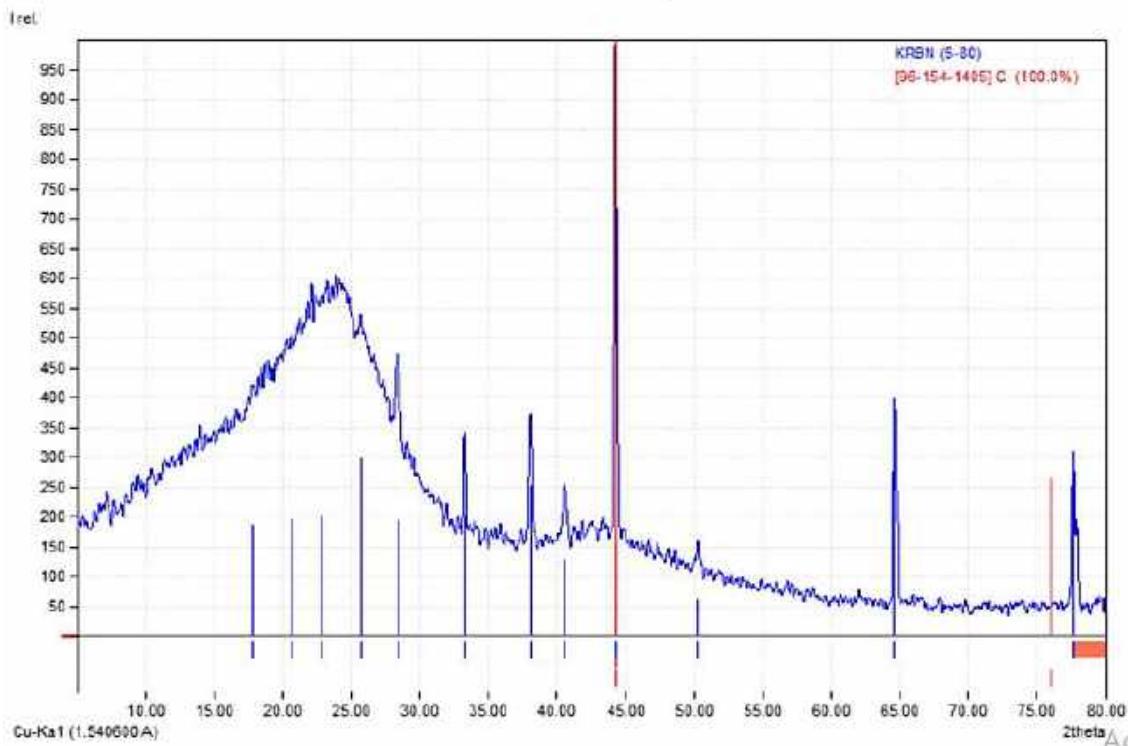
Profile area	Counts	Amount
Overall diffraction profile	350854	100.00%
Background radiation	245580	69.90%
Diffraction peaks	105274	30.01%
Peak area belonging to selected phases	7226	2.06%
Peak area of phase A (C)	7226	2.06%
Unidentified peak area	98049	27.95%

#### Peak Residuals

Peak data	Counts	Amount
Overall peak intensity	2420	100.00%
Peak intensity belonging to selected phases	124	5.12%
Unidentified peak intensity	2296	94.88%

Activate W

#### Diffraction Pattern Graphics



**b. RGO****Match! Phase Analysis Report****Sample: RGO (5-70)****Sample Data**

File name: RGO.RAW  
 File path: C:/Users/TAMBANG/Documents/XRD KIMIA 120922/RGO  
 Date collected: Sep 16, 2022 17:21:55  
 Data range: 5.000° - 80.000°  
 Original data range: 5.000° - 80.000°  
 Number of points: 3751  
 Step size: 0.020  
 Rietveld refinement converged: No  
 Alpha2 subtracted: No  
 Background subtr.: No  
 Data smoothed: Yes  
 Radiation: X-rays  
 Wavelength: 1.540600 Å

**Peak List**

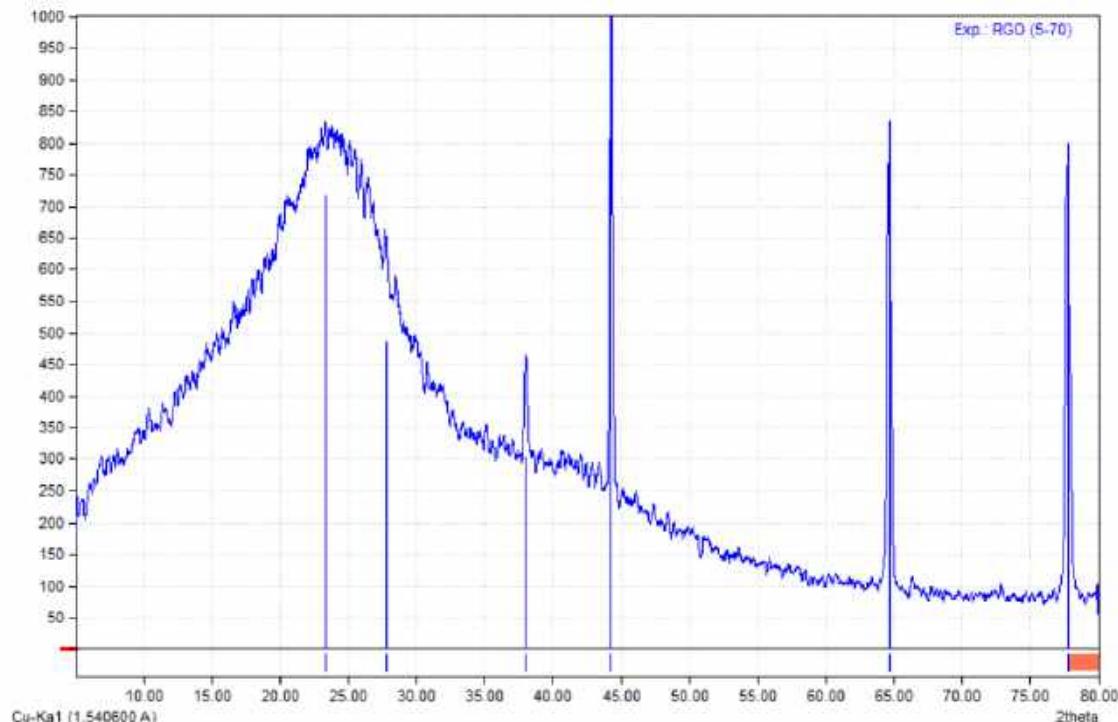
No.	2theta [°]	d [Å]	I/I0	FWHM
1	23.30	3.8146	716.79	0.2400
2	27.78	3.2088	484.67	1.3256
3	38.02	2.3648	303.04	0.4122
4	44.24	2.0457	1000.00	0.2604
5	64.64	1.4406	855.07	0.3042
6	77.72	1.2277	798.69	0.4097

**Integrated Profile Areas****Based on calculated profile**

Profile area	Counts	Amount
Overall diffraction profile	439492	100.00%
Background radiation	243864	55.49%
Diffraction peaks	195020	44.51%
Peak area belonging to selected phases	0	0.00%
Unidentified peak area	195020	44.51%

**Peak Residuals**

Peak data	Counts	Amount
Overall peak intensity	735	100.00%
Peak intensity belonging to selected phases	457	62.16%
Unidentified peak intensity	278	37.84%

**Diffraction Pattern Graphics**

### c. ZnO-RGO

#### Peak List

No.	2theta [°]	d [Å]	I/I0	FWHM	Matched
1	31.77	2.8142	547.95	0.2422	A
2	34.46	2.6005	412.68	0.2309	A
3	36.28	2.4741	1000.00	0.2294	A
4	47.57	1.9009	230.48	0.2434	A
5	56.62	1.6242	341.04	0.2628	A
6	56.76	1.6206	187.10	0.9376	
7	62.86	1.4773	282.95	0.2651	A
8	63.10	1.4722	48.36	0.2400	
9	66.31	1.4084	33.73	0.4400	A
10	67.94	1.3765	235.14	0.2686	A
11	68.18	1.3743	52.01	0.2400	
12	69.05	1.3592	114.21	0.2574	A
13	69.34	1.3541	29.67	0.2400	

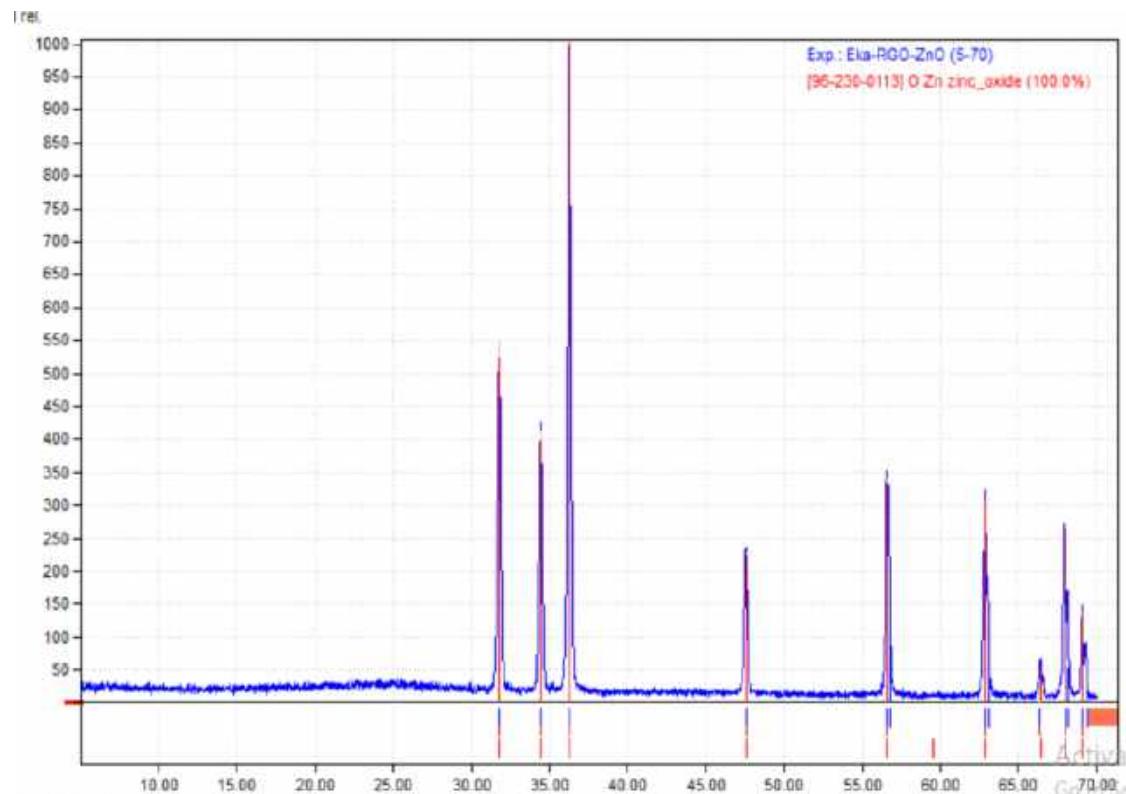
#### Integrated Profile Areas

Based on calculated profile

Profile area	Counts	Amount
Overall diffraction profile	274852	100.00%
Background radiation	119160	43.35%
Diffraction peaks	155692	56.65%
Peak area belonging to selected phases	113152	41.17%
Peak area of phase A (zinc_oxide)	113152	41.17%
Unidentified peak area	42540	15.48%

#### Peak Residuals

Peak data	Counts	Amount
Overall peak intensity	3348	100.00%
Peak intensity belonging to selected phases	3138	93.71%
Unidentified peak intensity	211	6.29%



### d. ZnO

No.	2theta [°]	d [Å]	$\nu/\text{I}$	FWHM	Matched
1	31.82	2.8100	565.30	0.2327	A
2	34.47	2.5997	390.91	0.2242	A
3	36.32	2.4715	1000.00	0.2235	A
4	47.62	1.9081	224.54	0.2323	A
5	56.67	1.6230	350.01	0.2505	A
6	62.93	1.4758	274.25	0.2656	A
7	63.14	1.4713	31.47	0.2400	
8	66.29	1.4088	22.27	0.9715	A
9	68.00	1.3776	254.74	0.2383	A
10	68.24	1.3733	53.87	0.2400	
11	69.10	1.3582	126.10	0.2258	A
12	69.38	1.3535	31.31	0.2400	

### Integrated Profile Areas

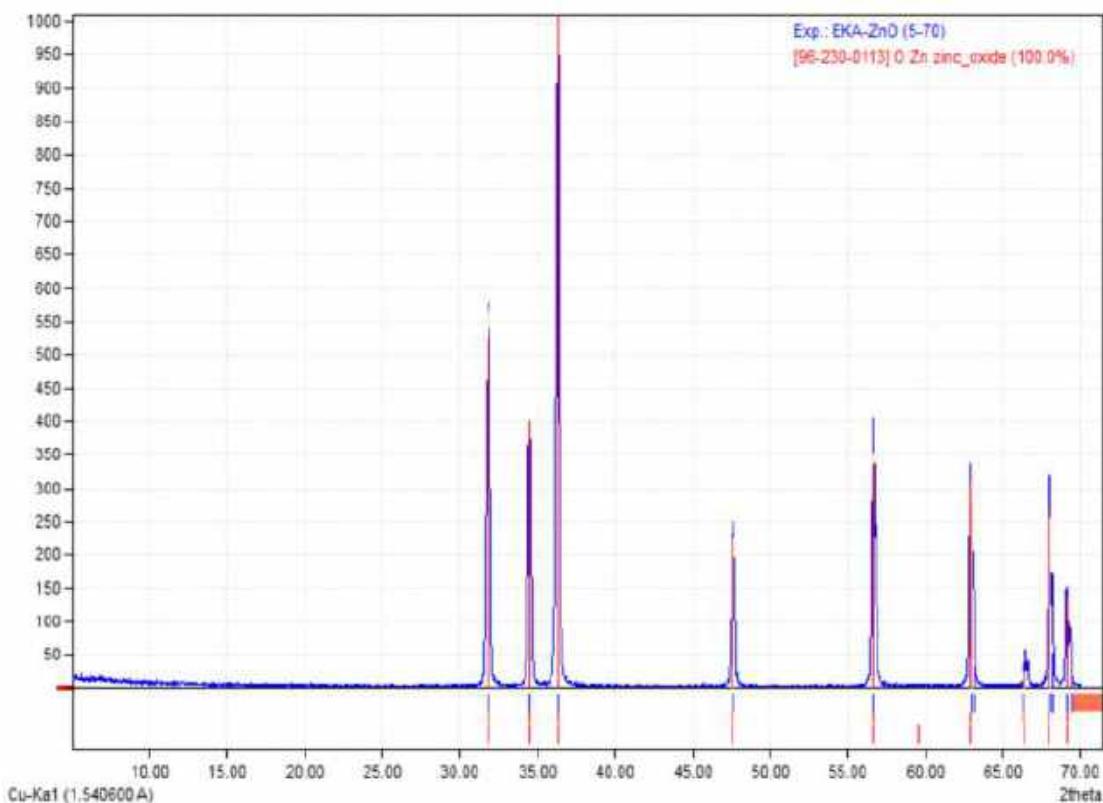
Based on calculated profile

Profile area	Counts	Amount
Overall diffraction profile	235196	100.00%
Background radiation	42814	18.20%
Diffraction peaks	192382	81.80%
Peak area belonging to selected phases	147798	62.84%
Peak area of phase A (zinc_oxide)	147798	62.84%
Unidentified peak area	44584	18.96%

### Peak Residuals

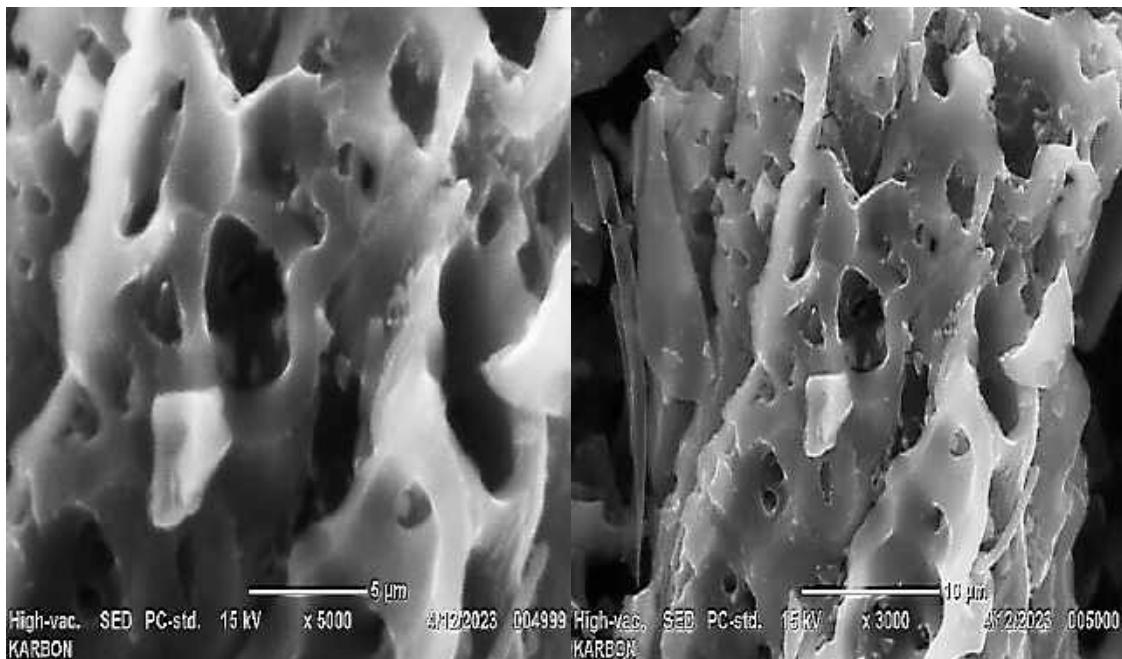
Peak data	Counts	Amount
Overall peak intensity	4133	100.00%
Peak intensity belonging to selected phases	4078	98.67%
Unidentified peak intensity	55	1.33%

### Diffraction Pattern Graphics

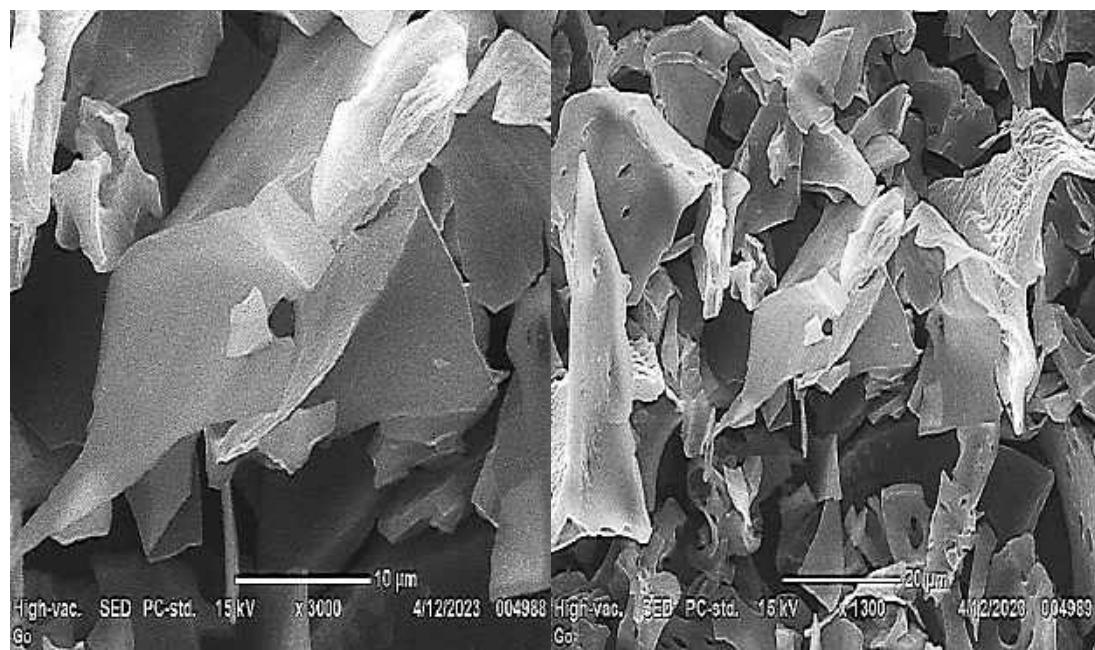


**Lampiran 15. Hasil Analisis SEM**

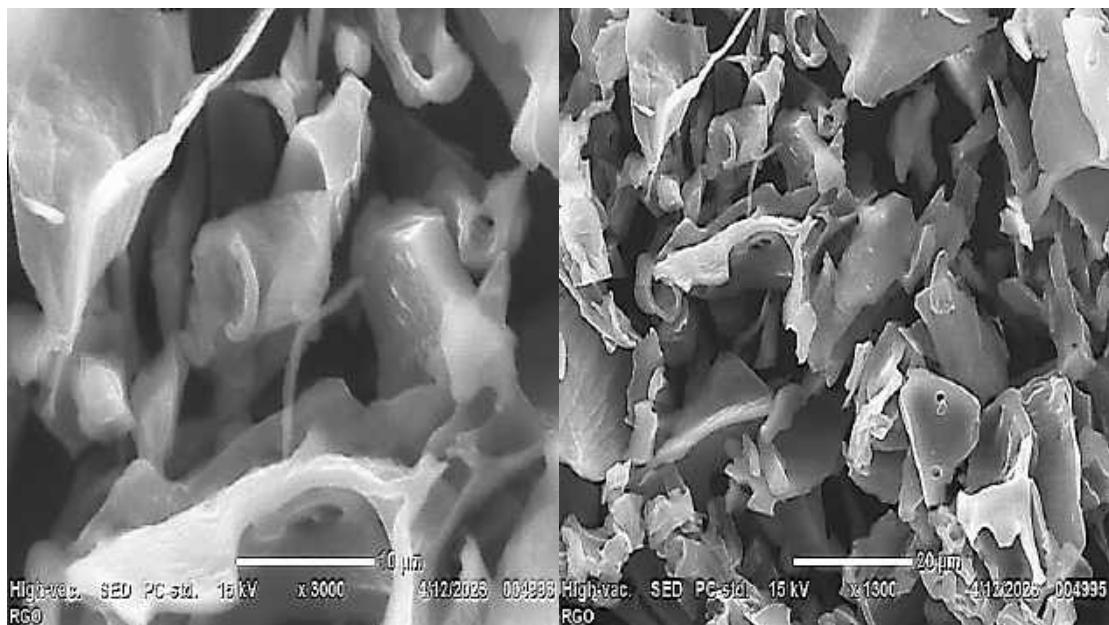
a) Karbon tongkol jagung



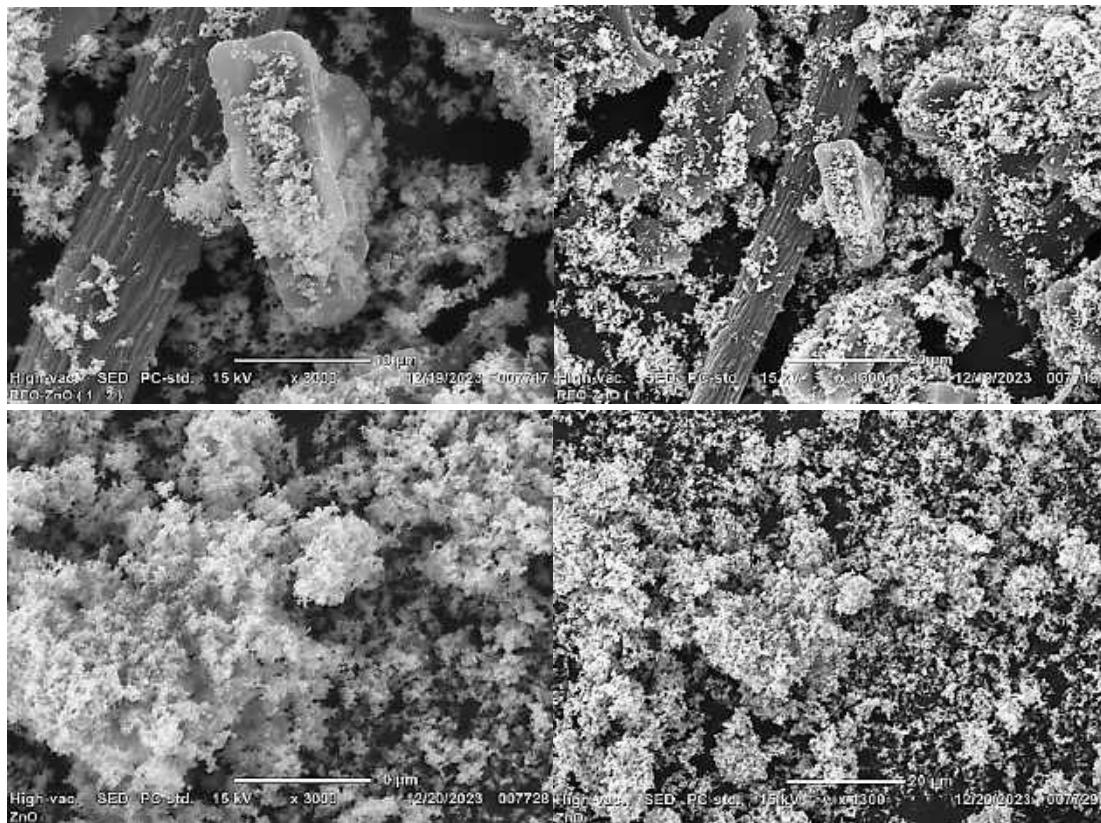
b) GO



c) RGO

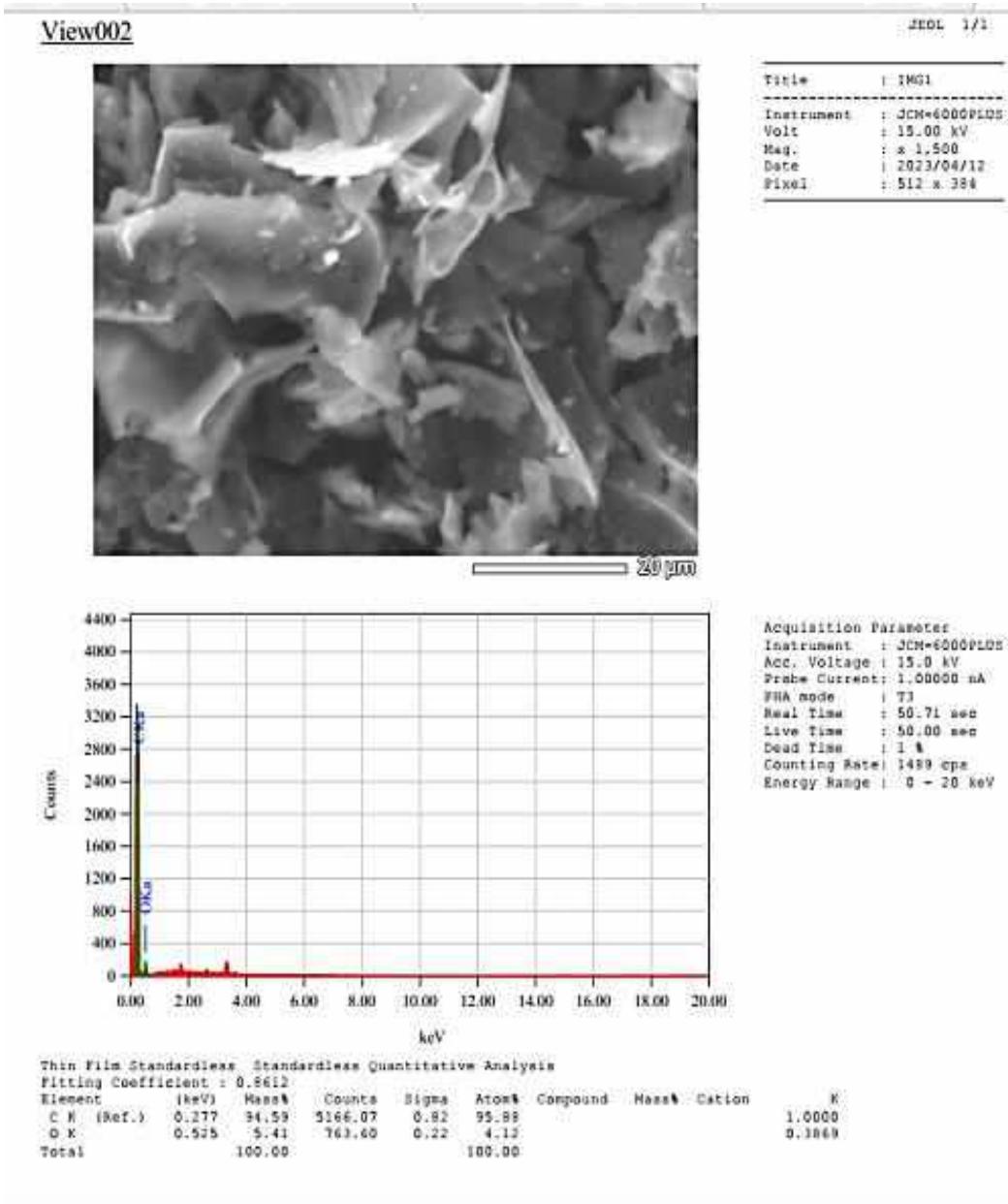


d) ZnO-RGO dan ZnO

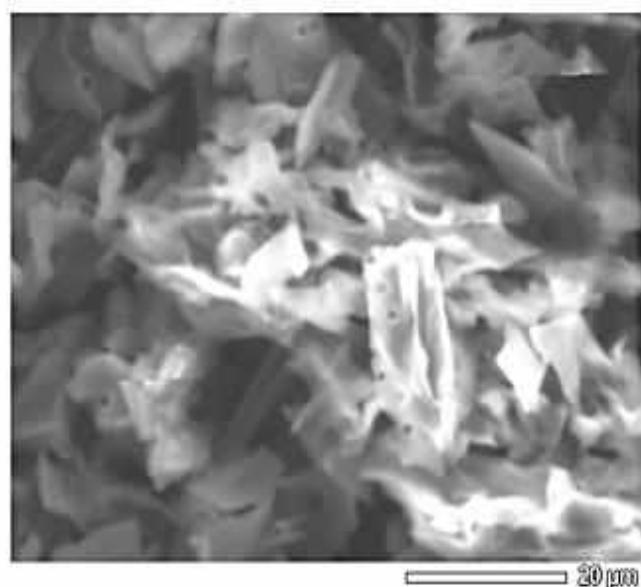


### Lampiran 16. Hasil analisis EDS

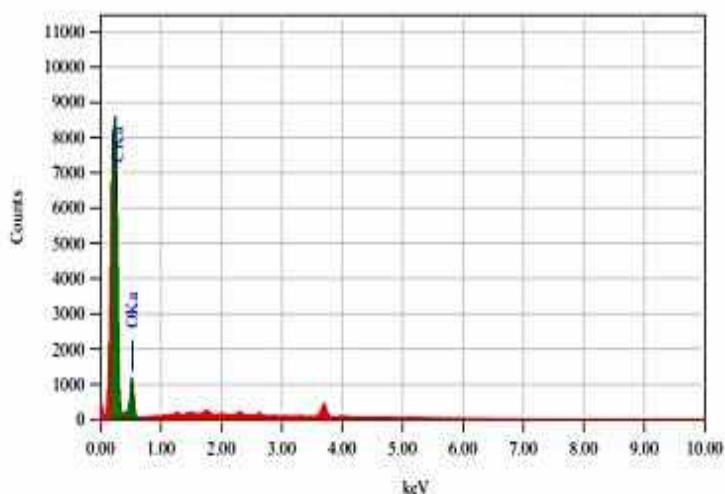
#### a) Karbon tongkol jagung



b) GO



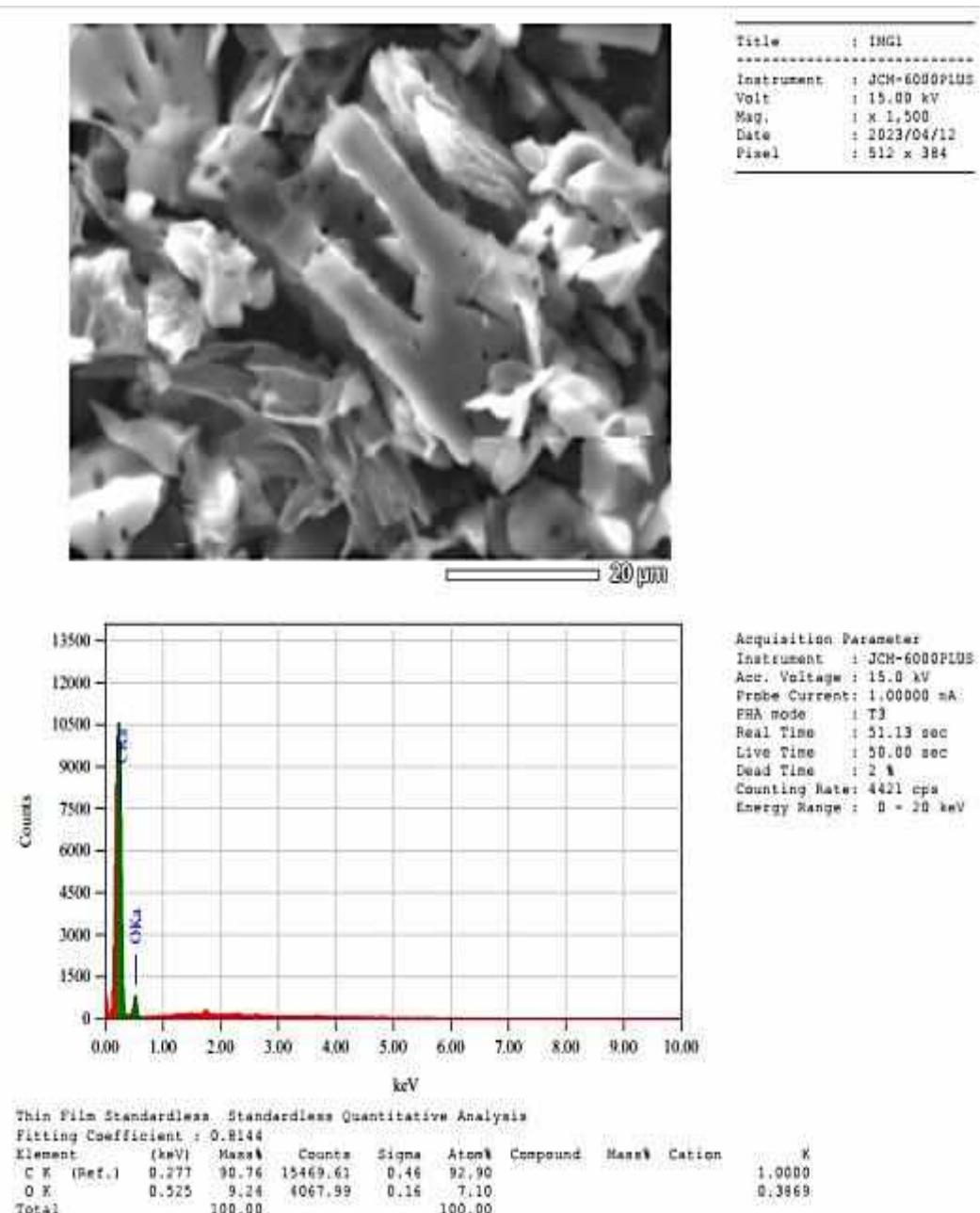
Title : IM01  
 -----  
 Instrument : JCM-6000PLUS  
 Volt : 15.00 kV  
 Mag. : X 1,500  
 Date : 2023/04/12  
 Pixel : 512 x 384



Acquisition Parameter  
 Instrument : JCM-6000PLUS  
 Acc. Voltage : 15.0 kV  
 Probe Current: 1.00000 nA  
 PHA mode : T3  
 Real Time : 51.00 sec  
 Live Time : 50.00 sec  
 Dead Time : 1 %  
 Counting Rate: 3957 cps  
 Energy Range : 0 - 20 keV

Thin Film Standardless Quantitative Analysis							
Fitting Coefficient : 0.7893							
Element	(keV)	Mass%	Counts	Sigma	Atom%	Compound	Mass% Cation
C K (Ref.)	0.271	84.48	11998.24	0.48	87.88		1.0000
O K*	0.525	15.52	5697.30	0.23	12.12		0.3869
Total		100.00			100.00		

c) RGO



d) ZnO-RGO

