

## DAFTAR PUSTAKA

- Abdullah, M., Khairurrijal. 2010. Karakterisasi Nanomaterial: Teori, Penerapan, dan Pengolahan Data. Bandung: CV. Rezeki Putera Bandung
- Abdullah, M., Virgus, Y., Nirmin, & Khairurrijal. 2008. Review: Sintesis nanomaterial. *Nanosains & Nanoteknologi*, 1.2: 33-57 ISSN 1979-0880.
- Ahmed, S., Ikram, S. 2015. Silver Nanoparticles: One Pot Green Synthesis Using *Terminalia arjuna* Extract for Biological Application. *J. Nanomed Nanotechnol* 6(4): 1-6, ISSN: 2157-7439 JNMNT
- Ali, M., Taleb, A.A., Remali N., Abdullah M., Srikanth V., Labhasetwar N., 2016. Dragon's blood aided synthesis of Ag/Ag<sub>2</sub>O core/shell nanostructures and Ag/Ag<sub>2</sub>O decked multi-layered graphene for efficient As(III) uptake from water and antibacterial activity. *J. Royal society of Chem.* 6: 44145-44153
- Amir, M.J.S., Wungouw, H., Pangemanan, D. 2015. Kadar glukosa darah sewaktu pada pasien diabetes mellitus tipe 2 di puskesmas Bahu Kota Manado. *J. e-Biomedik* 3 (1): 32-40
- Amooaghiae, R., Saeri, M.R., Azizi, M. 2015. Synthesis, characterization and biocompatibility of silver nanoparticles synthesised from *Nigella sativa* leaf extract in comparison with chemical silver nanoparticles. *Ecotoxicol. Environ. Saf.* 120: 400–408
- Anjum, S., Abbasi, H. 2016. Thidiazuron-enhanced biosynthesis and antimicrobial efficacy of silver nanoparticles via improving phytochemical reducing potential in callus culture of *Linum usitatissimum L.* *Intl. J. Nanomedicine* 11: 715-728
- Anwar, I.B., Patmasari, R., Fauzi, H. 2016. Perancangan dan implementasi alat pengukur kadar glukosa dalam darah secara *non-invasive* berbasis arduino. *J. e-Proceeding of engineering* 3 (3): 4665-4668
- Baru, D.O.B., Wahyuni, S., Hadisaputro, S., dan Harjono. 2013. Sintesis nanopartikel perak menggunakan metode poliol dengan



- agen stabilisator *Polivinilalkohol* (PVA). Jurnal MIPA 36 (2): 157-168. Universitas Negeri Semarang
- Ashokkumar, S., Ravi, S., Kathiravan, V., and Velmuruqan, S. 2015. Synthesis of silver nanoparticles using *A. indicum* leaf extract and their antibacterial activity. *Spectrochim. Acta A Mol. Biomol. Spectrosc.* 134: 34–39
- Asnawati, Indarti, D., Mulyono, T., Kesuma, G. 2013. Biosensor amperometri untuk deteksi glukosa berbasis immobilisasi glukosa oksidase dalam membran selulosa asetat dengan ferrocene sebagai mediator. *J. Ilmu dasar.* 14(1): 45-51
- Balavigneswaran, C. K., Sujin,J. K. T., Moses, P. R., Prakash,S., *Rapid detection of Cr(VI) by AgNPs probe produced by Anacardium occidentale fresh leaf extracts.* 2014. *Applied Nanoscience.* 4: 367-378.
- Banerjee, A. I. 2011. *Biosensor Nanomaterial.* Wiley-VCH. Germany
- Balasubramanian, K. and Burghard, M. 2006. Biosensors Based on Carbon Nanotubes. *Analytical and Bioanalytical Chemistry,* 385, 452-468.
- Buzea, C., Blandino, I.I.P., dan Robbie, K. 2007. Nanomaterial and nanoparticles: sources and toxicity, *Biointerphases,* 2: MR170–MR172
- Caro, C., Castillo, P.M., Klippstein, R., Pozo, D., dan Zaderenko, A.P. 2010. Silver nanoparticle: sensing and imaging application. In Perez, D.P. (Ed). *Silver nanoparticles* (pp. 210-223). India: InTech
- Chandran, S., P., Chaundary M., Pasricha R., Ahmad A., Sastry M., 2006. Synthesis of gold nanotriangles and silver nanoparticles using Aloe Vera plant extract. *Biotechnology progress* 22 : 577-583
- Chook, S.W., et al. 2012. Antibacterial performance of Ag nanoparticle and AgGO Nanocomposite prepared via rapid microwave-assisted synthesis method. *Nanoscale Research Letter.* 7:541
- ha, S. 2005. Ramuan tradisional untuk pengobatan Diabetes mellitus. Penebar Swadaya: Jakarta



- Dangi,R.,Shakya, S. 2013. Preparation, Optimization and Characterization of PLGA Nanoparticle. *Int J of Pharm & Life Sci.* 4(7): 2810-2818.
- Edison, T. N. J. I., Atchudan, R., Lee, Y. R. 2016. Optical Sensor for Dissolved Ammonia Through the Green Synthesis of Silver Nanoparticles by Fruit Extract of Terminalia chebula. *Journal of Cluster Science.* 27: 683-690.
- El-Nour, K. M. M. A., Eftaiha, A., Al-Warthan, A. dan Ammar, R. A. A., 2010, Synthesis and Application of Silver Nanoparticles, *Arabian Journal of Chemistry*, 3: 135–140.
- Elzey, S.R. 2010. *Applications and physicochemical characterization of nanomaterials in environmental, health, and safety studies.* Iowa: University of Iowa.
- Engida, M.A., Kasim, S.N., Tsigie, A.Y., Ismadji, S., Huynh, H.L., and Ju, H.Y. 2012. Extraction, identification and quantitative HPLC analysis of flavonoids from sarang semut (*Myrmecodia pendans*). *Industrial Crops and Products* 41 (2013) 392– 396. Elsevier
- Engida, M.A., Faika, S., Nguyen-Thi, T.B., and Ju, H.Y. 2014. Analysis of major antioxidants from extracts of *Myrmecodia pendans* by UV/visible spectrophotometer, liquid chromatography/tandem mass spectrometry, and high-performance liquid chromatography /UV techniques. *J. food and drug analysis* 23 (2015) 303-309. Elsevier
- Fernandez, B.R. 2011. Sintesis Nanopartikel. Makalah diterbitkan. Program Studi Kimia Pascasarjana Universitas Andalas Padang
- Gupta V, Karar PK. Optimization of Process Variables for the Preparation of Chitosan-Alginate Nanoparticle. *Int J Pharm Pharm Sci.* 2011; 3(2): 78-80.
- Handayani, W., Bakir, Imawan, C., dan Purbaningsih, S. 2010. Potensi ekstrak beberapa jenis tumbuhan sebagai agen pereduksi untuk biosintesis nanopartikel perak. Seminar nasional biologi pada tanggal 24-25 September 2010: 558-567.Universitas Gadjah Mada



M.I. 2012. Modifikasi nanopartikel perak dengan polivinil alkohol untuk meningkatkan selektivitas dan stabilitas indikator logam tembaga (Cu): uji coba pada mikroalga merah (*Kappaphycus*

*alvarezii*). Skripsi diterbitkan (online). FMIPA: Universitas Indonesia.

Hosokawa, M., Nishino, J., and Kanno, Y. 2007. *Nanoparticle Technology Handbook*, (1<sup>st</sup> ed). UK: Elsevier Linacre House, Jordan Hill, Oxford

International Diabetes Federation (IDF). (2015). IDF Diabetes Atlas Seventh Edition. International Diabetes Federation (IDF). Diperoleh tanggal 30 Januari 2018 dari [https://www.oedg.at/pdf/1606\\_IDF\\_Atlas\\_2015\\_UK.pdf](https://www.oedg.at/pdf/1606_IDF_Atlas_2015_UK.pdf)

Jain, S., Mehata, S.M. 2017. Medicinal Plant Leaf Extract and Pure Flavonoid Mediated Green Synthesis of Silver Nanoparticles and their Enhanced Antibacterial Property. *Scientific report* 7: 15867

Kannaiyan, S., Gopal, A. Biogenic synthesized silver colloid for colorimetric sensing of dichromate ion and antidiabetic studies. 2017. *Research on Chemical Intermediates*. 43: 2693-2706.

Kawashima, Y., Yamamoto, H., Takeuchi, H., and Kuno, Y. 2000. Mucoadhesive DLLactide/glycolide copolymer nanospheres coated with chitosan to improve oral delivery of elcatonin. *Pharmaceutical Development and Technology*. 5(1): 77-85

Kementerian Kesehatan. 2014. InfoDATIN. Pusat data dan Informasi Kementerian Kesehatan Republik Indonesia

Keppy,N.K.,Allen,M. 2008. Understanding spectral bandwidth and resolution in the regulated laboratory. USA: Thermo Fisher Scientific

Khan, A. N., Niaz, A., Zaman, I. M., Khan, A. F., Nisar, M., Tariq, M. 2018. Sensitive and selective colorimetric detection of Pb<sup>2+</sup> by silver nanoparticles synthesized from *Aconitum violaceum* plant leaf extract. *Materials research bulletin*. S0025-5408(17)33882-5

Kurniawati, E., Sianturi Y.C. 2016. Manfaat Sarang Semut (*Myrmecodia pendans*) sebagai Terapi Antidiabetes. *J. Majority* 5(3)

., and Vivekanandan, M. 2008. Tapping the unexploited plant resources for the synthesis of silver nanoparticle. *African Journal of Biotechnology* 7(17), 3162-3165



- Lembang, Y.E. 2013. Sintesis nanopartikel perak dengan metode reduksi menggunakan bioreduktor ekstrak daun ketapang (*Terminalia catappa*). Tesis. Universitas Hasanuddin
- Mardany, P.M., Chrystomo, Y.L., Karim, K.A. 2016. Skrining Fitokimia dan Uji Aktivitas Sitotoksik dari Tumbuhan Sarang Semut (*Myrmecodia beccarii* Hook f.) Asal Kabupaten Merauke. J. Biologi Papua 8 (1): 13–22.
- Mo, Y., Tang, Y., Wang, S., Lin, J., Zhang, H. 2015. Green Synthesis of Silver Nanoparticles Using *Eucalyptus* Leaf Extract. *Mater. Lett.* 144: 165-167
- Moores, A., Goettmann, F. 2006. The plasmon band in Noble metal nanoparticles : an Introduction to theory and applications. *New J. Chem.* 30: 1121-1132
- Moudir, N., Boukennousb, Y., Moulaï-Mostefa, N., Bozetine, I., Maoudj, M., Kamel, N., Kamel, Z., and Moudir, D. 2013. Preparation of Silver Powder Used for Solar Cell Paste by Reduction Process. *Energy Procedia*. 36: 1184-1191.
- Mukunthan, K.S., Balaji, S. 2012. Cashew apple juice (*Anacardium occidentale* L.) speeds up the synthesis of silver nanoparticles. *Intl. J. Green Nanotechnology*. 4(2): 71–79.
- Muliadi, Arief, A., Khadijah. 2015. Biosintesis nanopartikel logam menggunakan media ekstrak tanaman. *JF FIK UINAM* 3(2): 64-72
- Nagarajan, R. 2008. *Nanoparticles: building blocks for nanotechnology*. Dalam: Nagarajan R., Hatton T.A., 2008. *Nanoparticles: Synthesis, stabilization, passivation and functionalization*. American Chemical Society. 3: 4-6



013. Sintesis nanopartikel perak menggunakan ekstrak kayu manis (*Cinnamomum sp.*) sebagai bioreduktor. Universitas Hasanuddin.

rg, S.J. 2014. *Silver Nanoparticles: Properties and Applications*. [www.sigmaaldrich.com/materials-science/nanomaterials/silver](http://www.sigmaaldrich.com/materials-science/nanomaterials/silver)

- Pandey, S., Goswami, G. K., Nanda, K. K. 2012. Green synthesis of biopolymer–silver nanoparticle nanocomposite: An optical sensor for ammonia detection, *International Journal of Biological Macromolecules*, 51: 583-589
- Panigrahi, T. 2013. *Synthesis And Characterization Of Silver Nanoparticles Using Leaf Extract Of Azadirachta Indica*. National Institute of Technology. India
- Philip, D. 2010. Green Synthesis of Gold and Silver Nanoparticles Using *Hibiscus rosa sinensis*. *Phys. E.* Vol., 42: 1417–1424.
- Philip, D., Unni, C., Aromal, S.S., and Vidhu, V.K. 2011. *Murayya keonigii* Leaf-Assisted Rapid Green Syntesis of Silver and Gold Particles. *Spectrochimica Acta Part A: Molecular and Biomolecular* 78: 899-904.
- Poopathi, S., De brito, L.J., Praba, V.L., Mani, C., and Praveen, M. 2015. Synthesis of silver nanoparticles from *Azadirachta indica*—a most effective method for mosquito control. *Environ. Sci. Pollut. Res. Int.* 22: 2956–2963
- Pratama, I., 2014. Sensor optik pestisida berbasis nanopartikel perak dari ubi jalar ungu. Skripsi. Universitas Hasanuddin
- Prathna, T.C., Matthew, L., Chandrasekaran, N., Raichur, A.M., dan Mukherjee, E. 2010. Biomimetic synthesis of nanoparticles: Science, Technology and Application. In Mukherjee, A (Ed). *Biomimetics Learning from Nature* (pp. 1-20). India: InTech
- Prokopovich, P. 2016. *Biological and Pharmaceutical applications of nanomaterials*. CRC press. New York
- Psaro, Rinaldo, Guidotti, M., and Sgobba, M. 2009. Nanosystem. In Bertini, Ivano. *Inorganic and Bio-organic Chemistry* (vol.2, pp. 256-307).



P.,Imawan,C.2010.Pengembangan instrument pengkarakterisasi sensor elektrokimia menggunakan metode voltametri siklik', Jurnal Ilmu Pengetahuan dan Teknologi Vol 28.

- Purwaningsih, V.N. 2017. Perbandingan kadar glukosa darah sebelum dan sesudah minum kopi. *Journal of Muhammadiyah Medical laboratory technologist*. 2 (1): 61-66
- Rai, M., Yadav, A., and Gade, A. 2008. CRC 675— current trends in phytosynthesis of metal nanoparticles. *Critical Reviews in Biotechnology*. 28(4): 277–284.
- Raya, K.M., Legowo, M.A., dan Wijayahadi, N. 2016. Efektivitas ekstrak umbi sarang semut (*myrmecodia pendens* merr.& perry) sebagai penurun kadar glukosa darah tikus sprague dawley yang diabetes mellitus. *Jurnal Gizi Indonesia* 4(2): 138-144.
- Retnowati, Y., Uno, W., dan Rahman, S. 2012. Isolasi mikroba endofit tanaman sarang semut (*Myrmecodia pendens*) dan analisis potensi sebagai antimikroba. Skripsi. Universitas Negeri Gorontalo
- Ristian, I. 2013. Kajian Pengaruh Konsentrasi Perak Nitrat (AgNO<sub>3</sub>) Terhadap Ukuran Nanopartikel Perak. Skripsi (online). Universitas Negeri Semarang
- Sadeghi, B., Rostami, A., and Momeni, S.S. 2015. Facile green synthesis of silver nano-particles using seed aqueous extract of *Pistacia atlantica* and its antibacterial activity. *Spectrochim. Acta A Mol. Biomol. Spectrosc.* 134: 326–332
- Saeb, A. T. M., Alshammary, A. S., Al-brahim, H. dan Al-rubeaan, K. A., 2014, Production of Silver Nanoparticles with Strong and Stable Antimicrobial Activity against Highly Pathogenic and Multidrug Resistant Bacteria, *The Scientific Journal*, 1-9.
- Singh, C., Baboota, R.K., Naik, P.K., and Singh, H. 2012. Biocompatible Synthesis of Silver and Gold Nanoparticles Using Leaf Extract of *Dalbergia Sissoo*. *Research Article* 3(4): 279-285.

Singh, P., Kim, Y.J., Wang, C., Mathiyalaqan, R., Yang, D.C. 2015. The development of a green approach for the biosynthesis of silver and gold nanoparticles by using *Panax ginseng* root extract, and their biological applications. *Artif. Cells Nanomed. Biotechnol* 44 (4):1150-1157



- Singh, P., Kim, Y.J., El-agamy F.M., Yang, D.C. 2015. Biogenic silver and gold nanoparticles synthesised using red ginseng root extract, and their applications. *Artif. Cells Nanomed. Biotechnol.* 44(3):811-816
- Singh P., Kim J.Y., Zhang D., Yang C.D., 2016. Biological Synthesis of Nanoparticles from Plants and Microorganisms. *Trends in Biotechnology* 34 (7): 588-599
- Soegondo, S., Soewondo, P., dan Subekti, I. (2011). Penatalaksanaan diabetes melitus terpadu. (2<sup>nd</sup>ed). Fakultas Kedokteran Universitas Indonesia.
- Solomon, S.D., Bahadory, M., Jeyarajasingan, A.V., Rutkowsky, S.A., Boritz, C., and Mulfinger, L. 2007. Synthesis and study of silver nanoparticle. *Journal of Chemical Education.* 84(2): 322-325
- Subroto, M.A., Saputro, H. 2006. Gempur penyakit dengan sarang semut. Swadaya. Jakarta.
- Thakkar, K.N., Mathre, S.S., dan Parikh, N.Y.2010. Biological synthesis of metallic nanoparticle. *Nanomedicine: nanotechnology, biology and medicine.* 6: 257-262.
- Tiwari, A., Turner, F.P.A. 2014. *Biosensor Nanotechnology*. Wiley. Canada
- Torresdey, J.L., Gomez, E., Peralta-Videa, J.R., Parsons, J.G., Troiani, H., Jose-Yacaman, M. 2003. *Alfalfa sprouts : A natural source for the synthesis of silver nanoparticles.* *Langmuir.* 1357-1361
- Trisatya. 2015. Efek antidiabetes ekstrak etanol sarang semut (*Myrmecodia pendans*) terhadap mencit jantan (*Mus musculus*) yang diinduksi glukosa Skripsi (online). Gorontalo: Universitas Negeri Gorontalo
- Tripathy, A., Raichur, A.M., Chandrasekaran, N., Prathna, T.C., and Mukherjee, A. 2010. Process variable in biomimetic synthesis of silver nanoparticles by aqueous extract of *Azadirachta indica* (Neem) leaves. *J. Nanoparticles Research* 12 (1): 237-246



- Udapudi, B., Naik P., Savadatti, S.T., Sharma, R., and Balgi, S. 2012. Synthesis and Characterization of Silver Nanoparticles. *Int. J. Pharma. Bio. Sci.* 3: 10-14.
- Uddin, I., Ahmad, K., Khan, A. A., Kazmi, A. M. 2017. Synthesis of silver nanoparticles using Matricaria recutita (Babunah) plant extract and its study as mercury ions sensor. *Sensing and Bio-Sensing Research*. S2214-1804(17)30141-1
- Velmurugan, P., Park, J.H., Lee, S.M., Jang, J.S., Han, S.S., Cho, M.M., and Oh, B.T. 2015. Synthesis and characterization of nanosilver with antibacterial properties using *Pinus densiflora* young cone extract. *J. Photochem. Photobiol. B* 147: 63–68
- Wang, J. 2000. *Analytical Electrochemistry*, 2nd Ed., Wiley-VCH. New York ISBN: 0-471- 22823-0 (Electronic)
- Wang, Y., Barbieri, L. R., Berg, B. P., McAllister, T. A. 2007. Effects of mixing sainfoin with alfalfa on ensiling, ruminal fermentation and total tract digestion of silage. *Anim. Feed Sci. Technol.*, 135 (3-4): 296-314
- Widegren, J., Finke, R., Mol, J. 2003. Preparation of A Multifunctional CoreShell Nanocatalyst and Its Characterization. *Catal. A: Chem.* 191: 187.
- Wulandari, Putri. 2011. Kajian Sebaran Logam Berat Kromium (Cr), Mangan (Mn) dan Kobalt (Co) pada Sedimen Di Muara Sungai Way Kuala Bandar Lampung. Universitas Lampung. Bandar Lampung
- Xia, B., Cui, Q., He, F., and Li, L. 2012. Preparation of Hybrid Hydrogel Containing Ag Nanoparticles by a Green In Situ Reduction Method. *Langmuir*. 28: 11188-11194.
- Yao, Y., Tian, D.M., and Li, H.B. 2010. Cooperative Binding of Bifunctionalized and Click Synthesized Silver Nanoparticles for Colorimetric  $\text{Co}^{2+}$  Sensing. *ACS App. Mater. Interfa.* 2(3): 684-690.
- I. 2012. Efek pemberian ekstrak etanol 70% umbi sarang semut (*Hydnophytum moseleyanum* Becc.) terhadap kadar asam urat



tikus putih jantan yang diinduksi kalium oksonat. Skripsi (online). Universitas Indonesia.

- Zahir, A.A., Chauhan, I.S., Baqavan, A., Kamaraj, C., Shankar, J., Roopan, S.M., and Singh, N. 2015. Green synthesis of silver and titanium dioxide nanoparticles using *Euphorbia prostrata* extract shows shift from apoptosis to G0/G1 arrest followed by necrotic cell death in *Leishmania donovani*. *Antimicrob. Agents Chemother.* 59: 4782–4799
- Zhang, Y., Wang, L., Tian, J., Li, H., Lou, Y., and Sun, X. 2011. Ag poly (m-phenylenediamine) core–shell nanoparticles for highly selective, multiplex nucleic acid detection. *Langmuir*. 27: 2170–2175.
- Zuas, O., Hamim, N., Sampora, Y., 2014. Bio-synthesis of silver nanoparticles using water extract of *Myrmecodia pendan* (Sarang Semut plant). *Mater. Lett.* 123 (2014) 156–159. Elsevier



## Lampiran 1

### Pembuatan ekstrak umbi Sarang Semut (*Myrmecodia Pendans*)

#### a. Persiapan umbi Sarang Semut



Diambil dari hutan

- dibersihkan dari kotoran
- dibuang bagian ujung sarang semut yang berdaun dengan menggunakan pisau
- dikupas kulit luar sarang semut dengan menggunakan pisau
- dibelah menjadi 4 bagian
- dicuci dengan akuabides
- dirajang tipis-tipis
- dikering-anginkan selama 1 hari
- dioven selama  $\pm 30$  menit pada suhu  $50^{\circ}\text{C}$

Sampel sarang semut kering

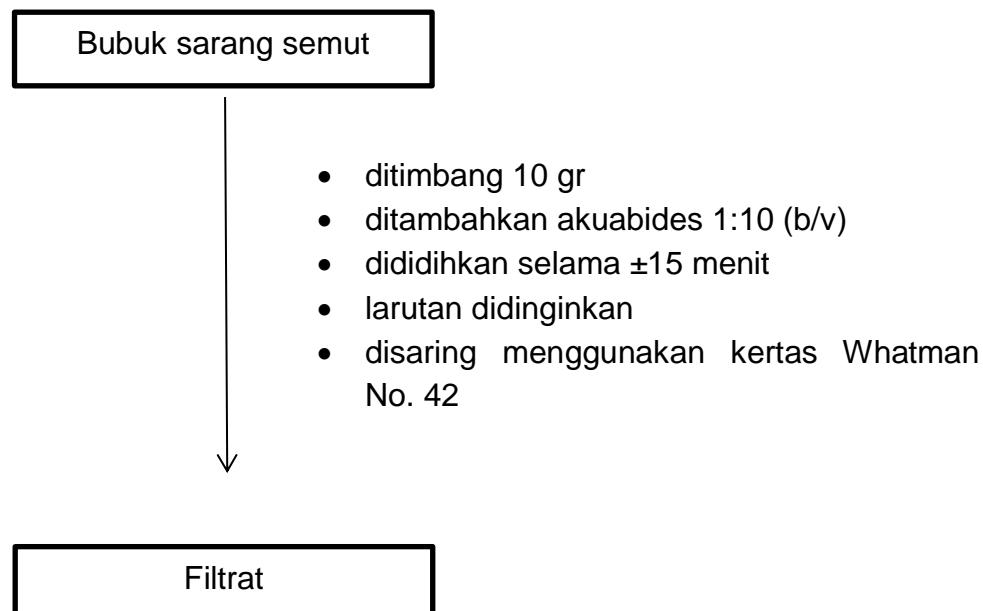
- diblender

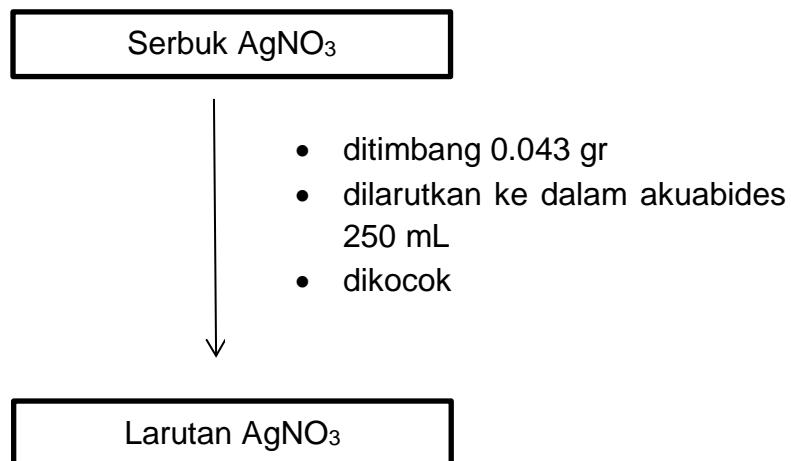
Bubuk sarang semut



Optimization Software:  
[www.balesio.com](http://www.balesio.com)

b. Ekstraksi Sarang Semut



**Lampiran 2****Pembuatan larutan induk AgNO<sub>3</sub> 1 mM**

Optimization Software:  
[www.balesio.com](http://www.balesio.com)

**Lampiran 3 Variasi konsentrasi dan lama pengadukan****a. Variasi konsentrasi larutan AgNO<sub>3</sub>**

Masing-masing 2 mL Ekstrak sampel Sarang Semut

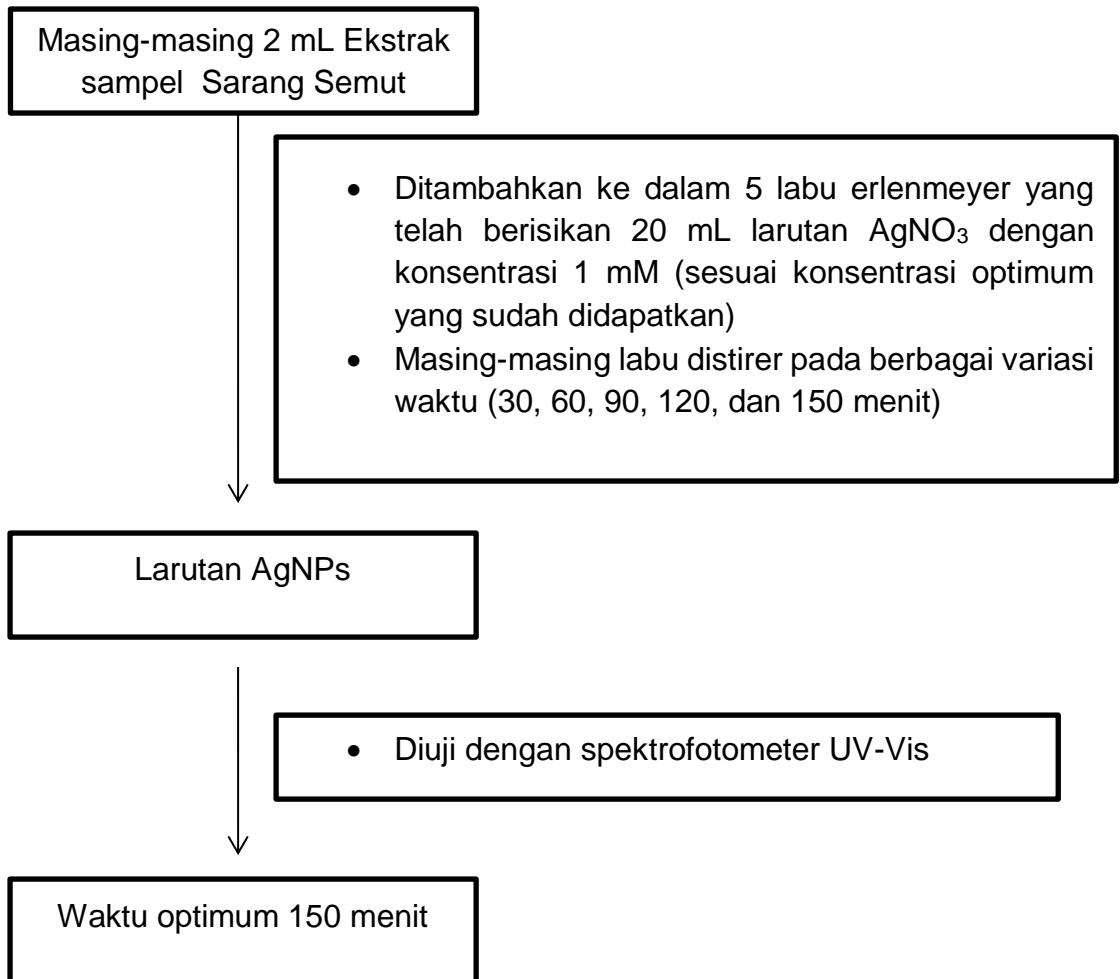
- Ditambahkan ke dalam 5 labu erlenmeyer yang telah berisikan 20 mL larutan AgNO<sub>3</sub> pada berbagai konsentrasi (0.5, 1, 1.5, 2, 2.5 mM)
- Masing-masing labu distirer selama 2 jam.

Larutan AgNPs

- Diuji dengan spektrofotometer UV-Vis

Konsentrasi optimum 1 mM

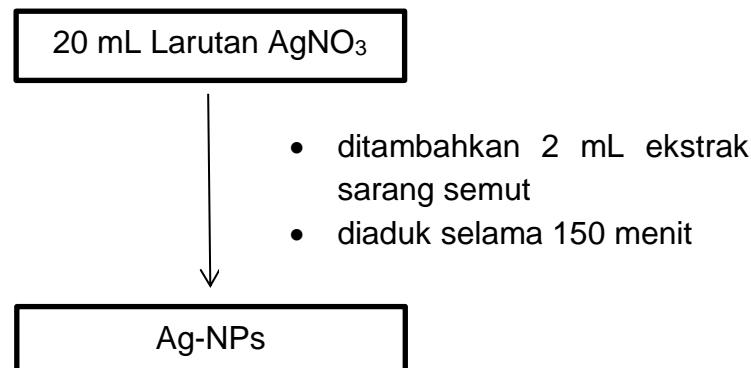


**b. Variasi lama pengadukan**

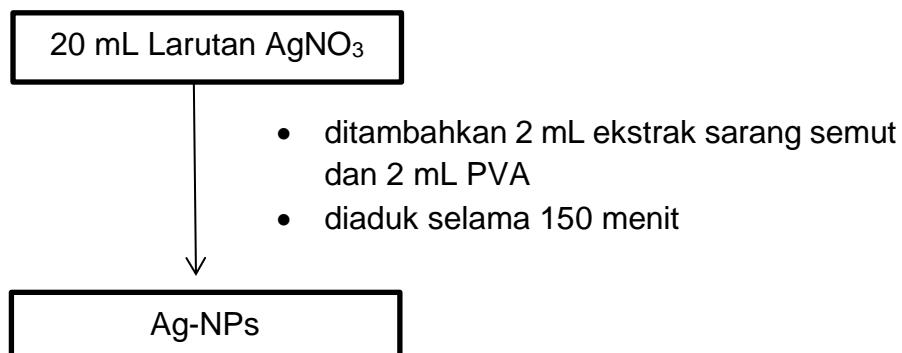
## Lampiran 4

### Sintesis nanopartikel perak

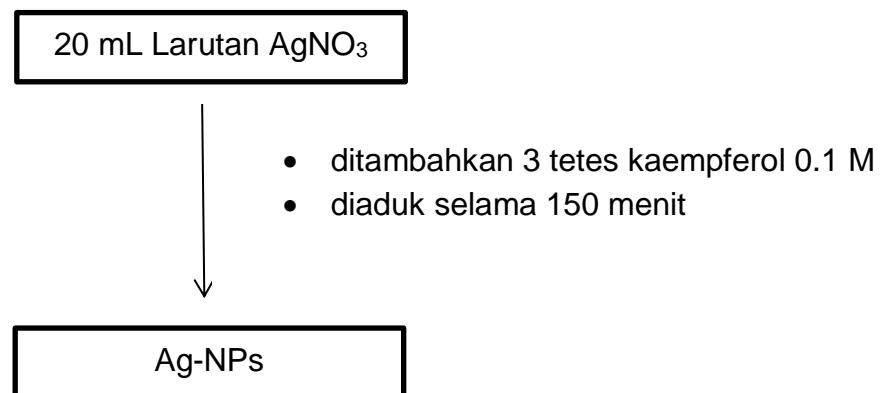
a. Dengan ekstrak umbi Sarang Semut tanpa penstabil



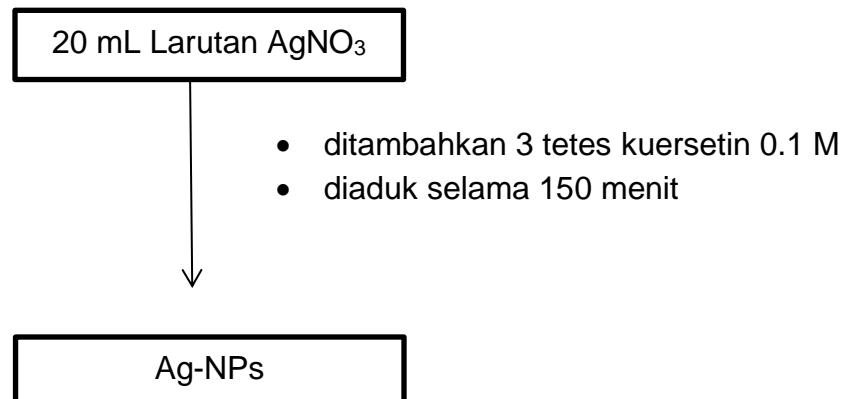
b. Dengan ekstrak umbi Sarang semut menggunakan penstabil



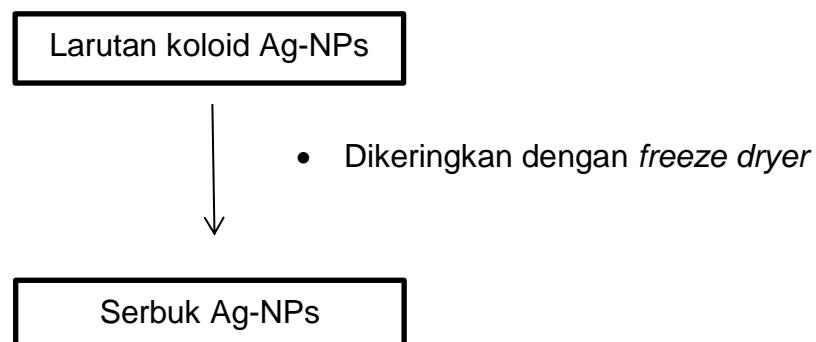
c. Dengan Kaempferol



d. Dengan kuersetin

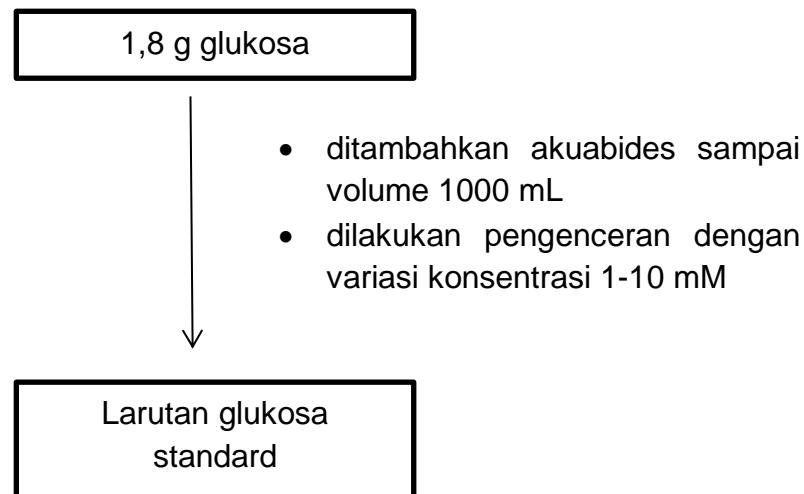


e. Pengeringan Nanopartikel Ag

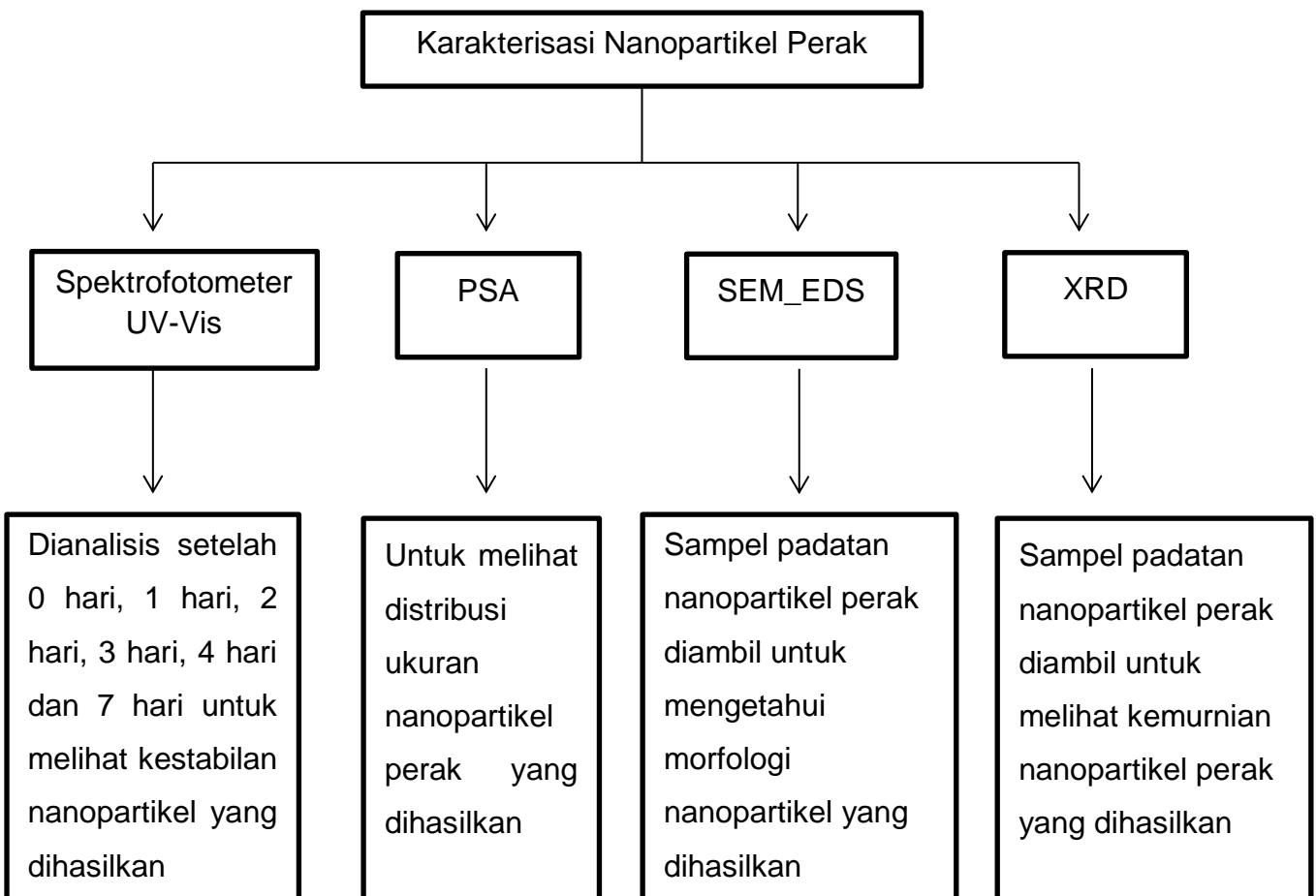


## Lampiran 5

### Pembuatan larutan glukosa standard

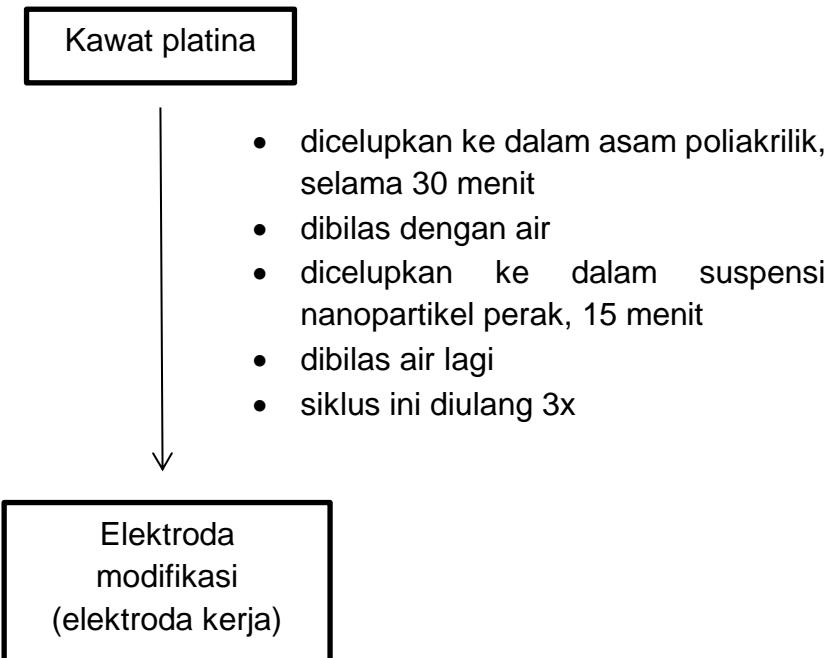


Optimization Software:  
[www.balesio.com](http://www.balesio.com)

**Lampiran 6****Karakterisasi Nanopartikel Perak**

## Lampiran 7

### Persiapan elektroda modifikasi (elektroda kerja)



## Lampiran 8

### Perhitungan XRD

#### a. 3 puncak tanpa penstabil

Peak 1.  $2\theta = 37.80$

$$\begin{aligned}
 d_s &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(1.032^\circ) \cos(\frac{37.80}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180^\circ}\right) \times 1.032^\circ \cos(18.9)} \\
 &= \frac{0.15092 \text{ nm}}{0.018002 \times 0.946085} \\
 &= \frac{0.15092}{0.01703} \\
 &= 8.862 \text{ nm}
 \end{aligned}$$

Peak 2.  $2\theta = 64.24$

$$\begin{aligned}
 d_s &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(0.660^\circ) \cos(\frac{64.24}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180^\circ}\right) \times 0.660^\circ \cos(32.12)} \\
 &= \frac{0.15092}{0.011513 \times 0.846936}
 \end{aligned}$$



$$\begin{array}{r}
 0.15092 \\
 \hline
 0.00975 \\
 \\ 
 5.478 \text{ nm}
 \end{array}$$

Peak 3.  $2\theta = 77.46$

$$\begin{aligned}
 ds &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(0.365^\circ) \cos(\frac{77.46}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180^\circ}\right) \times 0.365^\circ \cos(38.73)} \\
 &= \frac{0.15092}{0.006367 \times 0.780103} \\
 &= \frac{0.15092}{0.00496} \\
 &= 30,427 \text{ nm}
 \end{aligned}$$

#### b. 4 puncak dengan penstabil

Peak 1.  $2\theta = 37.775$

$$\begin{aligned}
 ds &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(0.55660^\circ) \cos(\frac{37.775}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180^\circ}\right) \times 0.55660^\circ \cos(18.8875)} \\
 &= \frac{0.15092}{0.009709 \times 0.946156} \\
 &= \frac{0.15092}{0.009186}
 \end{aligned}$$



43 nm

Optimization Software:  
[www.balesio.com](http://www.balesio.com)

Peak 2.  $2\theta = 44.047$

$$\begin{aligned}
 ds &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(0.2897^\circ) \cos(\frac{44.047}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180}\right) \times 0.2897^\circ \cos(22.023)} \\
 &= \frac{0.15092}{0.005053 \times 0.927033} \\
 &= \frac{0.15092}{0.00468} \\
 &= 32.24 \text{ nm}
 \end{aligned}$$

Peak 3.  $2\theta = 64.38$

$$\begin{aligned}
 ds &= \frac{k\lambda}{B \cos \theta} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{(0.3159^\circ) \cos(\frac{64.38}{2})} \\
 &= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180}\right) \times 0.3159^\circ \cos(32.19)} \\
 &= \frac{0.15092}{0.005510 \times 0.846286} \\
 &= \frac{0.15092}{0.00466}
 \end{aligned}$$

38 nm



Peak 4.  $2\theta = 77.49$

$$ds = \frac{k\lambda}{B \cos \theta}$$

$$= \frac{0.98 \times 0.154 \text{ nm}}{(0.3405^\circ) \cos(\frac{77.49}{2})}$$

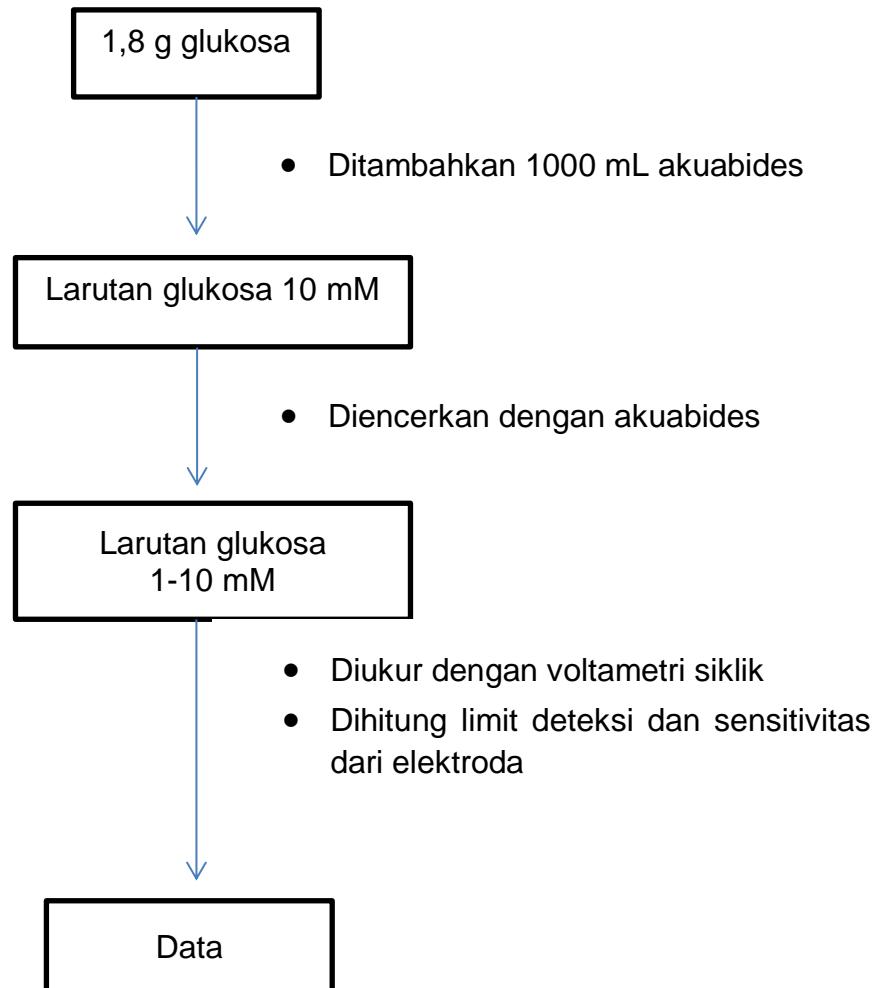
$$= \frac{0.98 \times 0.154 \text{ nm}}{\left(\frac{3.14}{180^\circ}\right) \times 0.3405^\circ \cos(38.745)}$$

$$= \frac{0.15092}{0.005939 \times 0.779939}$$

$$= \frac{0.15092}{0.00463}$$

$$= 32.59 \text{ nm}$$



**Lampiran 9. Bagan kerja pengujian terhadap larutan gula standar**

## Lampiran 10. Hasil pengukuran nanopartikel perak dengan menggunakan XRD

```
*** Basic Data Process ***

Group      : Standard
Data       : tp101

# Strongest 3 peaks
no. peak    2Theta      d          I/I1      FWHM      Intensity   Integrated Int
no.          (deg)     (A)        (deg)      (deg)      (Counts)   (Counts)
  1   1       31.9351  2.80015   100  0.33590      765      15518
  2   16      45.9440  1.97370   60   0.34220      461       8736
  3   7       37.8000  2.37808   33   1.03200      255       9124

# Peak Data List
peak    2Theta      d          I/I1      FWHM      Intensity   Integrated Int
no.          (deg)     (A)        (deg)      (deg)      (Counts)   (Counts)
  1   31.9351  2.80015   100  0.33590      765      15518
  2   33.4800  2.67438     3  0.31000      23        491
  3   33.9500  2.63843     7  0.62000      51       1796
  4   36.8200  2.43909     5  0.20000      42        545
  5   37.2000  2.41504    11  0.64000      86       1659
  6   37.4800  2.39764    24  0.50660     182       2002
  7   37.8000  2.37808    33  1.03200      255       9124
  8   38.6600  2.32713    10  0.00000      77         0
  9   38.9200  2.31218     6  0.00000      48         0
 10   39.4800  2.28066     6  0.40000      45      1805
 11   41.2497  2.18681    15  0.28850     113      2134
 12   43.6600  2.07151     4  0.44000      32      1102
 13   44.0016  2.05622    11  0.29670      81      1034
 14   44.4000  2.03869     3  0.28000      25        564
 15   45.5400  1.99027     7  0.18660      54       911
 16   45.9440  1.97370    60  0.34220      461      8736
 17   48.8595  1.86253     4  0.32900      34       701
 18   51.1893  1.78310     5  0.39470      35       947
 19   54.5469  1.68101    18  0.35200     136      2906
 20   57.2111  1.60889    21  0.35110     160      2852
 21   57.5000  1.60149     8  0.20400      60       756
 22   57.8250  1.59326     3  0.17000      23       224
 23   64.2400  1.44876    12  0.66000      93      3735
 24   67.1983  1.39199     7  0.30330      55       987
 25   67.9116  1.37910     4  0.31670      34       600
 26   68.3783  1.37082     3  0.26330      24       283
 27   68.7366  1.36455     4  0.28670      27       513
 28   74.1675  1.27748     4  0.33500      34       827
 29   75.5500  1.25750     3  0.38000      26       471
 30   76.0600  1.25034     5  0.34000      36       719
 31   76.5460  1.24360    20  0.50800     152      4051
 32   77.0000  1.23740    13  0.00000     102         0
 33   77.4600  1.23120    15  0.36500     115      3698
```



```

*** Basic Data Process ***

# Data Information
    Group          : Standard
    Data           : tp101
    Sample Name    : sarbuk
    Comment        :
    Date & Time   : 08-24-18 09:55:14

# Measurement Condition
    X-ray tube
        target       : Cu
        voltage      : 40.0 (kV)
        current      : 30.0 (mA)
    Slits
        Auto Slit    : Used
        divergence slit : 1.00000 (deg)
        scatter slit   : 1.00000 (deg)
        receiving slit : 0.30000 (mm)
    Scanning
        drive axis    : Theta-2Theta
        scan range     : 30.0000 - 80.0000 (deg)
        scan mode      : Continuous Scan
        scan speed     : 2.0000 (deg/min)
        sampling pitch : 0.0200 (deg)
        preset time    : 0.60 (sec)

# Data Process Condition
    Smoothing      [ AUTO ]
        smoothing points : 19
    B.G. Subtraction [ AUTO ]
        sampling points : 21
        repeat times   : 30
    Kal-a2 Separation [ MANUAL ]
        Kal a2 ratio   : 50 (%)
    Peak Search
        differential points : 19
        FWHM threshold    : 0.050 (deg)
        intensity threshold : 30 (par mil)
        FWHM ratio (n-1)/n : 2
    System error Correction [ NO ]
    Precise peak Correction [ NO ]

```



```

*** Basic Data Process ***

Group : Standard
Data  : p101

# Strongest 3 peaks
no. peak 2Theta      d      I/I1    FWHM      Intensity Integrated Int
      no. (deg)       (A)      (deg)      (Counts)   (Counts)
  1   3  31.9179  2.80161  100  0.44180     337     8122
  2  39  77.4990  1.23068   86  0.34050     290     5782
  3   9  37.7750  2.37959   64  0.55660     214     4778

# Peak Data List
peak 2Theta      d      I/I1    FWHM      Intensity Integrated Int
no. (deg)       (A)      (deg)      (Counts)   (Counts)
  1  30.8050  2.90025   3  0.29000     10     198
  2  31.3600  2.85017   7  0.14000     23     263
  3  31.9179  2.80161  100  0.44180     337     8122
  4  33.5600  2.66819   3  0.32000     11     299
  5  33.9606  2.63763   14  0.37470     47     827
  6  34.4200  2.60347   4  0.28000     12     291
  7  36.1600  2.48208   3  0.24000     10     196
  8  36.9800  2.42890   9  0.44000     32     1297
  9  37.7750  2.37959   64  0.55660     214     4778
 10 38.2600  2.35053   24  0.80000     80     2712
 11 38.6800  2.32597  12  0.00000     39      0
 12 38.8600  2.31561   8  0.80000     27     773
 13 39.4960  2.27978  16  0.31200     55     838
 14 41.1825  2.19023  12  0.40500     41     914
 15 43.5600  2.07604   5  0.12000     17     222
 16 44.0475  2.05418  54  0.28970    182     2699
 17 44.5567  2.03188   3  0.18000     11     161
 18 45.9191  1.97472  63  0.43830    212     5301
 19 48.8000  1.86466   4  0.28000     13     282
 20 51.1700  1.78372   5  0.38000     16     300
 21 51.5000  1.77307   4  0.24000     12     141
 22 54.5366  1.68130  18  0.44670     61     1620
 23 56.7200  1.62165   4  0.24000     12     202
 24 57.2753  1.60724  24  0.57730     81     2213
 25 57.8800  1.59188   3  0.16000     10     116
 26 63.7600  1.45851   6  0.32000     19     657
 27 64.3800  1.44595  59  0.31590    200     3064
 28 64.7200  1.43917   5  0.26000     18     533
 29 67.1900  1.39214   8  0.50000     26     739
 30 67.9433  1.37853   8  0.32670     26     416
 31 68.3400  1.37149   4  0.28000     14     219
 32 68.7788  1.36381  24  0.24960     81     1003
 33 69.2453  1.35576   5  0.16270     18     177
 34 71.8616  1.31270   3  0.18330     10     152
 35 74.1000  1.27848   6  0.36000     19     416
 36 75.4950  1.25828   5  0.29000     17     273
 37 75.9800  1.25145   6  0.33340     19     419
 38 76.4800  1.24451  24  0.56000     81     2956
 39 77.4990  1.23068   86  0.34050    290     5782

```



```
*** Basic Data Process ***

# Data Information
    Group          : Standard
    Data           : p101
    Sample Name    : serbuk
    Comment        :
    Date & Time   : 08-24-18 09:25:57

# Measurement Condition
    X-ray tube
        target      : Cu
        voltage     : 40.0 (kV)
        current     : 30.0 (mA)
    Slits
        Auto Slit   : Used
        divergence slit : 1.00000 (deg)
        scatter slit  : 1.00000 (deg)
        receiving slit : 0.30000 (mm)
    Scanning
        drive axis   : Theta-2Theta
        scan range   : 30.0000 - 80.0000 (deg)
        scan mode    : Continuous Scan
        scan speed   : 2.0000 (deg/min)
        sampling pitch : 0.0200 (deg)
        preset time  : 0.60 (sec)

# Data Process Condition
    Smoothing      [ AUTO ]
        smoothing points : 25
    B.G. Subtraction [ AUTO ]
        sampling points : 27
        repeat times   : 30
    Kal-a2 Separate [ MANUAL ]
        Kal a2 ratio   : 50 (%)
    Peak Search      [ AUTO ]
        differential points : 23
        FWHM threshold   : 0.050 (deg)
        intensity threshold : 30 (par mil)
        FWHM ratio (n-1)/n : 2
    System error Correction [ NO ]
    Precise peak Correction [ NO ]
```



## Lampiran 11. Hasil pengukuran nanopartikel perak dengan menggunakan PSA



Delsa™ Nano  
Common

### Condition Summary

S/N : 123909

User : Common	Group :	Repetition : 1/1
Date : 8/14/2018	File Name : SS - Ag 1_20180814_145558	
Time : 14:55:58	Sample Information :	
SOP Name : Sampel Uji PSA		Security : No Security

Version 2.31 / 2.03

### Measurement Condition

Sampling Time	: N/A	( $\mu$ s)		
Correlation Channel	: 440	(ch)	Correlation Method	: TD
Accumulation times	: 30	(times)	Attenuator 1	: 30.38 (%)
Cell Center	: Z : 3.000	(mm)	Pinhole	: 50 ( $\mu$ m)
	X : 7.500	(mm)		
Scattering Angle	: 165.0	( $^{\circ}$ )	Temperature	: 24.9 ( $^{\circ}$ C)
Diluent Name	: WATER		Viscosity	: 0.8898 (cP)
Refractive Index	: 1.328			
Intensity	: 8788	(cps)		

### Cumulants Results

Mean Diameter (d) : 76.1	(nm)	Diffusion Constant (D) : 6.447e-008	(cm <sup>2</sup> /sec)
Polydispersity Index (P.I.) : 0.324		Decay Constant ( $\Gamma$ ) : 4080.7	(1/sec)

### Fitting Parameter

Analysis Method	: CONTIN			
Histogram Range	: 10.0 - 4000.0	(nm)	Cut	Left : 0 Right : 0
Fitting Range	: 1.003 - 2			
Noise Cut Level	: 0.3	(%)		
Residual	: 8.713e-003 [OK]			



Optimization Software:  
[www.balesio.com](http://www.balesio.com)



Delsa™ Nano  
Common

#### Condition Summary

S/N : 123909

User : Common	Group :	Repetition : 1/1
Date : 9/10/2018	File Name :	HS - PVA_20180910_135101
Time : 13:51:01	Sample Information :	
SOP Name : Sampel Uji PSA		Security : No Security

Version 2.31 / 2.03

#### Measurement Condition

Sampling Time	: N/A	(μs)		
Correlation Channel	: 440	(ch)	Correlation Method	: TD
Accumulation times	: 30	(times)	Attenuator 1	: 65.58 (%)
Cell Center	: Z : 3.000	(mm)	Pinhole	: 50 (μm)
	X : 7.500	(mm)		
Scattering Angle	: 165.0	(°)	Temperature	: 25.0 (°C)
Diluent Name	: WATER		Viscosity	: 0.8878 (cP)
Refractive Index	: 1.3328			
Intensity	: 8096	(cps)		

#### Cumulants Results

Mean Diameter (d) : 78.3	(nm)	Diffusion Constant (D) : 6.286e-008	(cm²/sec)
Polydispersity Index (P.I.) : 0.303		Decay Constant (Γ) : 3978.8	(1/sec)

#### Fitting Parameter

Analysis Method	: CONTIN			
Histogram Range	: 10.0 - 4000.0 (nm)	Cut	Left : 0	Right : 0
Fitting Range	: 1.003 - 2			
Noise Cut Level	: 0.3 (%)			
Residual	: 1.042e-002 [OK]			



Optimization Software:  
[www.balesio.com](http://www.balesio.com)

Perhitungan konsentrasi glukosa dalam darah

Diketahui : Kuat arus (y)= 2,67 mA

$$y = 0,961x + 0,495$$

$$2,67 = 0,961x + 0,495$$

$$x = \frac{2,67 - 0,495}{0,961}$$

$$= 2,263 \text{ mM}$$

Faktor pengenceran 10/5=2

Konsentrasi glukosa =  $2,263 \times 2 = 4,53 \text{ mM}$

$$4,53 \text{ mM} = 4,53 \text{ mM} \times M_r$$

$$= 4,53 \text{ mmol/L} \times 180 \text{ mg/mmol}$$

$$= 815,4 \text{ mg/L}$$

$$= 81,54 \text{ mg/dL}$$

Jadi, konsentrasi glukosa dalam sampel darah berdasarkan analisis dengan sensor berbasis nanopartikel perak adalah 4,53 mM atau 81,54 mg/dL.



### Lampiran 12. Perhitungan limit deteksi

$$y = 0,961x + 0,495$$

$$\begin{aligned} y &= 0,184x + 4,016 \\ &\hline &= 0,777x - 3,521 \end{aligned}$$

$$x = \frac{3,521}{0,777}$$

$x = 4,53 \text{ mM}$  (garis singgung pada fungsi linear Nernstian dan non-Nernstian)

Limit deteksi untuk konsentrasi glukosa yang terukur:

Konsentrasi glukosa (x)	Kuat arus (y)	$y_i$	$(y-y_i)^2$
1	1,98	1,456	0,275
2	2,34	2,417	0,006
3	2,67	3,378	0,501
4	3,89	4,339	0,201
5	6,01	5,3	0,504
			$\Sigma = 1,487$

$y_i$  = nilai x yang dimasukkan dalam persamaan regresi.

$$S(y/x)^2 = \frac{\sum (y - y_i)^2}{n-2} = \frac{1,487}{3} = 0,4956 ; S(y/x) = \sqrt{0,4956} = 0,704$$



$$\text{eksi} = \frac{3 S(\frac{y}{x})}{b} = \frac{3 \times 0,704}{0,961} = 2,19 \text{ mM}$$

**Lampiran 13. Uji statistik perbedaan hasil pengukuran glukosa darah metode sensor berbasis nanopartikel perak dengan alat *Automated analyzed clinical chemistry***

Pengukuran ke-	Sensor		<i>Automated analyzed clinical chemistry</i>
	Kuat arus (mA)	Konsentrasi	
1	2,63	4,442 (79,95 mg/dL)	77 mg/dL
2	2,66	4,504 (81,07 mg/dL)	77 mg/dL
3	2,71	4,608 (82,94 mg/dL)	77 mg/dL
Rata-rata	2,67	4,53 (81,54 mg/dL)	77 mg/dL

Diketahui;

$$\bar{x}_1 = 81,54 \quad n_1 = 3$$

$$\bar{x}_2 = 77 \quad n_2 = 3$$

Ditanyakan;

$H_0$ = Ada perbedaan yang bermakna antara hasil pemeriksaan glukosa darah metode sensor berbasis nanopartikel perak dengan alat *Automated analyzed clinical chemistry*

$H_a$ = Tidak ada perbedaan yang bermakna antara hasil pemeriksaan glukosa darah metode sensor berbasis nanopartikel perak dengan alat *Automated analyzed clinical chemistry*

Kriteria penerimaan dan penolakan yaitu  $H_0$  diterima bila  $t_{hitung} < t_{tabel}$  dan sebaliknya  $H_a$

bila  $t_{hitung} > t_{tabel}$



Penyelesaian;

$$\bar{S} = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n}}$$

$$\bar{S}_{\text{sensor}} = \sqrt{\frac{\sum (x_i - \bar{x}_1)^2}{n}}$$

$$= \sqrt{\frac{(79,95 - 81,54)^2 + (81,07 - 81,54)^2 + (82,94 - 81,54)^2}{3}}$$

$$= 1,252$$

$$\bar{S}_2 = \sqrt{\frac{\sum (x_i - \bar{x}_2)^2}{n}}$$

$$= \sqrt{\frac{(77 - 77)^2 + (77 - 77)^2 + (77 - 77)^2}{3}}$$

$$= 0$$

$$S_{\text{gabungan}} = \sqrt{\frac{(n_1+1)\bar{S}_1^2 + (n_2+1)\bar{S}_2^2}{n_1+n_2-2}}$$

$$= \sqrt{\frac{(3+1)(1,252)^2 + (3+1)(0)^2}{3+3-2}}$$

$$= 1,251$$

Uji t



Optimization Software:  
[www.balesio.com](http://www.balesio.com)

$$\frac{\bar{X}_1 + \bar{X}_2}{ab \sqrt{\frac{1}{n_1} + \frac{1}{n_2}}}$$

$$= \frac{81,54 - 77}{1,251 \sqrt{\frac{1}{3} + \frac{1}{3}}}$$

$$= 4,45$$

Menunjukkan bahwa  $t_{\text{hitung}}$  4,45 >  $t_{\text{tabel}}$  2,13 dengan tingkat kepercayaan 95% maka  $H_a$  diterima.

Untuk data pada pengujian 1

Pengukuran ke-	Kuat arus	Konsentrasi	<i>Automated analyzed clinical chemistry</i>
1	0,60	0,218 (3,924 mg/dL)	77 mg/dL
2	3,10	5,421 ( 97,57 mg/dL)	77 mg/dL
3	3,53	6,3 (113,4 mg/dL)	77 mg/dL
Rata-rata	2,41	71,71 mg/dL	77 mg/dL

Diketahui;

$$\bar{x}_1 = 71,71 \quad n_1 = 3$$

$$\bar{x}_2 = 77 \quad n_2 = 3$$

Ditanyakan;

$H_0$ = Ada perbedaan yang bermakna antara hasil pemeriksaan glukosa darah metode sensor berbasis nanopartikel perak dengan alat

*Automated analyzed clinical chemistry*



$H_a$  = Tidak ada perbedaan yang bermakna antara hasil pemeriksaan glukosa darah metode sensor berbasis nanopartikel perak dengan alat

*Automated analyzed clinical chemistry*

Kriteria penerimaan dan penolakan yaitu  $H_0$  diterima bila  $t_{hitung} < t_{tabel}$  dan sebaliknya  $H_a$  diterima bila  $t_{hitung} > t_{tabel}$

Penyelesaian;

$$\bar{S} = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n}}$$

$$\bar{S}_{\text{sensor}} = \sqrt{\frac{\sum (x_i - \bar{x}_1)^2}{n}}$$

$$= \sqrt{\frac{(3,924 - 71,71)^2 + (97,57 - 71,71)^2 + (113,4 - 71,71)^2}{3}}$$

$$= 48,303$$

$$\bar{S}_2 = \sqrt{\frac{\sum (x_i - \bar{x}_2)^2}{n}}$$

$$= \sqrt{\frac{(77 - 77)^2 + (77 - 77)^2 + (77 - 77)^2}{3}}$$

$$= 0$$

$$\bar{S}_{\text{gabungan}} = \sqrt{\frac{(n_1+1)\bar{S}_1^2 + (n_2+1)\bar{S}_2^2}{n_1+n_2-2}}$$

$$\frac{(3+1)(48,303)^2 + (3+1)(0)^2}{3+3-2}$$

$$8,303$$

Uji t

$$\text{Uji } t = \frac{\bar{X}_1 + \bar{X}_2}{S_{gab} \sqrt{\frac{1}{n_1} + \frac{1}{n_2}}}$$

$$= \frac{71,71 - 77}{48,303 \sqrt{\frac{1}{3} + \frac{1}{3}}}$$

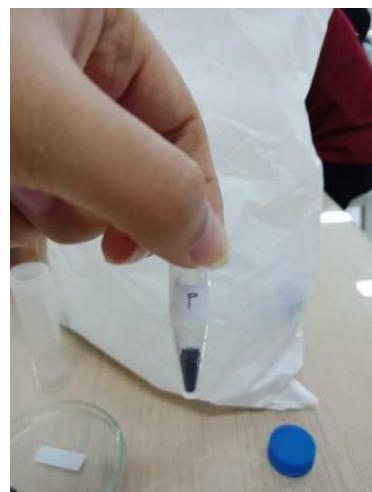
$$= -0,134$$

Menunjukkan bahwa  $t_{\text{hitung}} -0,134 < t_{\text{tabel}} 2,13$  dengan tingkat kepercayaan 95% maka  $H_0$  diterima.



Optimization Software:  
[www.balesio.com](http://www.balesio.com)

**Lampiran 14. Dokumentasi penelitian**



Optimization Software:  
[www.balesio.com](http://www.balesio.com)