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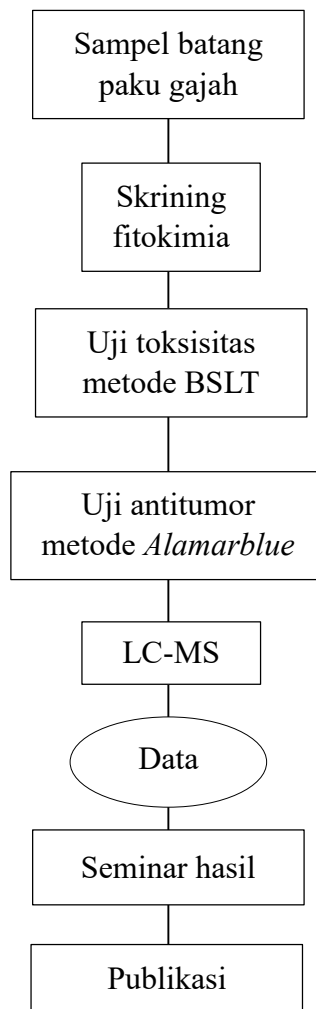
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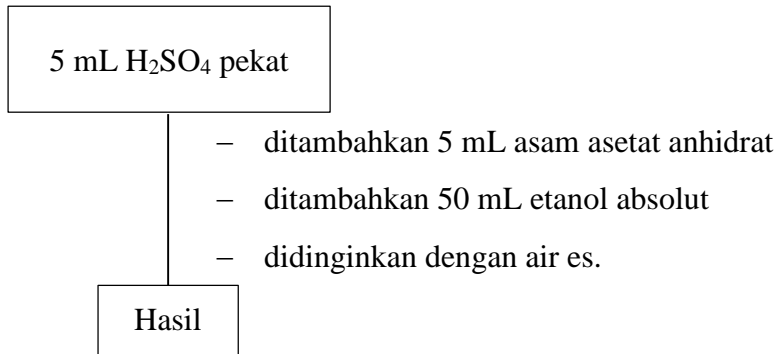
Lampiran 1. Diagram Alir Penelitian



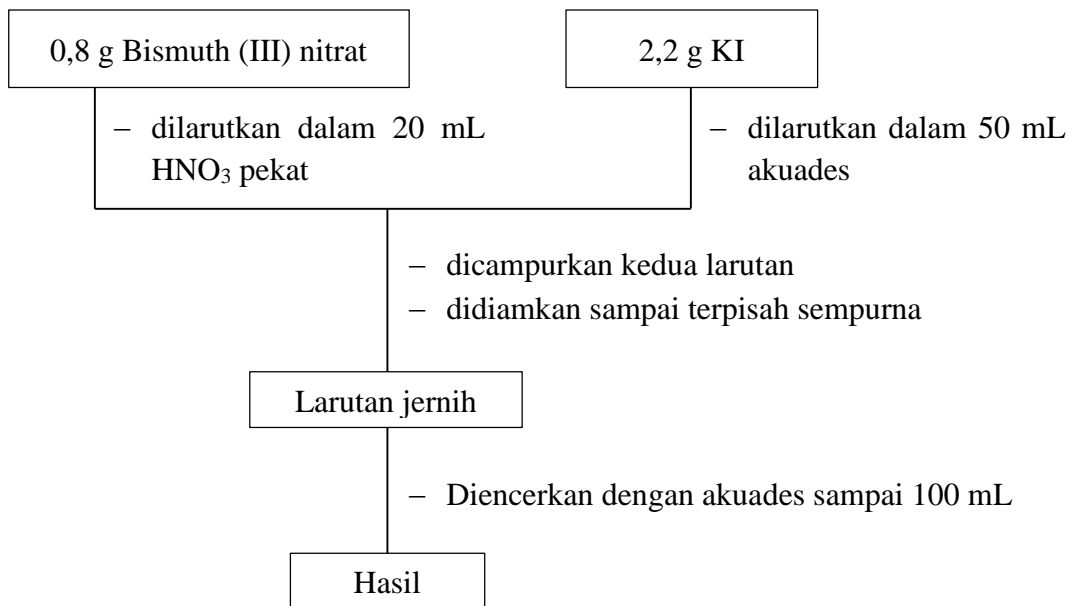
Lampiran 2. Bagan Prosedur Kerja

1. Pembuatan Reagen (Shaikh dan Patil, 2020)

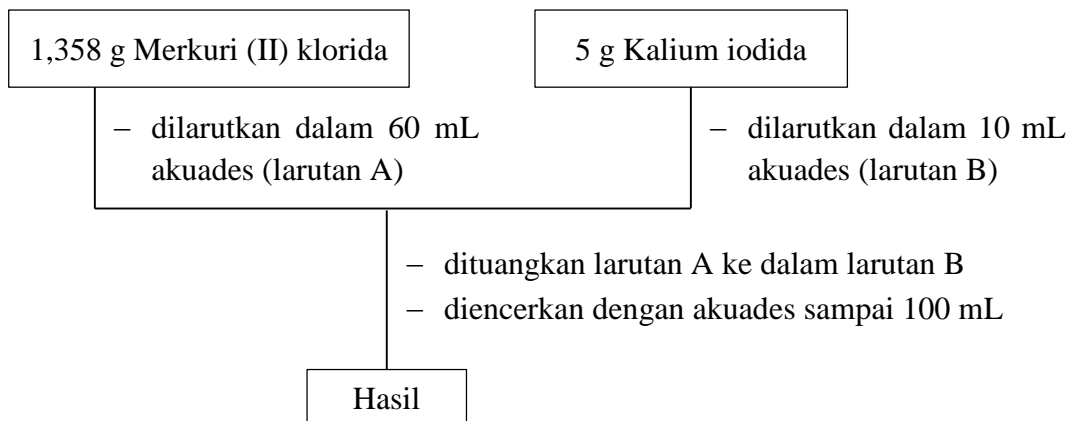
a. Pembuatan Reagen Libermann-Burchard



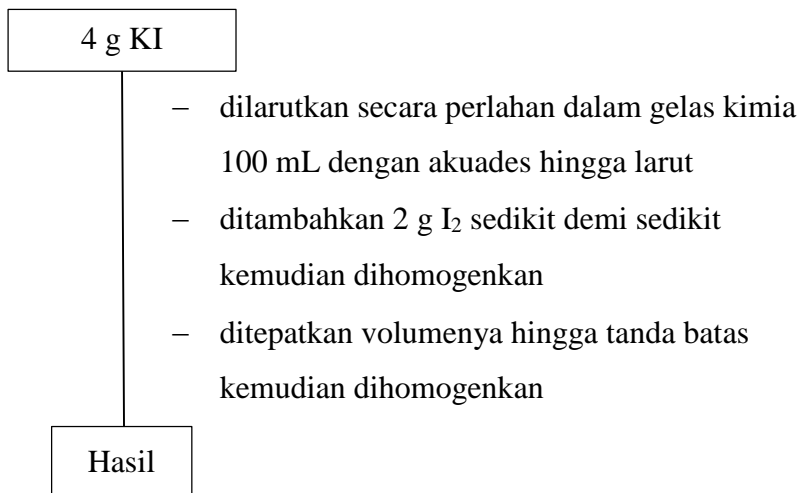
b. Pembuatan Reagen Dragendroff



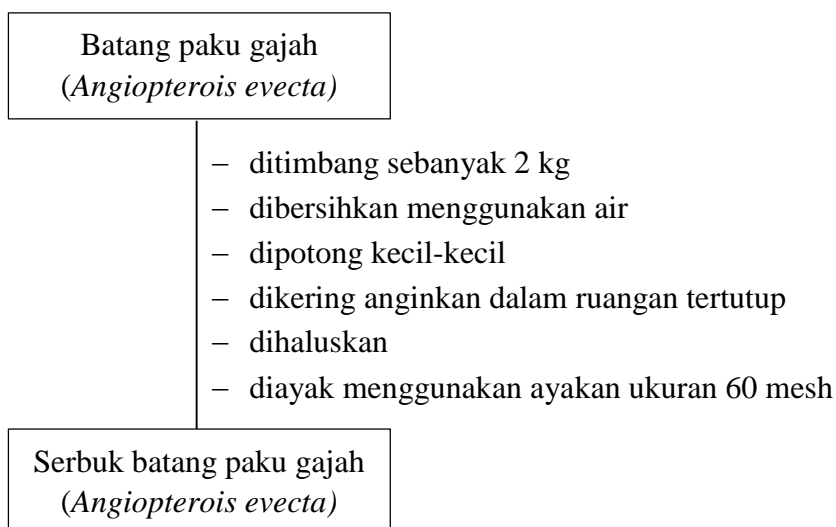
c. Pembuatan Reagen Mayer



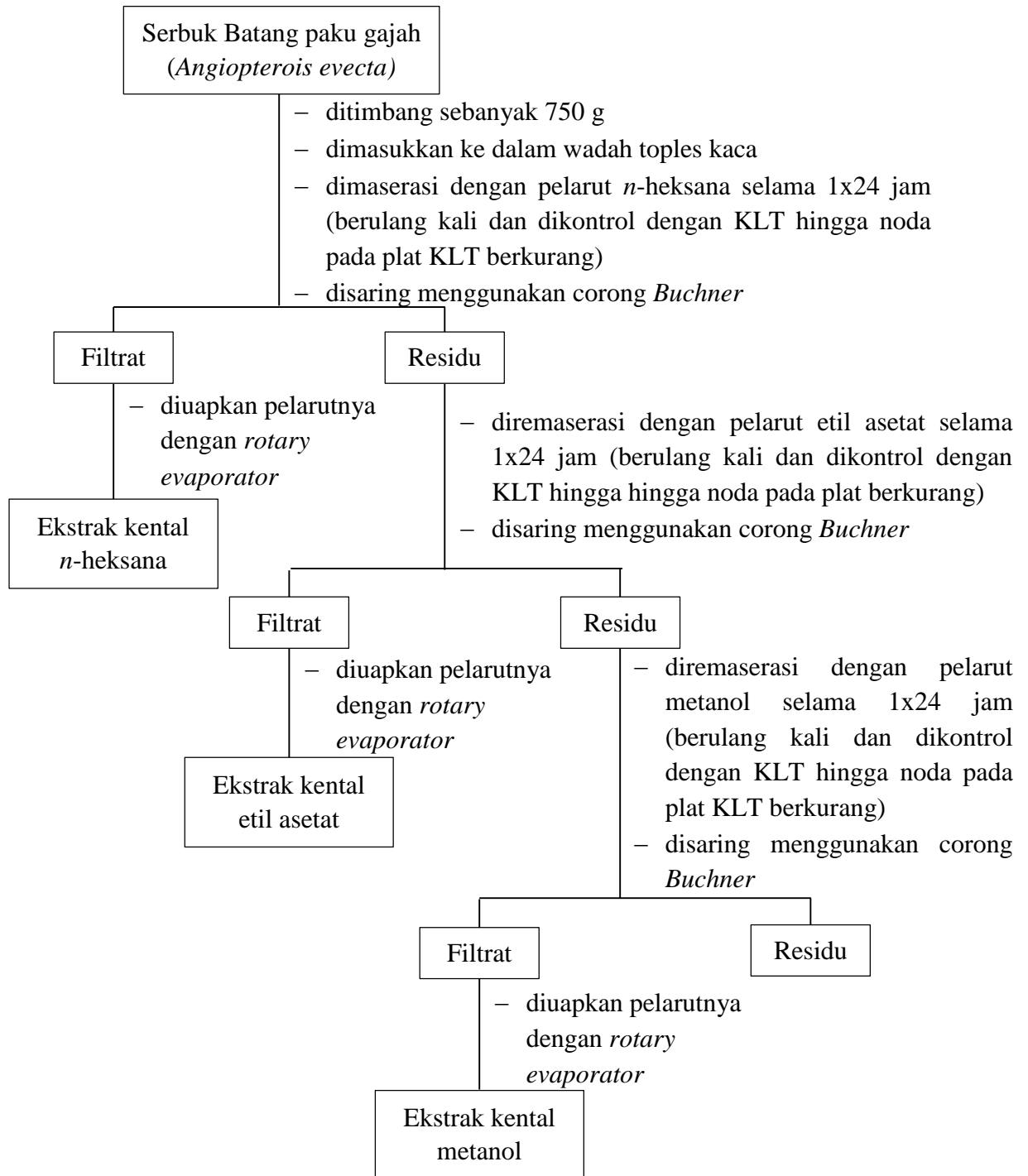
d. Pembuatan Reagen Wagner



2. Preparasi Sampel

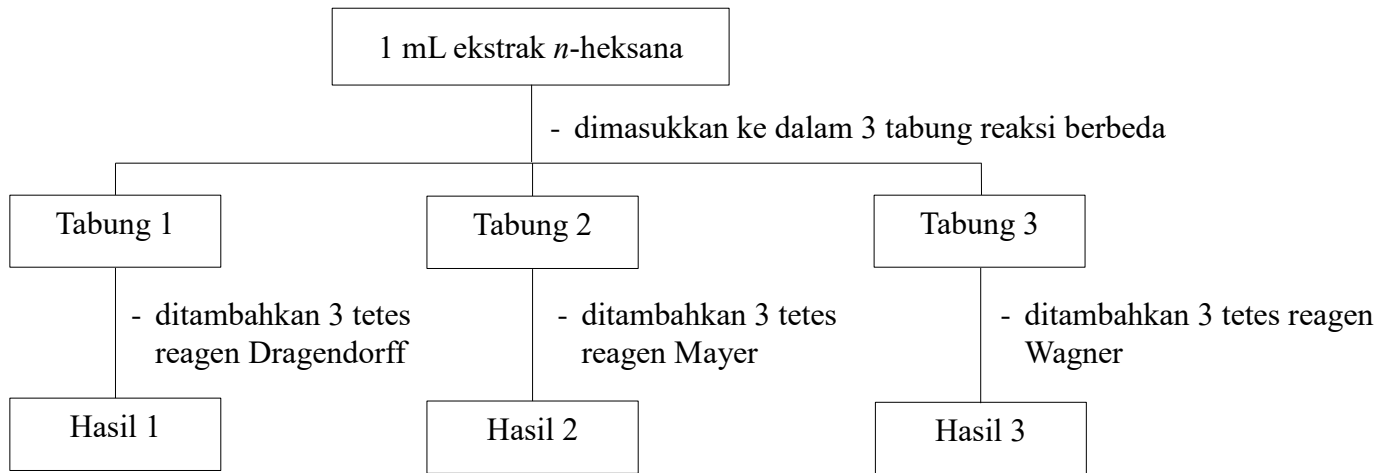


3. Ekstraksi Sampel

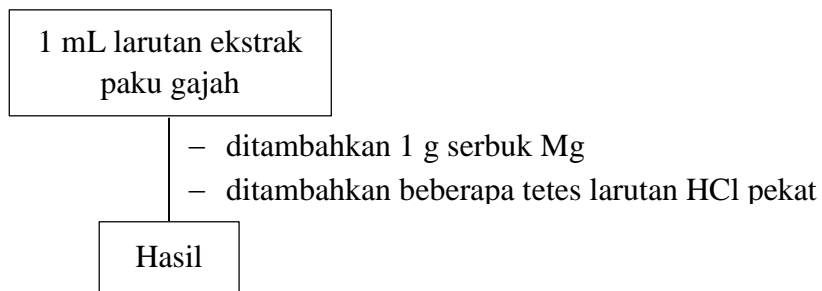


4. Skrining Fitokimia (Shaikh dan Patil, 2020)

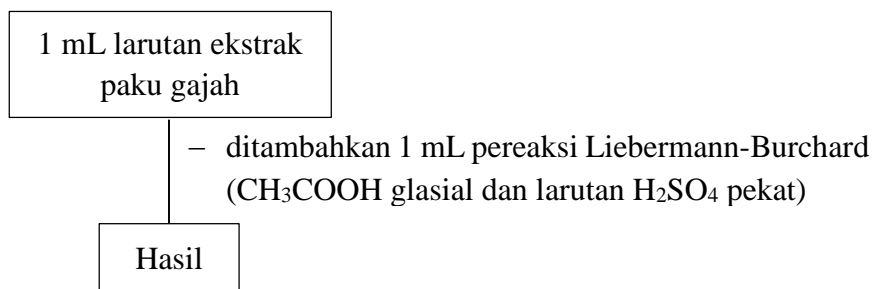
a. Uji Alkaloid



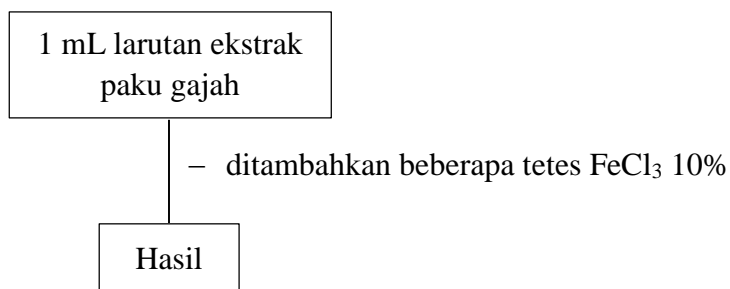
b. Uji Flavonoid



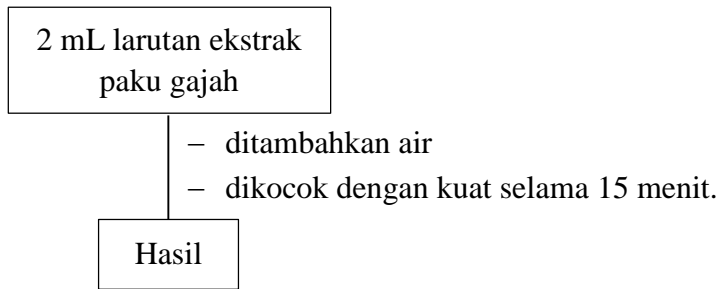
c. Uji Steroid dan Terpenoid



d. Uji Tanin

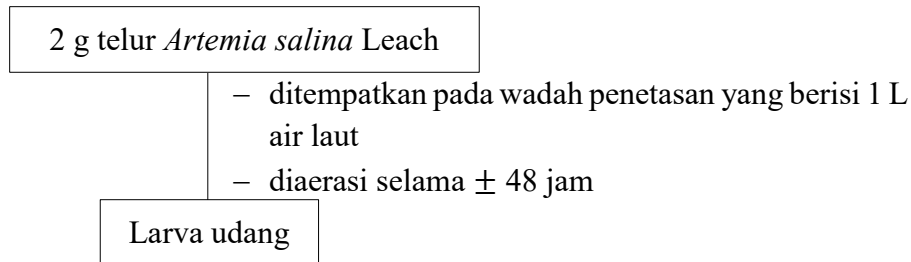


e. Uji Saponin

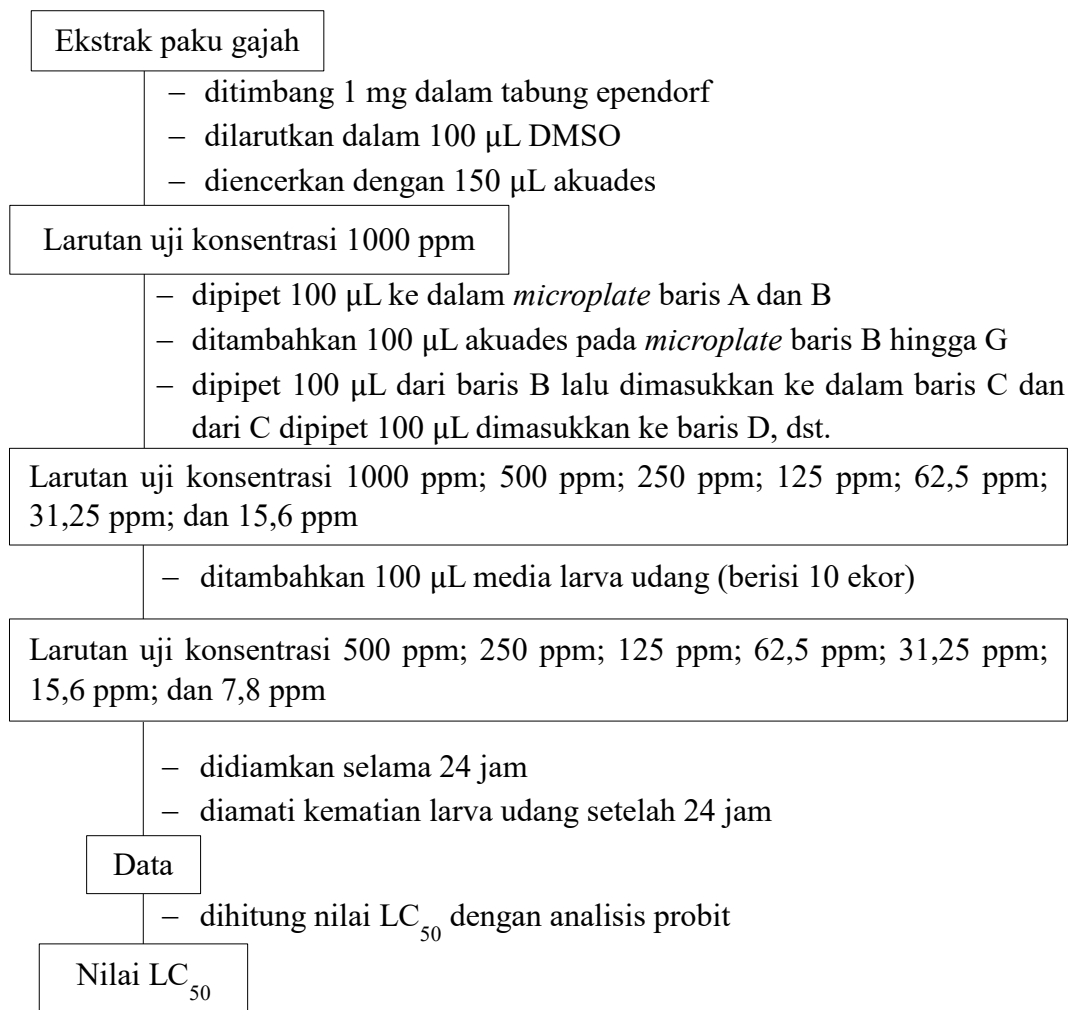


5. Uji Toksisitas metode *Brine Shrimp Lethality Test* (BSLT)

a. Penyiapan larva udang



b. Uji Toksisitas

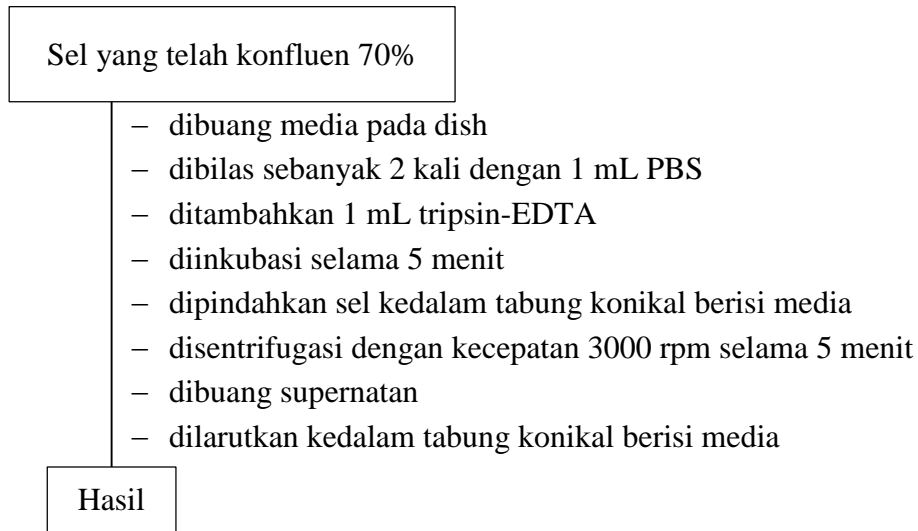


Catatan:

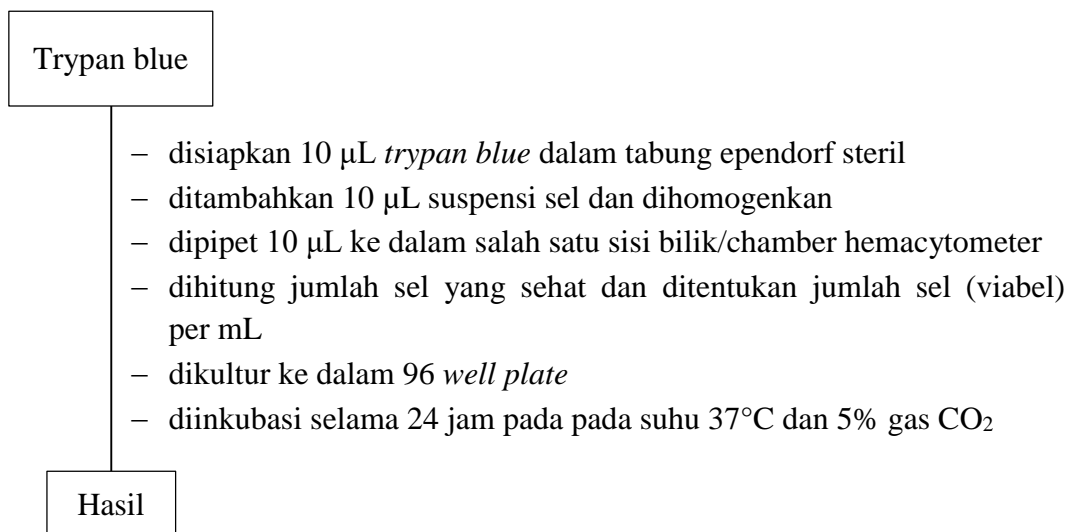
- dilakukan pengulangan sebanyak 3 kali (triplo) untuk setiap sampel
- dibuat kontrol negatif dengan menggunakan DMSO dan dilakukan pengerjaan yang sama tanpa menggunakan sampel

7. Uji Antitumor terhadap sel tumor payudara MCF-7 dengan Metode *Alamarblue*

a. Preparasi Sel



b. *Seeding* Sel ke dalam 96 *Well Plate*



b. Perlakuan Sel dengan Sampel, Kontrol Positif, dan Kontrol Negatif

96 *well plate* yang telah berisi sel hasil inkubasi

- diberi label pada sepanjang margin kiri *plate* untuk baris mana yang akan diberi perlakuan dengan sampel dan kontrol
- dibuang media dari setiap *well*
- ditambahkan 100 μL sampel dengan 8 konsentrasi berbeda dan kontrol positif cisplatin dari tabung ependorf ke dalam masing-masing *well* yang sesuai dengan menggunakan mikropipet
- di inkubasi kembali selama 48 jam

Hasil

c. Pemberian Reagen Presto Blue dan Pengukuran Absorbansi

96 *well plate* yang berisi sel yang telah diberi perlakuan

- dibuang media pada setiap *well*
- dimasukkan 100 μL larutan campuran (9 mL media dan 1 mL reagen *presto blue cell viability*) ke dalam tiap *well microplate*
- diinkubasi selama 1-2 jam sampai terlihat perubahan warna
- diukur absorbansinya pada panjang gelombang 570 nm menggunakan *multimode reader*

Hasil

Lampiran 3. Perhitungan

1. Persen Rendemen Ekstrak

$$\text{Persen Rendemen} = \frac{\text{berat ekstrak}}{\text{berat sampel}} \times 100\%$$

a. Ekstrak *n*-Heksana

$$\text{Berat sampel serbuk awal} = 750 \text{ g}$$

$$\text{Berat ekstrak yang dihasilkan} = 1,875 \text{ g}$$

$$\begin{aligned} \text{Persen rendemen ekstrak } n\text{-heksana} &= \frac{\text{berat ekstrak}}{\text{berat sampel}} \times 100\% \\ &= \frac{1,875}{750} \times 100 \\ &= 0,250\% \end{aligned}$$

b. Ekstrak Etil Asetat

$$\text{Berat sampel serbuk awal} = 748,125 \text{ g}$$

$$\text{Berat ekstrak yang dihasilkan} = 4,7 \text{ g}$$

$$\begin{aligned} \text{Persen rendemen ekstrak etil asetat} &= \frac{\text{berat ekstrak}}{\text{berat sampel}} \times 100\% \\ &= \frac{4,7}{748,125} \times 100 \\ &= 0,628\% \end{aligned}$$

c. Ekstrak Metanol

$$\text{Berat sampel serbuk awal} = 743,425 \text{ g}$$

$$\text{Berat ekstrak yang dihasilkan} = 5,675 \text{ g}$$

$$\begin{aligned} \text{Persen rendemen ekstrak metanol} &= \frac{\text{berat ekstrak}}{\text{berat sampel}} \times 100\% \\ &= \frac{5,675}{743,425} \times 100 \\ &= 0,763\% \end{aligned}$$

2. Perhitungan Konsentrasi untuk Uji Toksisitas

$$\begin{aligned} \text{Konsentrasi sampel} &= \frac{\frac{200 \mu\text{L}}{250 \mu\text{L}} \times 1 \text{ mg}}{800 \mu\text{L}} \\ &= \frac{0,8 \text{ mg}}{800 \mu\text{L}} \\ &= 1 \mu\text{g}/\mu\text{L} \\ &= 1000 \text{ ppm} \end{aligned}$$

Konsentrasi sampel setelah pengenceran pada *microplate*

$$\begin{aligned} \text{Baris A} &= \frac{100 \mu\text{g}}{100 \mu\text{L}} = 1 \mu\text{g}/\mu\text{L} = 1000 \text{ ppm} \\ \text{B} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{100 \mu\text{g}}{100 \mu\text{L}} = \frac{50 \mu\text{g}}{100 \mu\text{L}} = 0,5 \mu\text{g}/\mu\text{L} = 500 \text{ ppm} \\ \text{C} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{50 \mu\text{g}}{100 \mu\text{L}} = \frac{25 \mu\text{g}}{100 \mu\text{L}} = 0,25 \mu\text{g}/\mu\text{L} = 250 \text{ ppm} \\ \text{D} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{25 \mu\text{g}}{100 \mu\text{L}} = \frac{12,5 \mu\text{g}}{100 \mu\text{L}} = 0,125 \mu\text{g}/\mu\text{L} = 125 \text{ ppm} \\ \text{E} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{12,5 \mu\text{g}}{100 \mu\text{L}} = \frac{6,25 \mu\text{g}}{100 \mu\text{L}} = 0,0625 \mu\text{g}/\mu\text{L} = 62,5 \text{ ppm} \\ \text{F} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{6,25 \mu\text{g}}{100 \mu\text{L}} = \frac{3,125 \mu\text{g}}{100 \mu\text{L}} = 0,03125 \mu\text{g}/\mu\text{L} = 31,25 \text{ ppm} \\ \text{G} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{3,125 \mu\text{g}}{100 \mu\text{L}} = \frac{1,5625 \mu\text{g}}{100 \mu\text{L}} = 0,015625 \mu\text{g}/\mu\text{L} = 15,6 \text{ ppm} \end{aligned}$$

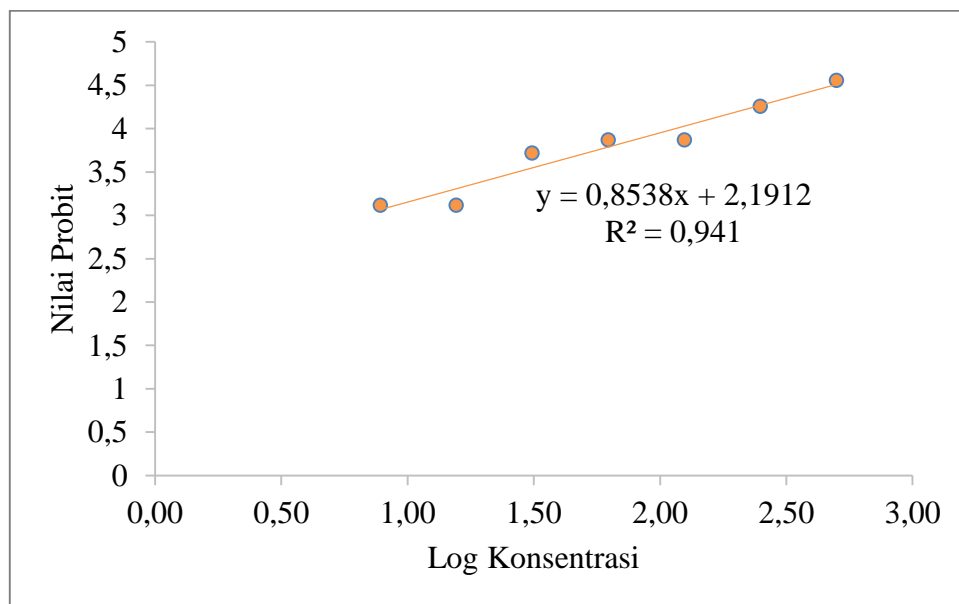
Konsentrasi sampel setelah penambahan media udang 100 μL , maka masing-masing lubang berisi 200 μL (pengenceran 2x)

$$\begin{aligned} \text{Baris A} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{100 \mu\text{g}}{100 \mu\text{L}} = 0,5 \mu\text{g}/\mu\text{L} = 500 \text{ ppm} \\ \text{B} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{50 \mu\text{g}}{100 \mu\text{L}} = \frac{25 \mu\text{g}}{100 \mu\text{L}} = 0,25 \text{ mg}/\mu\text{L} = 250 \text{ ppm} \\ \text{C} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{25 \mu\text{g}}{100 \mu\text{L}} = \frac{12,5 \mu\text{g}}{100 \mu\text{L}} = 0,125 \text{ mg}/\mu\text{L} = 125 \text{ ppm} \\ \text{D} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{12,5 \mu\text{g}}{100 \mu\text{L}} = \frac{6,25 \mu\text{g}}{100 \mu\text{L}} = 0,0625 \text{ mg}/\mu\text{L} = 62,5 \text{ ppm} \\ \text{E} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{6,25 \mu\text{g}}{100 \mu\text{L}} = \frac{3,125 \mu\text{g}}{100 \mu\text{L}} = 0,03125 \text{ mg}/\mu\text{L} = 31,25 \text{ ppm} \\ \text{F} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{3,125 \mu\text{g}}{100 \mu\text{L}} = \frac{1,5625 \mu\text{g}}{100 \mu\text{L}} = 0,015625 \text{ mg}/\mu\text{L} = 15,6 \text{ ppm} \\ \text{G} &= \frac{100 \mu\text{L}}{200 \mu\text{L}} \times \frac{1,5625 \mu\text{g}}{100 \mu\text{L}} = \frac{0,78125 \mu\text{g}}{100 \mu\text{L}} = 0,0078125 \text{ mg}/\mu\text{L} = 7,8 \text{ ppm} \end{aligned}$$

3. Perhitungan LC₅₀

a. Ekstrak *n*-Heksana

Konsentrasi Uji (µg/mL)	Log Konsentrasi	Jumlah Larva Udang Uji (Ekor)	Jumlah Larva Udang yang Mati				Kontrol Negatif				Persen Kematian (%)	Nilai Probit
			1	2	3	Rata-rata	1	2	3	Rata-rata		
500	2,70	10	5	4	4	4,33	1	1	1	1	33,33	4,56
250	2,40	10	2	2	3	2,33	0	0	0	0	23,33	4,26
125	2,10	10	1	2	1	1,33	0	0	0	0	13,33	3,87
62,5	1,80	10	0	2	2	1,33	0	0	0	0	13,33	3,87
31,2	1,49	10	1	1	1	1,00	0	0	0	0	10,00	3,72
15,6	1,19	10	1	0	0	0,33	0	0	0	0	3,33	3,12
7,8	0,89	10	0	1	0	0,33	0	0	0	0	3,33	3,12



$$y = 0,8538x + 2,1912$$

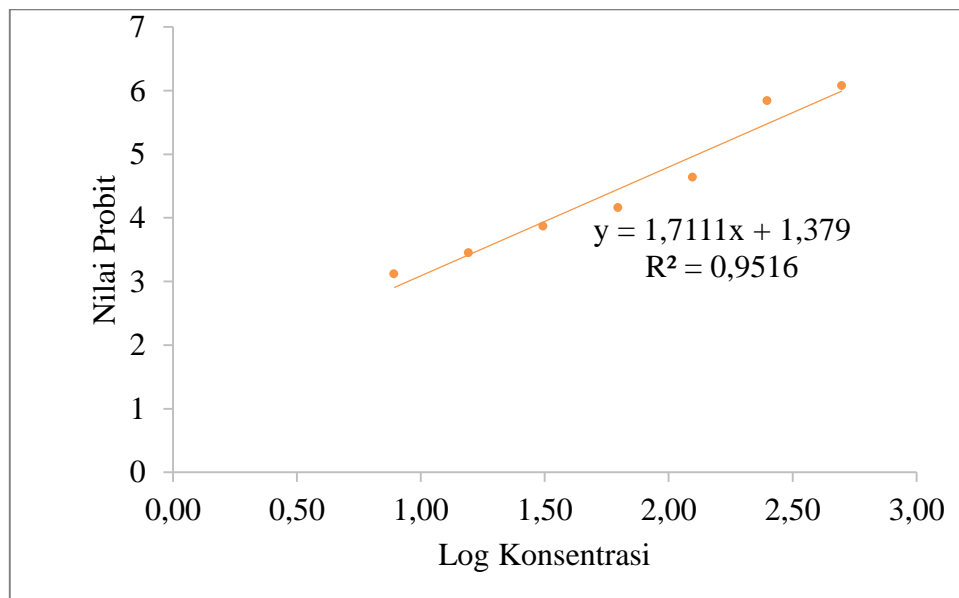
$$5 = 0,8538x + 2,1912$$

$$x = 3,2897634$$

$$LC_{50} = 10^x = 10^{(3,2897634)} = 1948,78 \mu\text{g/mL}$$

b. Ekstrak Etil Asetat

Konsentrasi Uji ($\mu\text{g/mL}$)	Log Konsentrasi	Jumlah Larva Udang Uji (Ekor)	Jumlah Larva Udang yang Mati				Kontrol Negatif				Persen Kematian (%)	Nilai Probit
			1	2	3	Rata-rata	1	2	3	Rata-rata		
500	2,70	10	10	10	9	9,67	1	1	1	1	86,67	6,08
250	2,40	10	8	7	9	8,00	0	0	0	0	80,00	5,84
125	2,10	10	3	4	4	3,67	0	0	0	0	36,67	4,64
62,5	1,80	10	2	1	3	2,00	0	0	0	0	20,00	4,16
31,2	1,49	10	1	2	1	1,33	0	0	0	0	13,33	3,87
15,6	1,19	10	0	1	1	0,67	0	0	0	0	6,67	3,45
7,8	0,89	10	0	0	1	0,33	0	0	0	0	3,33	3,12



$$y = 1,7111x + 1,379$$

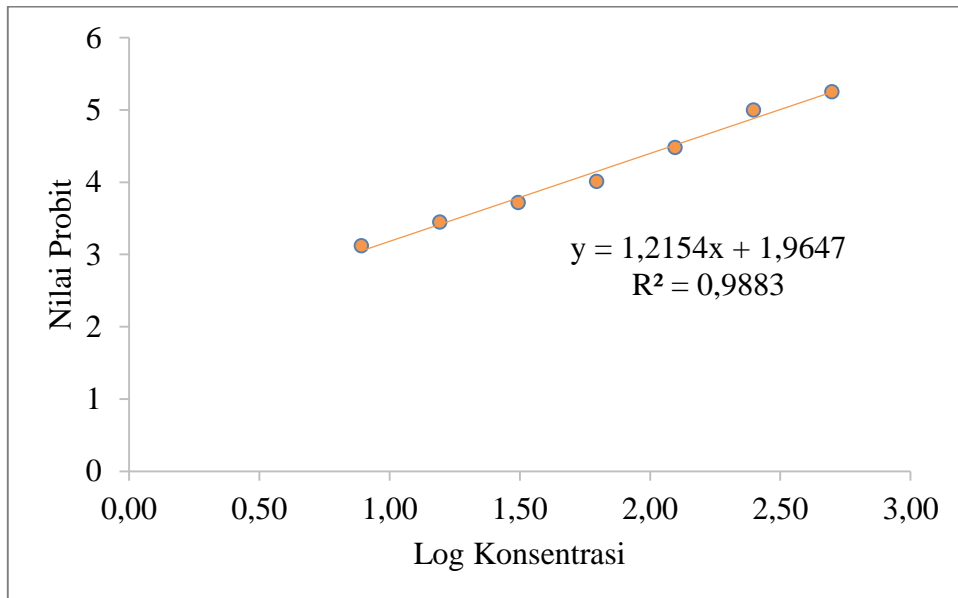
$$5 = 1,7111x + 1,379$$

$$x = 2,1161826$$

$$LC_{50} = 10^x = 10^{(2,1161826)} = 130,67 \mu\text{g/mL}$$

c. Ekstrak Metanol

Konsentrasi Uji ($\mu\text{g/mL}$)	Log Konsentrasi	Jumlah Larva Udang Uji (Ekor)	Jumlah Larva Udang yang Mati				Kontrol Negatif				Persen Kematian (%)	Nilai Probit
			1	2	3	Rata-rata	1	2	3	Rata-rata		
500	2,70	10	8	7	6	7,00	1	1	1	1	60,00	5,25
250	2,40	10	5	5	5	5,00	0	0	0	0	50,00	5
125	2,10	10	4	2	3	3,00	0	0	0	0	30,00	4,48
62,5	1,80	10	2	1	2	1,67	0	0	0	0	16,67	4,01
31,2	1,49	10	1	1	1	1,00	0	0	0	0	10,00	3,72
15,6	1,19	10	1	1	0	0,67	0	0	0	0	6,67	3,45
7,8	0,89	10	0	0	1	0,33	0	0	0	0	3,33	3,12



$$y = 1,2154x + 1,9647$$

$$5 = 1,2154x + 1,9647$$

$$x = 2,49736712$$

$$LC_{50} = 10^x = 10^{(2,49736712)} = 314,31 \mu\text{g/mL}$$

Lampiran 4. Dokumentasi



Pemotongan Sampel



Pengeringan Sampel



Penghalusan Sampel



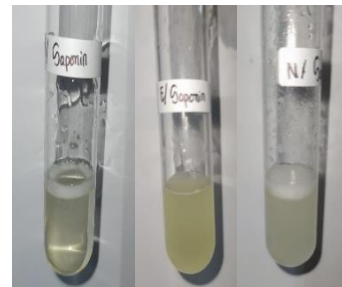
Maserasi Sampel



Penyaringan Hasil Maserasi



Pemekatan Maserat menjadi Ekstrak



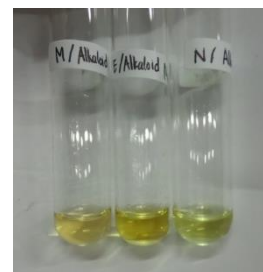
Uji Saponin



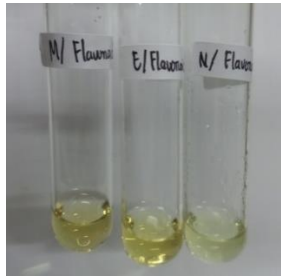
Uji Alkaloid
Dragendorff



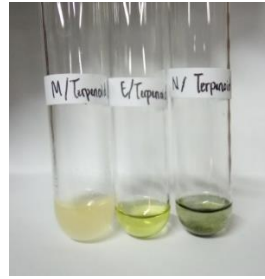
Uji Alkaloid Wagner



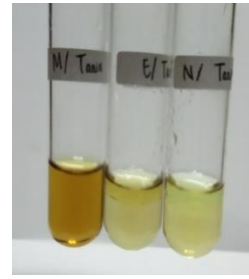
Uji Alkaloid Mayer



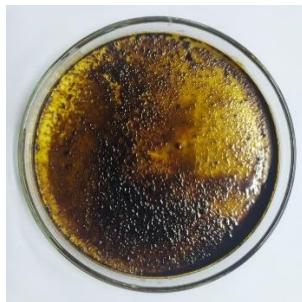
Uji Flavonoid



Uji Terpenoid/steroid



Uji Tanin



Ekstrak Etil Asetat



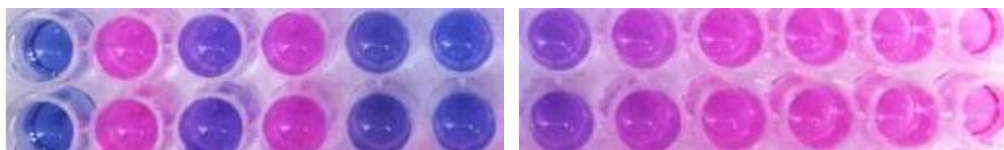
Ekstrak *n*-Heksana



Ekstrak Metanol



Uji Toksisitas Metode BSLT



Uji Aktivitas Antitumor Metode *Alamarblue*