

DAFTAR PUSTAKA

- Adumitrachioaie, A., Tertis M., Cernat A., Sandulescu R., dan Cristea C., 2018, Electrochemical Methods Based on Molecularly Imprinted Polymers for Drug Detection, *Electro. Sci. J.*, **13**: 2558-2562.
- Afgani, A.Q. dan Destiani, D.P., Review Artikel: Pengaruh Polaritas Porogen pada Sintesis Molecularly Imprinted Polymer (MIP), *Farm. Sup.*, **16**(3): 224-233.
- Alberty, R.A. dan Daniel, F., 1987, *Physical Chemistry, 5th ed, SI Version*, John Wiley and Sons Inc., New York.
- Alexander, C., Andersson, H.S., Andersson, L.I., Ansell, R.J., Kirsch, N., Nicholls, I.A., O'Mahony, J., dan Whitcombe, M.J., 2006, Molecular Imprinting Science and Technology: A Survey of the Literature for the Years Up to and Including 2003, *Mol. Rec. J.*, **19**: 106-180.
- Allen, S.J., Mekay, G., dan Porter, J.F., 2004, Adsorption Isotherm Models for Basic Dye Adsorption by Peat in Single and Binary Component Systems, *Coll. Interf Sci. J.*, **280**: 322-333.
- Amin, F., 2009, *Evaluasi Migrasi Di-(2-Etilheksil)Ftalat dari Botol Polietilena Tereftalat Menggunakan GC/MS*, Skripsi tidak diterbitkan, departemen Kimia, Fakultas Matematika dan Ilmu Pengetahuan Alam, Institut Pertanian Bogor, Bogor.
- Ansari, S. dan Ghoban, A., 2016, Molecularly Imprinted Polymers (MIP) for Selective Solid Phase Extraction of Celecoxib in Urine Samples Followed by High Performance Liquid Chromatography, *Chem. Health Risks. J.*, **7**:1-5.
- Apridinata, 2017, *Analisis Migrasi Ftalat dari Kemasan Plastik Makanan pada Simulan Makanan dengan Kromatografi Gas-Detektor Nyala Ionisasi (GC-FID)*, Skripsi tidak diterbitkan, program studi Farmasi, Fakultas Matematika dan Ilmu Pengetahuan Alam, Universitas Sriwijaya, Palembang.
- ATSDR (Agency for Toxic Substances and Disease Registry), 2002, *Toxicological Profile for Di(2-ethylhexyl)phthalate*, US Department of Health and Human Services, Atlanta.
- Atkins, P.W., 1990, *Kimia Fisika*, Erlangga, Jakarta.
- Azapagic, A., Emsley, A., dan Hamerton, I., 2003, *Polymers: The Environment and Sustainable Development*, John Wiley & Sons, Ontario.

- Biscardi, D., Monarca, S., De Fusco R., Senatore, F., Poli, P., Buschini, A., Rossi, C., dan Zani, C., 2003, Evaluation of the Migration of Mutagens / Carcinogens from PET Bottles into Mineral Water by *Tradescantia* / Micronuclei Test, Comet Assay on Leukocytes and GC/MS, *The Science of the Total Environment*, **302**(2003): 101-108.
- Bosnir, J., Puntaric, D., Galic, A., Skes, I., Dijanic, T., Klaric, M., Grgic, M., Curkovic, M., dan Smit, Z., 2007, Migration of Phthalates from Plastic Containers into Soft Drinks and Mineral Water, *Food Tech. Biotech.*, **45**(1): 91-95.
- Botahala, L., 2019, *Perbandingan Efektivitas Daya Adsorpsi Sekam Padi dan Cangkang Kemiri terhadap Logam Besi (Fe) pada Air Sumur Gali*, Deepublish, Sleman.
- Cacho, C., Turiel, E., Martin, A.E., Perez, C.C., dan Camara, C., 2004, Characterisation and quality assessment of binding sites on a propazine-imprinted polymer prepared by precipitation polymerisation, *Chrom. B. J.*, **802**: 347-353.
- Caro, E., Marce, R.M., Borrul, F., Cormack, P.A.G., dan Sherrington, D.C., 2006, Application of Molecularly Imprinted Polymer to Solid-Phase Extraction of Compounds from Environmental and Biological Samples, *Anal. Chem. T.*, **25**: 143-154.
- Chapuis, F., Pichon V., Lanza F., Sellergren B., dan Hennion M. C., 2004, Retention Mechanism of Analytes in the Solid-phase Extraction Process using Molecularly Imprinted Polymers Application to the Extraction of Triazines from Complex Matrices, *Chrom. B. J.*, **804**(1): 93-101.
- Chen, F., Zhan, Y., Yubin, T., dan Xingang, W., 2017, Selective Extraction and Determination of Di(2-ethylhexyl) Phthalate in Aqueous Solution by HPLC Coupled with Molecularly Imprinted Solid-phase Extraction, *Chem. Eng. I. J.*, **36**(3): 127-136.
- Cormack, P.A.G. dan Elorza, A.Z., 2004, Molecularly Imprinted Polymers: Synthesis and Characterisation, *Chrom. B. J.*, **804**: 173-182.
- Cowd, M.A., 1991, *Kimia Polimer*, Penerbit ITB, Bandung
- David, R.M., Moore, M.R., Finney, D.C., dan Guest, D., 2000, Chronic Toxicity of Di(2-ethylhexyl)phthalate in Mice, *Tox. Sci.*, **58**(2): 377-385.
- Day, R.A. dan Underwood, A.L., 2001, *Analisis Kimia Kuantitatif*, Edisi ke 6, Penerbit Erlangga, Jakarta.
- Departemen Kesehatan R.I, 2013, Serba Serbi Kemasan Pangan, Retrieved Maret 01, 2019 from <http://www.gizikia.depkes.go.id/serba-serbi-kemasan-pangan/>.

- European Council for Plastics and Intermediates (ECPI), 2009, *About DEHP*, European Plastics, Brussels.
- Farhoodi, M., Emam-Djomeh, Z., dan Ehsani, M.R., 2008, Effect of Environmental Conditions on the Migration of Di(2-Ethylhexyl)Phthalate from PET Bottles into Yogurt Drinks: Influence of Time, Temperature, and Food Simulant, *Sci & Eng, A. J.*, **33**(2B): 279-287.
- Fauziah, S., 2016, *Sintesis, Karakterisasi dan Penggunaan Polimer Bercetakan Molekul untuk Ekstraksi Fase Padat β -Sitosterol*, Disertasi tidak diterbitkan, Sekolah Pascasarjana, Universitas Hasanuddin, Makassar.
- Febrianto, J., Kosasih, A.N., Sunarso, J., Ju, Y., Indraswati, N., dan Ismadji, S., 2009, Equilibrium and Kinetic Studies in Adsorption of Heavy Metals Using Biosorbent: A Summary of Recent Studies, *Haz. Mat. J.*, **162**: 618.
- Freitas, A.F.F.L., 2015, *Synthesis and Characterization of Molecularly Imprinted Polymer Particles (MIPs) for Biomedical Applications*, Final Report of the Work Project presented to the Escola Superior de Tecnologia e Gestao, Biomedical Technology, Instituto Politecnico de Braganca, Braganca.
- Gafur, A.M.M., 2020, *Sintesis dan Karakterisasi Polimer Bercetakan Molekul Di-(2-etilheksil)ftalat dengan Metode Polimerisasi Presipitasi*, Skripsi tidak diterbitkan, departemen Kimia, Fakultas Matematika dan Ilmu Pengetahuan Alam, Universitas Hasanuddin, Makassar.
- Gunawan, A.I., 2019, *Sintesis dan Optimasi Molecularly Imprinted Polymers (MIP) sebagai Adsorben untuk Pemisahan β -Sitosterol*, Skripsi tidak diterbitkan, departemen Kimia, Fakultas Matematika dan Ilmu Pengetahuan Alam, Universitas Hasanuddin, Makassar.
- Hasanah, A.N., Utari, T.N.D., dan Pratiwi, R., 2019, Synthesis of Atenolol-Imprinted Polymers with Methyl Methacrylate as Functional Monomer in Propanol Using Bulk and Precipitation Polymerization Method, *Anal. Chem. Methods J.*, **2019**: 1-7.
- Hasanah, A.N., Soni, D., Pratiwi, R., Rahayu, D., Megantara, S., dan Mutakin, 2020, Synthesis of Diazepam-Imprinted Polymers with Two Functional Monomers in Chloroform Using a Bulk Polymerization Method, *Chem. J.*, **2020**: 1-8.
- Haginaka, J., Tabo, H., dan Kagawa, C., 2007, Uniformly Sized Molecularly Imprinted Polymers for D-Chlorpheniramine: Influence of A Porogen on Their Morphology and Enantioselectivity, *Pharm. & Biomed. Anal. J.*, **46**(2008): 877-881.
- Haupt, K., Molecularly Imprinted Polymers in Analytical Chemistry, *Analyst*, **126**: 747-756.

- He, C., Long, Y., Pan, J., Li, K., dan Liu, F., 2007, Application of Molecularly Imprinted Polymers to Solid-Phase Extraction of Analysis from Real Samples, *J. Biochem. Biophys. Methods*, **70**: 133-150.
- Hennion, M.C., 1999, Solid Phase Extraction: Method Development, Sorbents, and Coupling with Liquid Chromatography, *J. Chromatog. A*, **856**: 3-54.
- Hillman, L.S., Goodwin, S.L., dan Sherman, W.R., 1975, Identification and Measurement of Plasticizer in Neonatal Tissues after Umbilical Catheters and Blood Products, *Med. J.*, **292**(8), 381-386.
- Ilmiawati, C., Reza, M., Rahmatini, dan Rustam, E., 2017, Edukasi Pemakaian Plastik sebagai Kemasan Makanan dan Minuman serta Risikonya terhadap Kesehatan pada Komunitas di Kecamatan Bungus Teluk Kabung, Padang, *J. Ilm. Pengabd. Kpd. Masy.*, **1**(1): 20-28.
- Jing, T., Gao, X.D., Wang, P., Wang, Y., Lin, Y.F., Chen, X.Z., Zhou, Y.K., dan Mei, S.R., 2007, Preparation of High Selective Molecularly Imprinted Polymers for Tetracycline by Precipitation Polymerization, *Chinese. Chem. Lett.*, **18**(2007): 1535-1538.
- Jung, J., Park, B., dan Kim, J., 2012, Durability Test with Fuel Starvation Using A Pt/CNF Catalyst in PEMFC, *Nano. Res. Lett.*, **7**(1): 1-8.
- Kadhem, A.J., Gentile, G.J., dan de Cortalezzi, M.M.F., 2021, Molecularly Imprinted Polymers (MIPs) in Sensors for Environmental and Biomedical Applications: A Review, *Mol.*, **26**(6233): 1-34.
- Kaihatu, T.S., 2014, *Manajemen Pengemasan*, Penerbit Andi, Yogyakarta.
- Kirsch, N., Alexander, C., Lubke, M., Whitcombe, M.J., dan Vulfson, E.N., 2000, Enhancement of Selectivity of Imprinted Polymers Via Post-Imprinting Modification of Recognition Sites, *Polym.*, **14**: 5583-5590.
- Komiyama, M., Takeuchi, T., Mukawa, T., dan Asanuma, H., 2003, *Molecular Imprinting from Fundamental to Applications*, Wiley-VCH Verlag GmbH & Co. KgaA, Weinheim.
- Lee, W.C., Cheng, C.H., Pan, H.H., Chung, T.H., dan Hwang, C.C., 2008, Chromatographic Characterization of Molecularly Imprinted Polymers, *Anal. & Bioanal. Chem.*, **390**: 1101-1109.
- Li, W., dan Li, S., 2007, Molecular Imprinting: A Versatile Tool for Separation, Sensors and Catalysis, *Adv. Polym. Sci.*, **206**: 191-210.
- Li, L., Liu, J.C., Lai, F.N., Liu, H.Q., Zhang, X.F., Dyce, P.W., Shen, W., dan Chen, H., 2016, Di(2-ethylhexyl) Phthalate Exposure Impairs Growth of Antral Follicle in Mice, *J. Pone*, **11**(2): 1-18.

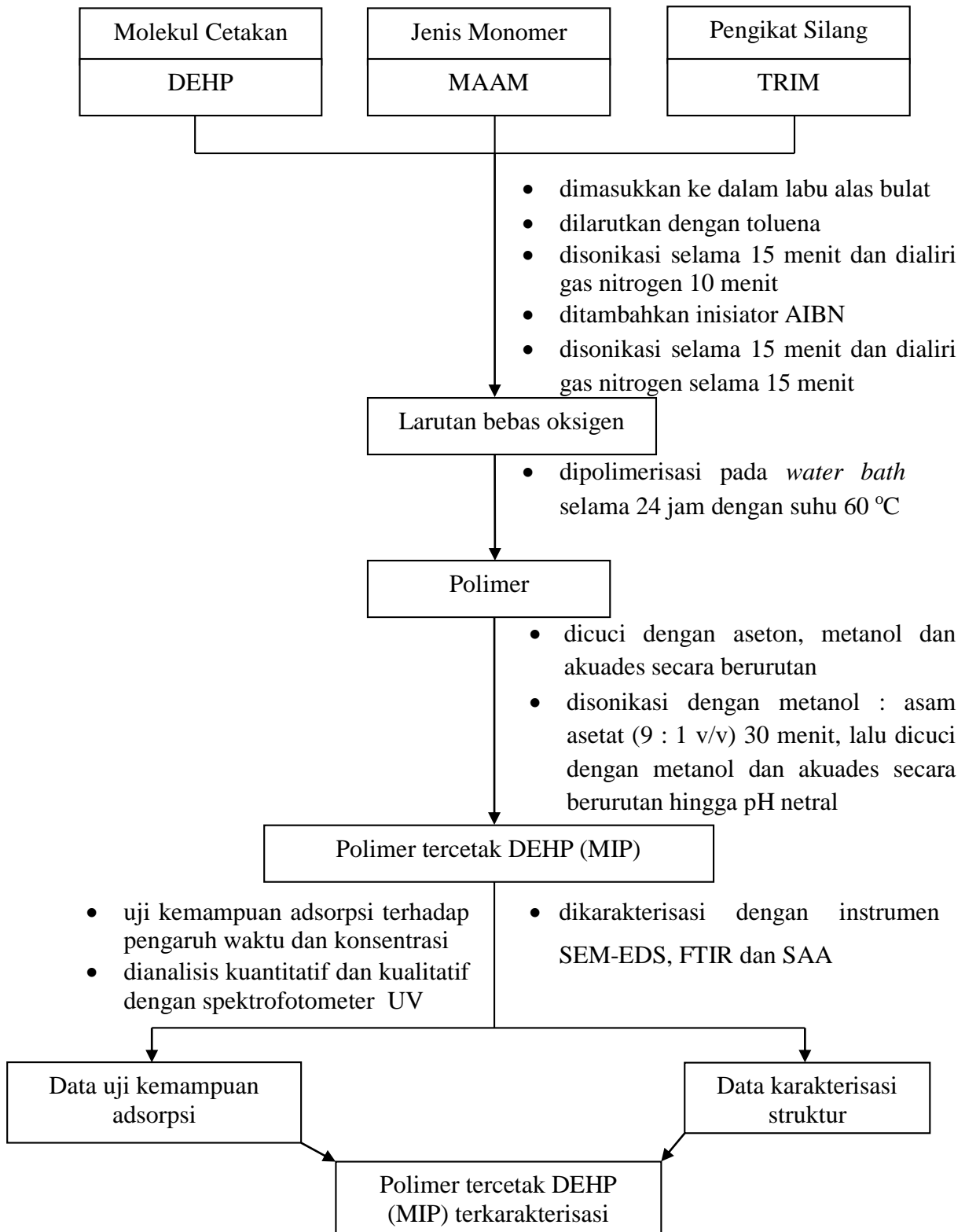
- Lu, F., Liyanan, W., Fuquan, W., Xiaofang, Y., Ren'ao, G., Jianlin, Y., Hailu, Z., dan Zongwu, D., 2011, Evaluation of the Binding Specificity of Electrosynthesized Poly-Ortho-Phenylenediamine Molecularly Imprinted with Metal Chelates, *Micr. Act.*, **174**: 47-54
- Madikizela, L.M., Mdluli, P.S., dan Chimuka, L., 2016, Experimental and Theoretical Study of Molecular Interactions between 2-Vinyl Pyridine and Acidic Pharmaceuticals Used as Multi-Template Molecules in Molecularly Imprinted Polymer, *Reac. & Func. Polym. J.*, **103**: 33-43.
- Mane, S., Ponrathnam, S., dan Chavan, N.N., 2015, Role of Interfacial Tension of Solvating Diluents and Hydrophilic-Hydrophobic Cross-Linkers in Hyper-Cross-Linked Solid Support, *Indust. & Eng. Chem. Res.*, **54**: 6893-6901.
- Montuori, P., Jover, E., Morgantini, M., Bayona, J.M., dan Triassi, M., 2008, Assessing Human Exposure to Phthalic Acid and Phthalate Esters from Mineral Water Stored in Polyethylene Terephthalate and Glass Bottles, *Food Addit. Contam.*, **25**(4):511-518.
- Muchtaridi, M. dan Musfiroh, I., 2012, OffLine SPE-GC/MS Analysis of Lead Compounds Aromatherapy in Blood Plasma of Mice of Essential Oils Materials from Indonesian Aromatic Plants, *Chem. Asian J.*, **24**(11): 5124-28.
- National Center for Biotechnology Information (NIH), 2022, PubChem Compound Summary for CID 18689, *Trimethylolpropane Trimethacrylate*, (Online), (<https://pubchem.ncbi.nlm.nih.gov/compound/Trimethylolpropane-trimethacrylate>, diakses 3 Februari 2022)
- Opik, A., Menaker, A., Reut, J., dan Syritski, V., 2009, *Molecularly Imprinted Polymers: A New Approach to the Preparation of Functional Materials*, Proceedings of the Estonian Academy of Sciences, Tallinn University of Technology, Estonia.
- Peraturan Kepala Badan Pengawas Obat dan Makanan Republik Indonesia (BPOM RI), 2011, *Pengawasan Kemasan Pangan*, Badan Pengawas Obat dan Makanan Republik Indonesia, Jakarta.
- Poole, C.F., 2003, New Trends in Solid-Phase Extraction, *Anal. Chem. T.*, **22**(6): 362-373.
- Rahmi, 2018, *Modifikasi Khitosan sebagai Adsorben*, Syiah Kuala University Press, Banda Aceh.
- Ramakrishna, D.M. dan Viraraghavan, T., 1997, Environmental Impact of Chemical Deicers-A Review, *Environ. Imp.*, **166**(1): 49-63.

- Reddy, B.S., Rozati, R., Reddy, B.V.R., dan Raman, N.V.V.S.S., 2006, Association of Phthalate Esters with Endometriosis in Indian Women, *BJOG: An Obst. & Gyn. Int. J.*, **113**(5): 515-520.
- Rochmadi, dan Permono, A., 2018, *Mengenal Polimer dan Polimerisasi*, Gadjah Mada University Press, Yogyakarta.
- Safitri, A., 2011, *Kontroversi Antimoni Trioksida*, Universitas Paramadina, Jakarta.
- Saldivar, G.E. dan Vivaldo, L.E., 2013, *Handbook of Polymer Synthesis, Characterization, and Processing*, John Wiley & Sons, Hoboken.
- Saputra A., Wijaya K., Ahmad M.N., dan Tahir I., 2013, Penggunaan Metode Semi Empirik AM 1 untuk Pemilihan Monomer Fungsional Efektif pada Prasintesis Polimer Tercetak Drazinon, *Valensi*, **3**(1): 1-9.
- Sellergen, B., 2001, *Molecularly Imprinted Polymers: Man-Made Mimics of Antibodies and Their Applications in Analytical Chemistry*, Elsevier, Amsterdam.
- Sembodo, B.S.T., 2006, Model Kinetika Langmuir untuk Adsorpsi Timbal pada Abu Sekam Padi, *Ekulibrium*, **5**: 28-33.
- Shaikh, H., Memon, N ., Khan, H., Bhangar, M.I., dan Nizamani, S.M., 2012, Preparation and Characterization of Molecularly Imprinted Polymer for Di(2-Ethylhexyl)Phthalate: Application to Sample Clean-Up Prior to Gas Chromatographic Determination, *Journal of Chromatography A*, **1247**: 125-133.
- Shi, X., Wu, A., Qu, G., Li, R., dan Azhang, D., 2007, Development and Characterization of MIP based on Metacrylic Acid for Selective Recognition of Drugs, *Biomaterials*, **28**: 3741-3749.
- Sole, R.D., Luca, A.D., Catalano, M., Mele, G., dan Vasapollo, G., 2007, Noncovalent Imprinted Microspheres: Preparation, Evaluation, and Selectivity of DBU Template, *App. Polym. Sci. J.*, **105**: 2190-2197.
- Sucipta, I.N., Suriasih, K., dan Kencana, P.K.D., 2017, *Pengemasan Pangan: Kajian Pengemasan yang Aman, Nyaman, Efektif dan Efisien*, Udayana University Press, Denpasar.
- Sulchan, M. dan Endang, N.W., 2007, *Keamanan Pangan Kemasan Plastik dan Styrofoam*, UI Press, Jakarta.
- Syawali, F., 2018, *Pengaruh Perlakuan dan Pelarut pada Proses Pembuangan Template terhadap Jumlah Rongga Molecularly Imprinted Polymer (MIP) Melamin*, Tesis tidak diterbitkan, jurusan Fisika, Fakultas Matematika dan Ilmu Pengetahuan Alam, Universitas Sriwijaya, Palembang.

- Syu, M.J., Deng J.H., Nian Y.M., Chiu T.C., dan Wu A.H., 2005, Binding Specificity of α -Bilirubin-Imprinted Poly (methacrylicacid-co-ethyleneglycol dimethylacrylate) Toward α -Bilirubin, *Biomaterials*, **26**: 4684- 4685.
- Tabarestani, M.S., Rahnama, K., Jahanshahi, M., Nasrollanejad, S., dan Fatemi, M.H., 2016, Synthesis of a Nanoporous Molecularly Imprinted Polymers for Dibutyl Phthalate Extracted from *Trichoderma harzianum*, *J. Nanostruct*, **6**(3): 245-249.
- Tristi, J. dan Muchtaridi M., 2018, Molecularly Imprinted Polymer (MIP) untuk Isolasi Atenolol dalam Sampel Biologis, *Farm. Sup.*, **16**(1): 304- 312.
- Turiel, E. dan Martin, E.A., 2005, A Molecularly Imprinting Technology in Capillary Electrochromatography, *J. Sep. Sci.*, **28**: 719-728.
- United State Environmental Protection Agency (US EPA), 1997, *Public Health Goal for Di(2-ethylhexyl)Phthalate (DEHP) in Drinking Water*, California Environmental Protection Agency, California.
- United State Environmental Protection Agency (US EPA), 2020, *Methacrylamide*, California Environmental Protection Agency, California.
- Vasapollo, G., Sole, R.D., Margola, L., Lazzoi, M.R., Scardino, A., Scoranno, S., dan Mele, G., 2011, Molecularly Imprinted Polymers: Present and Future Prospective, *Int. J. Mol. Sci.*, **12**: 5909-5945.
- WHO, 1992, *Diethylhexyl Phthalate (Environmental Health Criteria 131)*, International Programme on Chemical Safety, Geneva.
- Walsh, R., 2010, *Development and Characterization of Molecularly Imprinted Suspension Polymers*, Tesis tidak diterbitkan, pusat penelitian Bioteknologi Molekuler dan Farmasi, Waterford Institute of Technology, Waterford.
- Wei, S. dan Mizaikoff, B., 2007, Recent Advances on Noncovalent Molecular Imprints for Affinity Separations, *J. Sep. Sci.*, **30**: 1794-1805.
- Widayatno, T., Yuliawati, T., dan Susilo, A.A., 2017, Adsorpsi Logam Berat (Pb) dari Limbah Cair dengan Adsorben Arang Bambu Aktif, *J. Tekno. B. Alam*, **1**(1): 17-23.
- Wulff, G., 2002, Enzyme-like Catalysis by Molecularly Imprinted Polymers, *Chem. Rev.*, **102**: 1-27.
- Yalkowsky, S.H. dan He, Y., 2003, *Handbook of Aqueous Solubility Data*, CRC Press, New York.
- Yan, H. dan Row, K.H., 2006, Characteristic and Synthetic Approach of Molecularly Imprinted Polymer, *Int. J. Mol. Sci.*, **7**: 155-178.

- Yang, Y., Jianlong, Y., Jie, Y., Bing, S., dan Jing, Z., 2014, Molecularly Imprinted Solid-Phase Extraction for Selective Extraction of Bisphenol Analogues in Beverages and Canned Food, *J. Agric. & Food Chem.*, **62**: 11130-11137.
- Yang, Z., Chen, F., Tang, Y., dan Li, S., 2015, Selective Adsorption of Di(2-ethylhexyl) Phthalate by Surface Imprinted Polymers with Modified Silica Gel as Functional Support, *J. Chem. Soc. Pakistan*, **37**(5): 939-949.
- Yusof, N.A., Appribeyan, M.D., dan Harson, J., 2010, Synthesis and Characterization of a Molecularly Imprinted Polymer for Pb²⁺ Uptake Using 2-vinylpyridine as the Complexing Monomer, *Sains Malay.*, **39**(5): 829-835.
- Yuyun, A. dan Gunarsa, D., 2011, *Cerdas Mengemas Produk Makanan dan Minuman*, AgroMedia Pustaka, Jakarta.
- Zhang, M.L., Xie, J.P., Zhou, Q., Chen, G.Q., dan Liu, Z., 2003, On-line Solid-Phase Extraction of Ceramides from Yeast with Ceramide III Imprinted Monolith, *J. Chrom. A.*, **984**: 83-173.
- Zhang, H., Song T., Zhang W., Hua W., dan Pan C., 2007, Retention Behavior of Phenoxyacetic Herbicides on a Molecularly Imprinted Polymer with Phenoxyacetic Acid as a Dummy Template Molecule, *Bioorg. & Med. Chem.*, **15**(1): 6080-6089.
- Zhang, W., Li, Q., Cong, J., Wei, B., dan Wang, S., 2018, Mechanism Analysis of Selective Adsorption and Specific Recognition by Molecularly Imprinted Polymers of Ginsenoside Re, *Polym.*, **10**(216): 1-14.
- Zwir-Ferenc, A. dan Biziuk, M., 2006, Solid Phase Extraction Technique-Trends, Opportunities and Application, *Polish J. Environ. Stud.*, **15**(5): 677-690.

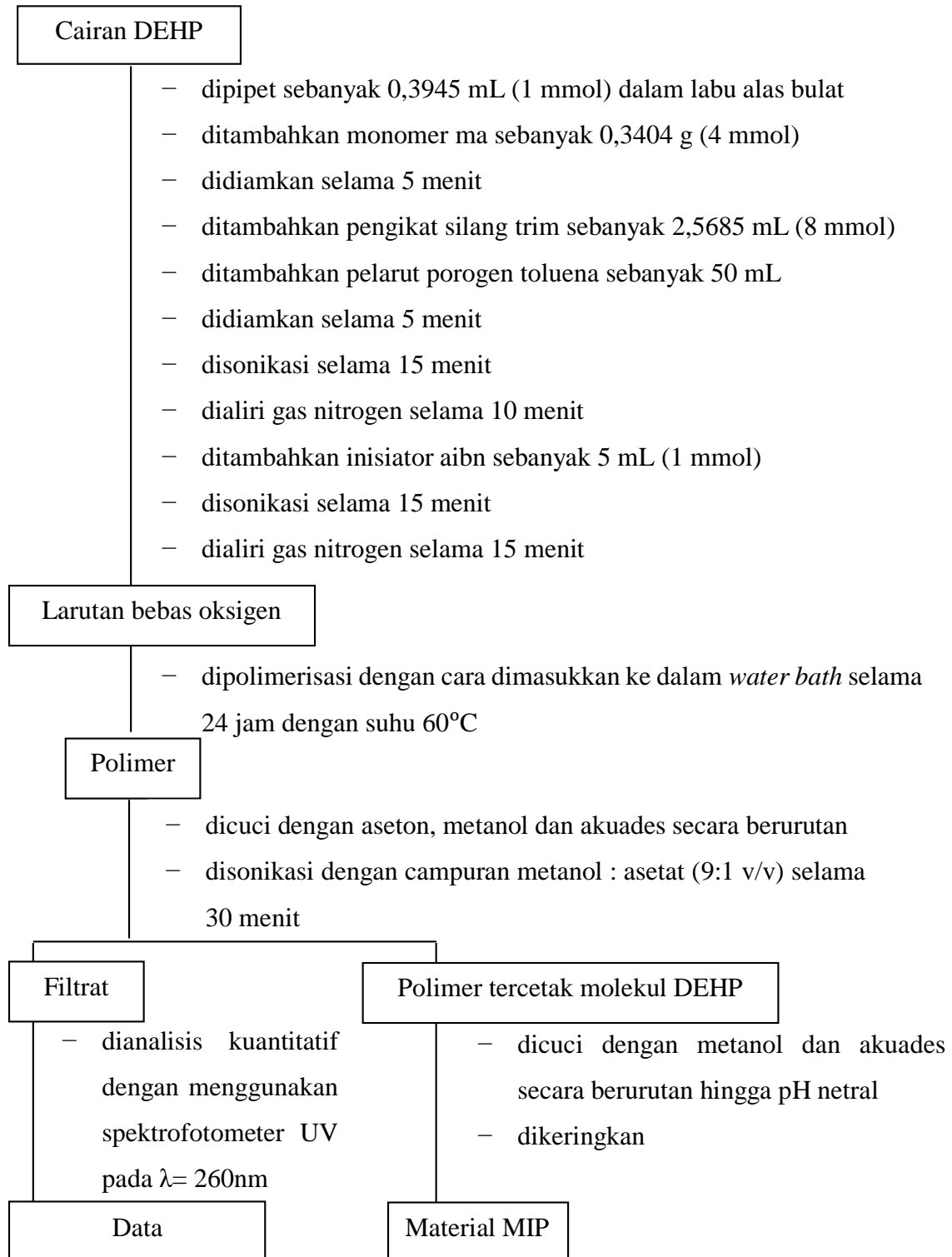
Skema 1. Sintesis Polimer Bercetakan Molekul DEHP menggunakan Metode Polimerisasi Presipitasi

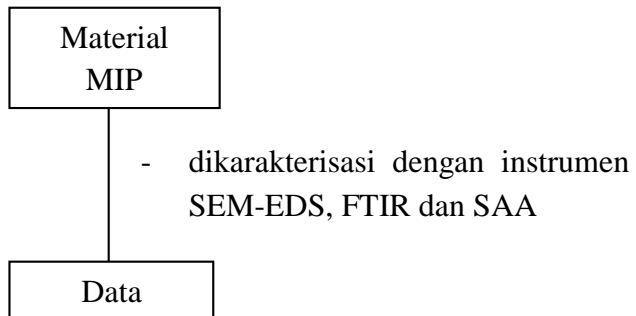


Catatan: Sintesis NIP dibuat dengan metode yang sama dengan MIP, tapi tanpa DEHP dan proses ekstraksi.

Lampiran 2. Bagan Kerja

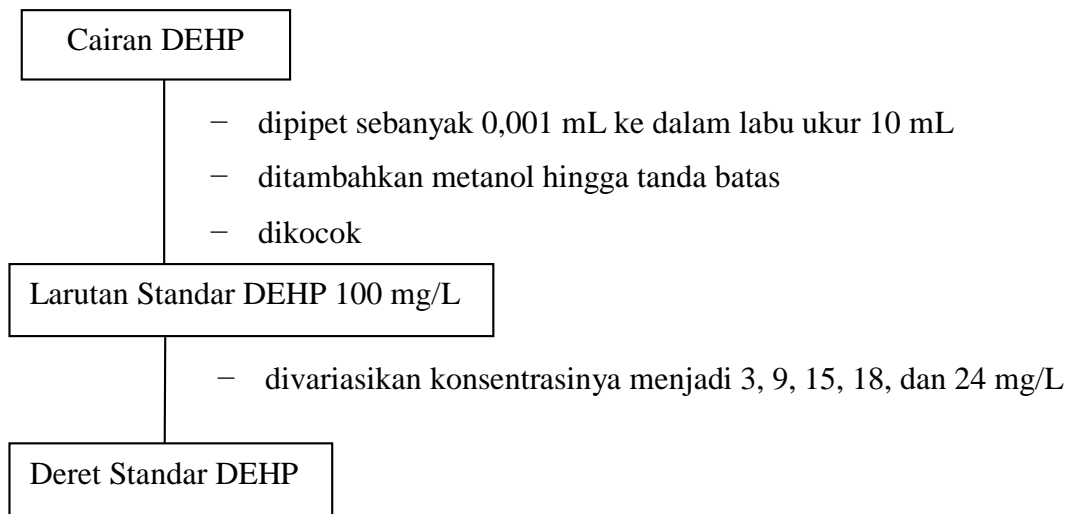
1. Sintesis MIP dan NIP



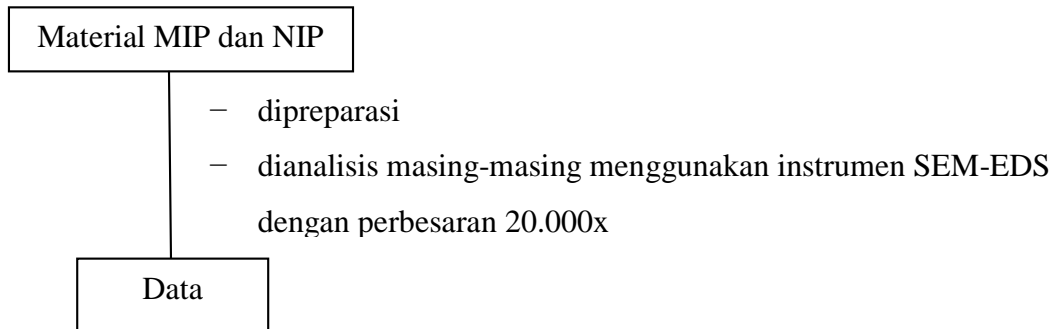


Catatan: Sintesis NIP dibuat dengan metode yang sama dengan MIP, tetapi tanpa menggunakan molekul cetakan (DEHP).

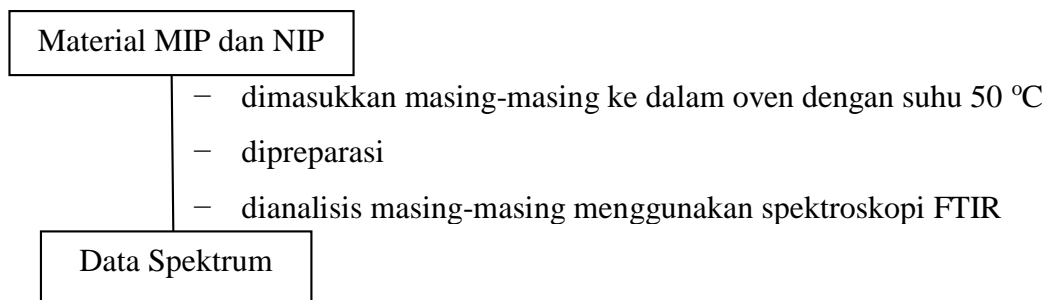
2. Pembuatan Larutan Standar DEHP 100 mg/L



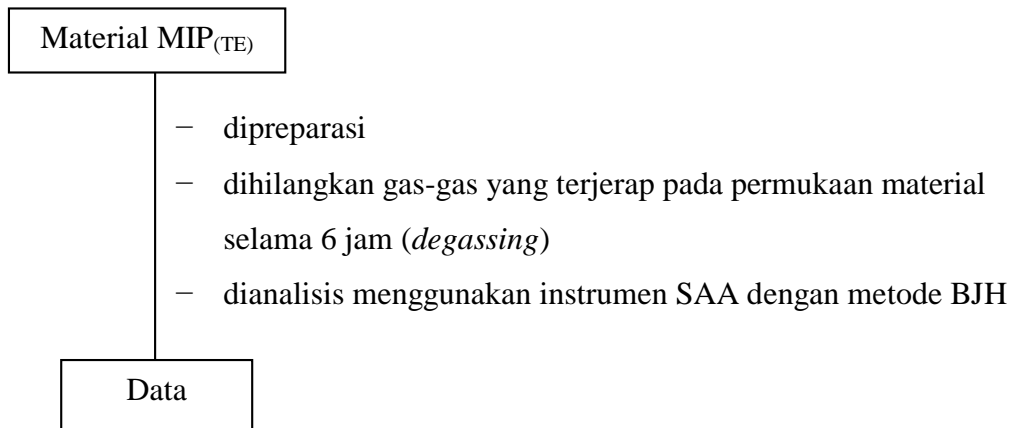
3. Karakterisasi MIP dan NIP Menggunakan SEM-EDS



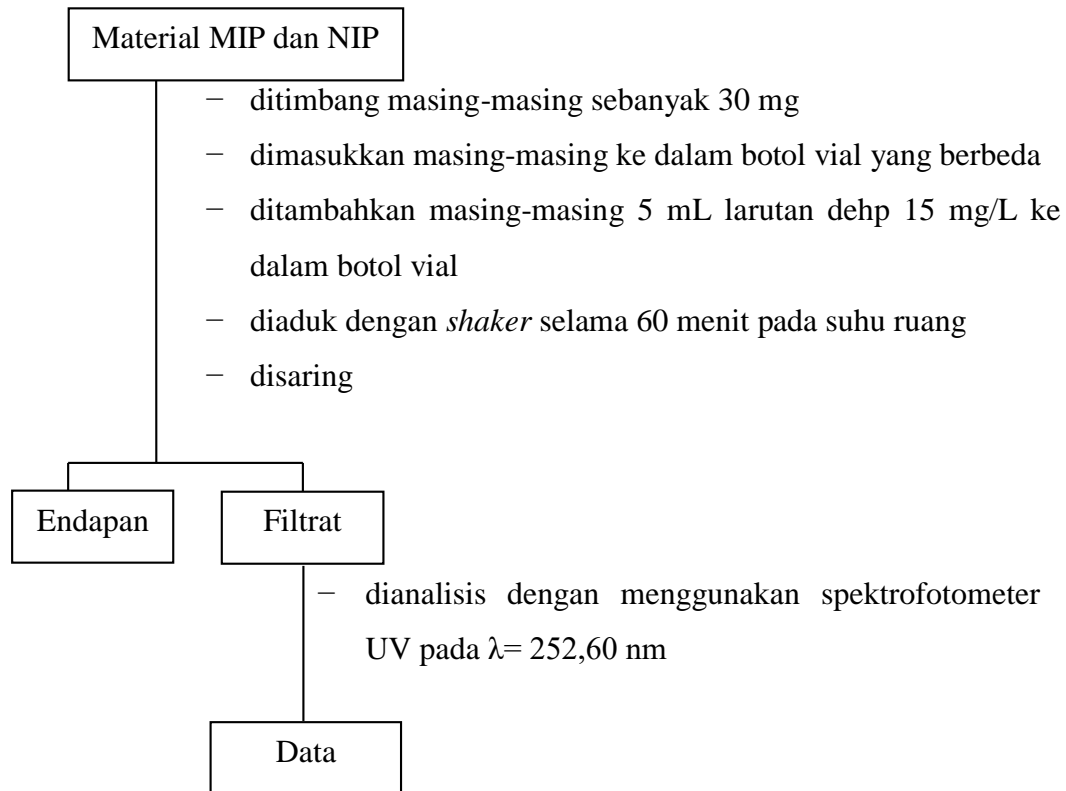
4. Karakterisasi MIP dan NIP Menggunakan FTIR



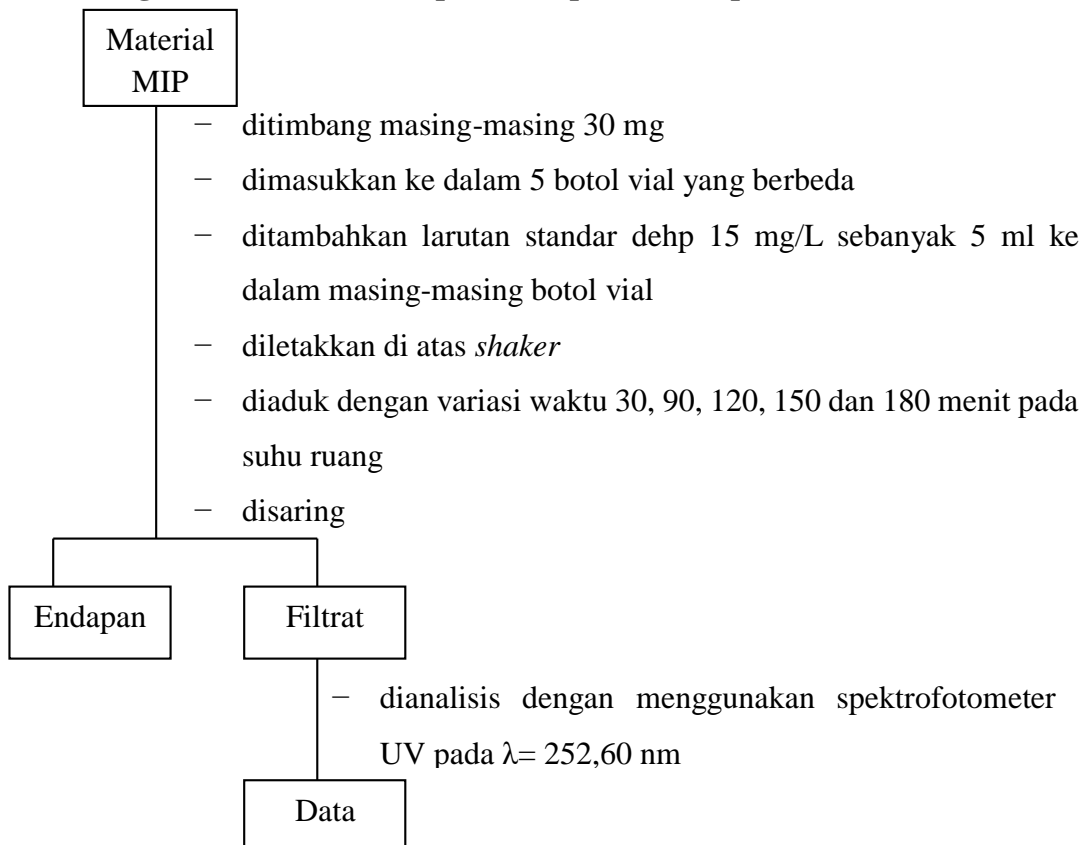
5. Karakterisasi MIP Menggunakan SAA



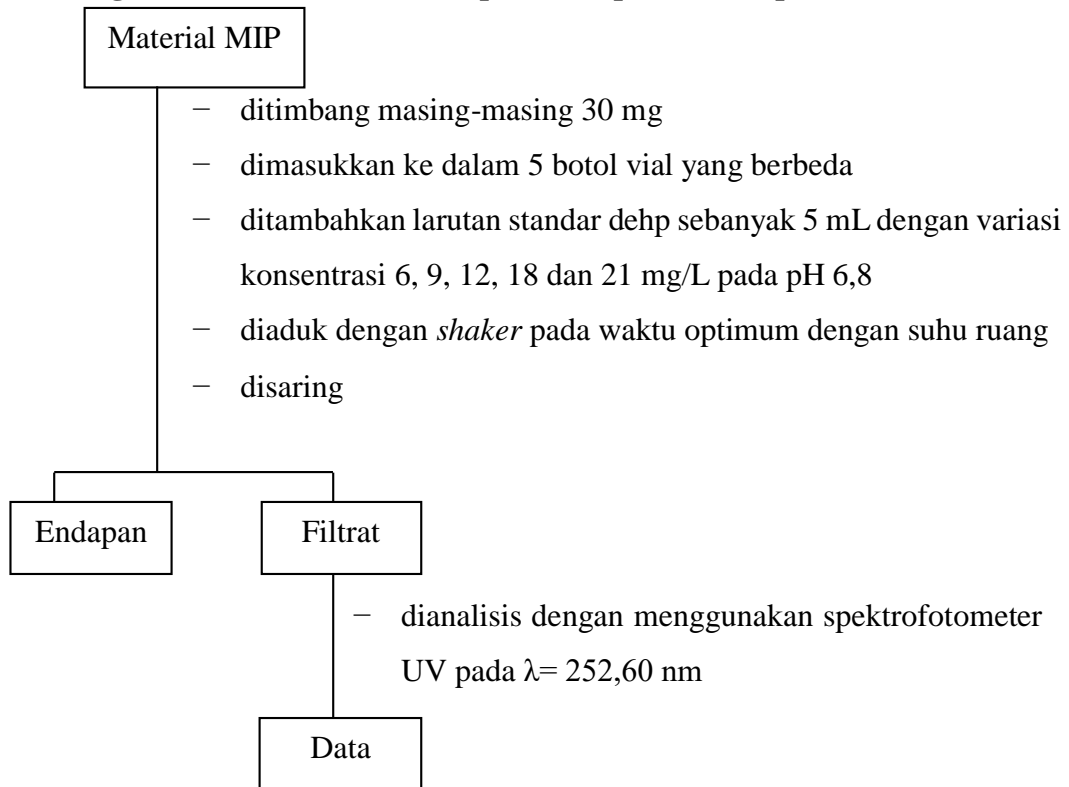
6. Uji Kemampuan Adsorpsi MIP dan NIP



7. Pengaruh Waktu terhadap Kemampuan Adsorpsi MIP



8. Pengaruh Konsentrasi terhadap Kemampuan Adsorpsi MIP

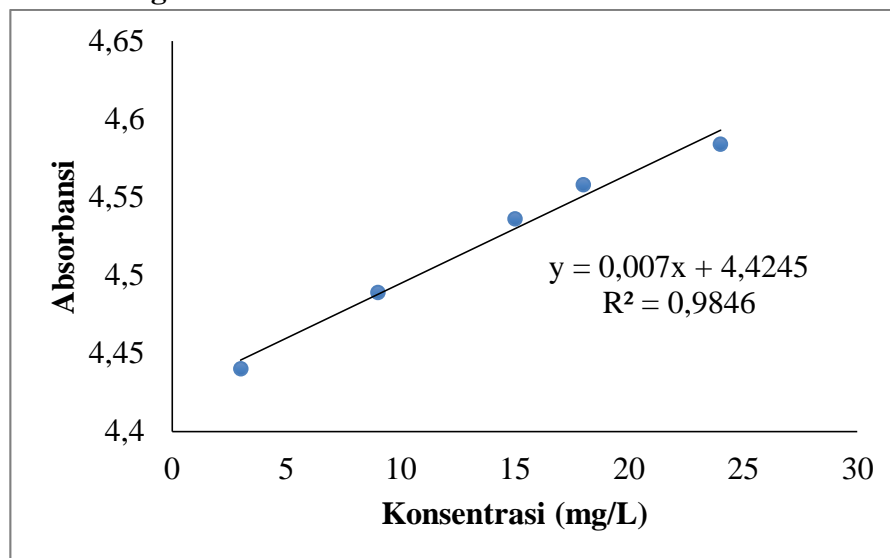


Lampiran 3. Data Spektrofotometer UV-Vis

1. Data absorbansi larutan standar DEHP

No.	Sampel	Konsentrasi (mg/L)	Absorbansi
1	DEHP 1	3	4,440
2	DEHP 2	9	4,489
3	DEHP 3	15	4,536
4	DEHP 4	18	4,558
5	DEHP 5	24	4,584

2. Kurva hubungan antara absorbansi vs konsentrasi larutan standar DEHP



3. Data absorbansi kemampuan adsorpsi MIP dan NIP

No.	Sampel	Absorbansi	q_e (mg/g)	Δq_e (mg/g)
1	Adsorpsi DEHP oleh MIP	4,499	0,73	0,50
2	Adsorpsi DEHP oleh NIP	4,520	0,23	

4. Data absorbansi adsorpsi DEHP oleh MIP terhadap pengaruh waktu

No.	Sampel	Waktu (menit)	Absorbansi
1	DEHP	30	4,502
2	DEHP	90	4,450
3	DEHP	120	4,461
4	DEHP	150	4,496
5	DEHP	180	4,490

5. Data penentuan waktu optimum adsorpsi MIP terhadap DEHP

No.	Waktu	C_e (mg/L)	q_t (mg/g)	$q_e - q_t$
1	0	0	0	1,89
2	30	11,07	0,65	1,24
3	90	3,64	1,89	0
4	120	5,21	1,63	0,24
5	150	10,21	0,80	1,1
6	180	9,36	0,94	0,86

Catatan:

q_t adalah q_e pada waktu t

q_e adalah q_t pada waktu optimum

6. Data absorbansi adsorpsi DEHP oleh MIP terhadap pengaruh konsentrasi

No.	Sampel	Konsentrasi (mg/L)	Absorbansi
1	DEHP	6	4,441
2	DEHP	9	4,451
3	DEHP	12	4,468
4	DEHP	18	4,487
5	DEHP	21	4,500

7. Data persamaan isoterma adsorpsi Langmuir dan Freundlich

No.	Sampel	C _o (mg/L)	C _e (mg/L)	q _e (mg/g)	Log C _e	Log q _e	1/C _e	1/q _e
1	MIP_DEHP	6	2,36	0,61	0,37	-0,22	0,42	1,65
2	MIP_DEHP	9	3,78	0,87	0,58	-0,06	0,26	1,15
3	MIP_DEHP	12	6,21	0,96	0,79	-0,02	0,16	1,04
4	MIP_DEHP	18	8,93	1,51	0,95	0,18	0,11	0,66
5	MIP_DEHP	21	10,79	1,70	1,03	0,23	0,09	0,59

8. Data kurva non linear untuk persamaan isoterma adsorpsi Langmuir

No.	C _o	1/K _L	1/C _e	1/q _m K _L	1/C _e x 1/q _m K _L	1/q _e	q _e
1	6	0,3767	0,42	3,0359	1,29	1,66	0,60
2	9	0,3767	0,26	3,0359	0,80	1,18	0,85
3	12	0,3767	0,16	3,0359	0,49	0,87	1,16
4	18	0,3767	0,11	3,0359	0,34	0,72	1,40
5	21	0,3767	0,09	3,0359	0,28	0,66	1,52

9. Data kurva non linear untuk persamaan isoterma adsorpsi Freundlich

No.	C _o	log K _F	1/n	log C _e	1/n x log C _e	log q _e	q _e
1	6	-0,467	0,6578	0,37	0,24	-0,22	0,60
2	9	-0,467	0,6578	0,58	0,38	-0,09	0,82
3	12	-0,467	0,6578	0,79	0,52	0,05	1,13
4	18	-0,467	0,6578	0,95	0,63	0,16	1,44
5	21	-0,467	0,6578	1,03	0,68	0,21	1,63

Lampiran 4. Perhitungan

1. Nilai konsentrasi adsorpsi DEHP oleh MIP dan NIP

$$y = 0,007x + 4,4245$$

a. Adsorpsi DEHP oleh MIP

$$y = 4,499$$

$$y = 0,007x + 4,4245$$

$$4,499 = 0,007x + 4,4245$$

$$x = \frac{4,499 - 4,4245}{0,007}$$

$$x = 10,64 \text{ mg/L}$$

b. Adsorpsi DEHP oleh NIP

$$y = 4,520$$

$$y = 0,007x + 4,4245$$

$$4,520 = 0,007x + 4,4245$$

$$x = \frac{4,520 - 4,4245}{0,007}$$

$$x = 13,64 \text{ mg/L}$$

2. Nilai Kemampuan Adsorpsi DEHP oleh MIP dan NIP

$$q_e = \frac{(C_o - C_e) V}{W}$$

Diketahui: $C_o = 15 \text{ mg/L}$ $W = 0,03 \text{ g}$

$V = 0,005 \text{ L}$ $C_e = \text{Konsentrasi setelah adsorpsi}$

a. Kemampuan Adsorpsi DEHP oleh MIP

$$q_e = \frac{(15 - 10,64) 0,005}{0,03}$$

$$q_e = \frac{0,021785}{0,03}$$

$$q_e = 0,73 \text{ mg/g}$$

b. Kemampuan Adsorpsi DEHP oleh NIP

$$q_e = \frac{(15 - 13,64) 0,005}{0,03}$$

$$q_e = \frac{0,006785}{0,03}$$

$$q_e = 0,23 \text{ mg/g}$$

3. Nilai konsentrasi adsorpsi dan kemampuan adsorpsi DEHP oleh MIP terhadap pengaruh waktu

Waktu (menit)	y (absorbansi)	x (konsentrasi) (mg/L)	q _e (mg/g)
30	4,502	11,07	0,65
90	4,450	3,64	1,89
120	4,461	5,21	1,63
150	4,496	10,21	0,80
180	4,490	9,36	0,94

Contoh perhitungan konsentrasi adsorpsi dan kemampuan adsorpsi DEHP oleh MIP terhadap pengaruh waktu:

a. Konsentrasi Adsorpsi DEHP oleh MIP 30 menit

$$y = 0,007x + 4,4245$$

$$y = 4,502$$

$$y = 0,007x + 4,4245$$

$$4,502 = 0,007x + 4,4245$$

$$x = \frac{4,502 - 4,4245}{0,007}$$

$$x = 11,07 \text{ mg/L}$$

b. Kemampuan Adsorpsi DEHP oleh MIP 30 menit

$$q_e = \frac{(C_o - C_e) V}{W}$$

Diketahui: $C_o = 15 \text{ mg/L}$ $W = 0,03 \text{ g}$

$C_e = \text{Konsentrasi setelah adsorpsi}$ $V = 0,005 \text{ L}$

$$q_e = \frac{(15 - 11,0714) 0,005}{0,03}$$

$$q_e = \frac{0,0196}{0,03}$$

$$q_e = 0,65 \text{ mg/g}$$

4. Nilai konsentrasi adsorpsi dan kemampuan adsorpsi DEHP oleh MIP terhadap pengaruh konsentrasi

Konsentrasi awal (mg/L)	y (absorbansi)	x (konsentrasi) (mg/L)	q _e (mg/g)
6	4,441	2,36	0,61
9	4,451	3,79	0,87
12	4,468	6,21	0,96
18	4,487	8,93	1,51
21	4,500	10,79	1,70

Contoh perhitungan konsentrasi adsorpsi dan kemampuan adsorpsi DEHP oleh MIP terhadap pengaruh konsentrasi:

a. Konsentrasi Adsorpsi DEHP oleh MIP 6 mg/L

$$y = 0,007x + 4,4245$$

$$y = 4,441$$

$$y = 0,007x + 4,4245$$

$$4,441 = 0,007x + 4,4245$$

$$x = \frac{4,441 - 4,4245}{0,007}$$

$$x = 2,3571 \text{ mg/L}$$

b. Kemampuan Adsorpsi DEHP oleh MIP 6 mg/L

$$q_e = \frac{(C_o - C_e) V}{W}$$

Diketahui: C_o = konsentrasi awal W = 0,03 g

C_e = Konsentrasi setelah adsorpsi V = 0,005 L

$$q_e = \frac{(6 - 2,3571) 0,005}{0,03}$$

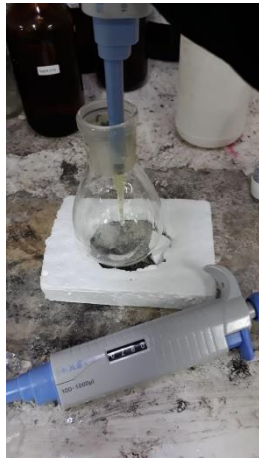
$$q_e = \frac{0,0182}{0,03}$$

$$q_e = 0,6071 \text{ mg/g}$$

Lampiran 5. Foto Hasil Penelitian



Proses persiapan alat dan bahan



Proses pencampuran bahan dan prapolimerisasi



Sonikasi



Pengaliran gas nitrogen untuk menghilangkan gas oksigen



Polimerisasi dalam *waterbath*



Polimer terbentuk berwarna putih



Pengeringan polimer



Proses pencucian polimer dengan aseton, metanol, dan akuades agar bersih dari pengotor



Proses ekstraksi (sonikasi)



Pencucian polimer dengan akuades



Penentuan pH



Penimbangan polimer hasil sintesis



Pembuatan deret standar DEHP



Uji kemampuan adsorpsi MIP dan NIP



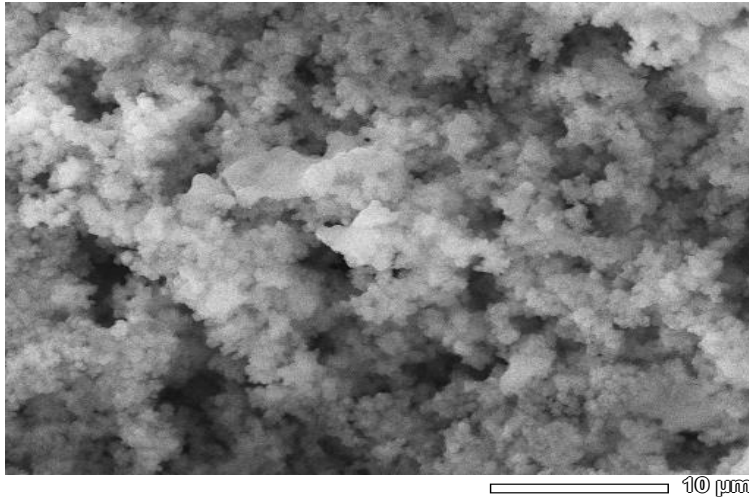
Pengocokan dengan alat *shaker* untuk pengaruh waktu dan konsentrasi terhadap adsorpsi DEHP



Polimer hasil sintesis MIP_DEHP_MAA-co-EGDMA dan NIP_MAA-co-EGDMA disimpan

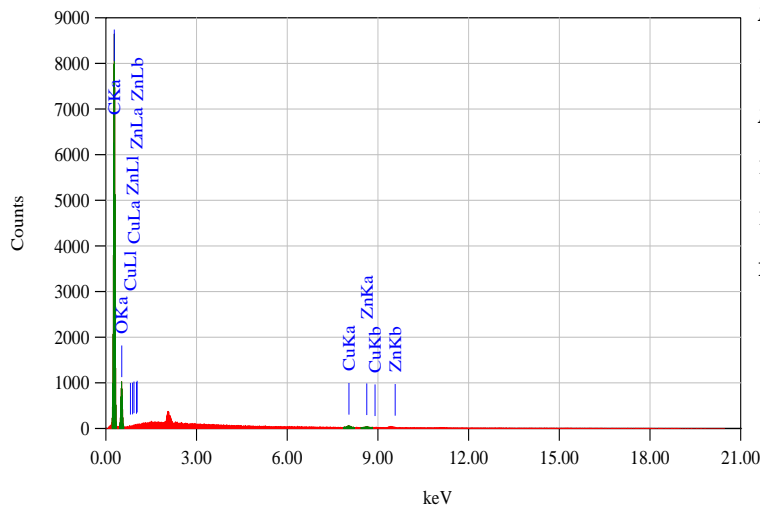
Lampiran 6. Karakterisasi EDS

1. NIP_MA-co-_TRIM



Title : IMG1

Instrument : 6510 (LA)
Volt : 20.00 kV
Mag. : x 3,000
Date : 2021/12/10
Pixel : 512 x 384



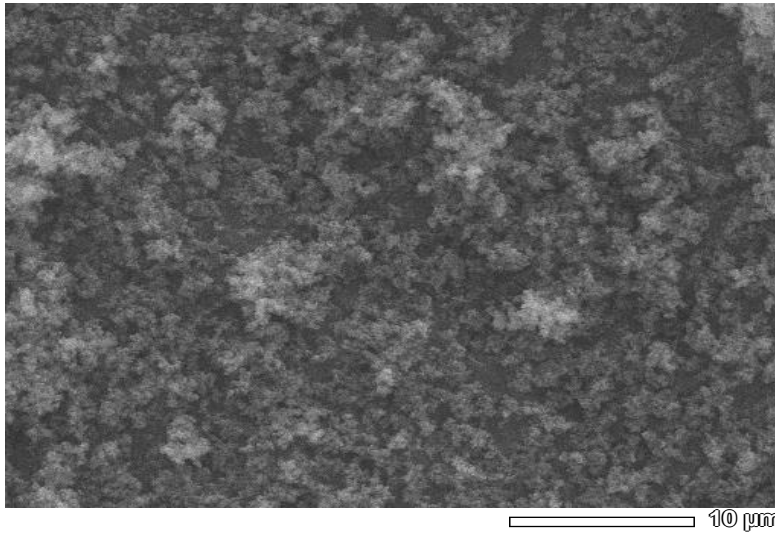
Acquisition Parameter
Instrument : 6510 (LA)
Acc. Voltage : 20.0 kV
Probe Current: 1.00000 nA
PHA mode : T3
Real Time : 51.04 sec

ZAF Method Standardless Quantitative Analysis

Fitting Coefficient : 0.0317

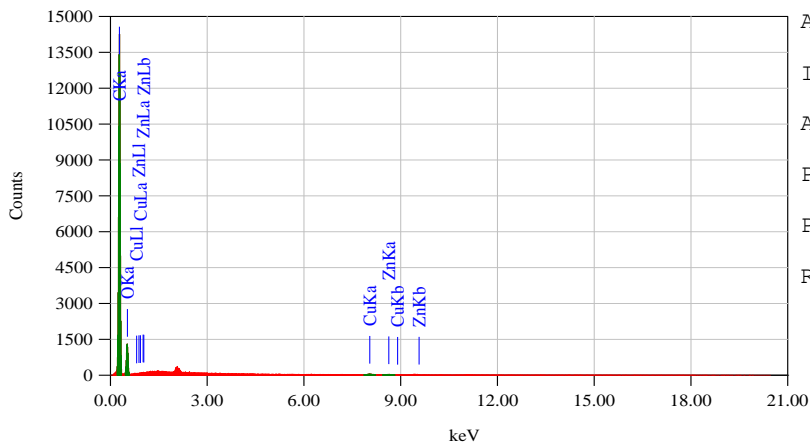
Element	(keV)	Mass%	Sigma	Atom%	Compound	Mass%	Cation	K
C K	0.277	76.13	0.03	81.49				85.2900
O K	0.525	22.76	0.10	18.29				13.2582
Cu K	8.040	0.67	0.03	0.14				0.8801
Zn K	8.630	0.44	0.03	0.09				0.5716
Total		100.00		100.00				

2. MIP_DEHP_MAAM-co-TRIM_(BE)



Title : IMG1

Instrument : 6510 (LA)
Volt : 20.00 kV
Mag. : x 3,000
Date : 2021/12/10
Pixel : 512 x 384



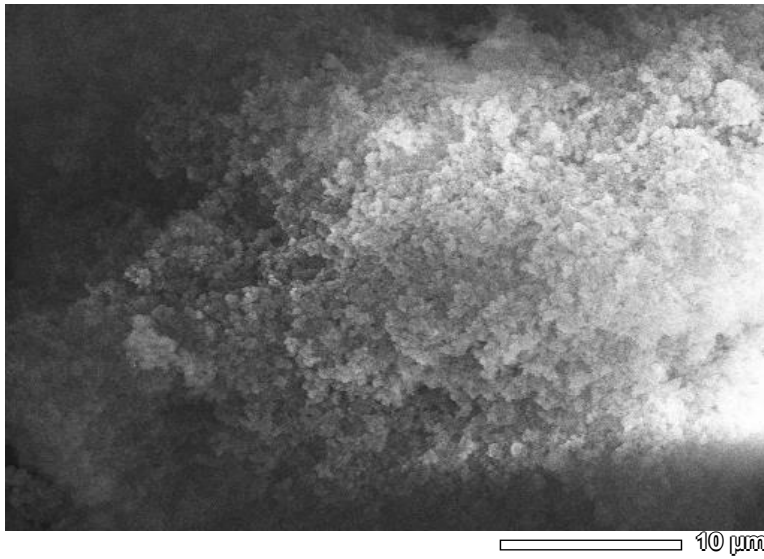
Acquisition Parameter
Instrument : 6510 (LA)
Acc. Voltage : 20.0 kV
Probe Current: 1.00000 nA
PHA mode : T3
Real Time : 51.24 sec

ZAF Method Standardless Quantitative Analysis

Fitting Coefficient : 0.0247

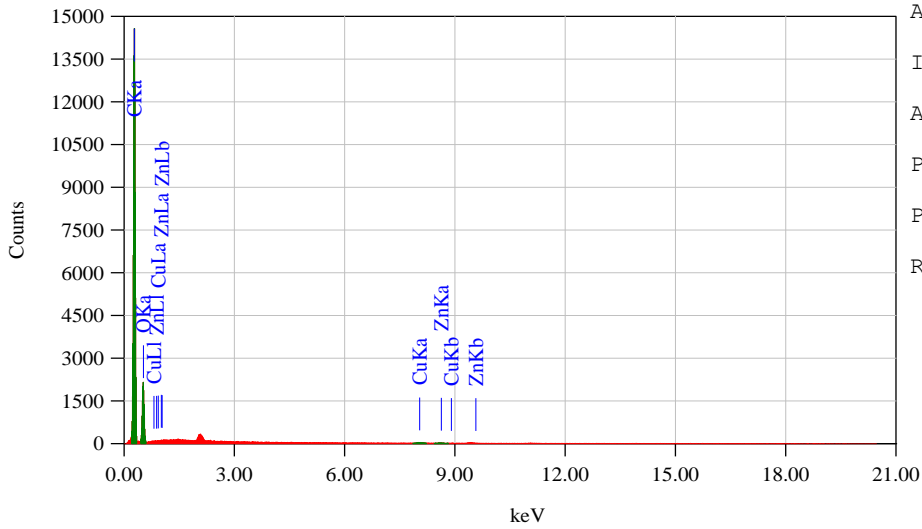
Element	(keV)	Mass%	Sigma	Atom%	Compound	Mass%	Cation	K
C K	0.277	78.76	0.17	83.48				88.2524
O K	0.525	20.61	0.26	16.40				10.9629
Cu K	8.040	0.39	0.03	0.08				0.4903
Zn K	8.630	0.24	0.03	0.05				0.2945
Total		100.00		100.00				

3. MIP_DEHP_MAAM-co-TRIM_(TE)



Title : IMG1

 Instrument : 6510(LA)
 Volt : 20.00 kV
 Mag. : x 3,000
 Date : 2021/12/10
 Pixel : 512 x 384



Acquisition Parameter
 Instrument : 6510(LA)
 Acc. Voltage : 20.0 kV
 Probe Current: 1.00000 nA
 PHA mode : T3
 Real Time : 51.21 sec

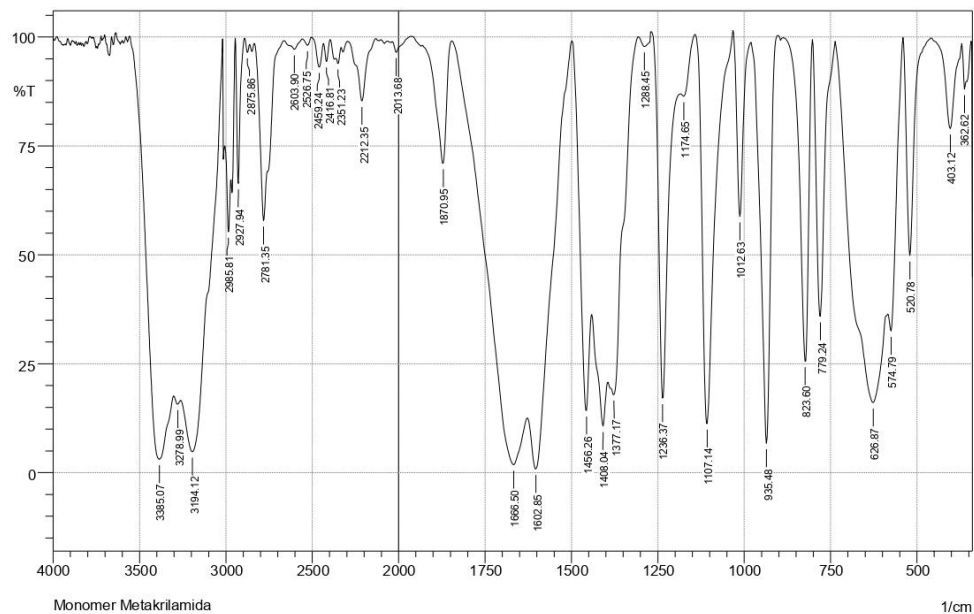
ZAF Method Standardless Quantitative Analysis

Fitting Coefficient : 0.0262

Element	(keV)	Mass%	Sigma	Atom%	Compound	Mass%	Cation	K
C K	0.277	73.25	0.15	78.68				83.0763
O K	0.525	26.34	0.26	21.24				16.3838
Cu K	8.040	0.26	0.02	0.05				0.3393
Zn K	8.630	0.15	0.02	0.03				0.2006
Total		100.00		100.00				

Lampiran 7. Karakterisasi FTIR

1. Spektrum Monomer MAAM

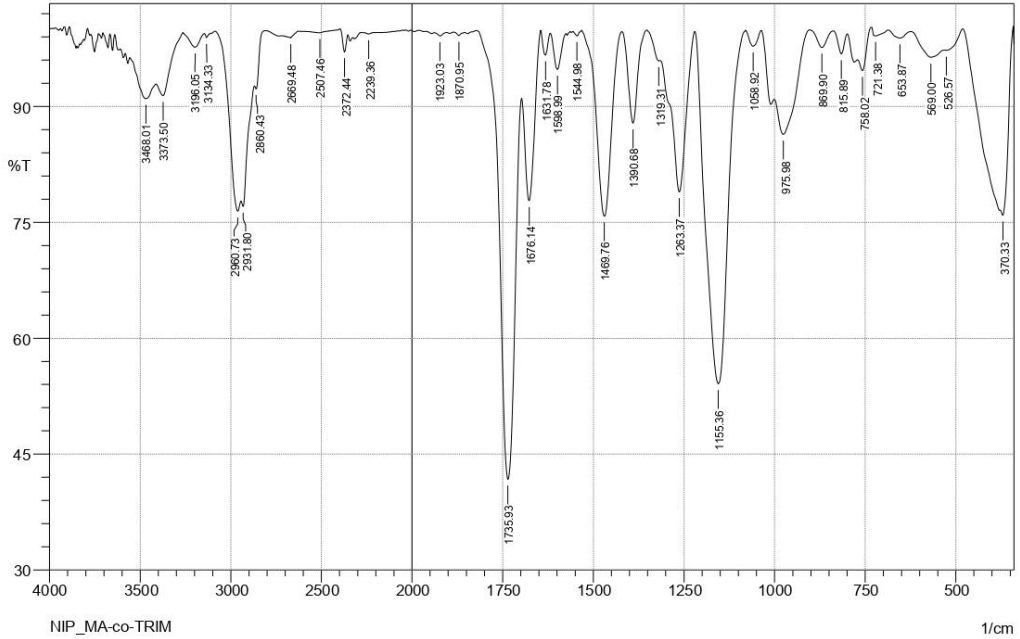


No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	362.62	88.048	10.596	370.33	343.33	0.91	0.764
2	403.12	78.997	19.43	457.13	372.26	3.148	2.655
3	520.78	49.845	49.998	540.07	459.06	6.427	6.363
4	574.79	32.578	15.506	582.5	542	10.177	1.807
5	626.87	16.088	37.739	734.88	584.43	68.962	34.936
6	779.24	35.88	62.593	800.46	736.81	10.16	9.774
7	823.6	25.622	73.87	887.26	802.39	13.525	13.377
8	935.48	6.775	92.943	979.84	902.69	23.22	23.091
9	1012.63	58.842	41.206	1029.99	981.77	4.295	4.23
10	1107.14	11.282	89.593	1141.86	1031.92	26.148	26.647
11	1174.65	86.399	3.789	1184.29	1143.79	1.476	0.378
12	1236.37	17.137	78.434	1269.16	1186.22	18.715	16.351
13	1288.45	97.805	2.798	1305.81	1269.16	0.192	0.292
14	1377.17	17.931	18.692	1394.53	1307.74	28.455	3.84
15	1408.04	10.746	13.942	1440.83	1396.46	31.462	6.244
16	1456.26	14.286	37.353	1496.76	1442.75	21.957	9.561
17	1602.85	0.884	27.322	1625.99	1498.69	77.902	26.359
18	1666.5	1.836	26.049	1845.88	1627.92	129.427	40.037
19	1870.95	71.016	27.771	1930.74	1847.81	4.957	4.442
20	2013.68	96.5	2.816	2036.83	1963.53	0.559	0.411
21	2212.35	85.333	13.986	2295.29	2133.27	4.305	3.821
22	2351.23	93.905	2.268	2364.73	2333.87	0.666	0.163
23	2416.81	94.397	4.506	2434.17	2395.59	0.552	0.375
24	2459.24	93.082	5.864	2503.6	2436.09	1.034	0.787
25	2526.75	98.106	1.676	2557.61	2503.6	0.249	0.192
26	2603.9	97.193	2.204	2663.69	2557.61	0.806	0.516
27	2781.35	57.904	19.719	2833.43	2760.14	9.425	3.251
28	2875.86	96.472	2.466	2891.3	2862.36	0.287	0.159
29	2927.94	66.43	33.271	2945.3	2893.22	3.591	3.521
30	2985.81	55.314	14.934	3005.1	2972.31	6.479	1.521
31	3194.12	4.821	34.455	3261.63	3020.53	158.506	63.967
32	3278.99	15.748	1.254	3300.2	3263.56	28.866	0.709
33	3385.07	3.132	41.189	3558.67	3302.13	168.031	75.404

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No. of Scans;

2. Spektrum NIP_MAAM-co-TRIM

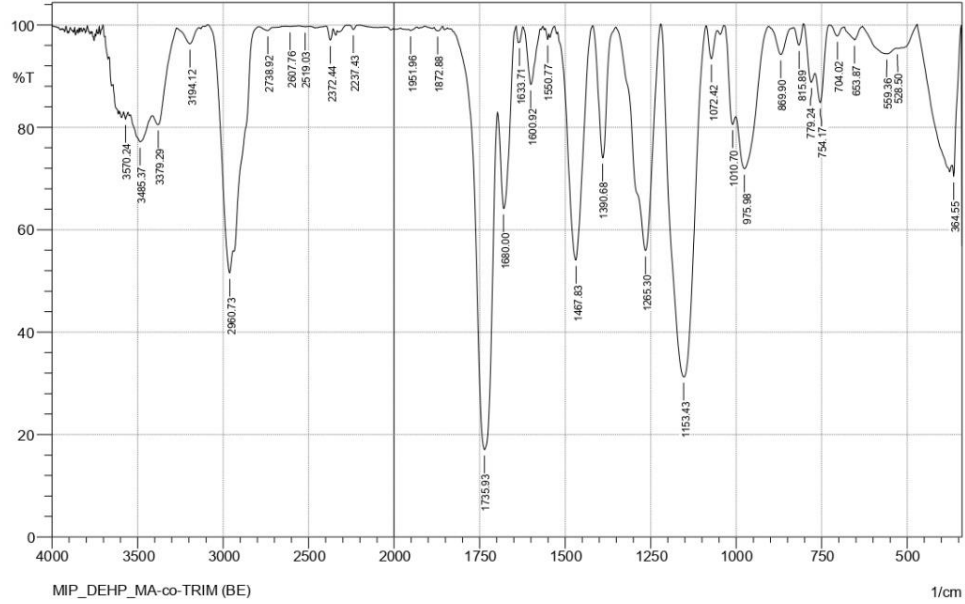


No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	370.33	75.914	4.737	376.12	343.33	2.363	0.578
2	526.57	97.281	0.334	532.35	480.28	0.411	0.11
3	569	96.382	1.864	623.01	534.28	0.998	0.406
4	653.87	98.871	0.931	686.66	623.01	0.191	0.137
5	721.38	99.153	1.027	732.95	686.66	0.104	0.121
6	758.02	94.655	2.885	771.53	732.95	0.537	0.224
7	815.89	96.825	2.759	837.11	798.53	0.284	0.217
8	869.9	97.651	2.216	902.69	837.11	0.32	0.282
9	975.98	86.392	6.703	999.13	904.61	3.608	1.677
10	1058.92	97.808	2.054	1080.14	1035.77	0.247	0.22
11	1155.36	54.117	45.664	1219.01	1080.14	16.329	16.202
12	1263.37	78.944	19.033	1313.52	1220.94	4.725	3.847
13	1319.31	95.917	0.456	1352.1	1315.45	0.349	0.013
14	1390.68	87.842	11.923	1419.61	1354.03	1.568	1.505
15	1469.76	75.796	23.917	1529.55	1421.54	5.297	5.163
16	1544.98	99.149	0.552	1556.55	1537.27	0.048	0.024
17	1598.99	94.81	4.643	1618.28	1573.91	0.545	0.439
18	1631.78	96.667	3.008	1645.28	1618.28	0.231	0.192
19	1676.14	77.816	16.806	1697.36	1647.21	3.274	2.235
20	1735.93	41.748	51.765	1820.8	1699.29	16.194	13.567
21	1870.95	99.187	0.376	1886.38	1861.31	0.062	0.016
22	1923.03	99.13	0.443	1938.46	1905.67	0.088	0.027
23	2239.36	99.409	0.168	2279.86	2196.92	0.174	0.022
24	2372.44	97.052	2.56	2397.52	2353.16	0.285	0.22
25	2507.46	99.542	0.336	2601.97	2397.52	0.243	0.145
26	2669.48	98.897	0.607	2717.7	2630.91	0.31	0.109
27	2860.43	92.28	1.497	2870.08	2808.36	0.938	-0.107
28	2931.8	77.066	2.672	2941.44	2870.08	5.086	0.279
29	2960.73	76.44	3.993	3086.11	2943.37	7.785	0.92
30	3134.33	98.914	0.605	3143.97	3086.11	0.153	0.055
31	3196.05	97.678	1.838	3263.56	3143.97	0.737	0.489
32	3373.5	91.418	3.217	3412.08	3263.56	3.259	0.796
33	3468.01	91.033	3.168	3556.74	3414	4.585	1.13

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No. of Scans;

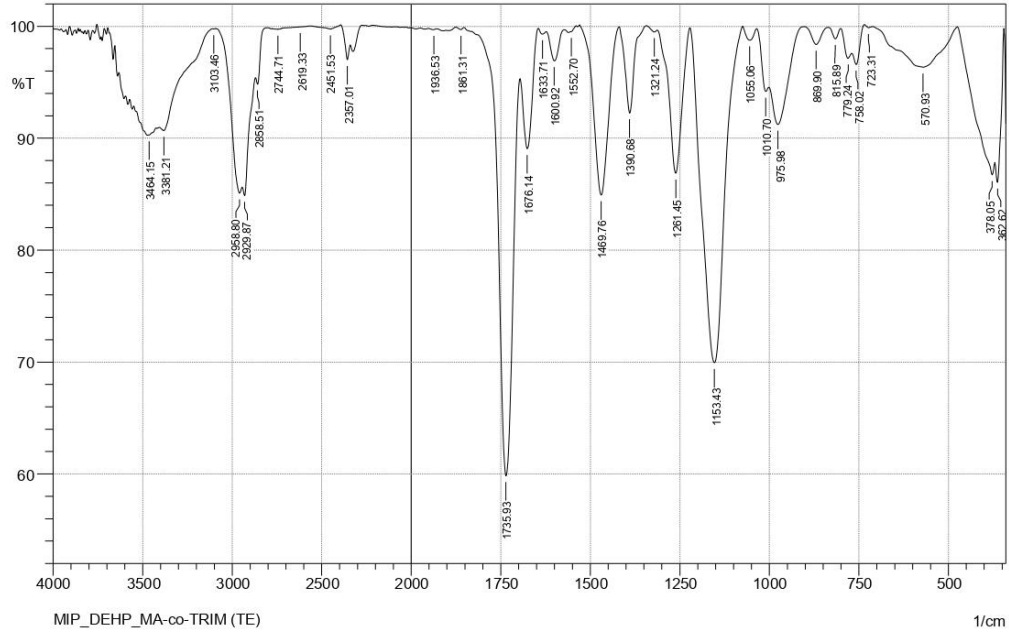
3. Spektrum MIP_DEHP_MAAM-co-TRIM_(BE)



No.	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	364.55	70.426	6.306	368.4	343.33	1.899	0.274
2	376.12	95.428	0.041	532.35	514.99	0.349	0.002
3	559.36	94.37	2.251	630.72	532.35	1.802	0.732
4	653.87	97.078	2.595	682.8	630.72	0.385	0.309
5	704.02	97.837	1.946	727.16	682.8	0.196	0.156
6	754.17	84.833	9.304	769.6	729.09	1.661	0.832
7	779.24	88.835	4.055	802.39	771.53	1.018	0.324
8	815.89	96.063	3.834	833.25	802.39	0.27	0.25
9	869.9	94.232	5.288	904.61	839.03	0.805	0.672
10	975.98	71.93	14.752	1001.06	904.61	7.952	3.88
11	1010.7	80.588	5.879	1033.85	1002.98	1.819	0.441
12	1072.42	93.386	6.023	1087.85	1055.06	0.519	0.431
13	1153.43	31.269	68.714	1219.01	1087.85	32.517	32.506
14	1265.3	55.933	44.027	1350.17	1220.94	14.32	14.238
15	1390.68	74.027	25.65	1417.68	1352.1	3.317	3.219
16	1467.83	54.08	45.206	1521.84	1419.61	11.729	11.4
17	1550.77	97.153	1.472	1558.48	1546.91	0.091	0.03
18	1600.92	88.372	11.024	1622.13	1573.91	1.349	1.215
19	1633.71	96.602	0.547	1635.64	1624.06	0.118	0.023
20	1680	64.136	23.499	1695.43	1645.28	5.636	3.553
21	1735.93	17.059	70.669	1824.66	1697.36	34.226	28.623
22	1872.88	98.87	0.113	1882.52	1870.95	0.048	0.007
23	1951.96	98.918	0.341	1957.75	1932.67	0.088	0.019
24	2237.43	99.111	0.856	2266.36	2200.78	0.084	0.075
25	2372.44	97.042	2.406	2397.52	2357.01	0.288	0.202
26	2519.03	99.656	0.102	2561.47	2492.03	0.084	0.014
27	2607.76	99.684	0.057	2621.26	2582.68	0.046	0.004
28	2738.92	98.909	0.661	2791	2706.13	0.271	0.118
29	2960.73	51.592	10.103	3088.03	2941.44	18.654	2.687
30	3194.12	96.319	3.111	3271.27	3140.11	1.134	0.804
31	3379.29	80.547	5.75	3412.08	3273.2	7.568	1.605
32	3485.37	77.155	5.189	3541.31	3414	12.732	1.984
33	3570.24	81.63	1.291	3579.88	3556.74	1.961	0.079

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No. of Scans;

4. Spektrum MIP_DEHP_MAAM-co-TRIM_(TE)



	Peak	Intensity	Corr. Intensity	Base (H)	Base (L)	Area	Corr. Area
1	362.62	86.053	5.471	370.33	345.26	1.083	0.421
2	378.05	86.726	1.608	474.49	372.26	3.579	0.591
3	570.93	96.358	3.609	711.73	476.42	2.167	2.137
4	723.31	99.881	0.192	734.88	711.73	0.003	0.01
5	758.02	96.596	1.849	769.6	734.88	0.307	0.135
6	779.24	97.15	1.136	800.46	769.6	0.259	0.083
7	815.89	98.885	0.975	837.11	800.46	0.089	0.068
8	869.9	98.386	1.562	902.69	837.11	0.211	0.197
9	975.98	91.226	4.742	1001.06	904.61	2.034	0.91
10	1010.7	94.181	1.831	1035.77	1001.06	0.549	0.112
11	1055.06	98.757	1.143	1074.35	1035.77	0.134	0.117
12	1153.43	69.944	29.948	1220.94	1076.28	10.106	10.04
13	1261.45	86.87	12.895	1311.59	1222.87	2.406	2.311
14	1321.24	99.494	0.27	1342.46	1313.52	0.035	0.017
15	1390.68	92.267	7.728	1419.61	1342.46	0.915	0.918
16	1489.76	84.931	15.065	1527.62	1419.61	3.16	3.159
17	1552.7	99.521	0.069	1554.63	1539.2	0.017	0.002
18	1600.92	96.925	2.684	1622.13	1571.99	0.368	0.285
19	1633.71	99.31	0.366	1645.28	1622.13	0.054	0.021
20	1676.14	89.066	8.041	1695.43	1647.21	1.473	0.968
21	1735.93	59.842	36.566	1851.66	1697.36	9.559	7.94
22	1861.31	99.707	0.208	1874.81	1851.66	0.019	0.01
23	1936.53	99.71	0.082	1951.96	1928.82	0.025	0.005
24	2357.01	97.031	1.885	2393.66	2339.65	0.349	0.169
25	2451.53	99.758	0.344	2559.54	2393.66	0.059	0.122
26	2619.33	99.945	0.009	2625.12	2559.54	0.003	0.001
27	2744.71	99.738	0.092	2777.5	2696.48	0.073	0.015
28	2858.51	94.833	1.397	2870.08	2806.43	0.625	-0.039
29	2929.87	84.869	2.567	2943.37	2870.08	3.343	0.254
30	2958.8	85.104	1.818	3091.89	2945.3	4.553	0.375
31	3103.46	99.776	0.019	3111.18	3093.82	0.016	0.001
32	3381.21	90.683	1.032	3406.29	3111.18	5.789	0.192
33	3464.15	90.285	0.07	3468.01	3427.51	1.752	0.019

Date/Time; 9/17/2021 2:11:10 PM

No. of Scans;

Lampiran 8. Karakterisasi SAA



TriStar II 3020 2.00 TriStar II 3020 Version 2.00 Unit Serial #: 1108 Page 1
1 Port 1

Sample: MIP_DEHP_MA_Co_TRIM
Operator: Sarah
Submitter: 25991
File: C:\TriStar II 3020\data\SAMPEL\20...MIP_DEHP_MA_Co_TRIM.SMP

Started: 1/7/2022 7:05:13 AM	Analysis Adsorptive: N2
Completed: 1/7/2022 11:08:05 AM	Analysis Bath Temp.: -195.850 °C
Report Time: 1/7/2022 1:11:21 PM	Thermal Correction: No
Sample Mass: 0.0551 g	Warm Free Space: 11.4392 cm ³ Measured
Cold Free Space: 33.0250 cm ³	Equilibration Interval: 5 s
Low Pressure Dose: None	Sample Density: 1.000 g/cm ³
Automatic Degas: No	

Summary Report

Surface Area

Single point surface area at P/Po = 0.303234883: 357.5634 m²/g
BET Surface Area: 362.0731 m²/g
t-Plot Micropore Area: 137.9785 m²/g
t-Plot External Surface Area: 224.0946 m²/g

BJH Adsorption cumulative surface area of pores
between 1.7000 nm and 300.0000 nm diameter: 193.092 m²/g

BJH Desorption cumulative surface area of pores
between 1.7000 nm and 300.0000 nm diameter: 174.2059 m²/g

D-H Adsorption cumulative surface area of pores
between 1.7000 nm and 300.0000 nm diameter: 164.879 m²/g

D-H Desorption cumulative surface area of pores
between 1.7000 nm and 300.0000 nm diameter: 154.4766 m²/g

Pore Volume

Single point adsorption total pore volume of pores
less than 240.3666 nm diameter at P/Po = 0.991973315: 0.442559 cm³/g
t-Plot micropore volume: 0.072140 cm³/g

BJH Adsorption cumulative volume of pores
between 1.7000 nm and 300.0000 nm diameter: 0.359706 cm³/g

BJH Desorption cumulative volume of pores
between 1.7000 nm and 300.0000 nm diameter: 0.330791 cm³/g

Pore Size

Adsorption average pore width (4V/A by BET): 4.88917 nm

BJH Adsorption average pore diameter (4V/A): 7.4515 nm

BJH Desorption average pore diameter (4V/A): 7.5954 nm

D-H Adsorption average pore diameter (4V/A): 8.0798 nm

D-H Desorption average pore diameter (4V/A): 8.1945 nm



TriStar II 3020 2.00

TriStar II 3020 Version 2.00 Unit
1 Port 1

Serial #: 1108

Page 3

Sample: MIP_DEHP_MA_Co_TRIM
Operator: Sarah
Submitter: 25991
File: C:\TriStar II 3020\data\SAMPEL\20...MIP_DEHP_MA_Co_TRIM.SMP

Started: 1/7/2022 7:05:13 AM
Completed: 1/7/2022 11:08:05 AM
Report Time: 1/7/2022 1:11:21 PM
Sample Mass: 0.0551 g
Cold Free Space: 33.0250 cm³
Low Pressure Dose: None
Automatic Degas: No
Analysis Adsorptive: N2
Analysis Bath Temp.: -195.850 °C
Thermal Correction: No
Warm Free Space: 11.4392 cm³ Measured
Equilibration Interval: 5 s
Sample Density: 1.000 g/cm³

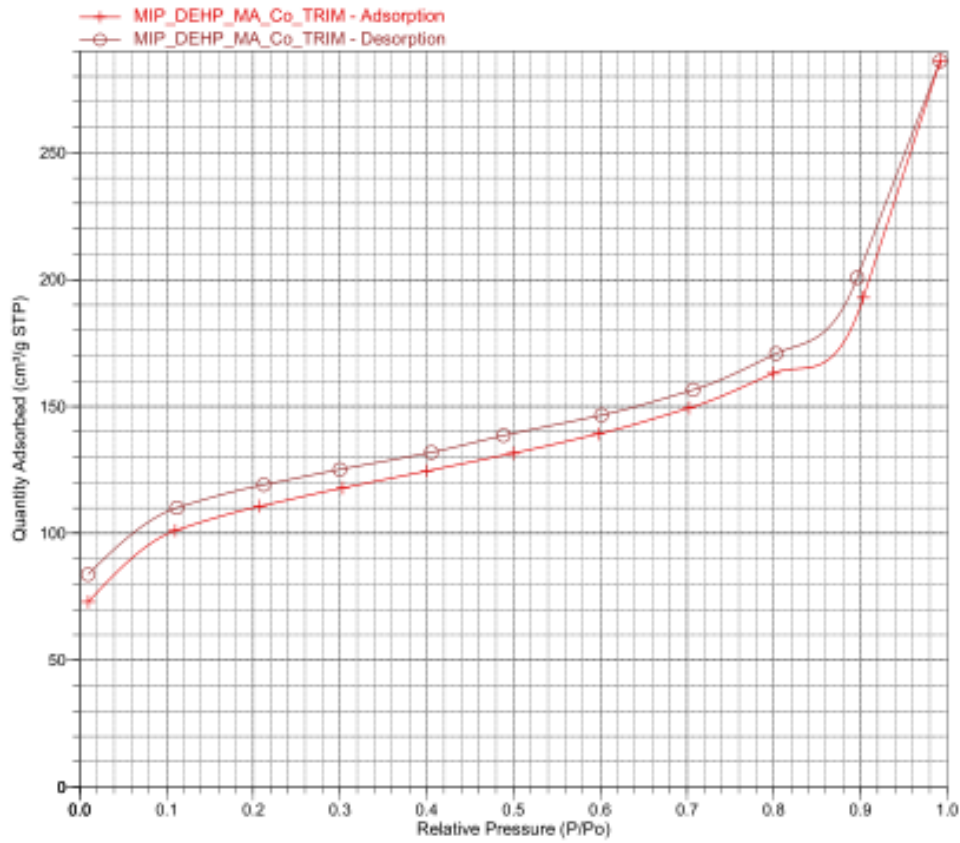
Isotherm Tabular Report

Relative Pressure (P/Po)	Absolute Pressure (mmHg)	Quantity Adsorbed (cm ³ /g STP)	Elapsed Time (h:min)	Saturation Pressure (mmHg)
				760.000000
0.010004170	7.603169	72.9769	01:37	
0.110378004	83.887283	101.0690	01:59	
0.207913991	158.014633	110.6729	02:06	
0.303234883	230.458511	117.8849	02:11	
0.400497959	304.378448	124.5734	02:14	
0.499579339	379.680298	131.5470	02:18	
0.597817752	454.341492	139.1830	02:22	
0.702672858	534.031372	149.3914	02:27	
0.799710324	607.779846	163.0793	02:32	
0.903295256	686.504395	192.8225	02:42	
0.991973315	753.899719	286.1125	02:49	
0.896344958	681.222168	200.6629	03:02	
0.802937879	610.232788	170.8213	03:09	
0.707111720	537.404907	156.5428	03:13	
0.601480584	457.125244	146.5617	03:16	
0.488706408	371.416870	138.5477	03:20	
0.405511073	308.188416	131.9219	03:24	
0.299760798	227.818207	125.1567	03:27	
0.212366787	161.398758	119.1374	03:30	
0.112060216	85.165764	110.0602	03:36	
0.009741831	7.403792	83.8048	04:01	

Sample: MIP_DEHP_MA_Co_TRIM
Operator: Sarah
Submitter: 25991
File: C:\TriStar II 3020\data\SAMPLE120...\MIP_DEHP_MA_Co_TRIM.SMP

Started: 1/7/2022 7:05:13 AM	Analysis Adsorptive: N2
Completed: 1/7/2022 11:08:05 AM	Analysis Bath Temp.: -195.850 °C
Report Time: 1/7/2022 1:11:21 PM	Thermal Correction: No
Sample Mass: 0.0551 g	Warm Free Space: 11.4392 cm ³ Measured
Cold Free Space: 33.0250 cm ³	Equilibration Interval: 5 s
Low Pressure Dose: None	Sample Density: 1.000 g/cm ³
Automatic Degas: No	

Isotherm Linear Plot





TriStar II 3020 2.00

TriStar II 3020 Version 2.00 Unit
1 Port 1

Serial #: 1108

Page 12

Sample: MIP_DEHP_MA_Co_TRIM
Operator: Sarah
Submitter: 25991
File: C:\TriStar II 3020\data\SAMPLE1\20...MIP_DEHP_MA_Co_TRIM.SMP

Started: 1/7/2022 7:05:13 AM
Completed: 1/7/2022 11:08:05 AM
Report Time: 1/7/2022 1:11:21 PM
Sample Mass: 0.0551 g
Cold Free Space: 33.0250 cm³
Low Pressure Dose: None
Automatic Degas: No

Analysis Adsorptive: N2
Analysis Bath Temp.: -195.850 °C
Thermal Correction: No
Warm Free Space: 11.4392 cm³ Measured
Equilibration Interval: 5 s
Sample Density: 1.000 g/cm³

BJH Adsorption Pore Distribution Report

Faas Correction

Harkins and Jura

$$t = [13.99 / (0.034 - \log(P/P_0))] ^{0.5}$$

Diameter Range: 1.7000 nm to 300.0000 nm

Adsorbate Property Factor: 0.95300 nm

Density Conversion Factor: 0.0015468

Fraction of Pores Open at Both Ends: 0.00

Pore Diameter Range (nm)	Average Diameter (nm)	Incremental Pore Volume (cm ³ /g)	Cumulative Pore Volume (cm ³ /g)	Incremental Pore Area (m ² /g)	Cumulative Pore Area (m ² /g)
240.5 - 21.5	22.9	0.187138	0.187138	32.619	32.619
21.5 - 10.7	12.6	0.054378	0.241517	17.247	49.866
10.7 - 7.3	8.3	0.022766	0.264283	11.029	60.894
7.3 - 5.3	5.9	0.016747	0.281030	11.291	72.186
5.3 - 4.2	4.6	0.012674	0.293704	11.097	83.283
4.2 - 3.3	3.6	0.012359	0.306063	13.564	96.847
3.3 - 2.7	3.0	0.012817	0.318880	17.339	114.186
2.7 - 2.2	2.4	0.015730	0.334610	26.145	140.332
2.2 - 1.7	1.9	0.025096	0.359706	52.761	193.092

Lampiran 9. Contoh Perhitungan Nilai Kapasitas Adsorpsi berdasarkan Model Persamaan Isothermal Adsorpsi Langmuir dan Isothermal Adsorpsi Freundlich

1. Isothermal Adsorpsi Langmuir

Persamaan:

$$\frac{1}{q_e} = \frac{1}{q_m K_L} \times \frac{1}{C_e} + \frac{1}{q_m}$$

Keterangan:

C_e = Konsentrasi saat kesetimbangan (mg/L)

q_e = Jumlah zat teradsorpsi saat kesetimbangan (mg/g)

q_m = Kapasitas adsorpsi monolayer (mg/g)

K_L = Konstanta afinitas adsorpsi atau konstanta kesetimbangan (L/mg)

Berdasarkan model isothermal Langmuir diperoleh persamaan garis:

$$y = 3,0359x + 0,3767$$

$$\frac{1}{q_m} = 0,3767 \quad \text{maka,} \quad q_m = \frac{1}{0,3767} = 2,65 \text{ mg/g}$$

$$\frac{1}{q_m K_L} = 3,0359 \quad \text{maka,} \quad K_L = \frac{1}{3,0359 \times 2,6545} = 0,12 \text{ L/mg}$$

2. Isothermal Adsorpsi Freundlich

Persamaan:

$$\log q_e = \frac{1}{n} \log C_e + \log K_F$$

Keterangan:

C_e = Konsentrasi saat kesetimbangan (mg/L)

q_e = Jumlah zat teradsorpsi saat kesetimbangan (mg/g)

K_F = Kapasitas adsorpsi (mg/g)

$\frac{1}{n}$ = Konstanta Freundlich menyatakan faktor heterogenitas

n = Intensitas adsorpsi

Berdasarkan model isothermal Freundlich diperoleh persamaan garis:

$$\frac{1}{n} = 0,6578 \quad \text{maka,} \quad y = 0,6578x - 0,467$$
$$n = \frac{1}{0,6578} = 1,52$$

$$\log K_F = -0,467$$
$$K_F = \text{Inv. log} (-0,467)$$
$$K_F = 0,34 \text{ mg/g}$$