

DAFTAR PUSTAKA

- Advincula, R.C., 2011, Engineering Molecularly Imprinted Polymer (MIP) Materials: Developments and Challenges for Sensing and Separation Technologies, *Korean Journal of Chemical Engineering*, **28**(6): 1313-1321.
- Agency for Toxic Substances and Disease Registry, 2001, *Toxicological Profile for di-n-butyl phthalate (DBP): Final Report*, ATDSR/TP-135, Public Health Service, US Department of Health and Human Services, Atlanta.
- Al-Ghouti, M.A. dan Da'ana, D.A., 2020, Guidelines for The Use and Interpretation of Adsorption Isotherm Models: A Review, *Journal of Hazardous Materials*, **393**: 122383.
- Amin, S., Damayanti, S. dan Ibrahim, S., 2018, Synthesis and Characterization Molecularly Imprinted Polymers for Analysis of Dimethylamylamine Using Acrylamide as Monomer Functional, *Jurnal Kefarmasian Indonesia*, **8**(2): 76-84.
- Ayob, S., Othman, N., Altowayti, W.A.H., Khalid, F.S., Bakar, N.A., Tahir, M., dan Soedjono, E.D., 2021, A Review on Adsorption of Heavy Metals from Wood-Industrial Waste-Water by Oil Palm Waste, *Journal of Ecological Engineering*, **22**(3): 249-265.
- Bitar, M., Bou-Maroun, E., Lerbret, A., Ouaini, N., dan Cayot, P., 2015, Binding Characteristics of Molecularly Imprinted Polymers Based on Fungicides in Hydroalcoholic Media, *J. Sep. Sci.*, **38**(20): 3607-3614.
- Castillo, M. dan Barceló, D., 2001, Characterisation of Organic Pollutants in Textile Wastewaters and Landfill Leachate by Using Toxicity-Based Fractionation Methods Followed by Liquid and Gas Chromatography Coupled to Mass Spectrometric Detection, *Analytica Chimica Acta*, **426**(2): 253-264.
- Chen, L., Wang, X., Lu, W., Wu, X., dan Li, J., 2016, Molecular Imprinting: Perspectives and Applications, *Chem. Soc. Rev.*, **45**(8): 2137-2211.
- Chen, L., Xu, S. dan Li, J., 2011, Recent Advances in Molecular Imprinting Technology: Current Status, Challenges and Highlighted Applications, *Chemical Society Reviews*, **40**(5): 2922-2942.
- Chen, W., Bin, Q., Bai, Z.W., Zhou, X.P., dan Xie, X.L., 2013, Partial Carbamoylation of Cellulose Microspheres: A New Method to Prepare Adsorbents for Liquid Chromatography, *Chin. J. Polym. Sci.*, **31**(12): 1725-1732.
- Cheong, W.J., Ali, F., Choi, J.H., Lee, J.O., dan Sung, K.Y., 2013, Recent Applications of Molecular Imprinted Polymers for Enantio-Selective

- Recognition, *Talanta*, **106**: 45-59.
- Cormack, P.A.G. dan Elorza, A.Z., 2004, Molecularly Imprinted Polymers: Synthesis and Characterisation, *Journal of Chromatography B*, **804**(1): 173-182.
- Dai, C., Zhang, J., Zhang, Y., Zhou, X., dan Liu, S., 2013, Application of Molecularly Imprinted Polymers to Selective Removal of Clofibric Acid from Water, *Plos One*, **8**(10): 1-8.
- David, R.M., McKee, R.H., Butala, J.H., Barter, R.A., Kayser, M., Bingham, E., Cohrssen, B., dan Powell, C.H., 2001, *Patty's Toxicology: p.635 (Chapter 802)*, Wiley, New York.
- Edet, U.A. dan Ifelebuegu, A.O., 2020, Kinetics, Isotherms, and Thermodynamic Modelling of The Adsorption of Phosphates from Model Wastewater Using Recycled Brick Waste, *Processes*, **8**(6): 665.
- Erkey, C., 2011, *Supercritical Fluids and Organometallic Compounds: From Recovery of Trace Metals to Synthesis of Nanostructured Materials*, Elsevier Science, Turkey.
- Esfandyari, M.M., Javanbakht, M., Shahmoradi, E., Dinarvand, R., dan Atyabi, F., 2013, The Control of Morphological and Size Properties of Carbamazepine-Imprinted Microspheres and Nanospheres under Different Synthesis Conditions, *J. Mater. Res.*, **28**: 2677-2686.
- European Union Council, 2001, *Decision No. 2455/2001/EC Establishing The List of Priority Substances in The Field of Water Policy Amending Directive 2000/60/EC Off. J. Eur. Commun. L331 1*.
- Fauziah, S., Sullahi, F.A., Soekamto, N.H., Taba, P., dan Sapar, A., 2021, Synthesis of Molecularly Imprinted Polymers Using Methacrylic Acid and Ethylene Glycol Dimethacrylate by Precipitation Polymerization Method, *Asian Journal of Chemistry*, **33**(4): 785-788.
- Figueroa, L. E., Brown, M.C., Briscoe, T., Chisam, J., Lewis, D., Westover, J., Brooks, E., Albahadily, F., Bowen, J., dan Ellis, S. B., 2019, Synthesis of A Potential New Internal Standard for The Analytical Determination of Dibutyl Phthalate (DBP) and Monobutyl Phthalate (MBP) in Water Samples, *Proc. Okla. Acad. Sci.*, **99**: 114-119.
- Gabelman, A., 2017, Adsorption Basics: Part 1, *American Institute of Chemical Engineers*, 48-53.
- Gedde, U.W. dan Hedenqvist, M., 2019, *Fundamental Polymer Science: Second Edition*, Springer, Switzerland.
- Gladis, J.M. dan Rao, T.P., 2004, Effect of Porogen Type on The Synthesis of

Uranium Ion Imprinted Polymer Materials for The Preconcentration/Separation of Traces of Uranium, *Microchimica Acta*, **146**(3): 251-258.

Hasanah, A.N., Safitri, N., Zulfa, A., Neli, N., dan Rahayu, D., 2021, Factors Affecting Preparation of Molecularly Imprinted Polymer and Methods on Finding Template-Monomer Interaction as the Key of Selective Properties of the Materials, *Molecules*, **26**(18): 5612.

Hasanah, A.N., Utari, T.N.D. dan Pratiwi, R., 2019, Synthesis of Atenolol-Imprinted Polymers with Methyl Methacrylate as Functional Monomer in Propanol Using Bulk and Precipitation Polymerization Method, *Journal of Analytical Methods in Chemistry*, **2019**: 1-7.

He, J., Lv, R., Zhu, J., dan Lu, K., 2010, Selective Solid-Phase Extraction of Dibutyl Phthalate from Soybean Milk Using Molecular Imprinted Polymers, *Analytica Chimica Acta*, **661**(2): 215-221.

Huang, X., Zhang, W., Wu, Z., Li, H., Yang, C., Ma, W., Hui, A., Zeng, Q., Xiong, B., dan Xian, Z., 2020, Computer Simulation Aided Preparation of Molecularly Imprinted Polymers for Separation of Bilobalide, *Journal of Molecular Modelling*, **26**: 198.

Inamuddin, Boddula, R. dan Asiri, A.M., 2020, *Green Sustainable Process for Chemical and Environmental Engineering and Science: Organic Synthesis in Water and Supercritical Water (Chapter 9)*, Elsevier, India.

Jones, R. G., Kahovec, J., Stepto, R., Wilks, E. S., Hess, M., dan Kitayama, T., 2009, *Compendium of Polymer Terminology and Nomenclature- IUPAC Recommendations 2008*, Royal Society of Chemistry, Cambridge.

Junko, S., 2001, Determination of Bisphenol A in Blood Using High-Performance Liquid Chromatography-Electrochemical Detection with Solid-Phase Extraction, *Journal of Chromatography B*, **755**: 9-15.

Karim, K., Breton, F., Rouillon, R., Piletska, E.V., Guerreiro, A., Chianella, I., dan Piletsky, S.A., 2005, How to Find Effective Functional Monomers for Effective Molecularly Imprinted Polymers ?, *Adv. Drug Deliv. Rev.*, **57**: 1795-1808.

Kitabatake, T., Tabo, H., Matsunaga, H., dan Haginaka, J., 2013, Preparation of Monodisperse Curcumin-Imprinted Polymer by Precipitation Polymerization and Its Application for The Extraction of Curcuminoids from Curcuma Longa L., *Anal. Bioanal. Chem.*, **405**: 6555-6561.

Lah, N.F.C., Ahmad, A.L., Low, S.C., dan Shoparwe, N.F., 2019, The Role of Porogen-Polymer Complexation in Atrazine Imprinted Polymer to Work as An Electrochemical Sensor in Water, *Journal of Environmental Chemical Engineering*, **103500**: 1-24.

- Lai, J.P., Yang, M.L., Niessner, R., dan Knopp, D., 2007, Molecularly Imprinted Microspheres and Nanospheres for di-(2-ethylhexyl) phthalate Prepared by Precipitation Polymerization, *Anal. Bioanal. Chem.*, **389**: 405-412.
- Li, S., Ge, Y., Tiwari, A., Wang, S., Turner, A.P.F., dan Piletsky S.A., 2011, “On/Off”-Switchable Catalysis by A Smart Enzyme-Like Imprinted Polymer, *J. Catal.*, **278**: 173-180.
- Li, Y., Peng, Z., Zhao, F., Han, Y., Zhao, N., Zhang, L., dan Liu, Y., 2017, Preparation of Molecularly Imprinted Polymeric Microspheres Based on Distillation-Precipitation Polymerization for An Ultrasensitive Electrochemical Sensor, *Analyst*, **142**(7): 1091-1098.
- Liu, Q., Shi, J., Sun, J., Wang, T., Zeng, L., dan Jiang, G., 2011, Graphene and Graphene Oxide Sheets Supported on Silica as Versatile and High-Performance Adsorbents for Solid-Phase Extraction, *Angewandte Chemie International Edition*, **50**: 5913-5917.
- Lorenzo, R.A., Carro, A.M., Alvarez-Lorenzo, dan Conheiro, A., 2011, The Challenge of Extracting The Template to Make The Cavities Available in Molecularly Imprinted Polymers (MIPs), *International Journal of Molecular Science*, **12**(12): 4327-4347.
- Lovell, P.A., 2011, *Introduction of Polymers 2nd*, Chapman and Hall, London.
- Lu, H., Tian, H., Wang, C., dan Xu, S., 2020, Designing and Controlling The Morphology of Spherical Molecularly Imprinted Polymers, *Material Advances*, **1**: 2182-2201.
- Meera, M.S. dan Ganesan, T.K., 2015, Adsorption Isotherm and Kinetics Studies of Cadmium (II) Ions Removal Using Various Activated Carbons Derived from Agricultural Bark Wastes: A Comparative Study, *Journal of Chemical and Pharmaceutical Research*, **7**(4): 1194-1200.
- Metanomski, W.V., 1999, *Compendium of Macromolecular Nomenclature*, Plastic Institute, London.
- Michael, J., Whitecome, M., Rodrigue, E., Villar, P., 1995, A New Method for The Introduction of Recognition Site Functionality into Polymers Prepared by Molecular Imprinting: Synthesis and Characterization of Polymeric Receptors for Cholesterol, *J. Am. Chem. Soc.*, **117**: 7105-7111.
- Mita, L., Stepto, R.F.T. dan Suter, U.W., 1994, Basic Classification and Definitions of Polymerization Reactions, *Pure & Applied Chem.*, **66**(12): 2483-2486.
- Mostafiz, B., Bigdeli, S.A., Banan, K., Afsharara, H., Hatamabadi, D., Mousavi, P., Hussain, C.M., Kecili, R., dan Bidkorbeh, F.G., 2021, Molecularly Imprinted Polymer-Carbon Paste Electrode (MIP-CPE)-Based Sensors for

The Sensitive Detection of Organic and Inorganic Environmental Pollutants: A Review, *Trends in Environmental Analytical Chemistry*, **32**(144): 1-12.

Muhammad, T., Nur, Z., Piletska, E.V., Yimit, O., dan Piletsky, S.A., 2012, Rational Design of Molecularly Imprinted Polymer: The Choice of Cross-Linker, *The Analyst*, **137**(11): 2623-2628.

Naarala, J. dan Korpi, A., 2009, Cell Death and Production of Reactive Oxygen Species by Murine Macrophages After Short Term Exposure to Phthalates, *Toxicology Letters*, **188**: 157-160.

Pengkamta, T., Mala, M., Klakasikit, C., Kanawuttikorn, P., Boonkorn, P., Chuaejedton, A., dan Karuehanon, W., 2020, Synthesis and Evaluation of Molecularly Imprinted Polymer as A Selective Material for Vanillin, *Suan Sunandha Sci. Technol. J.*, **7**: 1–6.

Prasad, B.B., Sharma, P.S. dan Lakshmi, D., 2007, Molecularly Imprinted Polymer-Based Solid-Phase Extraction Combined with Molecularly Imprinted Polymer-Based Sensor for Detection of Uric Acid, *Journal of Chromatography A*, **1173**(1-2): 18-26.

Puig, P., Borrull, F., Calull, M., dan Aguilar, C., 2007, Recent Advances in Coupling Solid-Phase Extraction and Capillary Electrophoresis (SPE-CE), *Trends in Analytical Chemistry*, **26**(7): 664-678.

Ramakrishna, D.M. dan Viraraghavan, T., 1997, Environmental Impact of Chemical Deicers-A Review, *Water Air and Soil Pollution*, **166**(1): 49-63.

Row, K.H., Wang, D., Hong, S.P., dan Yang G., 2003, Caffeine Molecular Imprinted Microgel Spheres by Precipitation Polymerization, *Korean Journal of Chemical Engineering*, **20**(6): 1073-1076.

Sajini, T. dan Mathew B., 2021, A Brief Overview of Molecularly Imprinted Polymers: Highlighting Computational Design, Nano and Photo-Responsive Imprinting, *Talanta Open*, **4**: 1-20.

Sellergren, B., Shen, X., Huang, C., Shinde, S., Switnicka-Plak, M., dan Cormack, P.A.G., 2016, Reflux Precipitation Polymerization: A New Synthetic Insight in Molecular Imprinting at High Temperature, *RSC Adv*, **6**: 81491-81499.

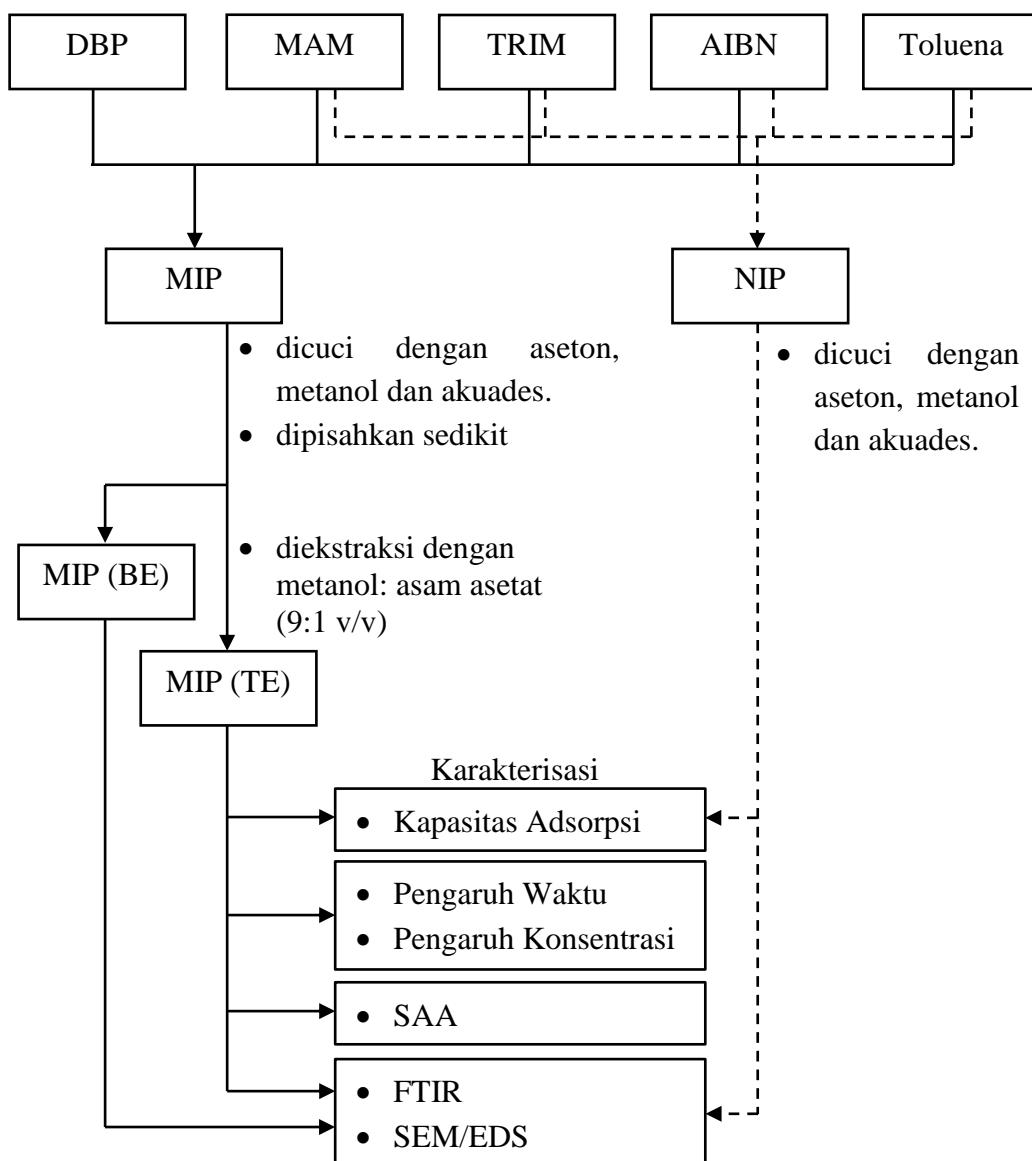
Sirumapea, L., Zulfikar, M.A., Amran, M.B., dan Alni, A., 2018, Studi Awal Sintesis Polimer Bercetakan Meropenem sebagai Sorben yang Selektif-Optimasi Monomer Fungsional, *JKPK (Jurnal Kimia dan Pendidikan Kimia)*, **3**(2): 103-108.

Svenson, J., Karlsson, J.G., Nicholls, I.A., 2004, 1H Nuclear Magnetic Resonance Study of The Molecular Imprinting of (-)-Nicotine: Template Self-

- Association, A Molecular Basis for Cooperative Ligand Binding, *J. Chromatogr. A*, **1024**: 39-44.
- Othman, Z., Habil, M. dan Hashem, A., 2013, Removal of Zinc(II) from Aqueous Solutions Using Modified Agricultural Wastes: Kinetics and Equilibrium Studies, *Arabian Journal Geoscience*, **6**: 4245-4255.
- Tadros, T., 2013, *Adsorption Isotherm: Encyclopedia of Colloid and Interface Science*, Springer, Berlin.
- Tokonami, S., Shiigi, H. dan Nagaoka, T., 2009, Micro and Nanosized Molecularly Imprinted Polymers for High-Throughput Analytical Applications, *Analytica Chimica Acta*, **641**(1-2): 7-13.
- Vasconcelos, I., Reis da Silva, P.H., Dias, D.R.D., Marques, M.B.F., Mussel, W.N., Pedrosa, T.A., Silva, M.E.S.R., Freitas, R.F.S., Sousa, Geraldo de Sousa, R., Fernandes, C., 2020, Synthesis and Characterization of A Molecularly Imprinted Polymer (MIP) for Solid-Phase Extraction of The Antidiabetic Gliclazide from Human Plasma, *Material Science & Engineering C*, **116**: 1-13.
- Verheyen, E., Schillemans, J.P., van Wijk, M., Demenix, M.A., Hennink, W.E., dan van Nostrum, C.F., 2011, Challenges for The Effective Molecular Imprinting of Proteins, *Biomaterials*, **32**(11): 3008-3020.
- Wagner, W.R., Elbert, S.E.S., Zhang, G., dan Yaszemski, M.J., 2020, *Biomaterial Science (Fourth Edition): An Introduction to Material Medicine*, Academic Press, United States of America.
- Wang, W., He, R., Yang, T., Hu, Y., Zhang, N., dan Yang, C., 2018, Three-Dimensional Mesoporous Calcium Carbonate-Silica Frameworks Thermally Activated from Porous Fossil Bryophyte: Adsorption Studies for Heavy Metal Uptake, *RSC Adv.*, **8**: 25754-25766.
- Whitcombe, M.J., Kirsch, N. dan Nicholls, I.A., 2014, Molecular Imprinting Science and Technology: A Survey of The Literature for Years 2004-2011, *Journal of Molecular Recognition*, **27**(6): 297-401.
- White, R.J., Budarin, V., Luque, R., Clark, J.H., dan Macquarrie, D.J., 2009, Tuneable Porous Carbonaceous Materials from Renewable Resources, *Chem. Soc. Rev.*, **38**: 3401-3418.
- Wloch, M. dan Datta, J., 2019, *Synthesis and Polymerisation Techniques of Molecularly Imprinted Polymers: Chapter Two*, p.23, Elsevier, Poland.
- World Health Organization, 2003, *Chapter 8: Chemical Aspects, Guidelines for Drinking Water Quality, Third Edition*, World Health Organization, Geneva.

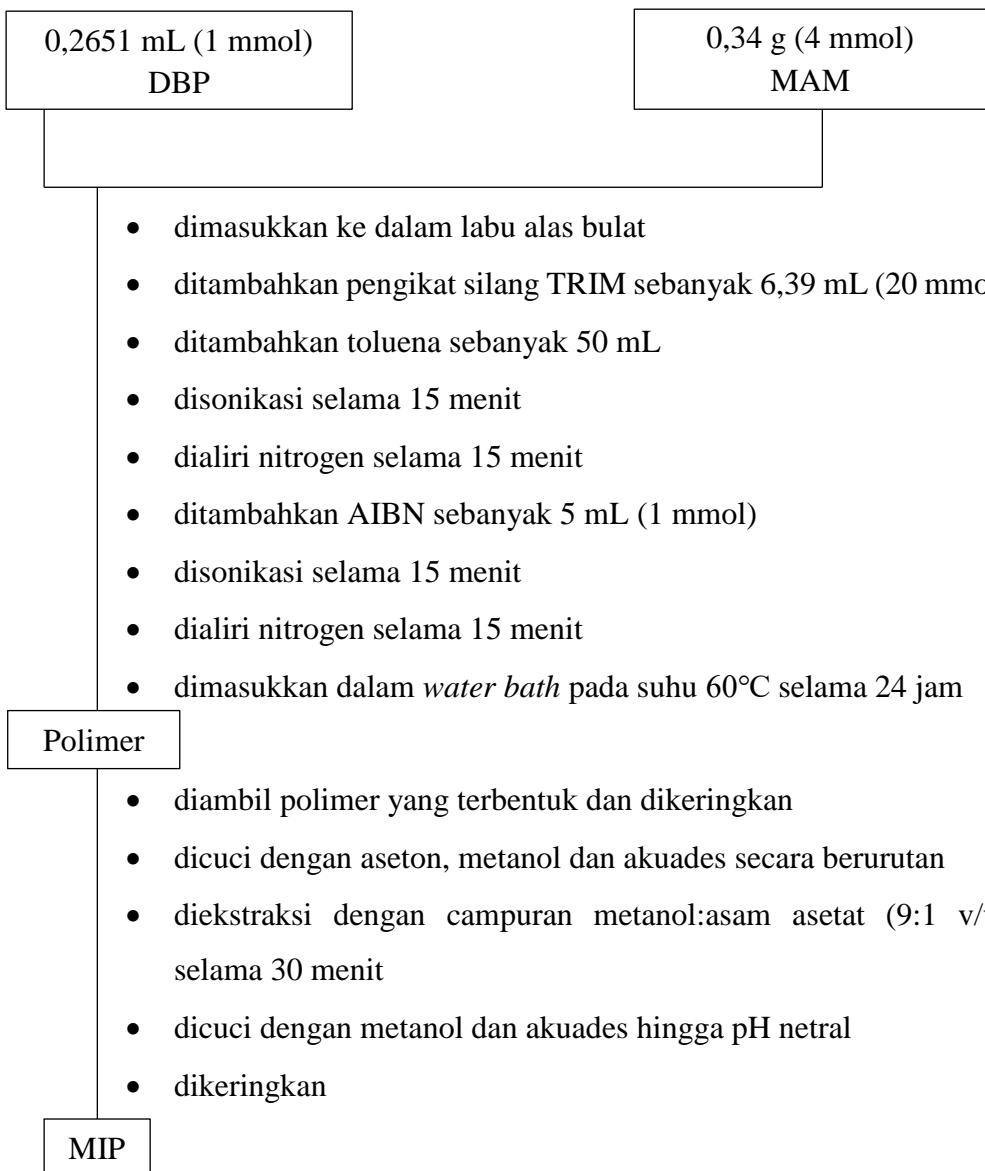
- Wulff, G., Sharhan, A. dan Zabrocki, K., 1973, Enzyme Analogue Built Polymers and Their Use for The Resolution of Racemates, *Tetrahedron Lett.*, **14**: 4329-4332.
- Wulff, G. dan Vietmeier, J., 1989, Enzyme-Analogue Built Polymer 26 Enantioselective Synthesis of Amino Acids Using Polymers Possessing Chiral Cavities Obtained by An Imprinting Procedure with Template Molecules, *Macromolecular Chemistry and Physics Journal*, **190**(1).
- Xie, X., Ma, X. dan Guo, L., 2019, Molecularly Imprinting Polymers for Detection and Removal of Environmental Endocrine Disruptors, *Progress in Chemistry*, **31**(12): 1749-1758.
- Xu, W., Zhang, X., Huang, W., Luan, Y., Yang, Y., Zhu, M., dan Yang, W., 2017, Synthesis of Surface Molecular Imprinted Polymers Based on Carboxyl-Modified Silica Nanoparticles with The Selective Detection of Dibutyl Phthalate from Tap Water Samples, *Applied Surface Science*, **426**: 1075-1083.
- Yan, H. dan Row, K.H., 2006, Characteristic and Synthetic Approach of Molecularly Imprinted Polymer, *Int. J. Mol. Sci.*, **7**(5): 155-178.
- Yu, C. dan Mosbach, K., 2001, Influence of Mobile Phase Composition and Cross-Linker Density on The Enantiomeric Recognition Properties of Molecularly Imprinted Polymers, *Journal of Chromatography*, **888**: 63-72.
- Yusof, N.A., Appribayan, M.D. dan Harson, J., 2010, Synthesis and Characterization of a Molecularly Imprinted Polymer for Pb²⁺ Uptake Using 2-vinylpyridine as The Complexing Monomer, *Sains Malaysiana*, **39**(5): 829-835.
- Zhang, W., She, X., Wang, L., Fan, H., Zhou, Q., Huang, X., dan Tang, J.Z., 2017, Preparation, Characterization and Application of A Molecularly Imprinted Polymer for Selective Recognition of Sulpiride, *Materials*, **10**(475): 1-16.
- Zhang, X., Li, H., Kang, J., Zhu, X., Peng, W., Zhou, H., Zhong, S., dan Wang, Y., 2016, The Synthesis of Temperature-Sensitive Molecularly Imprinted Film on Support Beads and Its Application for Bovine Serum Albumin Separation, *Colloids Surf. A*, **504**: 367-375.
- Zhang, Z., Luo, L., Cai, R., dan Chen, H., 2013, A Sensitive and Selective Molecularly Imprinted Sensor Combined with Magnetic Molecularly Imprinted Solid Phase Extraction for Determination of Dibutyl Phthalate, *Biosensors and Bioelectronics*, **49**: 367-373.

Lampiran 1. Diagram Alir Penelitian



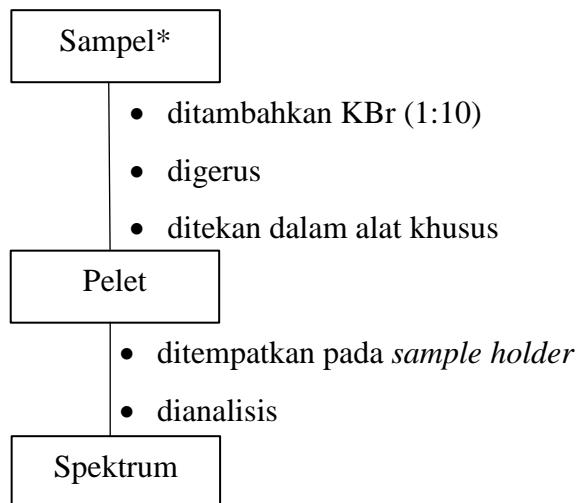
Lampiran 2. Prosedur Penelitian

1. Sintesis MIP dan NIP



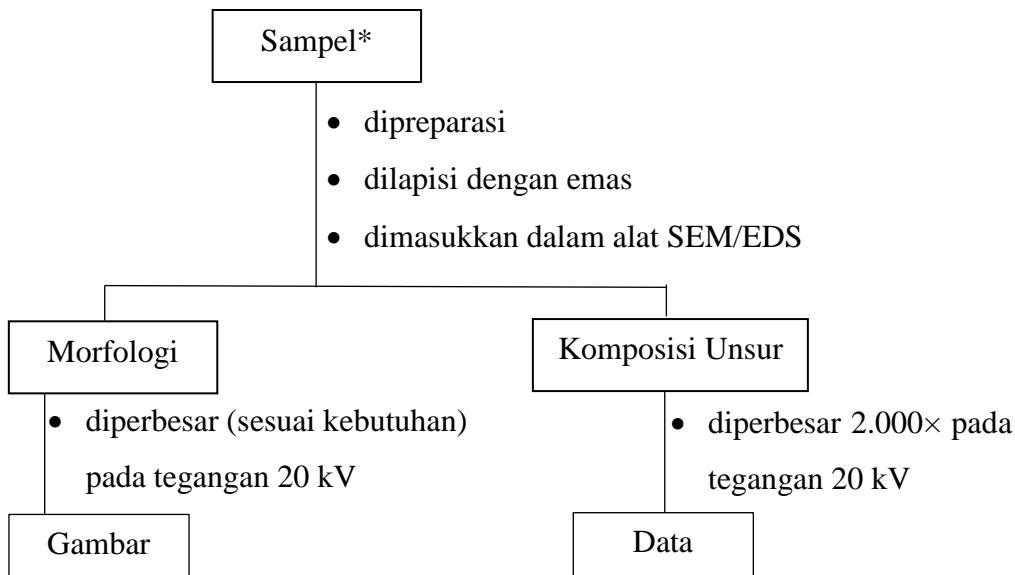
Catatan: Sintesis NIP dibuat dengan metode yang sama tanpa penggunaan DBP dan proses ekstraksi.

2. Analisis Struktur (Gugus Fungsi)



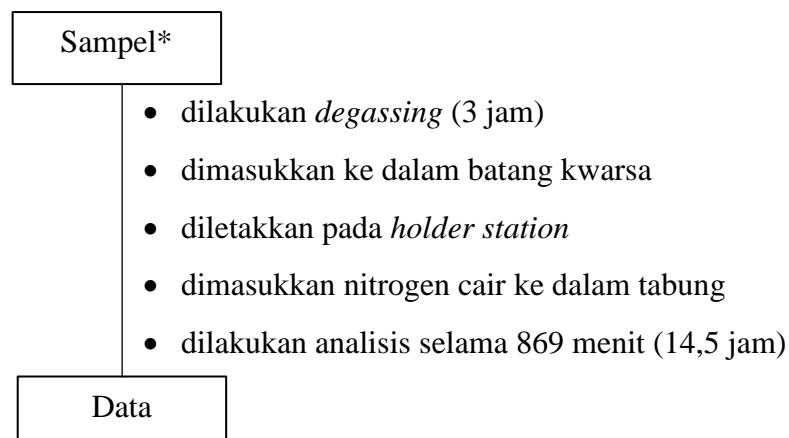
*Jenis sampel yaitu metakrilamida, MIP_DBP_MAM-co-TRIM_(TE), MIP_DBP_MAM-co-TRIM_(BE), dan NIP_MAM-co-TRIM. Analisis dilakukan secara bergantian pada setiap sampel.

3. Analisis Morfologi dan Komposisi Unsur



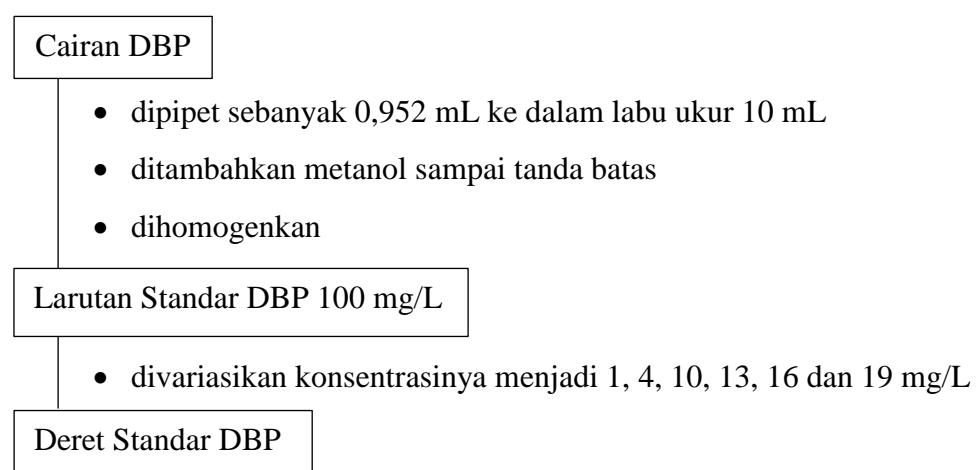
*Jenis sampel yaitu NIP_MAM-co-TRIM, MIP_DBP_MAM-co-TRIM_(BE) dan MIP_DBP_MAM-co-TRIM_(TE). Analisis dilakukan secara bergantian pada setiap sampel.

4. Analisis Luas dan Pori Permukaan

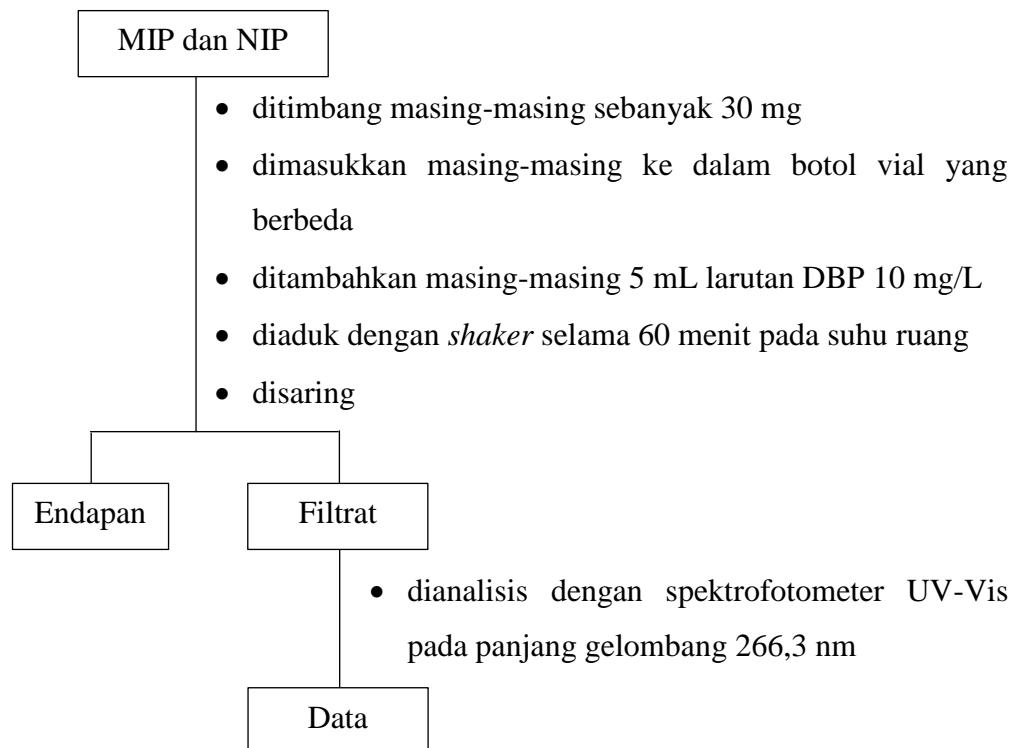


*Jenis sampel yaitu MIP_DBP_MAM-co-TRIM_(TE)

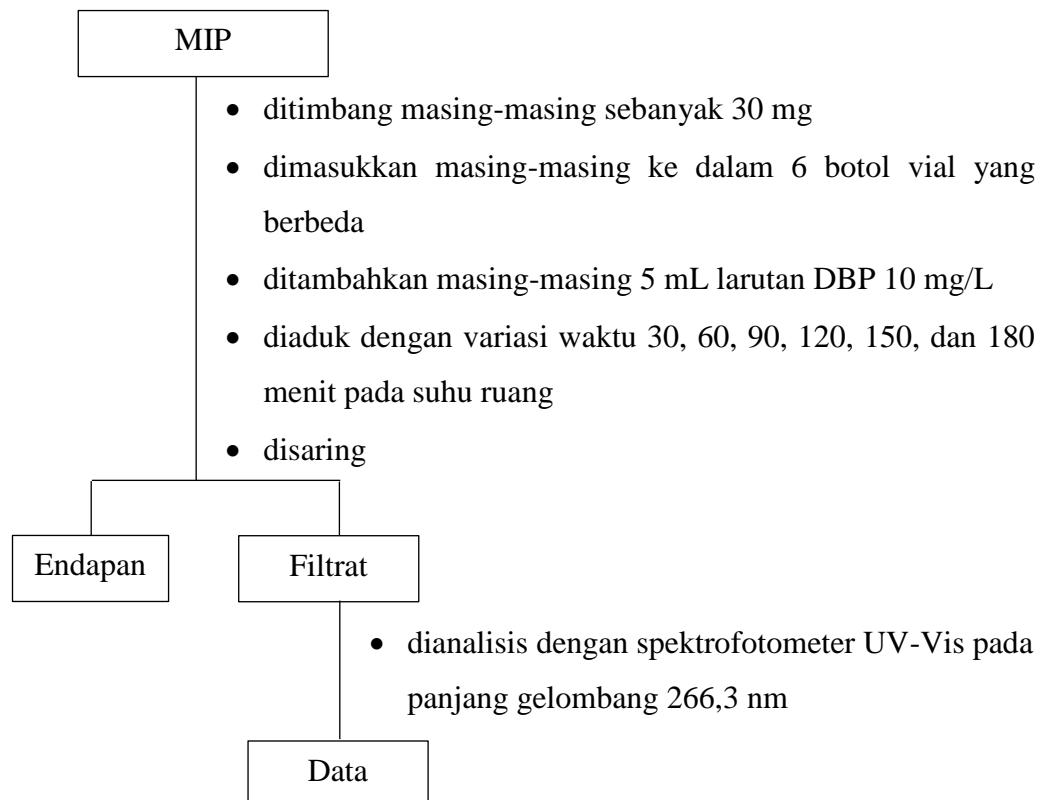
5. Pembuatan Larutan Standar DBP 100 mg/L



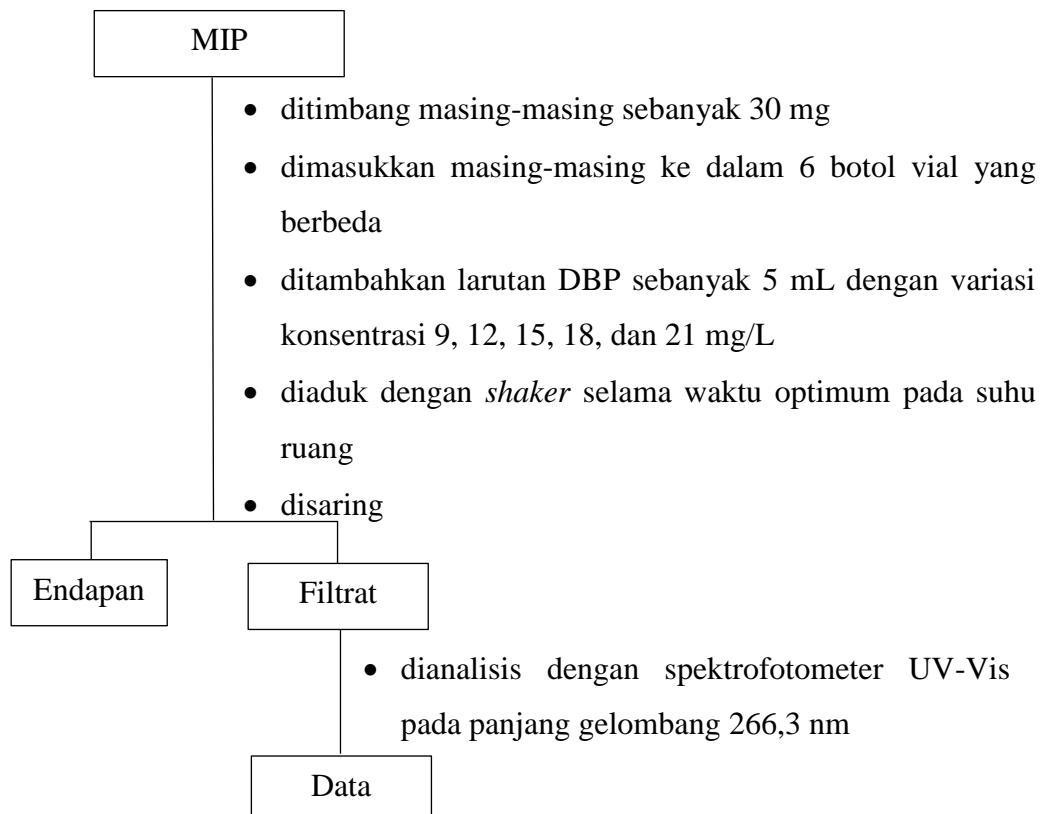
6. Uji Adsorpsi DBP pada MIP dan NIP



7. Uji Pengaruh Waktu terhadap Adsorpsi DBP pada MIP



8. Uji Pengaruh Konsentrasi terhadap Adsorpsi DBP pada MIP

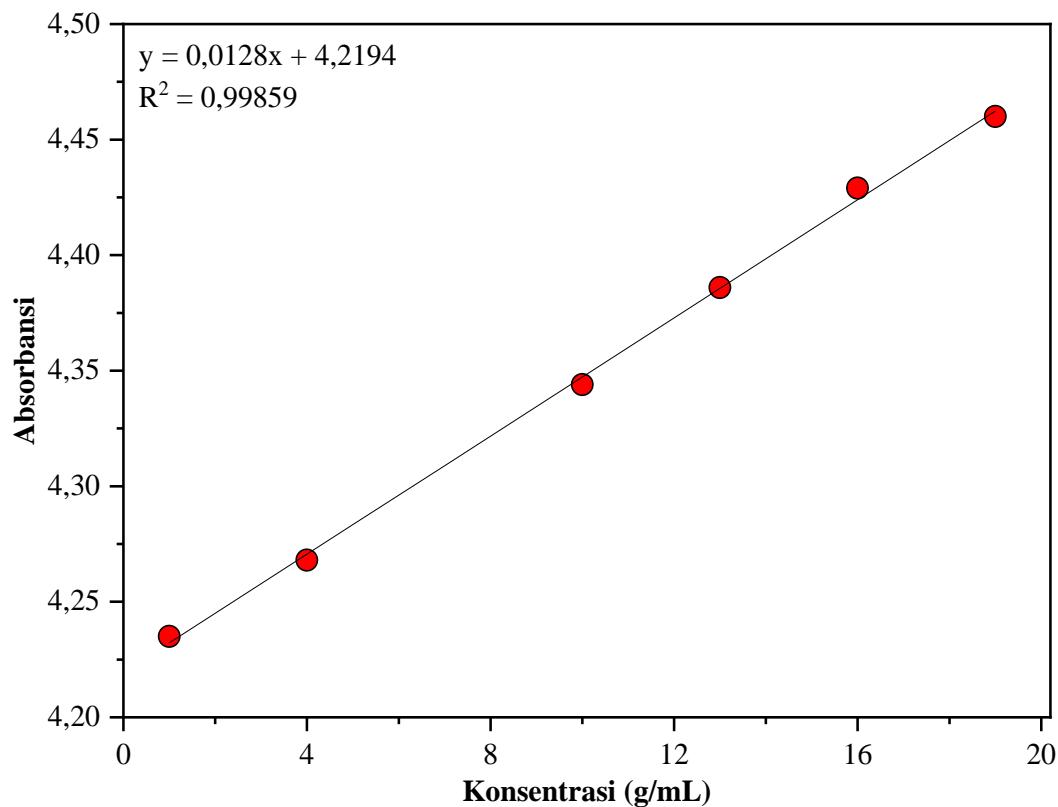


Lampiran 3. Data Spektrofotometer UV-Vis

1. Data Absorbansi Larutan Standar DBP

No.	Sampel	Konsentrasi (mg/L)	Absorbansi
1	DBP	1	4,235
2	DBP	4	4,268
3	DBP	10	4,344
4	DBP	13	4,386
5	DBP	16	4,429
6	DBP	19	4,460

2. Kurva Hubungan antara Absorbansi dan Konsentrasi Larutan Standar DBP



3. Data Absorbansi Larutan setelah Adsorpsi MIP dan NIP

No.	Sampel	Absorbansi	q_t (mg/g)	Δq_t (mg/g)
1	MIP	4,278	0,9036	
2	NIP	4,331	0,2135	0,6901

4. Absorbansi Larutan setelah Adsorpsi DBP pada MIP sebagai Fungsi Waktu

No.	Sampel	Waktu (menit)	Absorbansi
1	DBP	30	4,282
2	DBP	60	4,262
3	DBP	90	4,253
4	DBP	120	4,231
5	DBP	150	4,236
6	DBP	180	4,241

5. Data Penentuan Kinetika Adsorpsi Orde Satu Semu dan Orde Dua Semu

No.	Waktu (menit)	C _t (mg/L)	q _t (mg/g)	q _e -q _t	ln (q _e -q _t)	t/q _t
1	30	4,891	0,852	0,664	-0,409	35,229
2	60	3,328	1,112	0,404	-0,907	53,958
3	90	2,625	1,229	0,286	-1,250	73,220
4	120	0,906	1,516	0	0	79,175
5	150	1,297	1,451	0,065	-2,732	103,411
6	180	1,687	1,385	0,130	-2,039	129,925

Keterangan:

q_e adalah nilai q_t tertinggi

6. Adsorpsi Larutan setelah DBP pada MIP sebagai Fungsi Konsentrasi

No.	Sampel	Konsentrasi (mg/L)	Absorbansi
1	DBP	9	4,268
2	DBP	12	4,300
3	DBP	15	4,331
4	DBP	18	4,367
5	DBP	21	4,399

7. Data Penentuan Isoterm Langmuir dan Freundlich

No.	Sampel	C _o (mg/L)	C _e (mg/L)	q _e (mg/g)	Log C _e	Log q _e	1/C _e	1/q _e
1	DBP	9	3,79687	0,86719	0,57943	-0,06189	0,26337	1,15315
2	DBP	12	6,29687	0,95052	0,79913	-0,02204	0,15881	1,05205
3	DBP	15	8,71875	1,04688	0,94045	0,01989	0,11470	0,95522
4	DBP	18	11,53125	1,07813	1,06188	0,03267	0,08672	0,92754
5	DBP	21	14,03125	1,16146	1,14710	0,06500	0,07127	0,86099

Lampiran 4. Perhitungan

1. Nilai Konsentrasi DBP setelah Adsorpsi pada MIP dan NIP

Digunakan persamaan,

$$y = 0,0128x + 4,2194$$

dengan y adalah nilai absorbansi dan x adalah nilai konsentrasi sisa (mg/L).

a. Adsorpsi oleh MIP

$$y = 4,278$$

$$4,278 = 0,0128x + 4,2194$$

$$x = 4,5781 \text{ mg/L}$$

b. Adsorpsi oleh NIP

$$y = 4,331$$

$$4,331 = 0,0128x + 4,2194$$

$$x = 8,7188 \text{ mg/L}$$

2. Nilai Kapasitas Adsorpsi DBP pada MIP dan NIP

Digunakan persamaan,

$$q_t = \frac{(C_o - C_t)}{m} V$$

dengan, $C_o = 10 \text{ mg/L}$ $m = 0,03 \text{ g}$

$V = 0,005 \text{ L}$ $C_t = \text{Konsentrasi setelah adsorpsi}$

a. Kapasitas Adsorpsi pada MIP

$$q_t = \frac{(10 - 4,5781)}{0,03} 0,005$$

$$= 0,9036 \text{ mg/g}$$

b. Kapasitas Adsorpsi pada NIP

$$q_t = \frac{(10 - 8,7188)}{0,03} 0,005$$

$$= 0,2135 \text{ mg/g}$$

3. Nilai Konsentrasi dan Kapasitas Adsorpsi DBP pada MIP Akibat Pengaruh Waktu

Waktu (menit)	y (absorbansi)	x (mg/L)	q _t (mg/g)
30	4,282	4,891	0,852
60	4,262	3,328	1,112
90	4,253	2,625	1,229
120	4,231	0,906	1,516
150	4,236	1,297	1,451
180	4,241	1,687	1,385

Data contoh perhitungan yang akan diambil adalah data pada menit ke-30.

a. Nilai Konsentrasi DBP setelah Adsorpsi pada MIP

$$\begin{aligned}y &= 4,282 \\4,282 &= 0,0128x + 4,2194 \\x &= 4,891 \text{ mg/L}\end{aligned}$$

b. Nilai Kapasitas Adsorpsi DBP pada MIP

$$\begin{aligned}q_t &= \frac{(10 - 4,891)}{0,03} 0,005 \\&= 0,852 \text{ mg/g}\end{aligned}$$

4. Nilai Konsentrasi dan Kemampuan Adsorpsi DBP pada MIP Akibat Pengaruh Konsentrasi

C ₀ (mg/L)	y (absorbansi)	x (mg/L)	q _e (mg/g)
9	4,268	3,79687	0,86719
12	4,300	6,29687	0,95052
15	4,331	8,71875	1,04688
21	4,367	11,53125	1,07813
24	4,399	14,03125	1,16146

Data contoh perhitungan yang akan diambil adalah data pada konsentrasi awal 9 mg/L.

a. Nilai Konsentrasi DBP setelah Adsorpsi pada MIP

$$\begin{aligned}y &= 4,268 \\4,268 &= 0,0128x + 4,2194 \\x &= 3,79687 \text{ mg/L}\end{aligned}$$

b. Nilai Kapasitas Adsorpsi DBP pada MIP

$$\begin{aligned} q_t &= \frac{(9 - 3,79687)}{0,03} 0,005 \\ &= 0,86719 \text{ mg/g} \end{aligned}$$

5. Persamaan Kinetika Orde Satu Semu dan Orde Dua Semu

a. Persamaan Kinetika Orde Satu Semu

Jika persamaan kinetika orde satu semu pada persamaan (5),

$$\ln(q_e - q_t) = -k_1 t + \ln q_e$$

dikorelasikan dengan persamaan garis pada Gambar 15A,

$$y = -0,0118x + 0,0141$$

maka nilai q_e dan k_1 dapat ditentukan dengan cara berikut,

- $\ln q_e = 0,0141$

$$q_e = 1,0142 \text{ mg/g}$$

- $-k_1 = -0,0118$

$$k_1 = 0,0118 \text{ menit}^{-1}$$

b. Persamaan Kinetika Orde Dua Semu

Jika persamaan kinetika orde dua semu pada persamaan (8),

$$\frac{t}{q_t} = \frac{t}{q_e} + \frac{1}{k_2 q_e^2}$$

dikorelasikan dengan persamaan garis pada Gambar 15B,

$$y = 0,5979x + 16,374$$

maka nilai q_e dan k_2 dapat ditentukan dengan cara berikut,

- $1/q_e = 0,5979$

$$q_e = 1,6725 \text{ mg/g}$$

- $-1/k_2 q_e^2 = 16,374$

$$k_2 = 1/(16,374 \times (1,6725)^2)$$

$$= 0,0218 \text{ g/mg.menit}$$

6. Model Persamaan Isoterm Langmuir dan Isoterm Freundlich

a. Isoterm Langmuir

Jika persamaan laju orde dua semu pada persamaan (2),

$$\frac{1}{q_e} = \frac{1}{K_L q_m} \times \frac{1}{C_e} + \frac{1}{q_m}$$

dikorelasikan dengan persamaan garis pada Gambar 17,

$$y = 1,4458x + 0,7889$$

maka nilai q_m dan K_L dapat ditentukan dengan cara berikut,

- $1/q_m = 0,7889$

$$q_m = 1,2676 \text{ mg/g}$$

- $1/K_L q_m = 1,4458$

$$K_L = 1/(1,4458 \times 1,2676)$$

$$= 0,54565 \text{ L/mg}$$

b. Isoterm Freundlich

Jika persamaan laju orde dua semu pada persamaan (3),

$$\log q_e = \frac{1}{n} \log C_e + \log K_F$$

dikorelasikan dengan persamaan garis pada Gambar 18,

$$y = 0,2183x - 0,191$$

maka nilai K_F dan n dapat ditentukan dengan cara berikut,

- $\log K_F = -0,191$

$$K_F = 0,64417 \text{ mg/g}$$

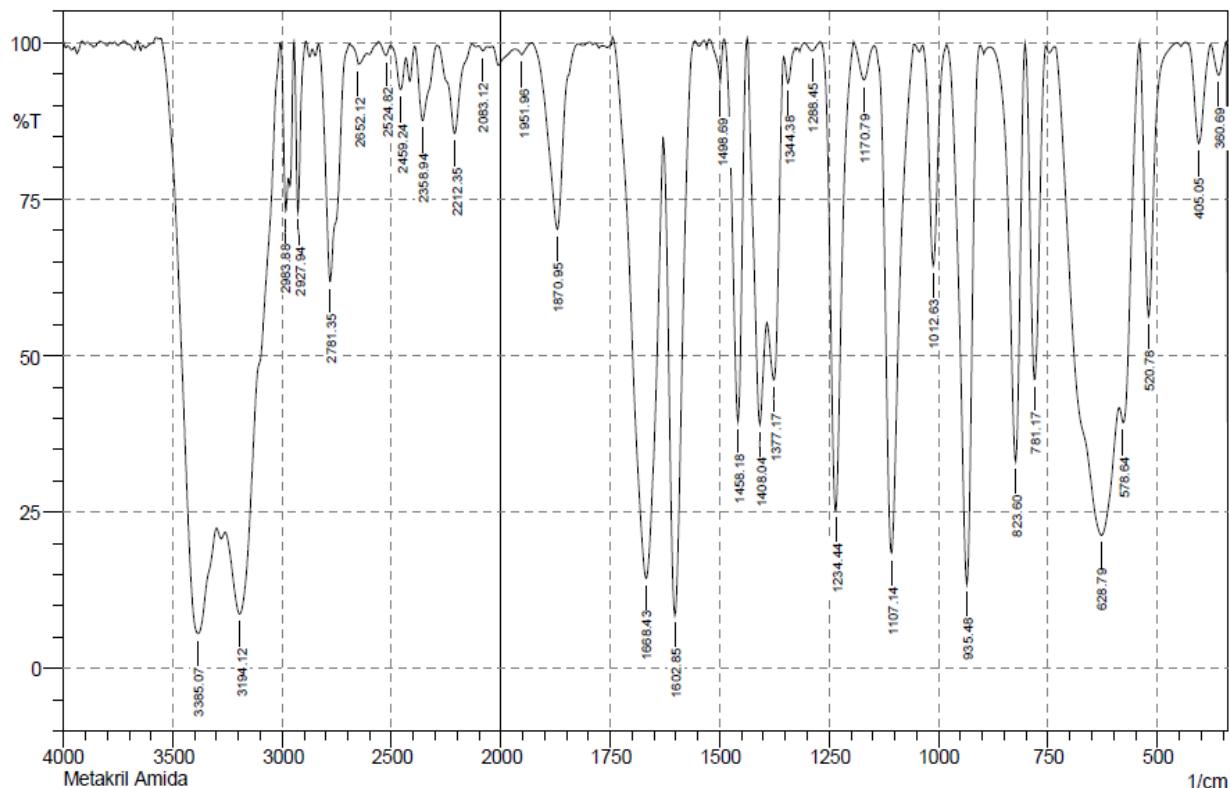
- $1/n = 0,2183$

$$n = 4,5809$$

Lampiran 5. Hasil Analisis FTIR

1. Metakrilamida (MAM)

 SHIMADZU



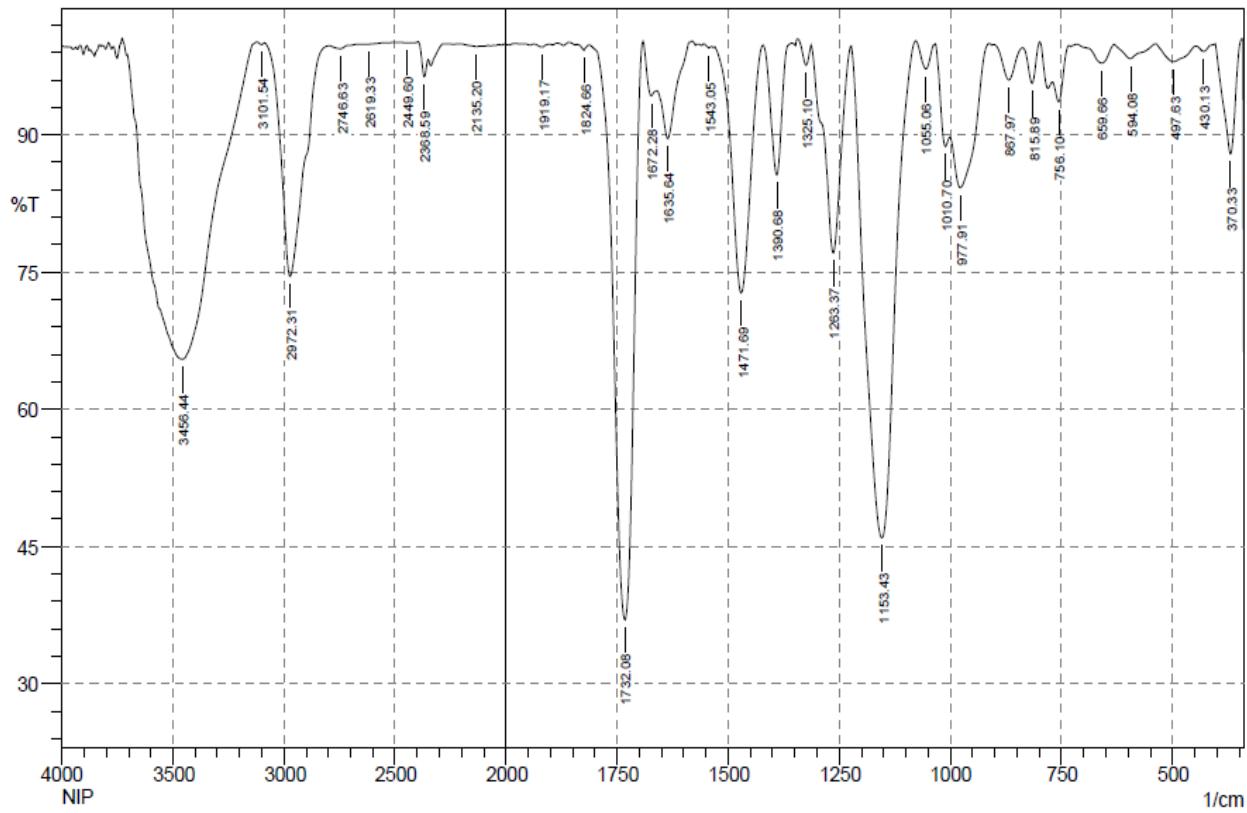
No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	360.69	94.836	5.125	378.05	341.4	0.429	0.425
2	405.05	83.92	15.941	433.98	379.98	1.75	1.719
3	520.78	56.248	43.885	540.07	457.13	5.663	5.691
4	578.64	39.307	12.474	586.36	542	9.943	2.043
5	628.79	21.321	36.483	732.95	588.29	57.625	29.555
6	781.17	46.194	52.961	800.46	756.1	6.872	6.719
7	823.6	33.083	66.628	883.4	802.39	11.387	11.241
8	935.48	13.451	85.972	979.84	906.54	19.987	19.802
9	1012.63	64.391	35.253	1035.77	981.77	3.944	3.864
10	1107.14	18.462	81.114	1145.72	1056.99	19.934	19.779
11	1170.79	94.086	5.762	1193.94	1147.65	0.621	0.59
12	1234.44	25.28	74.864	1269.16	1193.94	14.025	14.073
13	1288.45	98.717	1.335	1305.81	1271.09	0.099	0.107
14	1344.38	93.585	4.687	1354.03	1327.03	0.433	0.253
15	1377.17	46.105	24.578	1390.68	1354.03	7.917	3.229
16	1408.04	39.012	32.058	1436.97	1392.61	10.752	4.867
17	1458.18	39.453	60.86	1489.05	1438.9	8.269	8.343
18	1498.69	93.419	7.15	1523.76	1490.97	0.294	0.376
19	1602.85	8.464	81.318	1627.92	1558.48	25.435	22.893
20	1668.43	14.37	75.957	1743.65	1629.85	39.056	35.144
21	1870.95	70.221	29.711	1928.82	1824.66	5.615	5.586
22	1951.96	98.063	1.323	1967.39	1928.82	0.17	0.084
23	2083.12	98.742	0.692	2102.41	2063.83	0.151	0.056
24	2212.35	85.518	14.181	2289.5	2131.34	4.222	4.018
25	2358.94	87.574	11.924	2393.66	2291.43	2.934	2.722
26	2459.24	92.506	5.692	2503.6	2436.09	1.203	0.757
27	2524.82	97.98	1.8	2559.54	2503.6	0.245	0.209
28	2652.12	98.577	2.424	2692.63	2613.55	0.656	0.316
29	2781.35	61.892	37.875	2833.43	2694.56	12.248	12.102
30	2927.94	72.93	27.225	2945.3	2893.22	2.822	2.86
31	2983.88	73.097	12.454	3007.02	2972.31	2.928	1.005
32	3194.12	8.693	33.535	3259.7	3008.95	127.625	45.621
33	3385.07	5.567	42.518	3554.81	3302.13	143.779	64.221

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No. of Scans:

2. NIP_MAM-co-TRIM

 SHIMADZU



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	370.33	87.922	12.33	403.12	343.33	1.592	1.653
2	430.13	99.117	0.668	447.49	403.12	0.091	0.055
3	497.63	97.992	1.847	536.21	447.49	0.493	0.424
4	594.08	98.335	1.581	632.65	536.21	0.368	0.336
5	659.66	97.826	1.904	694.37	632.65	0.345	0.266
6	756.1	93.599	3.769	771.53	732.95	0.671	0.29
7	815.89	95.628	4.234	837.11	798.53	0.363	0.334
8	867.97	95.999	3.677	902.69	837.11	0.566	0.477
9	977.91	84.232	8.172	1001.06	912.33	4.087	2.055
10	1010.7	88.729	3.645	1031.92	1002.98	1.073	0.33
11	1055.06	97.222	2.787	1078.21	1037.7	0.258	0.265
12	1153.43	45.954	53.997	1222.87	1080.14	21.784	21.76
13	1263.37	77.096	22.498	1311.59	1224.8	4.526	4.371
14	1325.1	97.613	2.465	1344.38	1313.52	0.14	0.161
15	1390.68	85.607	14.254	1419.61	1357.89	1.734	1.701
16	1471.69	72.71	27.006	1529.55	1421.54	6.192	6.056
17	1543.05	99.505	0.259	1554.63	1537.27	0.024	0.01
18	1635.64	89.562	6.868	1658.78	1581.63	1.76	0.937
19	1672.28	94.281	2.558	1691.57	1660.71	0.561	0.218
20	1732.08	36.972	62.954	1813.09	1693.5	18.397	18.344
21	1824.66	99.237	0.608	1840.09	1813.09	0.044	0.026
22	1919.17	99.592	0.311	1938.46	1903.74	0.036	0.022
23	2135.2	99.665	0.161	2212.35	2081.19	0.139	0.05
24	2368.59	96.345	2.637	2395.59	2351.23	0.383	0.226
25	2449.6	100.048	0.027	2490.1	2434.17	-0.018	0.003
26	2619.33	99.895	0.004	2621.26	2544.11	0.005	-0.002
27	2746.63	99.43	0.357	2794.85	2673.34	0.185	0.081
28	2972.31	74.534	25.386	3086.11	2796.78	13.834	13.689
29	3101.54	99.867	0.218	3122.75	3088.03	0.002	0.016
30	3456.44	65.49	33.855	3705.26	3122.75	57.591	56.186

Comment;

NIP

Date/Time; 1/27/2023 4:19:34 PM

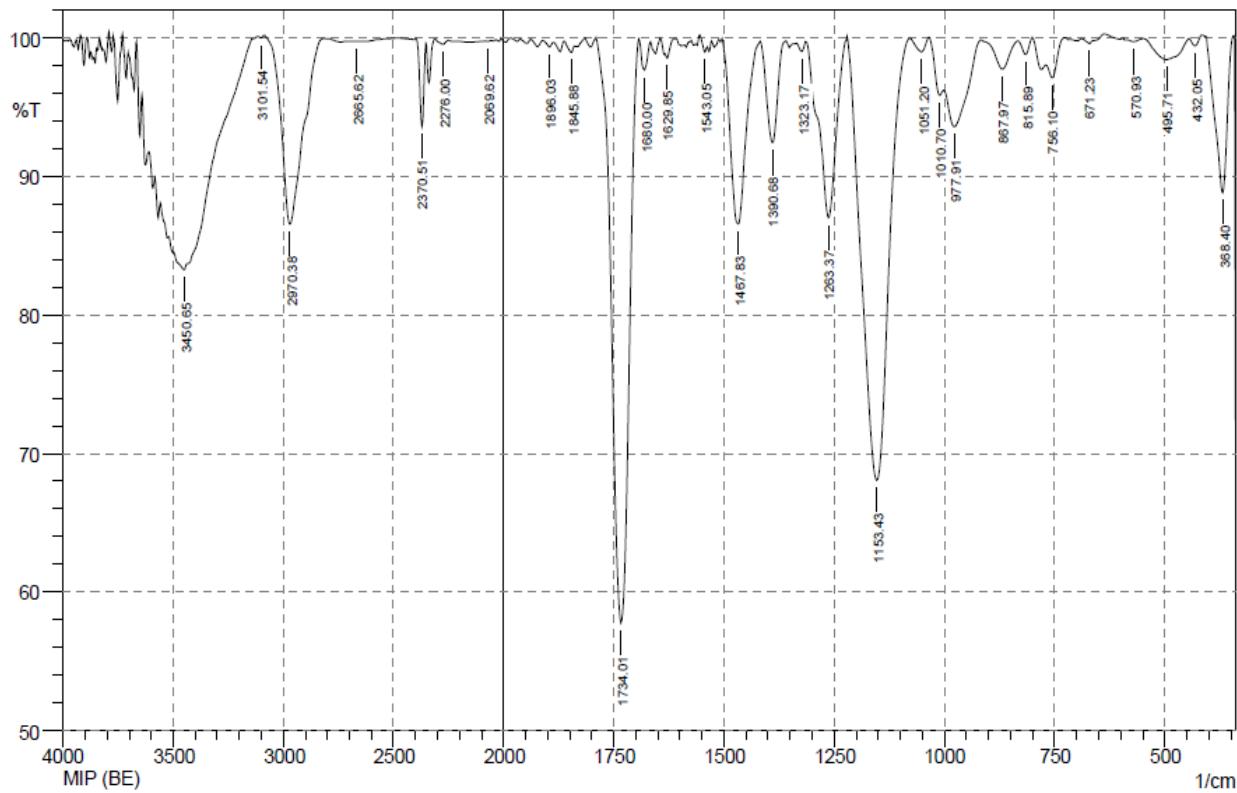
No. of Scans;

Resolution;

Apodization;

3. MIP_DBP_MAM-co-TRIM(BE)

 SHIMADZU



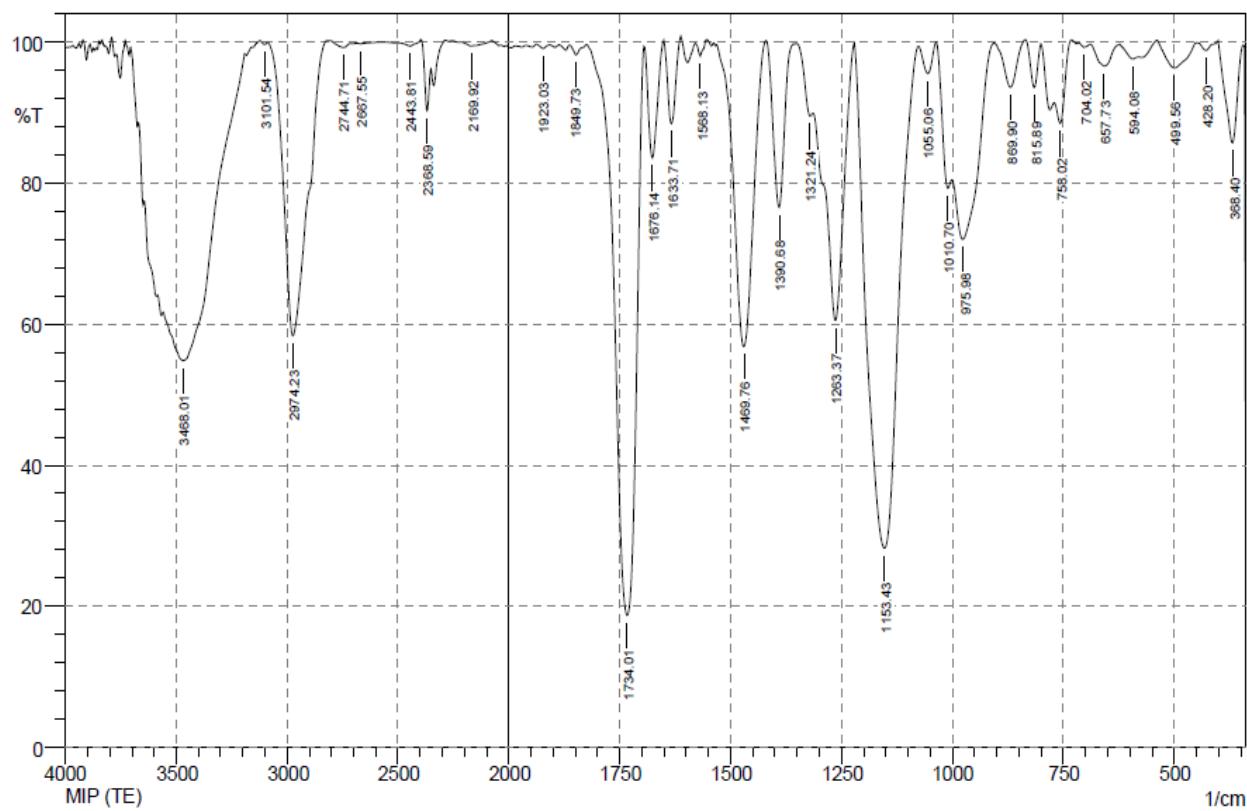
No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	368.4	88.836	11.33	405.05	343.33	1.5	1.543
2	432.05	99.447	0.654	445.56	416.62	0.032	0.046
3	495.71	98.462	1.518	549.71	445.56	0.424	0.415
4	570.93	99.754	0.173	588.29	549.71	0.026	0.015
5	671.23	99.619	0.463	688.59	638.44	0.024	0.053
6	756.1	97.145	1.554	769.6	727.16	0.295	0.12
7	815.89	98.819	1.12	833.25	800.46	0.09	0.081
8	867.97	97.77	2.062	916.19	833.25	0.375	0.317
9	977.91	93.609	3.639	1001.06	918.12	1.467	0.767
10	1010.7	95.877	1.485	1033.85	1001.06	0.384	0.115
11	1051.2	99.021	1.014	1078.21	1033.85	0.098	0.103
12	1153.43	68.078	31.844	1219.01	1078.21	10.373	10.328
13	1263.37	87.047	12.915	1313.52	1220.94	2.544	2.523
14	1323.17	99.05	0.661	1334.74	1315.45	0.058	0.033
15	1390.68	92.467	7.321	1417.68	1359.82	0.919	0.866
16	1467.83	86.596	13.273	1504.48	1419.61	2.425	2.368
17	1543.05	99.003	0.66	1554.63	1537.27	0.04	0.025
18	1629.85	98.567	0.498	1633.71	1612.49	0.067	0.013
19	1680	97.7	2.151	1693.5	1666.5	0.155	0.138
20	1734.01	57.776	42.218	1789.94	1693.5	9.443	9.444
21	1845.88	98.986	0.53	1863.24	1840.09	0.069	0.029
22	1896.03	99.362	0.385	1911.46	1886.38	0.043	0.017
23	2069.62	99.763	0.059	2085.05	2058.05	0.025	0.003
24	2276	99.547	0.353	2310.72	2250.93	0.075	0.051
25	2370.51	93.612	6.161	2397.52	2353.16	0.577	0.54
26	2665.62	99.761	0.013	2677.2	2655.98	0.021	0.001
27	2970.38	86.577	13.529	3086.11	2816.07	6.45	6.554
28	3101.54	100.03	0.137	3115.04	3088.03	-0.011	0.008
29	3450.65	83.283	0.66	3498.87	3435.22	4.916	0.143

Comment:
MIP (BE)

Date/Time: 1/27/2023 4:03:22 PM
No. of Scans:
Resolution:
Apodization:

4. MIP_DBP_MAM-co-TRIM(TE)

 SHIMADZU



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	368.4	85.72	14.181	401.19	343.33	1.916	1.899
2	428.2	98.86	1.215	449.41	401.19	0.103	0.122
3	499.56	96.359	3.793	538.14	449.41	0.822	0.871
4	594.08	97.665	0.831	623.01	580.57	0.304	0.085
5	657.73	96.63	3.094	688.59	623.01	0.541	0.462
6	704.02	99.293	0.521	717.52	688.59	0.053	0.029
7	758.02	88.494	5.723	769.6	732.95	1.115	0.472
8	815.89	93.556	6.474	835.18	800.46	0.485	0.494
9	869.9	93.61	6.525	906.54	835.18	0.874	0.914
10	975.98	72.057	13.586	1001.06	906.54	8.128	3.781
11	1010.7	79.304	5.712	1035.77	1002.98	2.133	0.511
12	1055.06	95.574	4.162	1076.28	1035.77	0.44	0.391
13	1153.43	28.274	71.383	1220.94	1078.21	35.674	35.449
14	1263.37	60.562	34.75	1313.52	1222.87	9.928	7.815
15	1321.24	89.536	2.486	1350.17	1313.52	0.976	0.117
16	1390.68	76.586	23.432	1419.61	1359.82	2.99	2.99
17	1469.76	56.868	43.044	1529.55	1421.54	10.907	10.851
18	1568.13	98.061	2.055	1579.7	1554.63	0.095	0.11
19	1633.71	88.462	11.932	1651.07	1614.42	0.983	1.047
20	1676.14	83.623	15.863	1693.5	1653	1.685	1.603
21	1734.01	18.714	80.801	1822.73	1695.43	31.044	30.792
22	1849.73	98.237	1.41	1863.24	1824.66	0.163	0.106
23	1923.03	99.127	0.561	1938.46	1907.6	0.076	0.034
24	2169.92	99.493	0.622	2222	2075.41	0.137	0.228
25	2368.59	90.288	7.376	2395.59	2353.16	0.989	0.643
26	2443.81	99.464	0.7	2495.89	2395.59	0.092	0.159
27	2667.55	99.804	0.087	2686.84	2617.4	0.036	0.008
28	2744.71	99.268	0.715	2810.28	2711.92	0.137	0.159
29	2974.23	58.345	41.741	3086.11	2819.93	25.194	25.323
30	3101.54	99.691	0.387	3122.75	3088.03	0.013	0.029
31	3468.01	54.888	15.894	3558.67	3192.19	58.886	19.368

Date/Time; 1/27/2023 4:14:04 PM

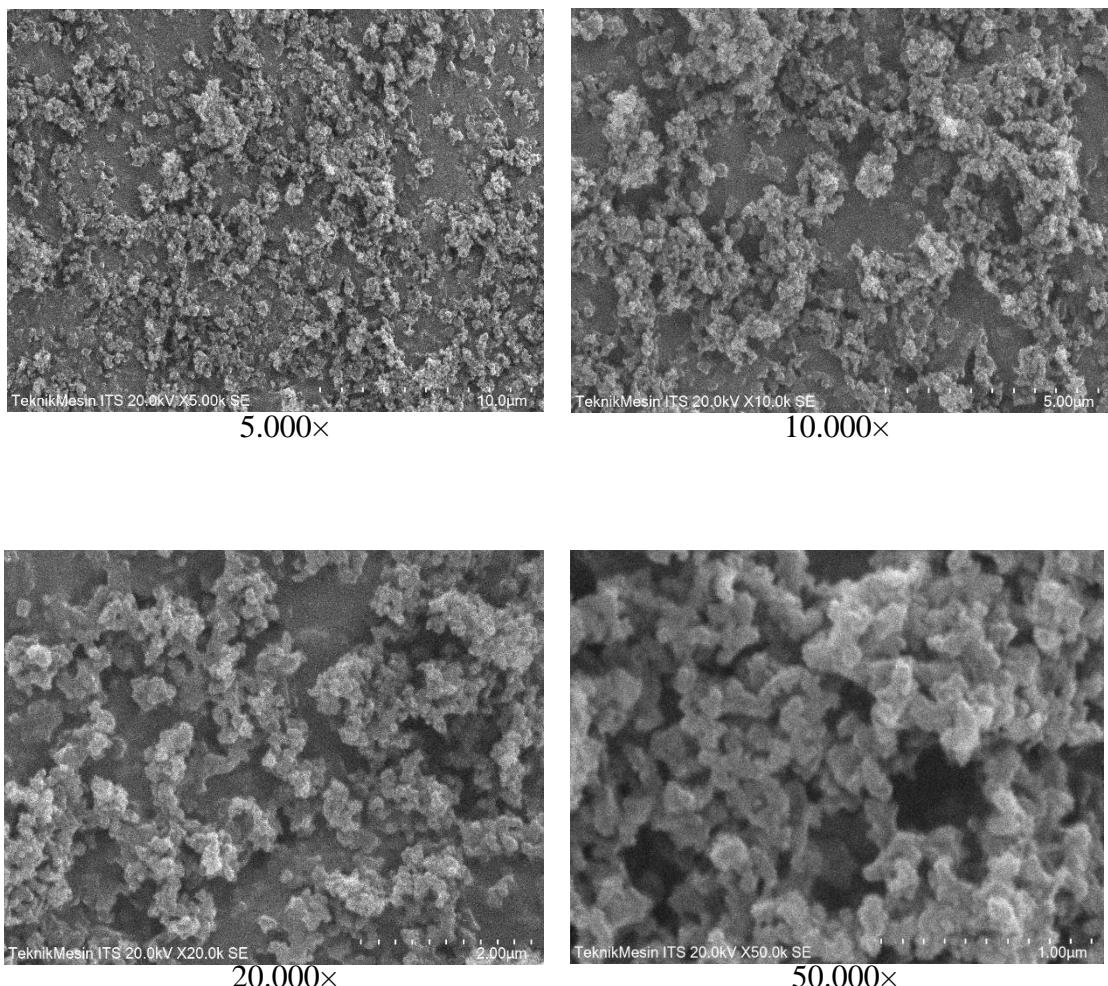
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Resolution;

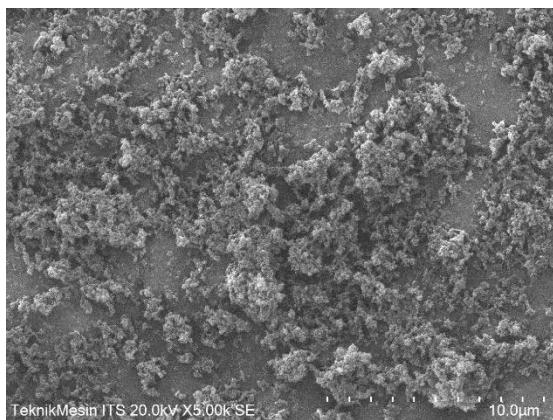
Apodization;

Lampiran 6. Hasil Analisis SEM

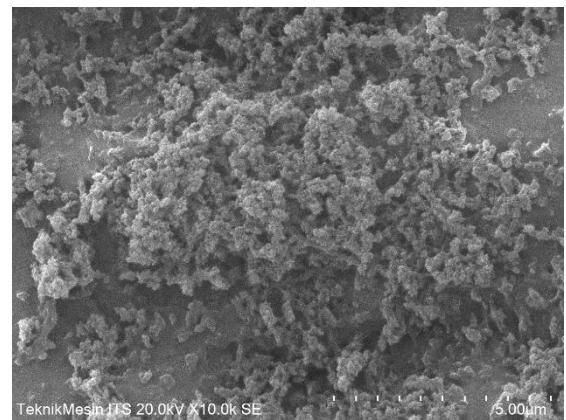
1. NIP_MAM-co-TRIM



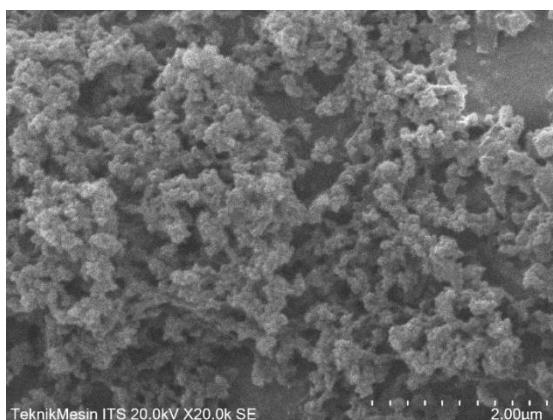
2. MIP_DBP_MAM-co-TRIM(BE)



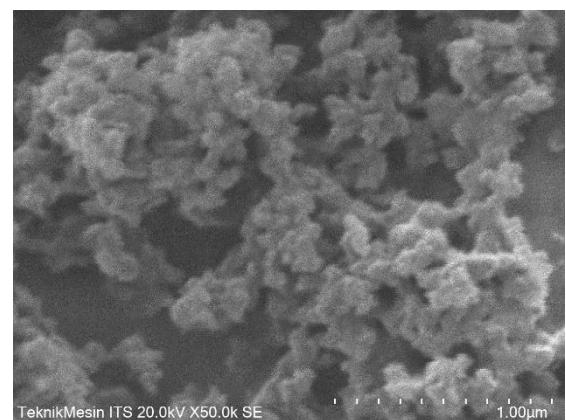
5.000 \times



10.000 \times

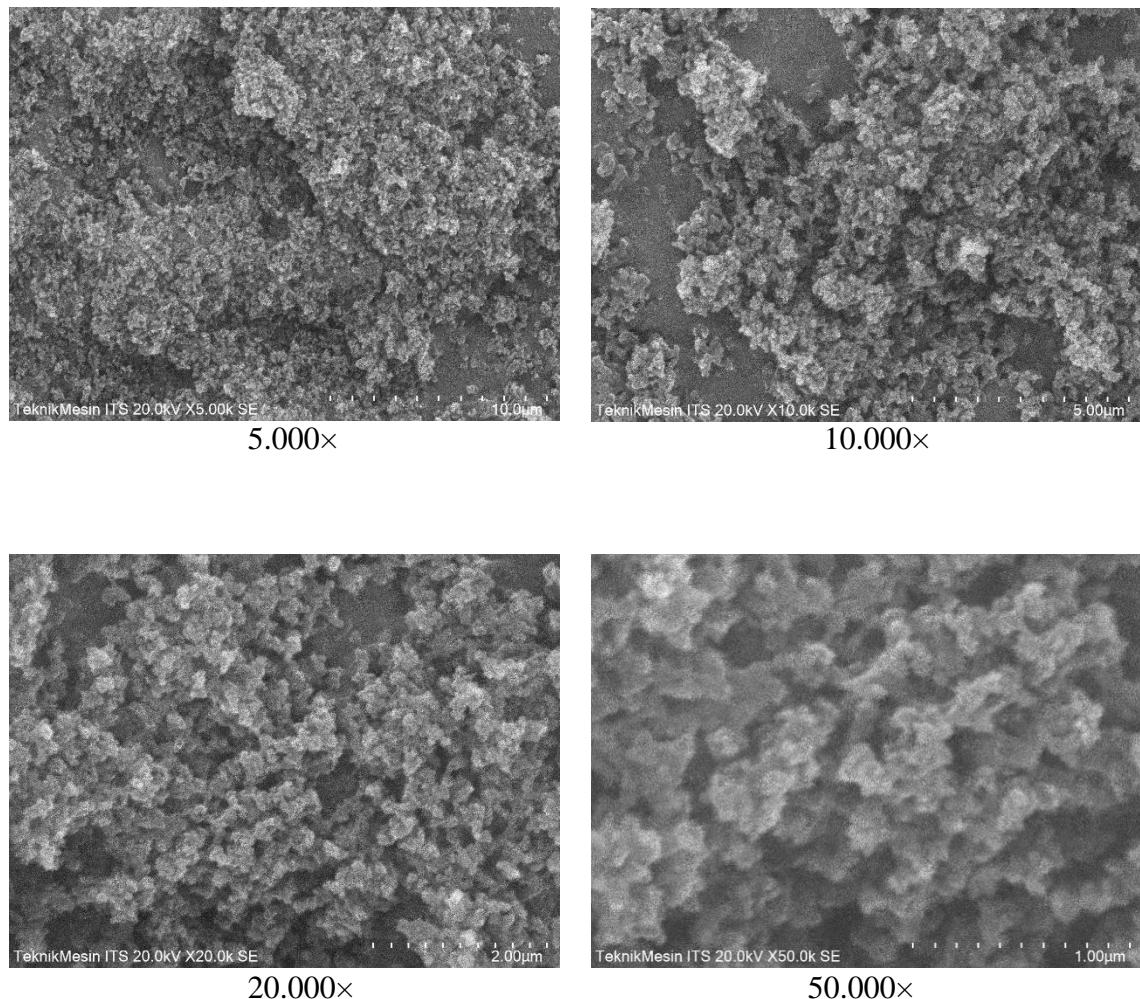


20.000 \times



50.000 \times

3. MIP_DBP_MAM-co-TRIM_(TE)



Lampiran 7. Hasil Analisis EDS

1. NIP_MAM-co-TRIM

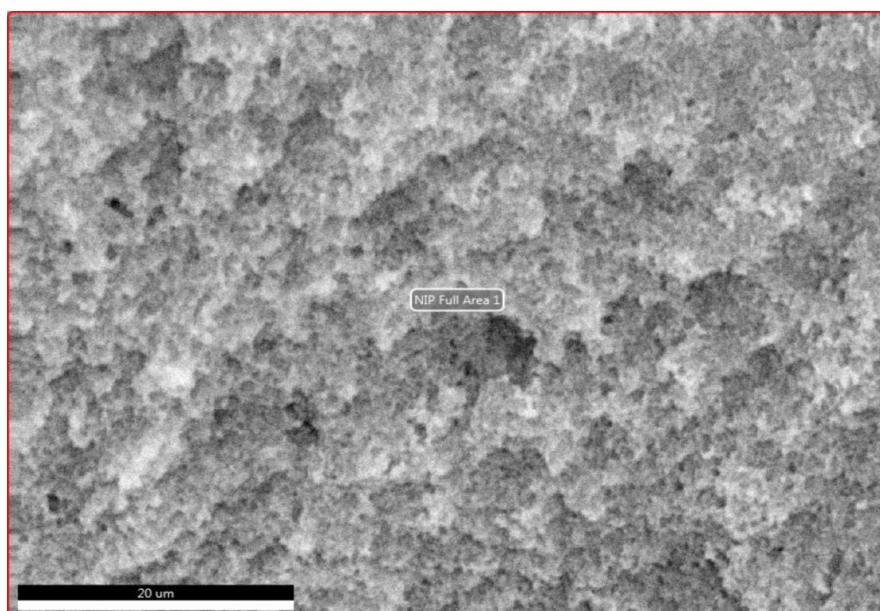
EDAX APEX

Page 1

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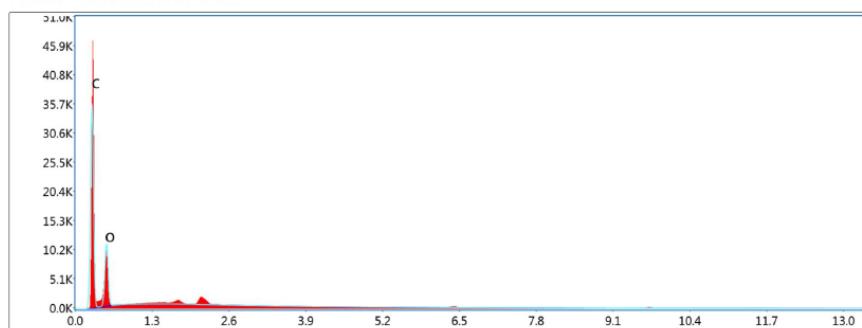
Author: Teknik Mesin ITS
Creation: 02/13/2023 2:56:37 PM
Sample Name: Hendrik Unhas|

NIP



Full Area 1

kV:20 Mag: 2000 Takeoff: 30 Live Time(s): 96.4 Amp Time(μs): 3.84 Resolution:(eV) 133.6



Smart Quant Results

Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	58.72	65.45	2323.34	4.82	0.3766	1.0182	0.6298	1.0000
O K	41.28	34.55	787.44	9.70	0.0652	0.9727	0.1622	1.0000

2. MIP_DBP_MAM-co-TRIM(BE)

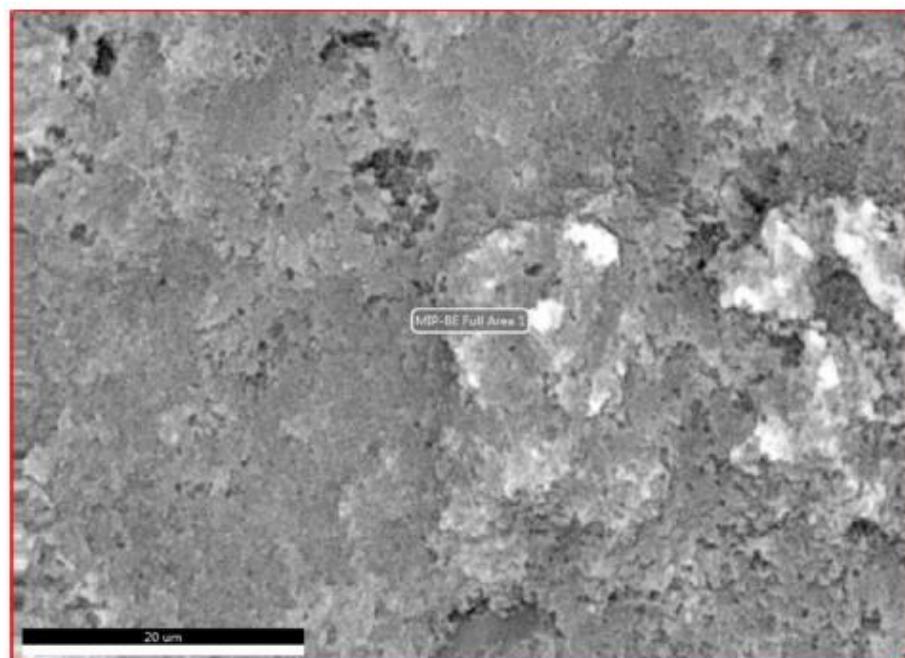
EDAX APEX

Page 1

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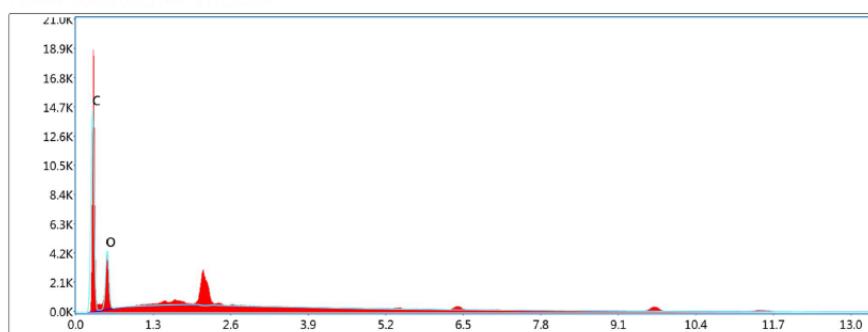
Author: Teknik Mesin ITS
Creation: 02/13/2023 2:59:05 PM
Sample Name: Hendrik Unhas

MIP-BE



Full Area 1

kV:20 Mag: 2000 Takeoff: 30 Live Time(s): 95.7 Amp Time(μs): 3.84 Resolution(eV) 133.6



Smart Quant Results

Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	59.65	66.32	955.37	4.91	0.3854	1.0178	0.6350	1.0000
O K	40.35	33.68	304.51	10.13	0.0627	0.9723	0.1599	1.0000

3. MIP_DBP_MAM-co-TRIM_(TE)

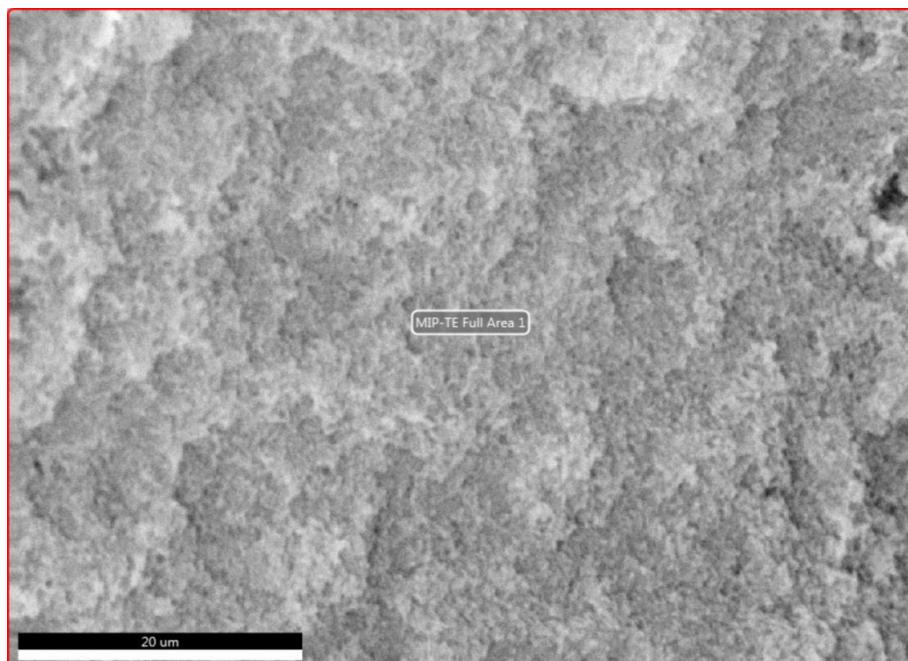
EDAX APEX

Page 1

10022023

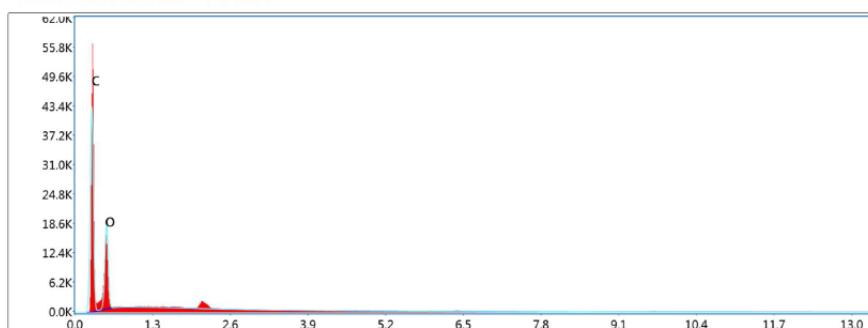
Author: Teknik Mesin ITS
Creation: 02/13/2023 3:01:39 PM
Sample Name: Hendrik Unhas

MIP-TE



Full Area 1

kV:20 Mag:2000 Takeoff: 30 Live Time(s): 96.8 Amp Time(μs): 3.84 Resolution:(eV) 133.6



Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	54.53	61.50	2816.07	5.04	0.3371	1.0201	0.6058	1.0000
O K	45.47	38.50	1265.17	9.49	0.0773	0.9745	0.1744	1.0000

Lampiran 8. Hasil Analisis SAA

**Quantachrome NovaWin - Data Acquisition and Reduction
for NOVA instruments**
 ©1994-2013, Quantachrome Instruments
 version 11.03



<u>Analysis</u>		<u>Report</u>		Date:
Operator:	quantachrome	Operator:	quantachrome	2023/03/16
Sample ID:	79012	Filename:	stn_A_15032023_7821_BET_EK_79012.qps	
Sample Desc:	MIP (TE) Ulang	Comment:		
Sample weight:	0.0882 g	Sample Volume:	0.02587 cc	
Outgas Time:	3.0 hrs	OutgasTemp:	250.0 C	
Analysis gas:	Nitrogen	Bath Temp:	273.0 K	
Press. Tolerance:	0.100/0.100 (ads/des)	Equil time:	60/60 sec (ads/des)	
Analysis Time:	869.0 min	End of run:	2023/03/16 8:18:54	
Cell ID:	9			

Area-Volume Summary

Data Reduction Parameters Data

<u>t-Method</u>	Thermal Transpiration: on	Eff. mol. diameter (D):	3.54 Å	Eff. cell stem diam. (d):	4.0000 mm
<u>BJH/DH method</u>	Calc. method: de Boer				
<u>HK method</u>	Moving pt. avg.: off				
<u>SF method</u>	Tabulated data interval: 1				
<u>DFT method</u>	Tabulated data interval: 1				
	Calc. Model: N2 at 77 K on carbon (slit pore, NLDFT equilibrium model)				
	Rel. press. range: 0.0000 - 1.0000				
<u>Adsorbate</u>	Nitrogen	Temperature	77.350K	Moving pt. avg:	off
	Molec. Wt.:	Cross Section:	16.200 Å ²	Liquid Density:	0.808 g/cc
<u>Adsorbent</u>	Carbon				

Surface Area Data

SinglePoint BET.....	3.340e+02 m ² /g
MultiPoint BET.....	3.455e+02 m ² /g
Langmuir surface area.....	5.609e+02 m ² /g
BJH method cumulative adsorption surface area.....	1.688e+02 m ² /g
BJH method cumulative desorption surface area.....	2.188e+02 m ² /g
DH method cumulative adsorption surface area.....	1.717e+02 m ² /g
DH method cumulative desorption surface area.....	2.226e+02 m ² /g
t-method external surface area.....	3.455e+02 m ² /g
DFT cumulative surface area.....	2.674e+02 m ² /g

Pore Volume Data

Total pore volume for pores with Radius less than 1362.57 Å at P/Po = 0.992921.....	1.035e+00 cc/g
BJH method cumulative adsorption pore volume.....	9.359e-01 cc/g
BJH method cumulative desorption pore volume.....	9.353e-01 cc/g
DH method cumulative adsorption pore volume.....	9.089e-01 cc/g
DH method cumulative desorption pore volume.....	9.124e-01 cc/g
HK method micropore volume.....	1.379e-01 cc/g
SF method micropore volume.....	9.200e-02 cc/g
DFT method cumulative pore volume.....	7.610e-01 cc/g

Pore Size Data

Average pore Radius.....	5.991e+01 Å
BJH method adsorption pore Radius (Mode D _v (r)).....	1.696e+01 Å
BJH method desorption pore Radius (Mode D _v (r)).....	1.898e+01 Å
DH method adsorption pore Radius (Mode D _v (r)).....	1.696e+01 Å
DH method desorption pore Radius (Mode D _v (r)).....	1.698e+01 Å
HK method pore Radius (Mode).....	1.838e+00 Å
SF method pore Radius (Mode).....	2.261e+00 Å
DFT pore Radius (Mode).....	9.660e+00 Å



Analysis		Date:	Report		Date:
Operator:	quantachrome		Operator:	quantachrome	
Sample ID:	79012		Filename:	stn_A_15032023_7821_BET_EK_79012.qps	
Sample Desc:	MIP (TE) Ulang		Comment:		
Sample weight:	0.0882 g		Sample Volume:	0.02587 cc	Sample Density:
Outgas Time:	3.0 hrs		OutgasTemp:	250.0 C	3.41 g/cc
Analysis gas:	Nitrogen		Bath Temp:	273.0 K	
Press. Tolerance:	0.100/0.100 (ads/des)		Equil time:	60/60 sec (ads/des)	Equil timeout:
Analysis Time:	869.0 min		End of run:	2023/03/16 8:18:54	Instrument:
Cell ID:	9				240/240 sec (ads/des) Nova Station A

BJH Pore Size Distribution Adsorption

Data Reduction Parameters Data

t-Method	Calc. method:	de Boer	Ignoring P-tags below 0.35 P/Po		
BJH/DH method	Moving pt. avg.:	off	Temperature	77.350K	
Adsorbate	Nitrogen		Cross Section:	16.200 Å ²	Liquid Density:
	Molec. Wt.:	28.013			0.808 g/cc

BJH Pore Size Distribution Adsorption Data

Radius [Å]	Pore Volume [cc/g]	Pore Surf Area [m ² /g]	dV(r) [cc/Å/g]	dS(r) [m ² /Å/g]	dV(logr) [cc/g]	dS(logr) [cc/g]
16.9635	1.2285e-02	1.4484e+01	6.1545e-03	7.2561e+00	2.4012e-01	2.8310e+02
19.0996	2.4570e-02	2.7348e+01	5.3973e-03	5.6517e+00	2.3708e-01	2.4826e+02
21.4629	3.8130e-02	3.9984e+01	5.5336e-03	5.1564e+00	2.7317e-01	2.5455e+02
24.2855	5.3577e-02	5.2705e+01	4.8353e-03	3.9821e+00	2.7000e-01	2.2235e+02
27.7970	6.9577e-02	6.4217e+01	4.1793e-03	3.0070e+00	2.6707e-01	1.9216e+02
32.2454	8.7806e-02	7.5523e+01	3.5964e-03	2.2306e+00	2.6647e-01	1.6528e+02
38.2920	1.1035e-01	8.7301e+01	3.2100e-03	1.6766e+00	2.8223e-01	1.4741e+02
47.2363	1.4275e-01	1.0102e+02	2.9816e-03	1.2624e+00	3.2286e-01	1.3670e+02
60.3757	1.7831e-01	1.1280e+02	2.3071e-03	7.6425e-01	3.1898e-01	1.0567e+02
84.4510	2.3724e-01	1.2675e+02	1.8001e-03	4.2632e-01	3.4562e-01	8.1851e+01
143.7029	3.5152e-01	1.4266e+02	1.3325e-03	1.8545e-01	4.2749e-01	5.9496e+01
262.8729	5.0911e-01	1.5465e+02	1.0329e-03	7.8585e-02	6.0724e-01	4.6200e+01
375.4530	5.8567e-01	1.5873e+02	1.0546e-03	5.6178e-02	9.0888e-01	4.8415e+01
455.5969	6.5328e-01	1.6169e+02	7.7100e-04	3.3846e-02	8.0632e-01	3.5396e+01
574.8518	7.4485e-01	1.6488e+02	6.0716e-04	2.1124e-02	7.9903e-01	2.7800e+01
813.8182	8.3805e-01	1.6717e+02	2.8491e-04	7.0018e-03	5.2662e-01	1.2942e+01
1169.9728	9.3594e-01	1.6884e+02	2.5415e-04	4.3445e-03	6.7843e-01	1.1597e+01

BJH adsorption summary

Surface Area = 168.843 m²/g
Pore Volume = 0.936 cc/g
Pore Radius Dv(r) = 16.963 Å

**Quantachrome NovaWin - Data Acquisition and Reduction
for NOVA instruments**
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version 11.03



Analysis
Operator: quantachrome
Sample ID: 79012
Sample Desc: MIP (TE) Ulang
Sample weight: 0.0882 g
Outgas Time: 3.0 hrs
Analysis gas: Nitrogen
Press. Tolerance: 0.100/0.100 (ads/des)
Analysis Time: 869.0 min
Cell ID: 9

Date: 2023/03/16 **Report**
Operator: quantachrome
Filename: stn_A_15032023_7821_BET_EK_79012.qps
Comment:
Sample Volume: 0.02587 cc
OutgasTemp: 250.0 C
Bath Temp: 273.0 K
Equil time: 60/60 sec (ads/des)
End of run: 2023/03/16 8:18:54

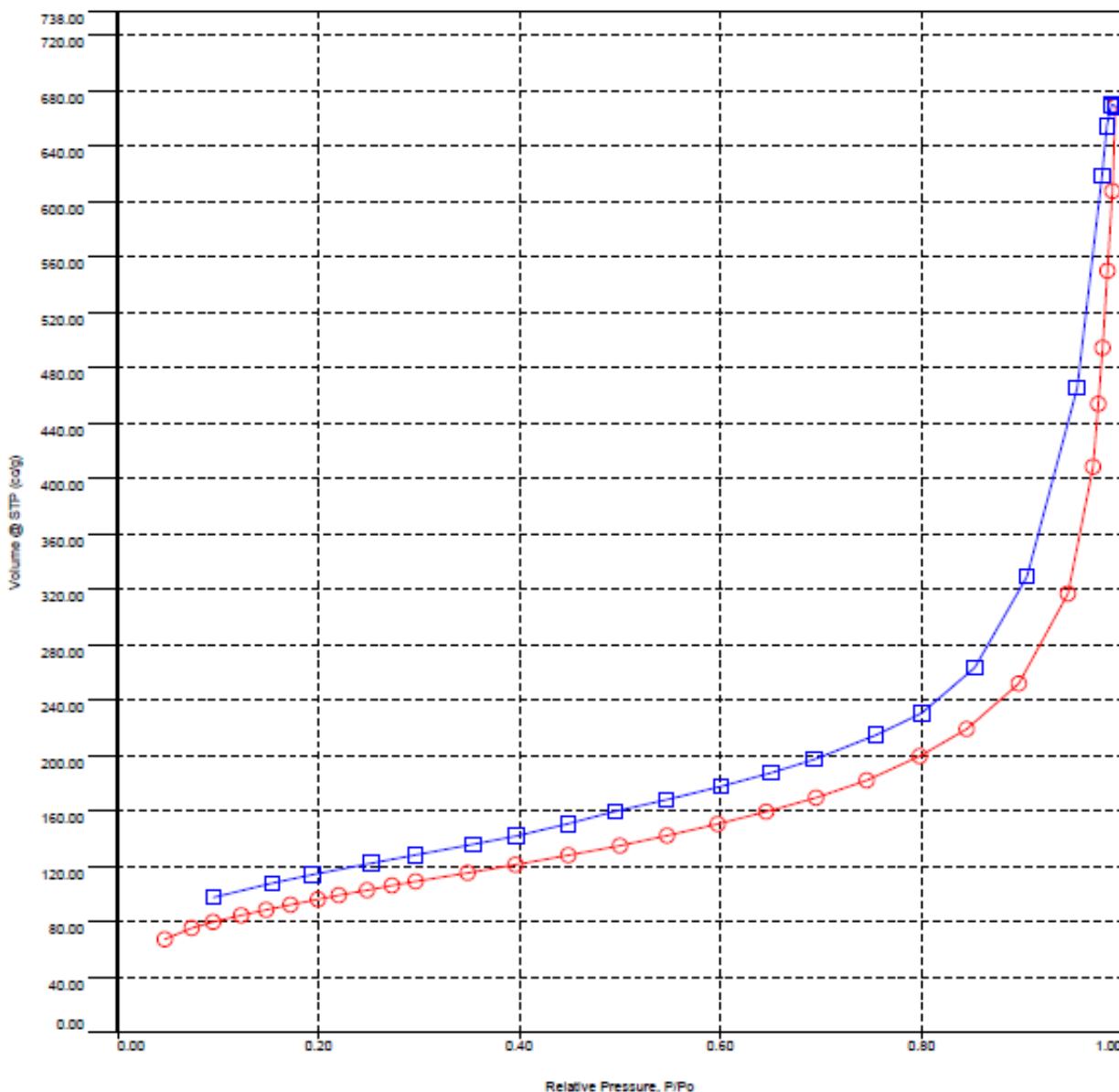
Date: 2023/03/16
Sample Density: 3.41 g/cc
Equil timeout:
Instrument: 240/240 sec (ads/des)
 Nova Station A

Isotherm : Linear

Data Reduction Parameters

Adsorbate	Nitrogen Molec. Wt.:	28.013	Temperature Cross Section:	77.350K 16.200 Å ²	Liquid Density:	0.808 g/cc
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Ads Des



Lampiran 9. Gambar Proses Penelitian



Persiapan Alat dan Bahan



Proses Pengaliran Gas Nitrogen



Proses Sonikasi



Proses Polimerisasi



Polimer yang Terbentuk



Pengeringan Polimer



Proses Pencucian Polimer yang Terbentuk



Pengukuran pH Hasil Pencucian



Deret Standar Larutan DBP



Rangkaian Alat dan Sampel untuk Melakukan Proses *Shaking*