

## DAFTAR PUSTAKA

- Arif, A. R., Erviani, A. E., Natsir, H., Haidir, I., & Affandy, M. A. (2018). Optimasi Pretreatment melalui Metode Hydrothermal Pressure dan Pelarut Alkali pada Produksi Bioetanol dari Lemma minor. *ALCHEMY Jurnal Penelitian Kimia*, 14(1), 95. <https://doi.org/10.20961/alchemy.14.1.15986.95-106>
- Aryanta, I. W. R. (2019). Manfaat Jahe Untuk Kesehatan. *Widya Kesehatan*, 1(2), 39–43.
- Badan Penelitian dan Pengembangan Pertanian Kementerian Pertanian. 2011. Pengenalan Bahan Baku Segar dan Bermutu Baik Untuk Jamu. Kementerian Pertanian, Jakarta.
- Badan Pusat Statistik (2021). Produksi Tanaman Biofarmaka Tahun 2018 - 2020.
- Bai, L., Zhu, P., Wang, W., & Wang, M. (2020). The Influence of Extraction pH on the Chemical Compositions, Macromolecular Characteristics, and Rheological Properties of Polysaccharide: The Case of Okra Polysaccharide. *Food Hydrocolloids*, 102. <https://doi.org/10.1016/j.foodhyd.2019.105586>
- Bera, K., Nosalova, G., Sivova, V., & Ray, B. (2016). Structural Elements and Cough Suppressing Activity of Polysaccharides from Zingiber officinale Rhizome. *Phytotherapy Research*, 30(1), 105–111. <https://doi.org/10.1002/ptr.5508>
- Birawidha, D. C., Sari, Y. M. A., Hendronursito, Y., Isnugroho, K., Amin, M., & Manurung, P. (2018). *Ekstraksi Asam Oksalat Dari Belimbing Wuluh (Averrhoa bilimbi L) Dengan Larutan NaOH Dan HNO3*. 1, 2655–2914.
- Chen, A., Lan, Y., Liu, J., Zhang, F., Zhang, L., Li, B., & Zhao, X. (2017). The Structure Property and Endothelial Protective Activity of Fucoidan from Laminaria japonica. *International Journal of Biological Macromolecules*, 105, 1421–1429. <https://doi.org/10.1016/j.ijbiomac.2017.07.148>
- Chen, G., & Kan, J. (2018). Characterization of a Novel Polysaccharide Isolated from Rosa roxburghii Tratt fruit and Assessment of its Antioxidant in Vitro and in Vivo. *International Journal of Biological Macromolecules*, 107, 166–174. <https://doi.org/10.1016/j.ijbiomac.2017.08.160>
- Chen, S. K., Tsai, M. L., Huang, J. R., & Chen, R. H. (2009). In Vitro Antioxidant Activities of Low-Molecular-Weight Polysaccharides with Various Functional Groups. *Journal of Agricultural and Food Chemistry*, 57(7), 2699–2704. <https://doi.org/10.1021/jf804010w>
- Chen, X., Chen, G., Wang, Z., & Kan, J. (2020). A Comparison of a Polysaccharide Extracted from Ginger (Zingiber officinale) Stems and Leaves Using Different Methods: Preparation, Structure Characteristics, and Biological Activities. *International Journal of Biological Macromolecules*, 151, 635–649. <https://doi.org/10.1016/j.ijbiomac.2020.02.222>
- Dewi Kusnadi, N., & Sukohar, A. (2018). Pengaruh Pemberian Ekstrak Jahe Merah (Zingiber officinale var rubrum) Terhadap Penurunan Kadar Asam Urat Darah Obesitas. *Majority*, 7(2), 203–208.
- Diniyah, N., & Lee, S.-H. (2020). Komposisi Senyawa Fenol dan Potensi Antioksidan dari Kacang-Kacangan: Review Phenolic Composition and Antioxidant Potential of Legumes-A Review. *Jurnal Agroteknologi*, 14(1).
- Dodgson, K. S., & Price, R. G. (1962). A Note on the Determination of the Ester Sulphate Content of Sulphated Polysaccharides. *Biochemical Journal*, 84(1), 106.

- Ega, L., Lopulalan, C. G. C., & Meiyasa, F. (2016). Kajian Mutu Karaginan Rumput Laut Eucheuma cottonii Berdasarkan Sifat Fisiko-Kimia pada Tingkat Konsentrasi Kalium Hidroksida (KOH) yang Berbeda. *Jurnal Aplikasi Teknologi Pangan*, 5(2), 38–44. [www.jatp.ift.or.id](http://www.jatp.ift.or.id).
- Faizah, F., Kusnandar, F., & Nurjanah, S. (2020). Senyawa Fenolik, Oryzanol, dan Aktivitas Antioksidan Bekatul yang Difermentasi dengan Rhizopus oryzae. *Jurnal Teknologi Dan Industri Pangan*, 31(1), 86–94. <https://doi.org/10.6066/jtip.2020.31.1.86>
- Gnanasambandam, R., & Proctor, A. (2000). Determination of Pectin Degree of Esterification by Diffuse Reflectance Fourier Transform Infrared Spectroscopy. *Food Chemistry*, 327–332. [www.elsevier.com/locate/foodchem](http://www.elsevier.com/locate/foodchem)
- Hari, N., Priya, C., Koshy, K. G., & Vandana, P. N. (2019). IR Spectroscopic Analysis of Critically Endangered Jasminum Species. In *International Journal of Innovative Science and Research Technology* (Vol. 4, Issue 4). [www.ijisrt.com/763](http://www.ijisrt.com/763)
- Harni, M., Anggraini, T., & Suliansyah, I. (2022). Review Artikel: Pati pada Berbagai Sumber Tanaman. *Agroteknika*, 5(1), 27–2022. <https://doi.org/10.55043/agroteknika.v5i1.118>
- Hastiana, Y., Nawawi, S., & Azizah, S. (2023). Pemanfaatan Tumbuhan Suku Zingiberaceae di Desa Sidorejo Kecamatan Muara Padang Kabupaten Banyuasin. *BEST Journal (Biology Education, Sains and Technology)*, 6(1), 288–294.
- Helmalia, A. W., Putrid, & Dirpan, A. (2019). *Potensi Rempah-Rempah Tradisional sebagai Sumber Antioksidan Alami untuk Bahan Baku Pangan Fungsional*. 2(1).
- Herawati, I. E., & Saptarini, N. M. (2019). Studi Fitokimia pada Jahe Merah (*Zingiber officinale Roscoe* Var. Sunti Val.). *Majalah Farmasetika.*, 4(1), 22–27. <https://doi.org/10.24198/mfarmasetika.v4i0.25850>
- Hu, W., Yu, A., Wang, S., Bai, Q., Tang, H., Yang, B., Wang, M., & Kuang, H. (2023). Extraction, Purification, Structural Characteristics, Biological Activities, and Applications of the Polysaccharides from *Zingiber officinale Roscoe* (Ginger): A Review. In *Molecules* (Vol. 28, Issue 9). MDPI. <https://doi.org/10.3390/molecules28093855>
- Huang, Y., Wu, X., Zhou, S., Lin, Y., Zhang, W., Fu, C., Luo, L., Wang, K., Xie, X., & Fan, H. (2018). Biphasic Extraction of Different Polysaccharides from *Radix Sophorae Tonkinensis* by Microwave-Assisted Aqueous Two-Phase Extraction: Process Optimization, Structural Characterization and Mechanism Exploration. *Separation and Purification Technology*, 207, 187–198. <https://doi.org/10.1016/j.seppur.2018.06.038>
- Jiménez-Escríg, A., Gómez-Ordóñez, E., & Rupérez, P. (2015). Infrared Characterisation, Monosaccharide Profile and Antioxidant Activity of Chemical Fractionated Polysaccharides from the Edible Seaweed Sugar Kombu (*Saccharina latissima*). *International Journal of Food Science and Technology*, 50(2), 340–346. <https://doi.org/10.1111/ijfs.12655>
- Khadijah, Jayali, A. M., Umar, S., & Sasmita, I. (2017). Penentuan Total Fenolik dan Aktivitas Antioksidan Ekstrak Etanolik Daunsamama (*Anthocephalus macrophyllus*) Asal Ternate, Maluku Utara. *Jurnal Kimia Mulawarman*, 15(1), 11–18.
- Khajouei, R. A., Keramat, J., Hamdami, N., Ursu, A. V., Delattre, C., Laroche, C., Gardarin, C., Lecerf, D., Desbrières, J., Djelveh, G., & Michaud, P. (2018). Extraction and Characterization of an Alginate from the Iranian Brown Seaweed *Nizimuddinia*

- zanardini. *International Journal of Biological Macromolecules*, 118, 1073–1081. <https://doi.org/10.1016/j.ijbiomac.2018.06.154>
- Li, J., & Huang, G. (2021). Extraction, Purification, Separation, Structure, Derivatization and Activities of Polysaccharide from Chinese Date. In *Process Biochemistry* (Vol. 110, pp. 231–242). Elsevier Ltd. <https://doi.org/10.1016/j.procbio.2021.08.018>
- Liao, D. wei, Cheng, C., Liu, J. ping, Zhao, L. yan, Huang, D. C., & Chen, G. tang. (2020). Characterization and Antitumor Activities of Polysaccharides Obtained from Ginger (*Zingiber officinale*) by Different Extraction Methods. *International Journal of Biological Macromolecules*, 152, 894–903. <https://doi.org/10.1016/j.ijbiomac.2020.02.325>
- Lu, J., Shi, K. K., Chen, S., Wang, J., Hassouna, A., White, L. N., Merien, F., Xie, M., Kong, Q., Li, J., Ying, T., White, W. L., & Nie, S. (2018). Fucoidan Extracted from the New Zealand Undaria pinnatifida—Physicochemical Comparison Against Five Other Fucoidans: Unique Low Molecular Weight Fraction Bioactivity in Breast Cancer Cell Lines. *Marine Drugs*, 16(12). <https://doi.org/10.3390/md16120461>
- Marjoni, M. R., Afrinaldi, & Novita, A. D. (2015). Kandungan Total Fenol dan Aktivitas Antioksidan Ekstrak Air Daun Kersen (*Muntingia calabura* L.). *Jurnal Kedokteran Yarsi*, 23(3), 187–196.
- Marnoto, T., Haryono, G., Gustinah, D., Fendy, D., & Putra, A. (2012). Ekstraksi Tannin sebagai Bahan Pewarna Alami dari Tanaman Putrimalu (*Mimosa pudica*) menggunakan Pelarut Organik. *Reaktor*, 14(1), 39–45.
- Mentari, V. A., Handika, G., & Maulina, S. (2018). Perbandingan Gugus Fungsi dan Morfologi Permukaan Karbon Aktif dari Pelepas Kelapa Sawit menggunakan Aktivator Asam Fosfat (H<sub>3</sub>PO<sub>4</sub>) dan Asam Nitrat (HNO<sub>3</sub>). *Jurnal Teknik Kimia USU*, 7(1).
- Munadi, R. (2020). Analisis Komponen Kimia dan Uji Aktivitas Antioksidan Ekstrak Rimpang Jahe Merah (*Zingiber officinale* Rosc. *Vsr rubrum*). In *Cokroaminoto Journal of Chemical Science* (Vol. 2, Issue 1).
- Muthukumar, J., Chidambaram, R., & Sukumaran, S. (2021). Sulfated Polysaccharides and its Commercial Applications in Food Industries-A Review. In *Journal of Food Science and Technology* (Vol. 58, Issue 7, pp. 2453–2466). Springer. <https://doi.org/10.1007/s13197-020-04837-0>
- Mutiara, H., & Pratiwi, L. A. (2017). Pengaruh Jahe terhadap Nyeri saat Menstruasi. *Majority*, 6(1).
- Nielsen, S. S. (2017). *Food Science Text Series Food Analysis Laboratory Manual* (S. S. Nielsen, Ed.; 3rd ed.). Springer International Publishing. [www.springer.com/series/5999](http://www.springer.com/series/5999)
- Nur, Y., Cahyotomo, A., Nanda, & Fistoro, N. (2020). Profil GC-MS Senyawa Metabolit Sekunder dari Jahe Merah (*Zingiber officinale*) dengan Metode Ekstraksi Etil Asetat, Etanol dan Destilasi. *Jurnal Sains Dan Kesehatan*, 2(3). <https://doi.org/10.25026/jsk.v2i3.115>
- Nurdyansyah, F., & Widayastuti, D. A. (2022). *Jahe Merah*. Widina Bhakti Persada Bandung. [www.penerbitwidina.com](http://www.penerbitwidina.com)
- Nurjanah, S., & Fathia, S. (2017). Aktivitas Antimikroba Ekstrak Jahe Kering Beku Terhadap Beberapa Bakteri Patogen. *Jurnal Mutu Pangan*, 4(1), 8–15.

- Orak, H. H. (2007). Total Antioxidant Activities, Phenolics, Anthocyanins, Polyphenoloxidase Activities of Selected Red Grape Cultivars and Their Correlations. *Scientia Horticulturae*, 111(3), 235–241. <https://doi.org/10.1016/j.scienta.2006.10.019>
- Oreopoulou, A., Tsimogiannis, D., & Oreopoulou, V. (2019). Extraction of Polyphenols from Aromatic and Medicinal Plants: An Overview of the Methods and the Effect of Extraction Parameters. In *Polyphenols in Plants* (pp. 243–259). Elsevier. <https://doi.org/10.1016/b978-0-12-813768-0.00025-6>
- Panjaitan, R. S., & Natalia, L. (2021). Ekstraksi Polisakarida Sulfat dari Sargassum polycystum dengan Metode Microwave Assisted Extraction dan Uji Toksisitasnya. *Jurnal Pascapanen dan Bioteknologi Kelautan Dan Perikanan*, 16(1). <https://doi.org/10.15578/jpbkp.v16i1.692>
- Pebiningrum, A., & Kusnadi, J. (2017). Pengaruh Varietas Jahe (*Zingiber officinale*) dan Penambahan Madu Terhadap Aktivitas Antioksidan Minuman Fermentasi Kombucha Jahe. *Journal of Food and Life Sciences*, 1(2), 33–42.
- Purwasih, R., & Rahayu, E. W. (2018). Potensi Tepung Ceker dan Leher Ayam sebagai Food Ingredien dan Sumber Pangan Fungsional. *Jurnal Ilmiah Ilmu Dan Teknologi Rekayasa*, 1(2), 124–132. <https://doi.org/10.31962/jiitr.v1i1.56>
- Sari, D., & Nasuha, A. (2021). Kandungan Zat Gizi, Fitokimia, dan Aktivitas Farmakologis pada Jahe (*Zingiber officinale Rosc.*): Review. *Tropical Bioscience: Journal of Biological Science*, 1(2), 11–18.
- Sedjati, S., Santosa, A., & Supriyantini dan Ali Ridlo, E. (2017). Aktivitas Antioksidan dan Kandungan Senyawa Fenolik Makroalga Coklat *Sargassum* sp. *Jurnal Kelautan Tropis*, 20(2), 117–123. [www.ejournal2.undip.ac.id/index.php/jkt](http://www.ejournal2.undip.ac.id/index.php/jkt)
- Setiati, R., Wahyuningrum, D., Siregar, S., & Marhaendrajana, T. (2016). Optimasi Pemisahan Lignin Ampas Tebu dengan menggunakan Natrium Hidroksida. *ETHOS (Jurnal Penelitian Dan Pengabdian)*, 257. <https://doi.org/10.29313/ethos.v0i0.1970>
- Sinurat, E., & Kusumawati, R. (2017). Optimasi Metode Ekstraksi Fukoidan dari Rumput Laut Cokelat *Sargassum* binderi Sonder. *Jurnal Pascapanen Dan Bioteknologi Kelautan Dan Perikanan*, 12(2). <https://doi.org/10.15578/jpbkp.v12i2.388>
- Sugiono. (2015). Isolasi dan Karakterisasi Fukoidan dari Alga Coklat *Sargassum* sp. *JURNAL AGROSAINS: Karya Kreatif Dan Inovatif*, 2(1), 97–107.
- Supu, R. D., Diantini, A., & Levita, J. (2018). Artikel Riset Red Ginger (*Zingiber officinale* var. *rubrum*): Its Chemical Constituents Pharmacological Activities and Safety. *Fitofarmaka Jurnal Ilmiah Farmasi*, 8(1), 25–31. <https://doi.org/10.33751/jf.v8i1.11768>
- Susanti, T. M. I., & Panunggal, B. (2015). Analisis Antioksidan, Total Fenol dan Kadar Kolesterol pada Kuning Telur Asin dengan Penambahan Ekstrak Jahe. *Journal of Nutrition College*, 4(2), 636–644.
- Ukeh, D. A., Birkett, M. A., Pickett, J. A., Bowman, A. S., & Jennifer Mordue, A. (2009). Repellent activity of alligator pepper, Aframomum melegueta, and ginger, *Zingiber officinale*, against the maize weevil, *Sitophilus zeamais*. *Phytochemistry*, 70(6), 751–758. <https://doi.org/10.1016/j.phytochem.2009.03.012>
- Wang, Y., Wei, X., Wang, F., Xu, J., Tang, X., & Li, N. (2018). Structural Characterization and Antioxidant Activity of Polysaccharide from Ginger. *International Journal of*

- Biological Macromolecules*, 111, 862–869.  
<https://doi.org/10.1016/j.ijbiomac.2018.01.087>
- Wijaya, H., Novitasari, & Jubaidah, S. (2018). Perbandingan Metode Ekstraksi Terhadap Rendemen Ekstrak Daun Rambai Laut (*Sonneratia caseolaris* L. Engl). *Jurnal Ilmiah Manuntung*, 4(1), 79–83.
- Yang, W., Pei, F., Shi, Y., Zhao, L., Fang, Y., & Hu, Q. (2012). Purification, Characterization and Anti-Proliferation Activity of Polysaccharides from *Flammulina velutipes*. *Carbohydrate Polymers*, 88(2), 474–480. <https://doi.org/10.1016/j.carbpol.2011.12.018>
- Yang, X., Wei, S., Lu, X., Qiao, X., Simal-Gandara, J., Capanoglu, E., Woźniak, Ł., Zou, L., Cao, H., Xiao, J., Tang, X., & Li, N. (2021). A Neutral Polysaccharide with a Triple Helix Structure from Ginger: Characterization and Immunomodulatory Activity. *Food Chemistry*, 350. <https://doi.org/10.1016/j.foodchem.2021.129261>
- Zhang, Z., Wang, X., Zhang, J., & Zhao, M. (2011). Potential Antioxidant Activities in Vitro of Polysaccharides Extracted from Ginger (*Zingiber officinale*). *Carbohydrate Polymers*, 86(2), 448–452. <https://doi.org/10.1016/j.carbpol.2011.04.062>

## LAMPIRAN

Lampiran 1. Data Hasil Penelitian Jumlah Rendemen

Konsentrasi NaOH	Ulangan 1	Ulangan 2	Ulangan 3	$\Sigma$	$\bar{x}$
0,1 M	2,2	1,9	2,3	6,4	2,13
0,2 M	3,2	2	4,8	10	3,33
0,3 M	4,6	1,9	9,7	16,2	5,4

Lampiran 2. Perhitungan Jumlah Rendemen

$$\% \text{ Yield} = \frac{\text{Ekstrak yang dihasilkan (g)}}{\text{Bahan baku yang digunakan (g)}} \times 100\%$$

- Konsentrasi NaOH 0,1 M ulangan 1 =  $\frac{0,22}{10} \times 100 = 2,2\%$
- Konsentrasi NaOH 0,1 M ulangan 2 =  $\frac{0,19}{10} \times 100 = 1,9\%$
- Konsentrasi NaOH 0,1 M ulangan 3 =  $\frac{0,23}{10} \times 100 = 2,3\%$
- Konsentrasi NaOH 0,2 M ulangan 1 =  $\frac{0,32}{10} \times 100 = 3,2\%$
- Konsentrasi NaOH 0,2 M ulangan 2 =  $\frac{0,20}{10} \times 100 = 2,0\%$
- Konsentrasi NaOH 0,2 M ulangan 3 =  $\frac{0,48}{10} \times 100 = 4,8\%$
- Konsentrasi NaOH 0,3 M ulangan 1 =  $\frac{0,46}{10} \times 100 = 4,6\%$
- Konsentrasi NaOH 0,3 M ulangan 2 =  $\frac{0,19}{10} \times 100 = 1,9\%$
- Konsentrasi NaOH 0,3 M ulangan 3 =  $\frac{0,97}{10} \times 100 = 9,7\%$

Lampiran 3. Hasil Uji Statistik Jumlah Rendemen

Descriptives									
	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum	
					Lower Bound	Upper Bound			
rendemen	0,1	3	2.133	.2082	.1202	1.616	2.650	1.9	2.3
	0,2	3	3.333	1.4048	.8110	-.156	6.823	2.0	4.8
	0,3	3	5.400	3.9611	2.2869	-4.440	15.240	1.9	9.7
	Total	9	3.622	2.5445	.8482	1.666	5.578	1.9	9.7
antioksidan	0,1	3	4.4838E2	79.678642	4.6002E1	250.45161	646.31705	357.910	508.095
	0,2	3	3.9259E2	278.372364	1.6071E2	-298.92029	1084.11029	217.981	713.619
	0,3	3	8.0727E2	212.022283	1.2241E2	280.58178	1333.96688	641.176	1046.086
	Total	9	5.4941E2	264.919810	8.8306E1	345.78250	753.05328	217.981	1046.086
totalgula	0,1	3	1.1069E3	375.07340	2.1654E2	175.1994	2038.6673	780.00	1516.40
	0,2	3	1.3133E3	901.77224	5.2063E2	-926.7931	3553.4598	288.00	1983.20
	0,3	3	6.9740E2	296.15354	1.7098E2	-38.2862	1433.0862	357.60	900.60
	Total	9	1.0392E3	578.01954	1.9287E2	594.9171	1483.5274	288.00	1983.20
totalfenolik	0,1	3	2.7526E1	1.580630	.912577	23.59950	31.45250	25.711	28.600
	0,2	3	4.6526E1	6.465891	3.733084	30.46384	62.58816	40.711	53.489
	0,3	3	1.2970E1	.420812	.242956	11.92531	14.01602	12.489	13.267
	Total	9	2.9007E1	14.949016	4.983005	17.51672	40.49839	12.489	53.489
sulfat	0,1	3	6.7600	3.70340	2.13816	-2.4398	15.9598	3.32	10.68
	0,2	3	91.6133	30.86055	17.81734	14.9515	168.2752	64.28	125.08
	0,3	3	18.8133	7.17031	4.13978	1.0013	36.6254	14.28	27.08
	Total	9	39.0622	42.83721	14.27907	6.1346	71.9898	3.32	125.08

ANOVA

		Sum of Squares	df	Mean Square	F	Sig.
rendemen	Between Groups	16.382	2	8.191	1.388	.320
	Within Groups	35.413	6	5.902		
	Total	51.796	8			
antioksidan	Between Groups	303874.755	2	151937.378	3.539	.097
	Within Groups	257586.367	6	42931.061		
	Total	561461.122	8			
totalgula	Between Groups	589692.382	2	294846.191	.849	.473
	Within Groups	2083160.293	6	347193.382		
	Total	2672852.676	8			
totalfenolik	Between Groups	1698.818	2	849.409	57.285	.000
	Within Groups	88.966	6	14.828		
	Total	1787.785	8			
sulfat	Between Groups	12645.211	2	6322.606	18.642	.003
	Within Groups	2035.004	6	339.167		
	Total	14680.215	8			

Lampiran 4. Data Hasil Penelitian Total Gula

Perlakuan	Total Gula $\mu\text{g/ml}$	Rata-rata
0,1 M	780.0	1.106,93
	1024.4	
	1516.4	
0,2 M	288.0	1.313,33
	1668.8	
	1983.2	
0,3 M	357.6	697,4
	834.0	
	900.6	

Lampiran 5. Perhitungan Total Gula

$$Y = 0,0063x + 0,0883$$

$$FP = 20$$

Konsentrasi NaOH 0,1 M ulangan 1

$$0,334 = 0,0063x + 0,0883$$

$$0,334 - 0,0883 = 0,0063x$$

$$0,2457 = 0,0063x$$

$$x = \frac{0,2457}{0,0063}$$

$$x = 39 \times 20 = 780 \mu\text{g/ml}$$

Konsentrasi NaOH 0,1 M ulangan 2

$$0,411 = 0,0063x + 0,0883$$

$$0,411 - 0,0883 = 0,0063x$$

$$0,3227 = 0,0063x$$

$$x = \frac{0,3227}{0,0063}$$

$$x = 51,22 \times 20 = 1.024,4 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,1 M ulangan 3

$$0,566 = 0,0063x + 0,0883$$

$$0,566 - 0,0883 = 0,0063x$$

$$0,4777 = 0,0063x$$

$$x = \frac{0,4777}{0,0063}$$

$$x = 75,82 \times 20 = 1.516,4 \mu\text{g}/\text{ml}$$

$$\text{Rata-rata} = \frac{780+1.024,4+1.516,4}{3} = 1.106,93 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,2 M ulangan 1

$$0,179 = 0,0063x + 0,0883$$

$$0,179 - 0,0883 = 0,0063x$$

$$0,0907 = 0,0063x$$

$$x = \frac{0,0907}{0,0063}$$

$$x = 14,40 \times 20 = 288 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,2 M ulangan 2

$$0,614 = 0,0063x + 0,0883$$

$$0,614 - 0,0883 = 0,0063x$$

$$0,5257 = 0,0063x$$

$$x = \frac{0,5257}{0,0063}$$

$$x = 83,44 \times 20 = 1.668,8 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,2 M ulangan 3

$$0,713 = 0,0063x + 0,0883$$

$$0,713 - 0,0883 = 0,0063x$$

$$0,6247 = 0,0063x$$

$$x = \frac{0,6247}{0,0063}$$

$$x = 99,16 \times 20 = 1.983,2 \mu\text{g}/\text{ml}$$

$$\text{Rata-rata} = \frac{288+1.668,8+1.983,2}{3} = 1.313,33 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,3 M ulangan 1

$$0,201 = 0,0063x + 0,0883$$

$$0,201 - 0,0883 = 0,0063x$$

$$0,1127 = 0,0063x$$

$$x = \frac{0,1127}{0,0063}$$

$$x = 17,88 \times 20 = 357,6 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,3 M ulangan 2

$$0,351 = 0,0063x + 0,0883$$

$$0,351 - 0,0883 = 0,0063x$$

$$0,2627 = 0,0063x$$

$$x = \frac{0,2627}{0,0063}$$

$$x = 41,70 \times 20 = 834 \mu\text{g}/\text{ml}$$

Konsentrasi NaOH 0,3 M ulangan 3

$$0,372 = 0,0063x + 0,0883$$

$$0,372 - 0,0883 = 0,0063x$$

$$0,2837 = 0,0063x$$

$$x = \frac{0,2837}{0,0063}$$

$$x = 45,03 \times 20 = 900,6 \mu\text{g}/\text{ml}$$

$$\text{Rata-rata} = \frac{357,6 + 834 + 900,6}{3} = 697,4 \mu\text{g}/\text{ml}$$

#### Lampiran 6. Hasil Uji Statistik Total Gula

Descriptives									
	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum	
					Lower Bound	Upper Bound			
rendemen	0,1	3	2.133	.2082	.1202	1.616	2.650	1.9	2.3
	0,2	3	3.333	1.4048	.8110	-.156	6.823	2.0	4.8
	0,3	3	5.400	3.9611	2.2869	-4.440	15.240	1.9	9.7
	Total	9	3.622	2.5445	.8482	1.666	5.578	1.9	9.7
antioksidan	0,1	3	4.4838E2	79.678642	4.6002E1	250.45161	648.31705	357.910	508.095
	0,2	3	3.9259E2	278.372364	1.6071E2	-298.92029	1084.11029	217.981	713.619
	0,3	3	8.0727E2	212.022283	1.2241E2	280.58178	1333.96688	641.176	1046.086
	Total	9	5.4941E2	264.919810	8.8306E1	345.78250	753.05328	217.981	1046.086
totalgula	0,1	3	1.1069E3	375.07340	2.1654E2	175.1994	2038.6673	780.00	1516.40
	0,2	3	1.3133E3	901.77224	5.2063E2	-926.7931	3553.4598	288.00	1983.20
	0,3	3	6.9740E2	296.15354	1.7098E2	-38.2862	1433.0862	357.60	900.60
	Total	9	1.0202E3	579.01064	1.0267E2	504.0171	1493.5274	288.00	1983.20
totalfenolik	0,1	3	2.7526E1	1.580630	.912577	23.59950	31.45250	25.711	28.600
	0,2	3	4.6526E1	6.465891	3.733084	30.46384	62.58816	40.711	53.489
	0,3	3	1.2970E1	.420812	.242956	11.92531	14.01602	12.489	13.267
	Total	9	2.9007E1	14.949016	4.983005	17.51672	40.49839	12.489	53.489
sulfat	0,1	3	6.7600	3.70340	2.13816	-2.4398	15.9598	3.32	10.68
	0,2	3	91.6133	30.86055	17.81734	14.9515	168.2752	64.28	125.08
	0,3	3	18.8133	7.17031	4.13978	1.0013	36.6254	14.28	27.08
	Total	9	39.0622	42.83721	14.27907	6.1346	71.9898	3.32	125.08

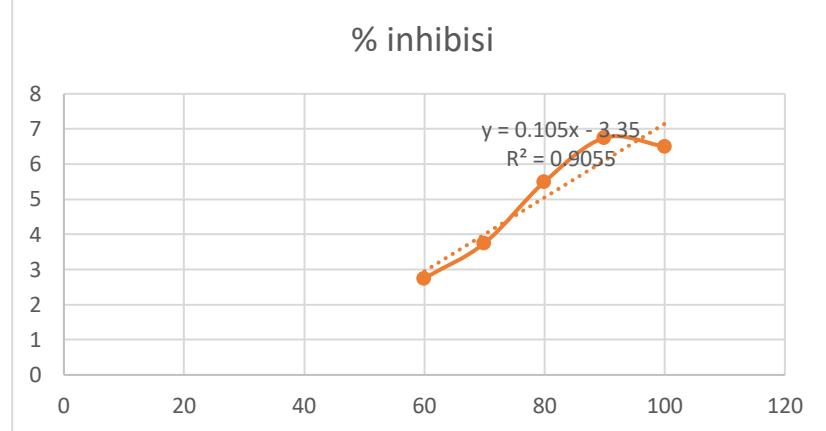
**ANOVA**

		Sum of Squares	df	Mean Square	F	Sig.
rendemen	Between Groups	16.382	2	8.191	1.388	.320
	Within Groups	35.413	6	5.902		
	Total	51.796	8			
antioksidan	Between Groups	303874.755	2	151937.378	3.539	.097
	Within Groups	257586.367	6	42931.061		
	Total	561461.122	8			
totalgula	Between Groups	589692.382	2	294846.191	.849	.473
	Within Groups	2083160.293	6	347193.382		
	Total	2672852.676	8			
totalfenolik	Between Groups	1698.818	2	849.409	57.285	.000
	Within Groups	88.966	6	14.828		
	Total	1787.785	8			
sulfat	Between Groups	12645.211	2	6322.606	18.642	.003
	Within Groups	2035.004	6	339.167		
	Total	14680.215	8			

#### Lampiran 7. Data Hasil Penelitian Antioksidan

- Konsentrasi NaOH 0.1 M Ulangan 1

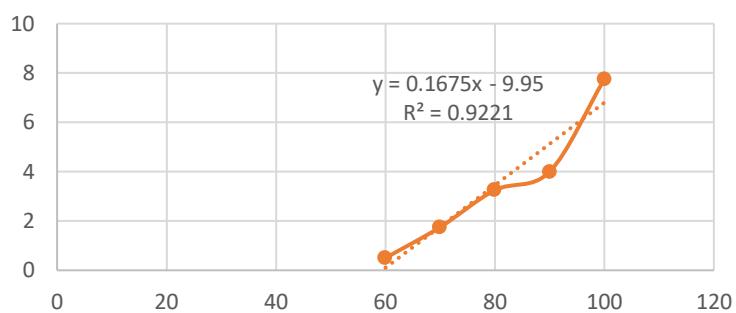
0,1.1			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,389	0,400	2,75
70	0,385	0,400	3,75
80	0,378	0,400	5,5
90	0,373	0,400	6,75
100	0,374	0,400	6,5
IC50	508,095238		



- Konsentrasi NaOH 0.1 M Ulangan 2

0,1.2			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,398	0,400	0,5
70	0,393	0,400	1,75
80	0,387	0,400	3,25
90	0,384	0,400	4
100	0,369	0,400	7,75
IC50	357,9104478		

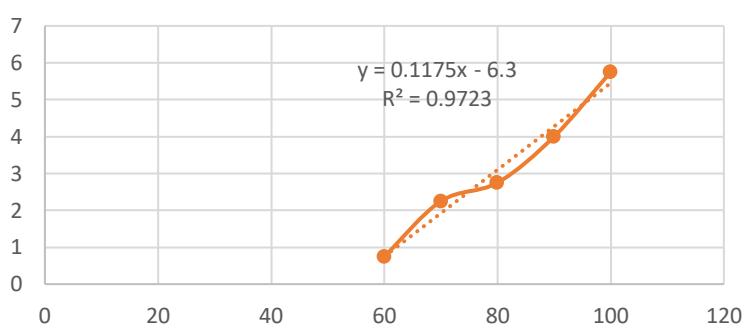
% inhibisi



- Konsentrasi NaOH 0.1 M Ulangan 3

0,1.3			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,397	0,400	0,75
70	0,391	0,400	2,25
80	0,389	0,400	2,75
90	0,384	0,400	4
100	0,377	0,400	5,75
IC50	479,1489362		

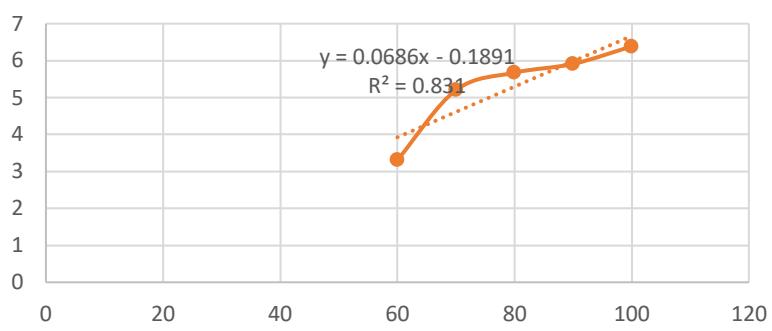
% inhibisi



- Konsentrasi NaOH 0.2 M Ulangan 1

0,2.1			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,409	0,423	3,3096927
70	0,401	0,423	5,2009456
80	0,399	0,423	5,6737589
90	0,398	0,423	5,9101655
100	0,396	0,423	6,3829787
IC50	731,619534		

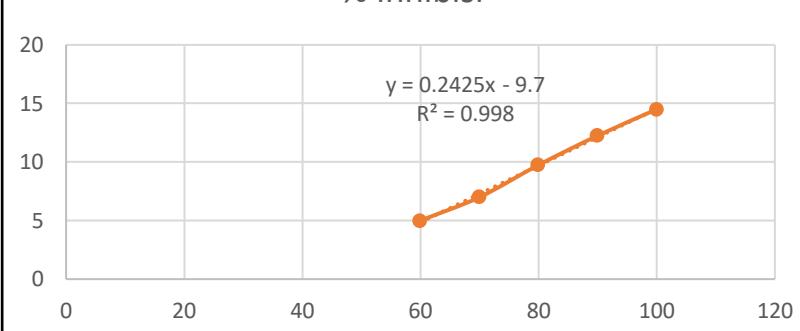
% inhibisi



- Konsentrasi NaOH 0.2 M Ulangan 2

0,2.2			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,380	0,400	5
70	0,372	0,400	7,000
80	0,361	0,400	9,750
90	0,351	0,400	12,250
100	0,342	0,400	14,5
IC50	246,185567		

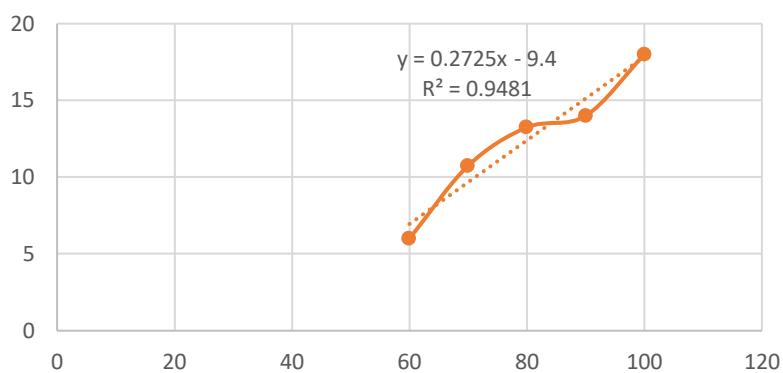
% inhibisi



- Konsentrasi NaOH 0.2 M Ulangan 3

0,2.3			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,376	0,400	6
70	0,357	0,400	10,75
80	0,347	0,400	13,25
90	0,344	0,400	14
100	0,328	0,400	18
IC50	217,981651		

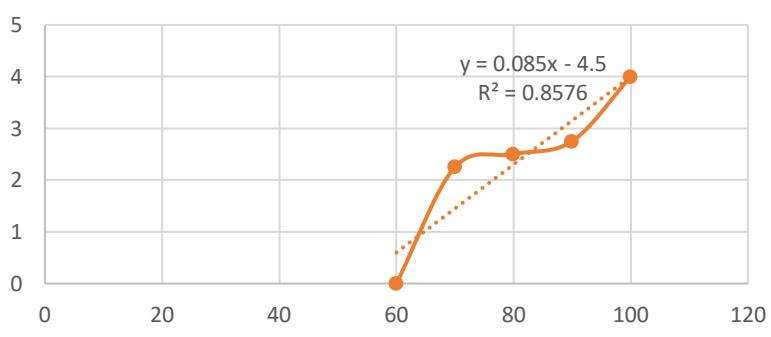
% inhibisi



- Konsentrasi NaOH 0.3 M Ulangan 1

0,3.1			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,400	0,400	0
70	0,391	0,400	2,25
80	0,390	0,400	2,5
90	0,389	0,400	2,75
100	0,384	0,400	4
IC50	641,176471		

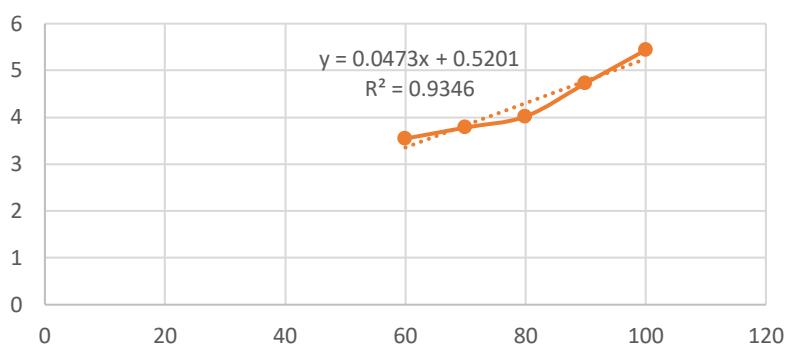
% inhibisi



- Konsentrasi NaOH 0.3 M Ulangan 2

0,3.2			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,408	0,423	3,5460993
70	0,407	0,423	3,7825059
80	0,406	0,423	4,0189125
90	0,403	0,423	4,7281324
100	0,400	0,423	5,4373522
IC50	1046,08668		

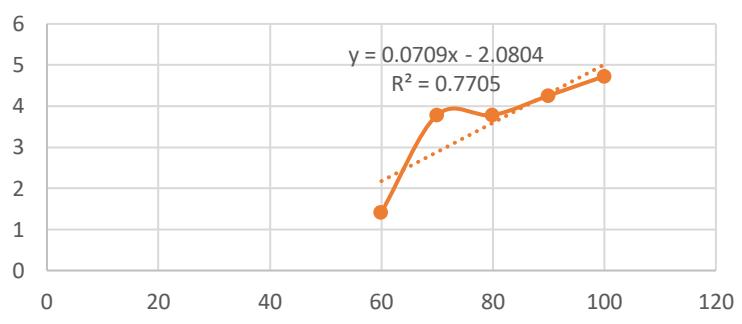
% inhibisi



- Konsentrasi NaOH 0.3 M Ulangan 3

0,3.3			
Konsentrasi (ppm)	Absorbansi	Kontrol	% inhibisi
60	0,417	0,423	1,418439716
70	0,407	0,423	3,78250591
80	0,407	0,423	3,78250591
90	0,405	0,423	4,255319149
100	0,403	0,423	4,728132388
IC50	734,561354		

% inhibisi



## Lampiran 8. Hasil Uji Statistik Antioksidan

**Descriptives**

		N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
						Lower Bound	Upper Bound		
rendemen	0,1	3	2.133	.2082	.1202	1.616	2.650	1.9	2.3
	0,2	3	3.333	1.4048	.8110	-.156	6.823	2.0	4.8
	0,3	3	5.400	3.9611	2.2869	-4.440	15.240	1.9	9.7
	Total	9	3.622	2.5445	.8482	1.666	5.578	1.9	9.7
antioksidan	0,1	3	4.4838E2	79.678842	4.6002E1	250.45161	646.31705	357.910	508.095
	0,2	3	3.9259E2	278.372364	1.6071E2	-298.92029	1084.11029	217.981	713.619
	0,3	3	8.0727E2	212.022283	1.2241E2	280.58178	1333.96688	641.176	1046.086
	Total	9	5.4941E2	264.919810	8.8306E1	345.78250	753.05328	217.981	1046.086
totalgula	0,1	3	1.1069E3	375.07340	2.1654E2	175.1994	2038.6673	780.00	1516.40
	0,2	3	1.3133E3	901.77224	5.2063E2	-926.7931	3553.4598	288.00	1983.20
	0,3	3	6.9740E2	296.15354	1.7098E2	-38.2862	1433.0862	357.60	900.60
	Total	9	1.0392E3	578.01954	1.9267E2	594.9171	1483.5274	288.00	1983.20
totalfenolik	0,1	3	2.7526E1	1.580630	.912577	23.59950	31.45250	25.711	28.600
	0,2	3	4.6526E1	6.465891	3.733084	30.46384	62.58816	40.711	53.489
	0,3	3	1.2970E1	.420812	.242956	11.92531	14.01602	12.489	13.267
	Total	9	2.9007E1	14.949016	4.983005	17.51672	40.49839	12.489	53.489
sulfat	0,1	3	6.7600	3.70340	2.13816	-2.4398	15.9598	3.32	10.68
	0,2	3	91.6133	30.86055	17.81734	14.9515	168.2752	64.28	125.08
	0,3	3	18.8133	7.17031	4.13978	1.0013	36.6254	14.28	27.08
	Total	9	39.0622	42.83721	14.27907	6.1346	71.9898	3.32	125.08

**ANOVA**

		Sum of Squares	df	Mean Square	F	Sig.
rendemen	Between Groups	16.382	2	8.191	1.388	.320
	Within Groups	35.413	6	5.902		
	Total	51.796	8			
antioksidan	Between Groups	303874.755	2	151937.378	3.539	.097
	Within Groups	257586.367	6	42931.061		
	Total	561461.122	8			
totalgula	Between Groups	589692.382	2	294846.191	.849	.473
	Within Groups	2083160.293	6	347193.382		
	Total	2672852.676	8			
totalfenolik	Between Groups	1698.818	2	849.409	57.285	.000
	Within Groups	88.966	6	14.828		
	Total	1787.785	8			
sulfat	Between Groups	12645.211	2	6322.606	18.642	.003
	Within Groups	2035.004	6	339.167		
	Total	14680.215	8			

### Lampiran 9. Data Hasil Penelitian Total Fenolik

Persamaan  $y = 0.0018x + 0.0296$

Panjang gelombang	Perlakuan	Absorbansi	Berat hasil evaporation (gram)	Berat sampel ditimbang (gr)/10 ml EtOH 70%	FP	Konsentrasi ( $\mu\text{g/mL}$ )	Total Fenol dalam 0,1 gr ( $\mu\text{g}$ )	Total Fenol dalam 1 gr (mg)	Total Fenol dalam berat hasil eva (mg)	Total Fenol mg GAE/gr	Rata-rata
782	0,1	0,287	2,2	0,1	2	286,000	2860,000	28,600	62,920	28,600	27,526
		0,261	1,9	0,1		257,111	2571,111	25,711	56,564	25,711	
		0,284	2,3	0,1		282,667	2826,667	28,267	62,187	28,267	
	0,2	0,511	3,3	0,1		534,889	5348,889	53,489	176,513	53,489	46,526
		0,438	2	0,1		453,778	4537,778	45,378	90,756	45,378	
		0,396	4,8	0,1		407,111	4071,111	40,711	195,413	40,711	
	0,3	0,149	4,6	0,1		132,667	1326,667	13,267	61,027	13,267	12,970
		0,148	1,9	0,1		131,556	1315,556	13,156	24,996	13,156	
		0,142	9,7	0,1		124,889	1248,889	12,489	121,142	12,489	

### Lampiran 10. Hasil Uji Statistik Total Fenolik

#### Descriptives

		N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
						Lower Bound	Upper Bound		
rendemen	0,1	3	2.133	.2082	.1202	1.616	2.650	1.9	2.3
	0,2	3	3.333	1.4048	.8110	-.156	6.823	2.0	4.8
	0,3	3	5.400	3.9611	2.2869	-4.440	15.240	1.9	9.7
	Total	9	3.622	2.5445	.8482	1.666	5.578	1.9	9.7
antioksidan	0,1	3	4.4838E2	79.678642	4.6002E1	250.45161	648.31705	357.910	508.095
	0,2	3	3.9259E2	278.372364	1.6071E2	-298.92029	1084.11029	217.981	713.619
	0,3	3	8.0727E2	212.022283	1.2241E2	280.58178	1333.96688	641.176	1046.086
	Total	9	5.4941E2	264.919810	8.8306E1	345.78250	753.05328	217.981	1046.086
totalgula	0,1	3	1.1069E3	375.07340	2.1654E2	175.1994	2038.6673	780.00	1516.40
	0,2	3	1.3133E3	901.77224	5.2063E2	-926.7931	3553.4598	288.00	1983.20
	0,3	3	6.9740E2	296.15354	1.7098E2	-38.2862	1433.0862	357.60	900.60
	Total	9	1.0392E3	578.01954	1.9267E2	594.9171	1483.5274	288.00	1983.20
totalfenolik	0,1	3	2.7526E1	1.580630	.912577	23.59950	31.45250	25.711	28.600
	0,2	3	4.6526E1	6.465891	3.733084	30.46384	62.58816	40.711	53.489
	0,3	3	1.2970E1	.420812	.242956	11.92531	14.01602	12.489	13.267
	Total	9	2.9007E1	14.949016	4.983005	17.51672	40.49839	12.489	53.489
sulfat	0,1	3	6.7600	3.70340	2.13816	-2.4398	15.9598	3.32	10.68
	0,2	3	91.6133	30.86055	17.81734	14.9515	168.2752	64.28	125.08
	0,3	3	18.8133	7.17031	4.13978	1.0013	36.6254	14.28	27.08
	Total	9	39.0622	42.83721	14.27907	6.1346	71.9898	3.32	125.08

**ANOVA**

		Sum of Squares	df	Mean Square	F	Sig.
rendemen	Between Groups	18.382	2	8.191	1.388	.320
	Within Groups	35.413	6	5.902		
	Total	51.796	8			
antioksidan	Between Groups	303874.755	2	151937.378	3.539	.097
	Within Groups	257586.367	6	42931.061		
	Total	561461.122	8			
totalgula	Between Groups	589692.382	2	294846.191	.849	.473
	Within Groups	2083160.293	6	347193.382		
	Total	2672852.676	8			
totalfenolik	Between Groups	1698.818	2	849.409	57.285	.000
	Within Groups	88.966	6	14.828		
	Total	1787.785	8			
sulfat	Between Groups	12645.211	2	6322.606	18.642	.003
	Within Groups	2035.004	6	339.167		
	Total	14680.215	8			

Lampiran 11. Hasil Uji Lanjut Duncan Total Fenolik

**totalfenolik**

Duncan		Subset for alpha = 0.05		
rendeme n	N	1	2	3
0,3	3	12.9707		
0,1	3		27.5260	
0,2	3			46.5260
Sig.		1.000	1.000	1.000

Lampiran 12. Data Hasil Penelitian Total Sulfat

Konsentrasi NaOH	Absorbansi		
	Ulangan 1	Ulangan 2	Ulangan 3
0,1 M	0,141	0,117	0,152
0,2 M	0,438	0,286	0,339
0,3 M	0,163	0,193	0,161

Lampiran 13. Perhitungan Total Sulfat

$$Y = 0,005x + 0,1253$$

$$FP = 2$$

Konsentrasi NaOH 0,1 M ulangan 1

$$0,141 = 0,005x + 0,1253$$

$$0,141 - 0,1253 = 0,005x$$

$$0,0157 = 0,005x$$

$$x = \frac{0,0157}{0,005} = 3,14 \times 2 = 6,28 \text{ mg/ml}$$

Konsentrasi NaOH 0,1 M ulangan 2

$$0,117 = 0,005x + 0,1253$$

$$0,117 - 0,1253 = 0,005x$$

$$0,0083 = 0,005x$$

$$x = \frac{0,0083}{0,005} = 1,66 \times 2 = 3,32 \text{ mg/ml}$$

Konsentrasi NaOH 0,1 M ulangan 3

$$0,152 = 0,005x + 0,1253$$

$$0,152 - 0,1253 = 0,005x$$

$$0,0152 = 0,005x$$

$$x = \frac{0,152}{0,005} = 5,34 \times 2 = 10,68 \text{ mg/ml}$$

$$\text{Rata-rata} = \frac{6,28+3,32+10,68}{3} = 6,67 \text{ mg/ml}$$

Konsentrasi NaOH 0,2 M ulangan 1

$$0,438 = 0,005x + 0,1253$$

$$0,438 - 0,1253 = 0,005x$$

$$0,3127 = 0,005x$$

$$x = \frac{0,3127}{0,005} = 62,54 \times 2 = 125,08 \text{ mg/ml}$$

Konsentrasi NaOH 0,2 M ulangan 2

$$0,286 = 0,005x + 0,1253$$

$$0,286 - 0,1253 = 0,005x$$

$$0,1607 = 0,005x$$

$$x = \frac{0,1607}{0,005} = 32,14 \times 2 = 64,28 \text{ mg/ml}$$

Konsentrasi NaOH 0,2 M ulangan 3

$$0,339 = 0,005x + 0,1253$$

$$0,339 - 0,1253 = 0,005x$$

$$0,2137 = 0,005x$$

$$x = \frac{0,2137}{0,005} = 42,74 \times 2 = 85,48 \text{ mg/ml}$$

$$\text{Rata-rata} = \frac{125,08+64,28+85,48}{3} = 91,613 \text{ mg/ml}$$

Konsentrasi NaOH 0,3 M ulangan 1

$$0,163 = 0,005x + 0,1253$$

$$0,163 - 0,1253 = 0,005x$$

$$0,0377 = 0,005x$$

$$x = \frac{0,0377}{0,005} = 7,54 \times 2 = 15,08 \text{ mg/ml}$$

Konsentrasi NaOH 0,3 M ulangan 2

$$0,193 = 0,005x + 0,1253$$

$$0,193 - 0,1253 = 0,005x$$

$$0,0677 = 0,005x$$

$$x = \frac{0,0677}{0,005} = 13,54 \times 2 = 27,08 \text{ mg/ml}$$

Konsentrasi NaOH 0,3 M ulangan 3

$$0,161 = 0,005x + 0,1253$$

$$0,161 - 0,1253 = 0,005x$$

$$0,0357 = 0,005x$$

$$x = \frac{0,0357}{0,005} = 7,14 \times 2 = 14,28 \text{ mg/ml}$$

$$\text{Rata-rata} = \frac{15,08+27,08+14,28}{3} = 18,813 \text{ mg/ml}$$

#### Lampiran 14. Data Hasil Statistik Total Sulfat

Descriptives								
	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
					Lower Bound	Upper Bound		
rendemen	0,1	3	2.133	.2082	.1202	1.616	2.650	1.9 2.3
	0,2	3	3.333	1.4048	.8110	-.156	6.823	2.0 4.8
	0,3	3	5.400	3.9611	2.2869	-4.440	15.240	1.9 9.7
	Total	9	3.622	2.5445	.8482	1.666	5.578	1.9 9.7
antioksidan	0,1	3	4.4838E2	79.678642	4.8002E1	250.45161	646.31705	357.910 508.095
	0,2	3	3.9259E2	278.372364	1.6071E2	-298.92029	1084.11029	217.981 713.619
	0,3	3	8.0727E2	212.022283	1.2241E2	280.58178	1333.96688	641.176 1046.086
	Total	9	5.4941E2	264.919810	8.8306E1	345.78250	753.05328	217.981 1046.086
totalgula	0,1	3	1.1069E3	375.07340	2.1654E2	175.1994	2038.6673	780.00 1516.40
	0,2	3	1.3133E3	901.77224	5.2063E2	-926.7931	3553.4598	288.00 1983.20
	0,3	3	6.9740E2	296.15354	1.7098E2	-38.2862	1433.0862	357.60 900.60
	Total	9	1.0392E3	578.01954	1.9267E2	594.9171	1483.5274	288.00 1983.20
totalfenolik	0,1	3	2.7526E1	1.580630	.912577	23.59950	31.45250	25.711 28.600
	0,2	3	4.6526E1	6.465891	3.733084	30.46384	62.58816	40.711 53.489
	0,3	3	1.2970E1	.420812	.242956	11.92531	14.01602	12.489 13.267
	Total	9	3.000754	11.040046	1.000005	17.54672	49.40000	13.400 50.400
sulfat	0,1	3	6.7600	3.70340	2.13816	-2.4398	15.9598	3.32 10.68
	0,2	3	91.6133	30.86055	17.81734	14.9515	168.2752	64.28 125.08
	0,3	3	18.8133	7.17031	4.13978	1.0013	36.6254	14.28 27.08
	Total	9	39.0622	42.83721	14.27907	6.1346	71.9898	3.32 125.08

**ANOVA**

		Sum of Squares	df	Mean Square	F	Sig.
rendemen	Between Groups	16.382	2	8.191	1.388	.320
	Within Groups	35.413	6	5.902		
	Total	51.796	8			
antioksidan	Between Groups	303874.755	2	151937.378	3.539	.097
	Within Groups	257586.367	6	42931.061		
	Total	561461.122	8			
totalgula	Between Groups	589692.382	2	294846.191	.849	.473
	Within Groups	2083160.293	6	347193.382		
	Total	2672852.676	8			
totalfenolik	Between Groups	1698.818	2	849.409	57.285	.000
	Within Groups	88.966	6	14.828		
	Total	1787.785	8			
sulfat	Between Groups	12645.211	2	6322.606	18.642	.003
	Within Groups	2035.004	6	339.167		
	Total	14680.215	8			

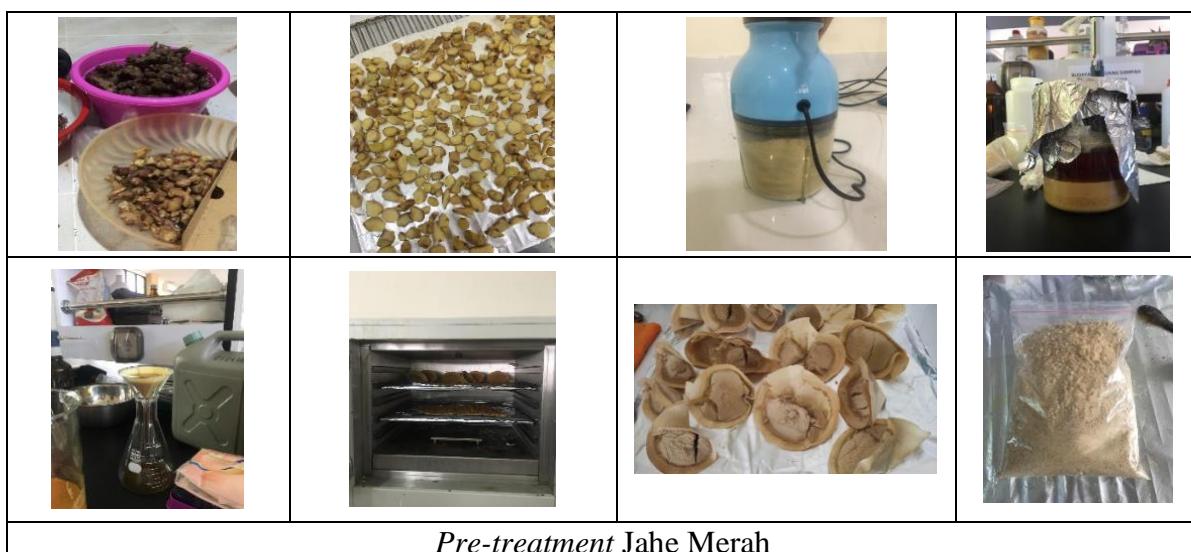
Lampiran 15. Hasil Uji Lanjut Duncan Total Sulfat

**sulfat**

Duncan

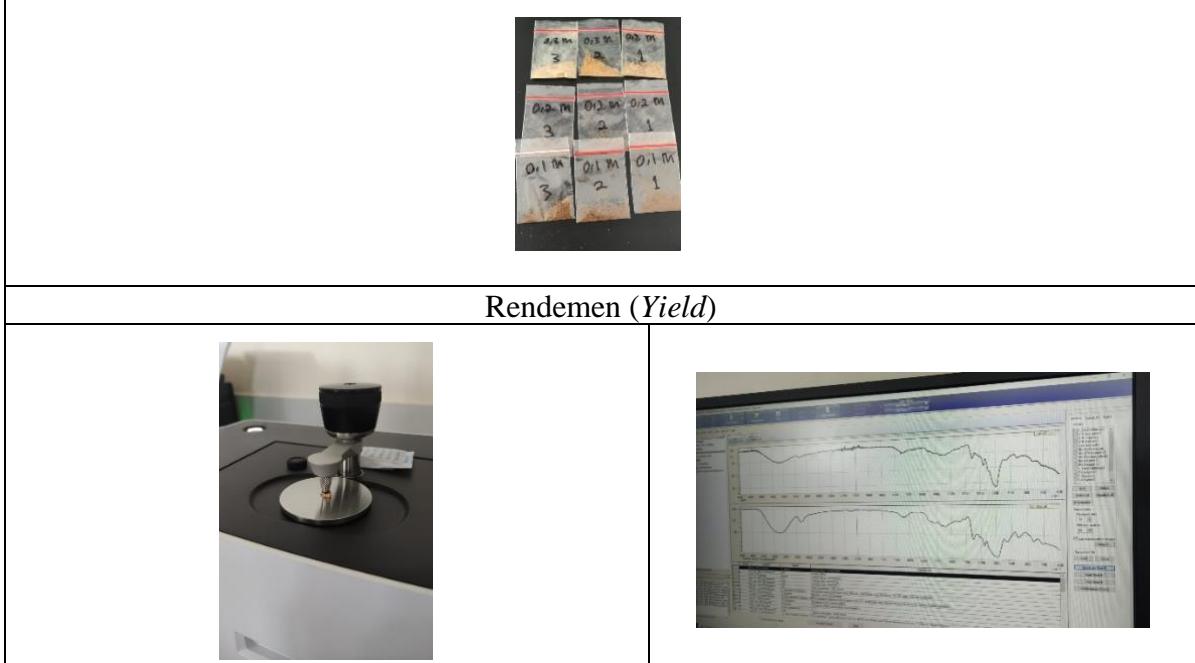
rend eme n	N	Subset for alpha = 0.05	
		1	2
0,1	3	6.7600	
0,3	3	18.8133	
0,2	3		91.6133
Sig.		.453	1.000

Lampiran 16. Dokumentasi Penelitian

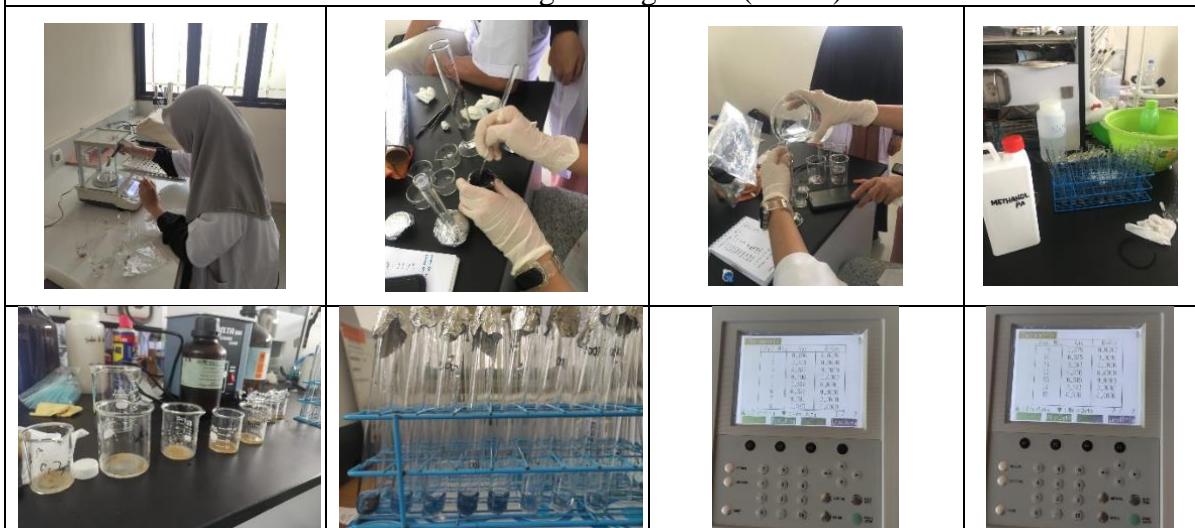




Ekstraksi Polisakarida Metode *Alkaline Solution Extraction*



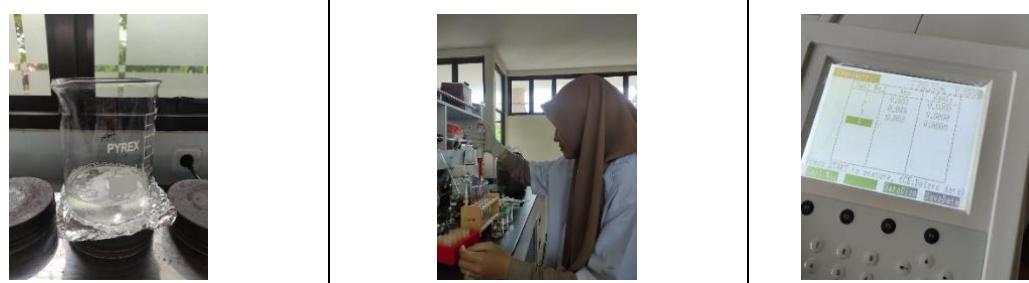
Penentuan Gugus Fungsional (FT-IR)



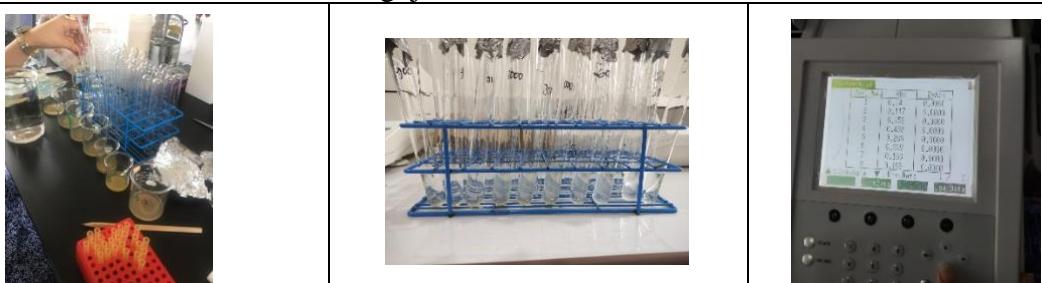
**Penentuan Aktivitas Antioksidan**



**Penentuan Total Gula**



**Pengujian Total Fenolik**



**Penentuan Sulfat**